TABLE 2-179 Enthalpies and Gibbs Energies of Formation, Entropies, and Net Enthalpies of Combustion of Inorganic and Organic Compounds at 298.15 K

						<u></u>	ga	
Cmpd.					Ideal gas enthalpy of formation,	Ideal gas Gibbs energy of formation,	Ideal gas entropy,	Standard net enthalpy of combustion,
no.	Name	Formula	CAS no.	Mol. wt.	J/kmol × 1E-07	J/kmol × 1E-07	J/(kmol·K) × 1E-05	J/kmol × 1E-09
1	Acetaldehyde	C ₂ H ₄ O	75-07-0	44.053	-16.64	-13.33	2.642	-1.1045
2	Acetamide	C ₂ H ₅ NO	60-35-5	59.067	-23.83	-15.96	2.722	-1.0741
3	Acetic acid	$C_2H_4O_2$	64-19-7	60.052	-46.11	-40.3	2.825	-0.7866
4	Acetic anhydride	$C_4H_6O_3$	108-24-7	102.089	-57.25	-47.34	3.899	-1.675
5	Acetone	$C_3H_6O_3$	67-64-1	58.079	-21.57	-15.13	2.954	-1.659
6	Acetonic	C ₂ H ₃ N	75-05-8	41.052	7.404	9.1868	2.4329	-1.19043
7	Acetylene	C_2H_3 1V C_2H_2	74-86-2	26.037	22.82	21.068	2.0081	-1.15045 -1.257
8	Acrolein	C_2H_2 C_3H_4O	107-02-8	56.063	-8.18	-5.68	2.97	-1.5468
9	Acrolem Acrylic acid	$C_3H_4O_2$	79-10-7	72.063	-35.591	-30.6 -30.6	3.15	-1.3466 -1.32717
10	Acrylic acid Acrylonitrile	$C_3H_4O_2$ C_3H_3N	107-13-1	53.063	-55.591 18.37	19.37	2.753	-1.52717 -1.69
11	Acrylomeme	Mixture	132259-10-0	28.960	0	0	1.99	0
12	Ammonia	H ₃ N	7664-41-7	17.031	-4.5898	-1.64	1.9266	-0.31683
13	Anisole	C_7H_8O	100-66-3	108.138	-6.79	2.27	3.61	-3.6072
14	Argon	Ar	7440-37-1	39.948	0	0	1.54737	0
15	Benzamide	C_7H_7NO	55-21-0	121.137	-10.09	-0.211	3.641	-3.39877
16	Benzene	C_6H_6	71-43-2	78.112	8.288	12.96	2.693	-3.136
17	Benzenethiol	C_6H_6S	108-98-5	110.177	11.15	14.76	3.369	-3.4474
18	Benzoic acid	$C_7H_6O_2$	65-85-0	122.121	-29.41	-21.42	3.69	-3.0951
19	Benzonitrile	C_7H_5N	100-47-0	103.121	21.57	25.78	3.21	-3.5238
20	Benzophenone	$C_{13}H_{10}O$	119-61-9	182.218	5.68	17.3	4.4	-6.2876
21	Benzyl alcohol	C_7H_8O	100-51-6	108.138	-9.025	-0.254	3.713	-3.56
22	Benzyl ethyl ether	$C_9H_{12}O$	539-30-0	136.191	-11.5	3.37	4.39	-4.83
23	Benzyl mercaptan	C_7H_8S	100-53-8	124.203	9.33	16.3	3.607	-4.06
24	Biphenyl	$C_{12}H_{10}$	92-52-4	154.208	17.849	27.63	3.9367	-6.248
25	Bromine	Br_2	7726-95-6	159.808	3.091	0.314	2.4535	
26	Bromobenzene	C_6H_5Br	108-86-1	157.008	10.5018	13.8532	3.24386	-3.01917
27	Bromoethane	C ₂ H ₅ Br	74-96-4	108.965	-6.36	-2.582	2.873	-1.285
28	Bromomethane	CH_3Br	74-83-9	94.939	-3.77	-2.819	2.458	-0.70542
29	1,2-Butadiene	C_4H_6	590-19-2	54.090	16.23	19.86	2.93	-2.4617
30	1,3-Butadiene	C_4H_6	106-99-0	54.090	10.924	14.972	2.7889	-2.409
