Reaction rates versus temperature - selected species:

Т	H2	H2	О	О
	$\operatorname{rxn} 0$	rxn 1	$\operatorname{rxn} 0$	rxn 1
300	-6.00e+01	1.59e + 13	-7.00e+01	-2.12e+22
1300	-6.00e+01	5.76e + 13	-7.00e+01	-6.96e+14
3000	-6.00e+01	6.17e + 13	-7.00e+01	-2.94e+13