

Quantum Physics 1

Class 23

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Hydrogen Atom

Last time :

Consider central force problem?

$$\hat{H} = \frac{\hat{p}^2}{2m_0} + V(r)$$

$$\Rightarrow \left(\frac{\hat{p}_r^2}{2m_0} + \frac{\hat{L}^2}{2m_0 r^2} \right) \psi(r, \theta, \phi) = (E - V(r)) \psi(r, \theta, \phi)$$

$$\text{where } \psi(r, \theta, \phi) = R(r) \underbrace{\psi_{\text{en}}(\theta, \phi)}_{\text{angular part}}$$

$$\begin{aligned} \hat{L}^2 \psi_{\text{en}}(\theta, \phi) &= l(l+1) h^2 \psi_{\text{en}}(\theta, \phi) \\ \hat{L}_z^2 \psi_{\text{en}}(\theta, \phi) &= m^2 h^2 \psi_{\text{en}}(\theta, \phi) \\ L_z \underline{\Phi}(\phi) &= m h \underline{\Phi}(\phi) \end{aligned} \quad \left. \right\} \quad \uparrow$$

Radial part:

$$\left(\frac{\hat{p}_r^2}{2m_0} + \underbrace{\frac{l(l+1) h^2}{2m_0 r^2}}_{\hat{L}^2} \right) R(r) = [E - V(r)] R(r)$$

Now consider the hydrogen-like atoms :