31	Butane	C_4H_{10}	106-97-8	58.122	-12.579	-1.67	3.0991	-2.65732
32	1,2-Butanediol	$C_4H_{10}O_2$	584-03-2	90.121	-44.58	-30.44	4.065	-2.2678
33	1,3-Butanediol	$C_4H_{10}O_2$	107-88-0	90.121	-43.32	-29.18	4.065	-2.2824
34	1-Butanol	$C_4H_{10}O$	71-36-3	74.122	-27.51	-15.07	3.618	-2.454
35	2-Butanol	$C_4H_{10}O$	78-92-2	74.122	-29.29	-16.7	3.566	-2.446
36	1-Butene	C_4H_8	106-98-9	56.106	-0.05	7.041	3.074	-2.5408
37	cis-2-Butene	C_4H_8	590-18-1	56.106	-0.74	6.536	3.012	-2.5339
38	trans-2-Butene	C_4H_8	624-64-6	56.106	-1.1	6.32	2.965	-2.53
39	Butyl acetate	$C_6H_{12}O_2$	123-86-4	116.158	-48.56	-31.26	4.425	-3.28
40	Butylbenzene	$C_{10}H_{14}$	104-51-8	134.218	-1.314	14.54	4.3949	-5.5644
41	Butyl mercaptan	$C_4H_{10}S$	109-79-5	90.187	-8.78	1.139	3.752	-2.9554
42	sec-Butyl mercaptan	$C_4H_{10}S$	513-53-1	90.187	-9.66	0.512	3.667	-2.949
43	1-Butyne	$C_4H_{10}S$ C_4H_6	107-00-6	54.090	16.52	20.225	2.9039	-2.4647
44	Butyraldehyde	C_4H_8O	123-72-8	72.106	-20.7	-11.63	3.4365	-2.3035
45	Butyric acid	$C_4H_8O_2$	107-92-6	88.105	-20.7 -47.58	-36	3.601	-2.008 -2.008
46	Butyronitrile	$C_4H_8O_2$ C_4H_7N	107-32-0	69.105	3.40578	10.8658	3.25432	-2.4148
47	Carbon dioxide	$C_4\Pi_7N$ CO_2	124-38-9	44.010	-39.351	-39.437	2.13677	-2.4146
48	Carbon disulfide	CS_2	75-15-0	76.141	11.69	6.68	2.379	-1.0769
49		CO	630-08-0		-11.053	-13.715	1.97556	-0.283
50	Carbon monoxide Carbon tetrachloride	CCl₄	56-23-5	28.010 153.823	-11.053 -9.581	-13.715 -5.354	3.0991	-0.283 -0.2653
50 51		CF ₄						
	Carbon tetrafluoride		75-73-0	88.004	-92.21 0	-87.76	2.62	0.5286
52	Chlorine	Cl ₂	7782-50-5	70.906		0	2.22972	2.050
53	Chlorobenzene	C ₆ H ₅ Cl	108-90-7	112.557	5.109	9.829	3.1403	-2.976
54	Chloroethane	C ₂ H ₅ Cl	75-00-3	64.514	-11.226	-6.0499	2.7578	-1.2849
55	Chloroform	CHCl ₃	67-66-3	119.378	-10.29	-7.01	2.956	-0.38
56	Chloromethane	CH ₃ Cl	74-87-3	50.488	-8.196	-5.844	2.3418	-0.67538
57	1-Chloropropane	C ₃ H ₇ Cl	540-54-5	78.541	-13.318	-5.261	3.1547	-1.867
58	2-Chloropropane	C ₃ H ₇ Cl	75-29-6	78.541	-14.477	-6.136	3.0594	-1.863

TABLE 2-179 Enthalpies and Gibbs Energies of Formation, Entropies, and Net Enthalpies of Combustion of Inorganic and Organic Compounds at 298.15 K (Continued)

Cmpd.		- 1	0.10		Ideal gas enthalpy of formation,	Ideal gas Gibbs energy of formation,	Ideal gas entropy,	Standard net enthalpy of combustion,
no.	Name	Formula	CAS no.	Mol. wt.	J/kmol×1E-07	J/kmol × 1E-07	J/(kmol·K)×1E-05	J/kmol×1E-09
