

PHYSICS HOMEWORK # 3 CHAPTERS 12 - 14

①

17) Mass Density same on Moon as Earth?

D.S
Density

$$D_m = \frac{m}{V} \quad D_w = \frac{F_w}{V}$$

mass density - same

weight density different $w = mg$, g on moon $\frac{1}{6}$

pg 2) Step Column $S = F/A$

EXAMPLE 4 Pg 317 / 318

$$A = 2500 \text{ cm}^2 \times \left(\frac{1\text{m}}{100\text{cm}}\right)^2 = .250 \text{ m}^2$$

$$F = 1.50 \times 10^5 \text{ N}$$

$$S = F/A = (1.5 \times 10^5 \text{ N})(.250 \text{ m}^2) = 600 \text{ k N/m}^2$$

$$= 600 \text{ kPa}$$

Homework

$$\# 2 \quad S = F/A \quad A = 3500 \text{ m}^2 = (3500) \left(\frac{1\text{m}}{100\text{cm}}\right) \left(\frac{1\text{m}}{100\text{cm}}\right) = .35 \text{ m}^2$$

$$F = 2.5 \times 10^5 \text{ N}$$

$$S = (2.5 \times 10^5 \text{ N})(.35 \text{ m}^2) = 714,285.7 \text{ N/m}^2$$

3) Compressed $3.46 \times 10^{-4} \text{ m}$ under weight of 6.42 E+5 N

Example Pg 320 what is compression with weight $5.8 \times 10^6 \text{ N}$

$$K = \frac{F}{A\Delta L} \quad \frac{F = 6.42 \text{ E+5 N}}{3.46 \times 10^{-4} \text{ m}} = 1.855 \text{ E+9 N/m}$$

If $F = 5.8 \times 10^6 \text{ N}$, what is ΔL

$$\Delta L = \frac{F}{K} = \frac{5.8 \times 10^6 \text{ N}}{\cancel{3.46 \times 10^{-4} \text{ m}} \cancel{1.855 \text{ E+9}}} =$$

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(2)

3) $K = F/\Delta l$ $F = 1.3 \times 10^5 N$ $\Delta l = 5.9 \times 10^{-3} cm$

$$\left[5.9 \times 10^{-3} cm \right] \frac{1m}{100cm} = 5.9 \times 10^{-5} m$$

$$K = \frac{1.3 \times 10^5 N}{5.9 \times 10^{-5} m} = 2.20 \times 10^9$$

a) If $F = 5.5 \times 10^5 N$ what is Δl ?

$$\Delta l = F/K = \frac{5.5 \times 10^5}{2.2 \times 10^9} = 2.5 \times 10^{-4} m$$

b) $F = (\Delta l)(K) = (0.0710 cm) \left(\frac{1m}{100cm} \right) 2.2 \times 10^9 = 15,620,000 N$
 $= 1.562 \times 10^7 N$

$$\left(0.0710 cm \frac{1m}{100cm} \right) (2.2 \times 10^9) = (0.0071) (2.2 \times 10^9) \\ = 15,620,000 = 1.562 \times 10^7$$

4) Specific gravity - ratio comparison of density of
substance to water

$\frac{\text{S.G.}}{\text{Water}}$ Gasoline has specific gravity of 0.68
Since unknown denser - will sink

(5) $D_m = \frac{m}{V}$ - Since water decreases volume, but mass
remains the same, $D \rightarrow$ increases

(6) On top of mountain g slightly lower therefore
 $w = mg$ less on mountain top

→ EASIER to SP on mountain top
Harder @ sea level

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(3)

7) Gas flows from larger diameter pipe to smaller
see what happens?

Pg 360

359 a) Speed increases where stream is narrow

b) Bernoulli's Principle - higher speed \rightarrow lower pressure

c) spacing reduced between streamlines

$$8) \text{ Pressure} = \frac{F}{A} \quad r = 250 \text{ cm}^2 \cdot \frac{1 \text{ m}}{100 \text{ cm}} \cdot \frac{1 \text{ m}}{100 \text{ cm}} = .025 \text{ m}$$

Pg 348/349

$$\text{Wind} \quad P_1 = P_2 = \frac{F_1}{A_1} = \frac{F_2}{A_2} \quad A_2 = 2.5 \cdot 10^{-3} \text{ m}^2$$

$$\frac{(1500 \text{ kg})(9.8 \text{ m/s}^2)}{0.025 \text{ m}} = \frac{F_2}{2.5 \cdot 10^{-3} \text{ m}} = \frac{36.75}{.025} = 1470 \text{ N}$$

$$9) Q = VA \quad r = 6.5 \text{ cm} = 6.5 \text{ cm} \cdot \frac{1 \text{ m}}{100 \text{ cm}} = .065 \text{ m}$$

Pg 354

$$A = \pi r^2 = 1.32 \cdot 10^{-2} \text{ m}^2$$

$$Q = (4.53 \text{ m/s}) (1.32 \cdot 10^{-2} \text{ m}) = [6.0127 \text{ m}^3] \left[\frac{10^3 \text{ L}}{\text{m}^3} \right] \left[\frac{60 \text{ s}}{\text{min}} \right]$$

$$= [6.0127 \text{ } \cancel{\frac{\text{m}^3}{\text{s}}}] \left[\frac{10^3 \text{ L}}{\cancel{\text{m}^3}} \right] \left[\frac{60 \text{ s}}{\text{min}} \right] = 3.67 \cdot 10^3 \text{ L/min}$$

$$= 3607 \text{ L/min}$$

b) In 25 minutes

$$(25)(3607) = 90,191.59 \text{ liters}$$

107 GNP becomes wider as metal expands

Pg 345) 11) $\Delta l = \alpha l \Delta T$ α = coefficient of linear expansion
Spec Example 1 page 346
 Steel railroad 40 ft long @ $0^\circ F$
 Heat to $1000^\circ F$
 $\alpha = 6.5 \times 10^{-6} / F^\circ$

$$\Delta l = \alpha l \Delta T = 6.5 \times 10^{-6} (40 \text{ ft}) (100^\circ F) = 0.26 \text{ ft}$$

NW

Problem $\Delta T = 70^\circ C - 20^\circ C = 50^\circ C$

$$l = 1 \text{ m} \quad \alpha = 2.3 \times 10^{-5} / C^\circ$$

$$\Delta l = \alpha l \Delta T = \left(2.3 \times 10^{-5} / C^\circ\right) (1 \text{ m}) (50^\circ C) = 1.15 \times 10^{-3} \text{ m}$$

$$L = 1 + 1.15 \times 10^{-3} = 1.00115 \text{ m}$$

12) $Q = cm \Delta T$ c = specific heat ΔT = temp change

Pg 379 m = mass

Q = heat

$$c = 0.092 \text{ cal/g} (10 \text{ g}) (100^\circ C) = 92 \text{ calories}$$

more heat for water - higher capacity

13) $\Delta T = \frac{Q}{cm} = \frac{1000 \text{ J}}{4.184 \text{ J/g}} = 0.23^\circ C$
 $(4.184 \text{ J})(1000 \text{ g})$