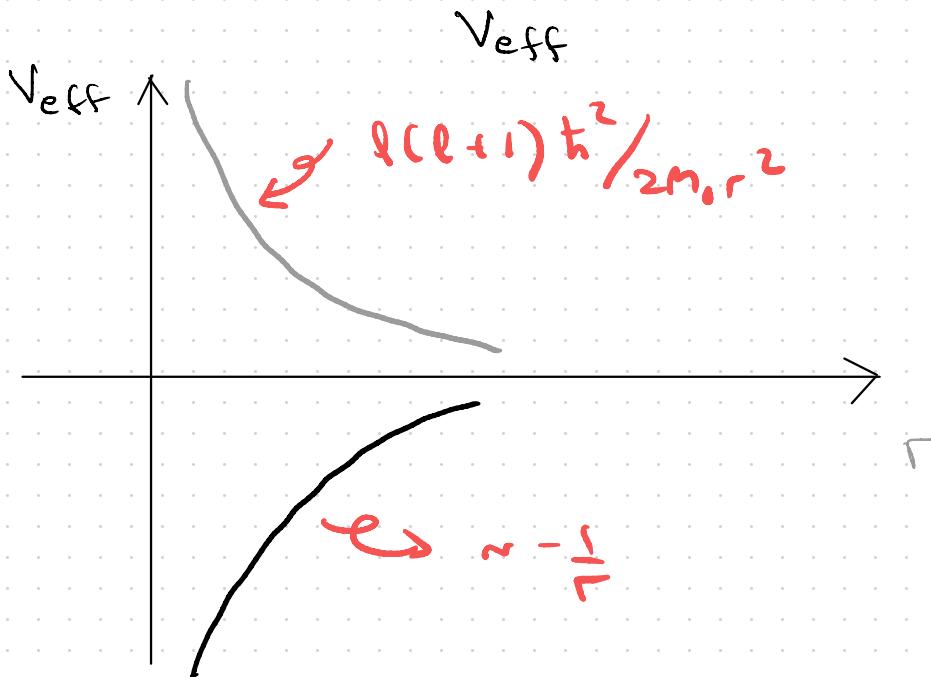
$$V = -\frac{Ze^2}{4\pi\epsilon_0 r}$$

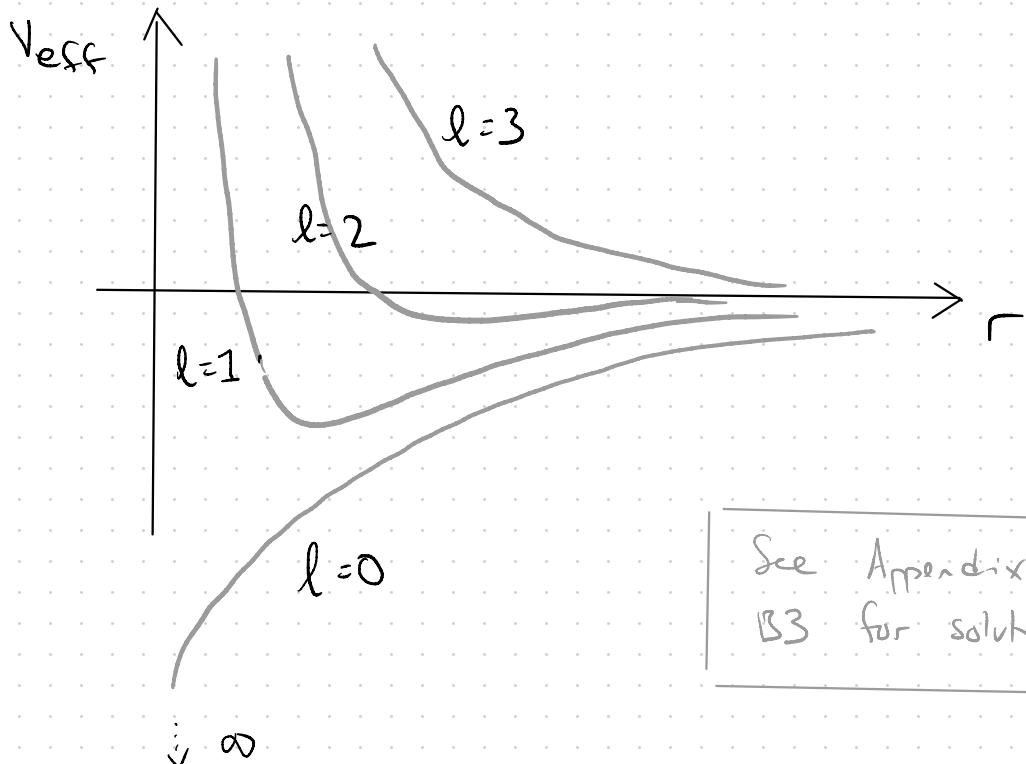
where
hydrogen :
 $Z=1$

Δ variables to simplify

$$u(r) \equiv r R(r)$$

$$\Rightarrow -\frac{\hbar^2}{2m_0} \frac{d^2 u}{dr^2} + \left[\frac{l(l+1)\hbar^2}{2m_0 r^2} - \frac{Ze^2}{4\pi\epsilon_0 r} \right] u(r) = E u(r)$$





See Appendix
B3 for solution

$$u_{n,l} = r^{-l-1} F_n(r) e^{-\sqrt{2m_0 E / \hbar^2} r}$$

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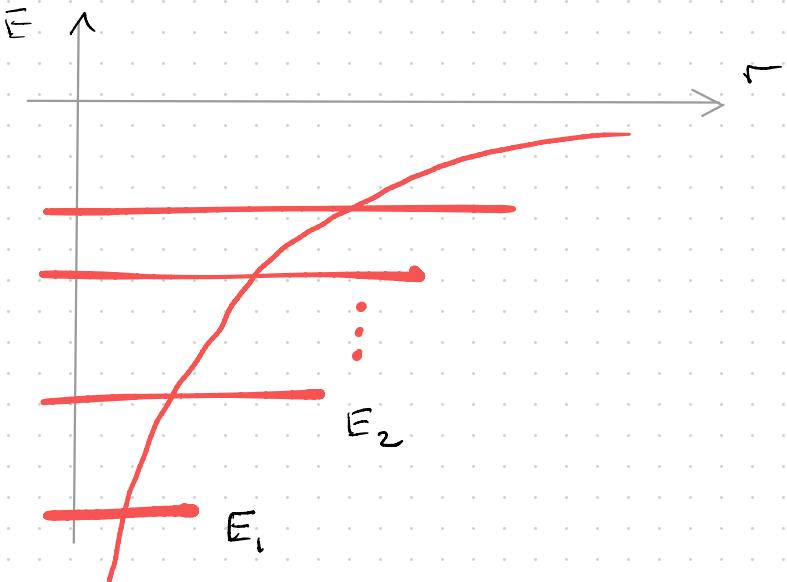
Associate Legendre  
Polynomials

$$R(r)_{n,l} = r^l F_n(r) e^{-\sqrt{2m_0 E / \hbar^2} r}$$

$$\text{w/t } E_n = \frac{-m_0 Z^2 e^4}{(4\pi \epsilon_0)^2 2\hbar^2 n^2} = \frac{-(13.6 \text{ eV}) Z^2}{n^2}, \quad n=1,2,3,\dots$$

$$l=0,1,2,\dots$$

$$n-1$$



Extreme limits of  $\sigma$ :