59	m-Cresol	C_7H_8O	108-39-4	108.138	-13.23	-4.019	3.5604	-3.52783
60	o-Cresol	C_7H_8O	95-48-7	108.138	-12.857	-3.543	3.5259	-3.528
61	p-Cresol	C_7H_8O	106-44-5	108.138	-12.535	-3.166	3.5075	-3.52256
62	Cumene	C_9H_{12}	98-82-8	120.192	0.4	13.79	3.86	-4.951
63	Cyanogen	C_2N_2	460-19-5	52.035	30.9072	29.7598	2.41463	-1.0961
64	Cyclobutane	C ₄ H ₈	287-23-0	56.106	2.85	11.22	2.64396	-2.5678
65	Cyclobatane	C ₆ H ₁₂	110-82-7	84.159	-12.33	3.191	2.97276	-3.656
66	Cyclohexanol	C ₆ H ₁₂ O	108-93-0	100.159	-28.62	-10.95	3.277	-3.4639
67	Cyclohexanore	C ₆ H ₁₀ O	108-94-1	98.143	-22.61	-9.028	3.3426	-3.299
68	Cyclohexene	C ₆ H ₁₀ O	110-83-8	82.144	-0.46	10.77	3.10518	-3.532
69	Cyclopentane	C ₅ H ₁₀	287-92-3	70.133	-7.703	3.885	2.929	-3.0709
70			142-29-0	68.117	3.23	11.05	2.91267	-3.0709 -2.9393
	Cyclopentene	C ₅ H ₈			5.33			
71	Cyclopropane	C ₃ H ₆	75-19-4	42.080		10.44	2.37378	-1.9593
72	Cyclohexyl mercaptan	$C_6H_{12}S$	1569-69-3	116.224	-9.602	4.886	3.646	-3.968
73	Decanal	$C_{10}H_{20}O$	112-31-2	156.265	-33.17	-6.739	5.7912	-5.959
74	Decane	$C_{10}H_{22}$	124-18-5	142.282	-24.946	3.318	5.457	-6.29422
75	Decanoic acid	$C_{10}H_{20}O_2$	334-48-5	172.265	-59.43	-30.5	5.99	-5.72
76	1-Decanol	$C_{10}H_{22}O$	112-30-1	158.281	-39.85	-10.02	5.971	-6.116
77	1-Decene	$C_{10}H_{20}$	872-05-9	140.266	-12.21	12.27	5.433	-6.1809
78	Decyl mercaptan	$C_{10}H_{22}S$	143-10-2	174.347	-21.09	6.165	6.116	-6.6161
79	1-Decyne	$C_{10}H_{18}$	764-93-2	138.250	4.1	25.16	5.263	-6.1037
80	Deuterium	D_2	7782-39-0	4.032	0	0	1.4486	-0.24625
81	1,1-Dibromoethane	$C_2H_4Br_2$	557-91-5	187.861	-4.08	-1.181	3.276	-1.16
82	1,2-Dibromoethane	$C_2H_4Br_2$	106-93-4	187.861	-3.89	-1.054	3.297	-1.1769
83	Dibromomethane	CH ₂ Br ₂	74-95-3	173.835			2.92964	
84	Dibutyl ether	C ₈ H ₁₈ O	142-96-1	130.228	-33.34	-8.827	5.014	-4.94691
85	m-Dichlorobenzene	C ₆ H ₄ Cl ₂	541-73-1	147.002	2.57	7.79	3.4353	-2.825
86	o-Dichlorobenzene	$C_6H_4Cl_2$	95-50-1	147.002	3.02	8.29	3.4185	-2.826
87	p-Dichlorobenzene	$C_6H_4Cl_2$	106-46-7	147.002	2.25	7.67	3.3674	-2.802
88	1,1-Dichloroethane	C ₂ H ₄ Cl ₂	75-34-3	98.959	-12.941	-7.259	3.0501	-1.1104
89	1,2-Dichloroethane	$C_2H_4Cl_2$ $C_2H_4Cl_2$	107-06-2	98.959	-12.979	-7.3945	3.0828	-1.105
90	Dichloromethane	CH ₂ Cl ₂	75-09-2	84.933	-9.552	-6.896	2.7018	-0.51388
91	1,1-Dichloropropane	C ₁ 1 ₂ C ₁₂ C ₃ H ₆ Cl ₂	78-99-9	112.986	-15.08	-6.52	3.448	-0.51566
92	1,2-Dichloropropane	C ₃ H ₆ Cl ₂ C ₃ H ₆ Cl ₂	78-87-5	112.986	-16.28	-8.018	3.548	-1.72