- ① for  $r \rightarrow \infty$ ;  $V_{\text{eff}} \rightarrow 0$

$$\Rightarrow -\frac{\hbar^2}{2m_0} \frac{d^2 u(r)}{dr^2} = E u(r); u = e^{-\sqrt{2m_0 E / \hbar^2} r}$$

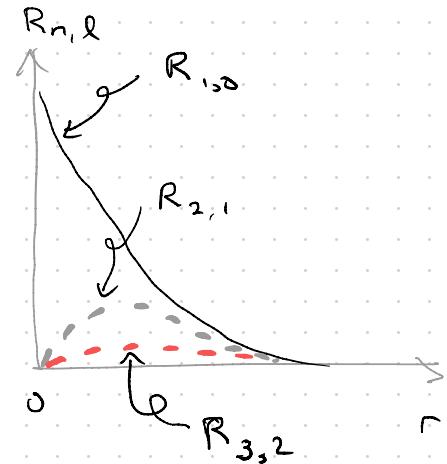
$$R = u/r$$

- ② for  $r \rightarrow 0$

$$\Rightarrow \frac{d^2 u(r)}{dr^2} = \frac{l(l+1)}{r^2} u(r)$$

$$u \sim r^{l+1}$$

$$R = \frac{u}{r} = r^l$$



In general (arbitrary  $r$ )

e.g)  $n=1, l=0, m=0$

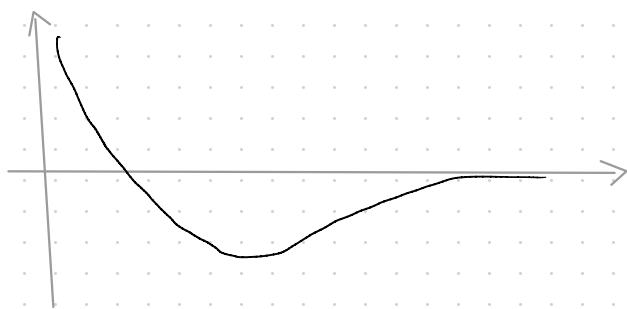
$$R_{1,0} = 2 \left( \frac{Z}{a_0} \right)^{3/2} e^{-Zr/a_0}$$

$$a_0 = \frac{4\pi \epsilon_0 \hbar^2}{m_e e^2}, \text{ Bohr Radius}$$

$$a_0 = 0.53 \text{ \AA} \text{ for Hydrogen.}$$

e.g)  $n=2 \begin{cases} l=1, m=\pm 1, 0 \\ l=0, m=0 \end{cases}$

$$R_{2,0}(r) = 2 \left( \frac{Z}{2a_0} \right)^{3/2} \left( 1 - \frac{Zr}{a_0} \right)^{-2} e^{-Zr/2a_0}$$



NB Are normalizable!

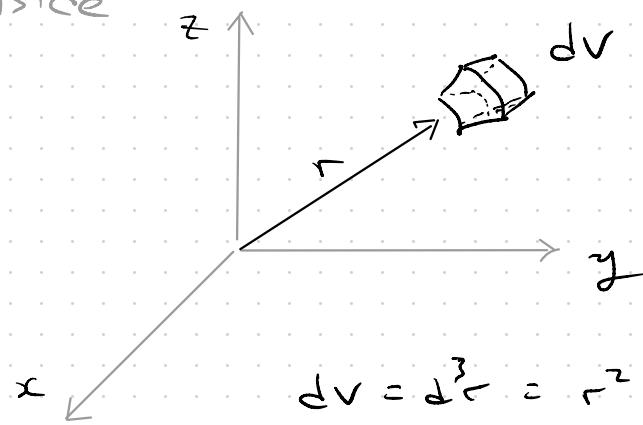
$$\int \Psi_{n,l,m}^*(r, \theta, \phi) \Psi_{n,l,m}(r, \theta, \phi) d^3r = 1$$

$\underbrace{dV}$

$$\Rightarrow \int \psi_{n,l,m}^* \psi_{n,l,m} r^2 \sin\theta d\phi d\theta dr$$

$$\Rightarrow \iiint dr r^2 R_{n,l}^* R_{n,l} Y_{l,m}^*(\theta, \phi) Y_{l,m}(\theta, \phi) \sin\theta d\theta d\phi = 1$$