93	Diethanol amine	C ₃ H ₆ C ₁₂ C ₄ H ₁₁ NO ₂	111-42-2	105.136	-10.23 -40.847	-3.013 -22.574	4.29	-1.707 -2.4105
94	Diethalor amine Diethyl amine	C ₄ H ₁₁ NO ₂ C ₄ H ₁₁ N	109-89-7	73.137	-40.547 -7.142	7.308	3.522	-2.4103 -2.8003
95	Diethyl annie Diethyl ether		60-29-7	74.122	-7.142 -25.21	-12.21	3.423	-2.5035 -2.5035
96 96	Diethyl ether Diethyl sulfide	C ₄ H ₁₀ O	352-93-2	90.187	-25.21 -8.356	1.774	3.681	-2.5055 -2.9607
90 97	1,1-Difluoroethane	$C_4H_{10}S$ $C_2H_4F_2$	75-37-6	66.050	-5.556 -49.7	-43.9485	2.824	-2.9007 -0.773662
98	1,2-Difluoroethane		624-72-6					
98 99		$C_2H_4F_2$		66.050	-44.77	-39.19	2.88194	-0.823 -0.183031
	Difluoromethane	CH_2F_2	75-10-5	52.023	-45.23	-42.4747	2.4658	
100	Di-isopropyl amine Di-isopropyl ether	$C_6H_{15}N$	108-18-9	101.190	-14.38	6.42	4.12	-3.99
101	Di-isopropyl ether	$C_6H_{14}O$	108-20-3	102.175	-31.92	-12.48	3.989	-3.70261
102	Di-isopropyl ketone 1,1-Dimethoxyethane	$C_7H_{14}O$	565-80-0	114.185	-31.14	-12.37	4.27	-4.095
103	1,1-Dimethoxyethane	$C_4H_{10}O_2$	534-15-6	90.121	-38.97	-23.8	3.726	-2.394
104	1,2-Dimethoxypropane	$C_5H_{12}O_2$	7778-85-0	104.148	-38.42	-20.11	4.038	-2.996
105	Dimethyl acetylene	C_4H_6	503-17-3	54.090	14.57	18.49	2.833	-2.4189
106	Dimethyl amine	C_2H_7N	124-40-3	45.084	-1.845	6.839	2.7296	-1.6146
107	2,3-Dimethylbutane	C_6H_{14}	79-29-8	86.175	-17.68	-0.3125	3.6592	-3.84761
108	1,1-Dimethylcyclohexane	C_8H_{16}	590-66-9	112.213	-18.1	3.52293	3.65012	-4.8639
109	cis-1,2-Dimethylcyclohexane	C_8H_{16}	2207-01-4	112.213	-17.2172	4.12124	3.7451	-4.87084
110	trans-1,2-Dimethylcyclohexane	C_8H_{16}	6876-23-9	112.213	-17.9996	3.44761	3.70912	-4.86436
111	Dimethyl disulfide	$C_2H_6S_2$	624-92-0	94.199	-2.42	1.516	3.35291	-2.0441
112	Dimethyl ether	C ₂ H ₆ O	115-10-6	46.068	-18.41	-11.28	2.667	-1.3284
113	N,N-Dimethyl formamide	C ₃ H ₇ NO	68-12-2	73.094	-19.17	-8.84	3.26	-1.78871
114	2,3-Dimethylpentane	C7H16	565-59-3	100.202	-19.41	0.5717	4.1455	-4.46075
115	Dimethyl phthalate	$C_{10}H_{10}O_4$	131-11-3	194.184	-60.5	-46.7749	6.6	-4.4662
116	Dimethylsilane	C ₂ H ₈ Si	1111-74-6	60.170	-9.47	-1.925	2.9953	-2.569

117	Dimethyl sulfide	C_2H_6S	75-18-3	62.134	-3.724	0.7302	2.8585	-1.7443
118	Dimethyl sulfoxide	C ₂ H ₆ OS	67-68-5	78.133	-15.046	-8.1441	3.0627	-1.6054
119	Dimethyl terephthalate	$C_{10}H_{10}O_4$	120-61-6	194.184	-64.4	-47.4	5.5	-4.4115
120	1,4-Dioxane	$C_4H_8O_2$	123-91-1	88.105	-31.58	-18.16	3.0012	-2.1863
120								
	Diphenyl ether	$C_{12}H_{10}O$	101-84-8	170.207	5.2	17.5	4.13	-5.8939
122	Dipropyl amine	$C_6H_{15}N$	142-84-7	101.190	-11.6	8.68	4.29	-4.0189
123	Dodecane	$C_{12}H_{26}$	112-40-3	170.335	-29.072	4.981	6.2415	-7.51368
124	Eicosane	$C_{20}H_{42}$	112-95-8	282.547	-45.646	11.57	9.3787	-12.3908
125	Ethane	C_2H_6	74-84-0	30.069	-8.382	-3.192	2.2912	-1.42864
126	Ethanol	C_2H_6O	64-17-5	46.068	-23.495	-16.785	2.8064	-1.235
127	Ethyl acetate	$C_4H_8O_2$	141-78-6	88.105	-44.45	-32.8	3.597	-2.061
128	Ethyl amine	C_2H_7N	75-04-7	45.084	-4.715	3.616	2.848	-1.5874
129	Ethylbenzene	C_8H_{10}	100-41-4	106.165	2.992	13.073	3.6063	-4.3448
130	Ethylbenzoate	C ₈ 11 ₁₀	93-89-0	150.175				
		$C_9H_{10}O_2$			-32.6	-19.05	4.55	-4.41
131	2-Ethyl butanoic acid	$C_6H_{12}O_2$	88-09-5	116.158	-53.78	-35.9	4.23	-3.21203
132	Ethyl butyrate	$C_6H_{12}O_2$	105-54-4	116.158	-48.55	-31.22	4.417	-3.284
133	Ethylcyclohexane	C_8H_{16}	1678-91-7	112.213	-17.15	3.955	3.826	-4.87051
134	Ethylcyclopentane	C ₇ H ₁₄	1640-89-7	98.186	-12.69	4.48	3.783	-4.2839
135	Ethylene	C_2H_4	74-85-1	28.053	5.251	6.844	2.192	-1.323
136	Ethylenediamine	$C_2H_8N_2$	107-15-3	60.098	-1.73	10.3	3.21833	-1.691
137	Ethylene glycol	$C_2H_6O_2$	107-21-1	62.068	-39.22	-30.18	3.04891	-1.0527
138	Ethyleneimine	C_2H_5N	151-56-4	43.068	12.3428	17.7987	2.5062	-1.481
139	Ethylene oxide	C_2H_4O	75-21-8	44.053	-5.263	-1.323	2.4299	-1.218
140	Ethyl formate	$C_3H_6O_2$	109-94-4	74.079	-38.83	-30.31	3.282	-1.50696
141	2-Ethyl hexanoic acid	$C_8H_{16}O_2$	149-57-5	144.211	-55.95	-32.5	5.1	-4.448
142	Ethylhexyl ether	$C_8H_{18}O$	5756-43-4	130.228	-33.37	-9.042	5.076	-4.943
143	Ethylisopropyl ether	C ₅ H ₁₂ O	625-54-7	88.148	-28.58	-12.64	3.8	-3.103
144								-3.4863
	Ethylisopropyl ketone	$C_6H_{12}O$	565-69-5	100.159	-28.61	-13.3	4.069	
145	Ethyl mercaptan	C_2H_6S	75-08-1	62.134	-4.63	-0.4814	2.961	-1.7366
146	Ethyl propionate	$C_5H_{10}O_2$	105-37-3	102.132	-46.36	-31.93	4.025	-2.674
147	Ethylpropyl ether	$C_5H_{12}O$	628-32-0	88.148	-27.22	-11.52	3.881	-3.12
148	Ethyltrichlorosilane	C ₂ H ₅ Cl ₃ Si	115-21-9	163.506	-59.54	-51.01	4.07	-1.671
					0	0		1.071
149	Fluorine	F ₂	7782-41-4	37.997			2.02682	
150	Fluorobenzene	C_6H_5F	462-06-6	96.102	-11.6566	-6.9036	3.02629	-2.81451
151	Fluoroethane	C_2H_5F	353-36-6	48.060	-26.44	-21.23	2.644	-1.127
152	Fluoromethane	CH ₃ F	593-53-3	34.033	-23.43	-21.04	2.22734	-0.5219
153	Formaldehyde	CH ₂ O	50-00-0	30.026	-10.86	-10.26	2.1866	-0.5268