Aside



$$dv = dr^3 = r^2 \sin\theta dr d\theta d\phi$$

$$\underbrace{\int dr r^2 R_{n,l}^* R_{n,l}}_P = 1$$

$P(r) \sim \text{probability density at } r$

In-class 23.1

Degeneracy:

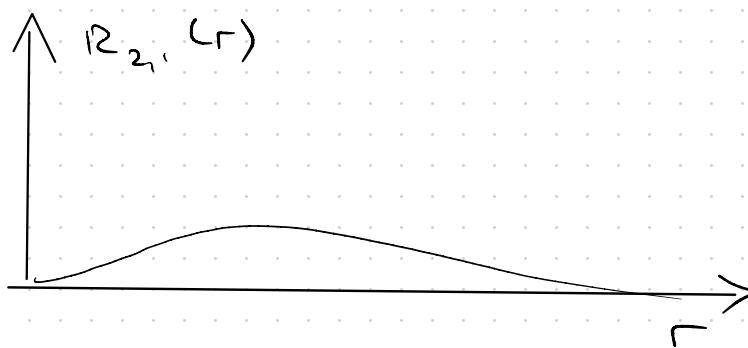
$n=1, l=0, m=0$  no degeneracy

$n=2 \left\{ \begin{array}{l} l=0, m=0 \\ l=1, m=0, \pm 1 \end{array} \right. \rightarrow 4$  fold degeneracy

$n=3 \left\{ \begin{array}{l} l=0, m=0 \\ l=1, m=0, \pm 1 \\ l=2, m=0, \pm 1, \pm 2 \end{array} \right. \rightarrow 9$

example:  $n=2, l=1$

$$R_{2,1}(r) = \frac{1}{\sqrt{3}} \left( \frac{z}{2a_0} \right)^{3/2} \left( \frac{ze}{a_0} \right) e^{-ze/2a_0}$$



Now, Total Solution:

$$\Psi_{nem}(r, \theta, \phi) = R_{ne} Y_{em}(\theta, \phi)$$

$(n \ l \ m)$ : example,  $(1, 0, 0)$

$$1S_{l=0} \quad \psi_{1,0,0} = \frac{2}{a_0^{3/2}} e^{-r/a_0} Y_{0,0}(\theta, \phi)$$

$$2S_{l=0} \quad \psi_{2,0,0} = \frac{2}{(2a_0)^{3/2}} \frac{\pi}{a_0} e^{-r/2a_0} Y_{0,0}(\theta, \phi)$$

$$2P_{l=1} \quad \begin{pmatrix} \psi_{2,1,1} \\ \psi_{2,1,0} \\ \psi_{2,1,-1} \end{pmatrix} = \frac{1}{\sqrt{3}(2a_0)^{3/2}} \frac{\pi}{a_0} e^{-r/2a_0} \begin{pmatrix} Y_{1,1} \\ Y_{1,0} \\ Y_{1,-1} \end{pmatrix}$$

## In-class 23-2

Atomic orbitals : \* can construct real functions out of  $\psi_{n,l,m}$ , which can have both real and imaginary parts.

$$\psi_{n,l=0} = \sqrt{\frac{3}{4\pi}} R_{n,1} \cos \theta$$

$$\psi_{n,l=1} = -\sqrt{\frac{3}{8\pi}} R_{n,1} \sin \theta e^{i\phi}$$

$$\psi_{n,l=1} = \sqrt{\frac{3}{8\pi}} R_{n,1} \sin \theta e^{-i\phi}$$

e.g.)

\* these are real  
functions

$$P_z = \sqrt{\frac{3}{4\pi}} R_{n,l} \cos\theta$$

$$P_x = -\frac{1}{\sqrt{2}} [\psi_{n,l,-1} - \psi_{n,l,+1}] = \sqrt{\frac{3}{4\pi}} R_{n,l} \frac{x}{r}$$

$$P_y = \frac{i}{\sqrt{2}} [\psi_{n,l,+1} + \psi_{n,l,-1}] = \sqrt{\frac{3}{4\pi}} R_{n,l} \frac{y}{r}$$

also have s-p hybrid orbitals etc.

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In-class 23-3

In-class 23-4