154	Formamide	CH ₃ NO	75-12-7	45.041	-19.22	-14.71	2.4857	-0.5021
155	Formic acid	CH_2O_2	64-18-6	46.026	-40.55	-37.78	2.487	-0.2115
156	Furan	C_4H_4O	110-00-9	68.074	-3.48	0.08225	2.6714	-1.9959
157	Helium-4	He	7440-59-7	4.003	0	0	1.26044	0
158	Heptadecane	C ₁₇ H ₃₆	629-78-7	240.468	-39.445	9.083	8.2023	-10.5618
		C ₁₇ 11 ₃₆						
159	Heptanal	$C_7H_{14}O$	111-71-7	114.185	-26.94	-9.191	4.6138	-4.136
160	Heptane	C_7H_{16}	142-82-5	100.202	-18.765	0.8165	4.2798	-4.46473
161	Heptanoic acid	$C_7H_{14}O_2$	111-14-8	130.185	-53.62	-33.4	4.8	-3.839
162	1-Heptanol	$C_7H_{16}O$	111-70-6	116.201	-33.68	-12.55	4.795	-4.285
163	2-Heptanol	C ₇ H ₁₆ O	543-49-7	116.201	-35.54	-14.25	4.74	-4.282
164			106-35-4	114.185	-30.1	-12.25		-4.098
	3-Heptanone	$C_7H_{14}O$					4.58	
165	2-Heptanone	$C_7H_{14}O$	110-43-0	114.185	-30.0453	-11.96	4.486	-4.09952
166	1-Heptene	C_7H_{14}	592-76-7	98.186	-6.289	9.482	4.252	-4.3499
167	Heptyl mercaptan	$C_7H_{16}S$	1639-09-4	132.267	-14.95	3.622	4.939	-4.7865
168	1-Heptyne	C ₇ H ₁₂	628-71-7	96.170	10.3	22.7	4.085	-4.2717
169					-37.417		7.8102	
	Hexadecane	$C_{16}H_{34}$	544-76-3	226.441		8.216		-9.95145
170	Hexanal	$C_6H_{12}O$	66-25-1	100.159	-24.86	-10.005	4.2214	-3.52
171	Hexane	C_6H_{14}	110-54-3	86.175	-16.694	-0.006634	3.8874	-3.8551
172	Hexanoic acid	$C_6H_{12}O_2$	142-62-1	116.158	-51.19	-33.8	4.41	-3.23
173	1-Hexanol	$C_6H_{14}O$	111-27-3	102.175	-31.62	-13.39	4.402	-3.675
174	2-Hexanol	$C_6H_{14}O$	626-93-7	102.175	-33.46	-15.06	4.349	-3.67
175	2-Hexanone	$C_6H_{12}O$	591-78-6	100.159	-27.9826	-13.0081	4.17856	-3.49
176	3-Hexanone	$C_6H_{12}O$	589-38-8	100.159	-27.76	-12.6	4.092	-3.492
177	1-Hexene	C ₆ H ₁₂	592-41-6	84.159	-4.167	8.7	3.863	-3.7397
178			928-49-4	82.144	10.6	19.9	3.76	-3.64
	3-Hexyne	C_6H_{10}						
179	Hexyl mercaptan	$C_6H_{14}S$	111-31-9	118.240	-12.92	2.759	4.546	-4.1762

TABLE 2-179 Enthalpies and Gibbs Energies of Formation, Entropies, and Net Enthalpies of Combustion of Inorganic and Organic Compounds at 298.15 K (Continued)

IADEL Z	177 Elimapies and Olbbs	Elicigies of To	imanon, Emopi	cs, and rich E	innaipies of combos	mon or morganic and or	game compounds at 2	i (Commoca)
Cmpd.		_		_	Ideal gas enthalpy of formation,	Ideal gas Gibbs energy of formation,	Ideal gas entropy,	Standard net enthalpy of combustion,
no.	Name	Formula	CAS no.	Mol. wt.	J/kmol×1E-07	J/kmol×1E-07	J/(kmol·K)×1E-05	J/kmol × 1E-09
180	1-Hexyne	C_6H_{10}	693-02-7	82.144	12.37	21.85	3.694	-3.661
181	2-Hexyne	C_6H_{10}	764-35-2	82.144	10.5	19.9	3.72	-3.64
182	Hydrazine	H_4N_2	302-01-2	32.045	9.5353	15.917	2.3861	-0.5342
183	Hydrogen	H_2	1333-74-0	2.016	0.0000	0	1.30571	-0.24182
184		HBr	10035-10-6	80.912	-3.629	-5.334	1.98591	-0.06904
	Hydrogen bromide							
185	Hydrogen chloride	HCl	7647-01-0	36.461	-9.231	-9.53	1.86786	-0.0286
186	Hydrogen cyanide	CHN	74-90-8	27.025	13.5143	12.4725	2.01719	-0.62329
187	Hydrogen fluoride	HF	7664-39-3	20.006	-27.33	-27.54	1.7367	0.1524
188	Hydrogen sulfide	H_2S	7783-06-4	34.081	-2.063	-3.344	2.056	-0.518
189	Isobutyric acid	$C_4H_8O_2$	79-31-2	88.105	-48.41	-36.21	3.412	-2.0004
190	Isopropyl amine	C_3H_9N	75-31-0	59.110	-8.38	3.192	3.124	-2.1566
191	Malonic acid	$C_3H_4O_4$	141-82-2	104.061	-76.68	-67	3.7	-0.7732
192	Methacrylic acid	$C_4H_6O_2$	79-41-4	86.089	-36.8	-28.8	3.5	-1.93
193	Methane	CH_4	74-82-8	16.042	-7.452	-5.049	1.8627	-0.80262
194	Methanol	CH ₄ O	67-56-1	32.042	-20.094	-16.232	2.3988	-0.6382
195	N-Methyl acetamide	C ₃ H ₇ NO	79-16-3	73.094	-24	-13.5	3.2	-1.71
196	Methyl acetate	$C_3H_6O_2$	79-20-9	74.079	-41.19	-32.42	3.198	-1.461
197	Methyl acetylene	C_3H_4	74-99-7	40.064	18.49	19.384	2.4836	-1.8487
198	Methyl acrylate	$C_4H_6O_2$	96-33-3	86.089	-33.3	-25.7	3.66	-1.9303
199	Methyl amine	CH_5N	74-89-5	31.057	-2.297	3.207	2.433	-0.97508
200	Methyl benzoate	$C_8H_8O_2$	93-58-3	136.148	-28.79	-18.1	4.14	-3.772
201	3-Methyl-1,2-butadiene	C_5H_8	598-25-4	68.117	12.908	19.75	3.2151	-3.032
202	2-Methylbutane	C_5H_{12}	78-78-4	72.149	-15.37	-1.405	3.4374	-3.23954
203	2-Methylbutanoic acid	$C_5H_{10}O_2$	116-53-0	102.132	-49.8	-34.99	3.9	-2.622
204	3-Methyl-1-butanol	$C_5H_{12}O$	123-51-3	88.148	-30.3	-14.1	3.869	-3.062
205	2-Methyl-1-butene	C_5H_{10}	563-46-2	70.133	-3.53	6.668	3.395	-3.1159
206	2-Methyl-1-butene	C_5H_{10}	513-35-9	70.133	-4.18	6.045	3.386	-3.1133
207	2-Methyl -1-butene-3-yne	C_5H_{10} C_5H_6	78-80-8	66.101	26	30.25	2.78	-3.1066 -2.93
	2-Metnyl -1-butene-3-yne	C_5H_6						
208	Methylbutyl ether	$C_5H_{12}O$	628-28-4	88.148	-25.81	-10.17	3.901	-3.12818
209	Methylbutyl sulfide	$C_5H_{12}S$	628-29-5	104.214	-10.2	2.691	4.118	-3.5723
210	3-Methyl-1-butyne	C_5H_8	598-23-2	68.117	13.8	20.72	3.189	-3.046
211	Methyl butyrate	$C_5H_{10}O_2$	623-42-7	102.132	-45.07	-30.53	3.988	-2.686
212	Methylchlorosilane	CH ₅ ClSi	993-00-0	80.589	-21.5	-16.61	2.98277	-1.693
213	Methylcyclohexane	C_7H_{14}	108-87-2	98.186	-15.48	2.733	3.433	-4.25714
214	1-Methylcyclohexanol	$C_7H_{14}O$	590-67-0	114.185	-33.2	-12.9	3.75	-4.058
215	cis-2-Methylcyclohexanol	$C_7H_{14}O$	7443-70-1	114.185	-32.7	-12.68	3.853	-4.0574
216	trans-2-Methylcyclohexanol	$C_7H_{14}O$	7443-52-9	114.185	-35.26	-15.24	3.853	-4.0318
217	Methylcyclopentane	C_6H_{12}	96-37-7	84.159	-10.62	3.63	3.399	-3.6741
218	1-Methylcyclopentene	C_6H_{10}	693-89-0	82.144	-0.38	10.38	3.264	-3.534
219	3-Methylcyclopentene	C_6H_{10}	1120-62-3	82.144	0.74	11.38	3.305	-3.5464
220	Methyldichlorosilane	CH ₄ Cl ₂ Si	75-54-7	115.034	-40.2	-34.83	3.287	-1.357
221	Methylethyl ether	C ₃ H ₈ O	540-67-0	60.095	-21.64	-11.71	3.0881	-1.9314
222	Methylethyl ketone	C_4H_8O	78-93-3	72.106	-23.9	-14.7	3.394	-2.268
223	Methylethyl sulfide	C_3H_8S	624-89-5	76.161	-5.96	1.147	3.332	-2.354
224	Methyl formate	$C_2H_4O_2$	107-31-3	60.052	-35.24	-29.5	2.852	-0.8924
225	Methylisobutyl ether		625-44-5	88.148	-35.24 -26.6	-29.5 -10.7	3.81	-0.3924 -3.122
		$C_5H_{12}O$						
226	Methylisobutyl ketone	$C_6H_{12}O$	108-10-1	100.159	-28.64	-13.51	4.129	-3.4762
227	Methyl Isocyanate	C_2H_3NO	624-83-9	57.051	-6.24	0.0244	1.955	-1.06
228	Methylisopropyl ether	$C_4H_{10}O$	598-53-8	74.122	-25.2	-12.18	3.416	-2.5311
229	Methylisopropyl ether Methylisopropyl ketone	$C_5H_{10}O$	563-80-4	86.132	-26.26	-13.93	3.699	-2.877
230	Methylisopropyl sulfide	$C_4H_{10}S$	1551-21-9	90.187	-8.96	1.4509	3.59	-2.957
231	Methyl mercaptan	CH_4S	74-93-1	48.107	-2.29	-0.98	2.55	-1.1517
232	Methyl methacrylate	$C_5H_8O_2$	80-62-6	100.116	-36	-25.4	4.01	-2.54
233	2-Methyloctanoic acid	C ₉ H ₁₈ O ₂	3004-93-1	158.238	-57.95	-31.8	5.533	-5.056
234	2-Methylpentane	C_6H_{14}	107-83-5	86.175	-17.455	-0.5338	3.8089	-3.84915
235	Methyl pentyl ether	$C_6H_{14}O$	628-80-8	102.175	-27.8	-9.35	4.32	-3.739
236	2-Methylpropane	$C_{6}H_{10}$	75-28-5	58.122	-13.499	-9.55 -2.144	2.955	-2.64895
237	2-Methyl-2-propanol	$C_4\Pi_{10}$ $C_4\Pi_{10}$ O	75-65-0	74.122	-31.24	-2.144 -17.76	3.263	-2.4239 -2.4239
201	- 2 11Ctily1-2-propanoi	· 0411100	10-00-0	17.122	01.27	11.10	0.200	2.4200

238	2-Methyl propene	C_4H_8	115-11-7	56.106	-1.71	5.808	2.9309	-2.5242
239	Methyl propionate	$C_4H_8O_2$	554-12-1	88.105	-42.75	-31.1	3.596	-2.078
240	Methylpropyl ether	$C_4H_{10}O$	557-17-5	74.122	-23.82	-11.1	3.52	-2.51739
241		$C_4H_{10}S$	3877-15-4	90.187	-8.23	1.793	3.717	-2.962
	Methylpropyl sulfide							
242	Methylsilane	CH ₆ Si	992-94-9	46.144	-2.91	1.853	2.565	-1.999
243	alpha-Methyl styrene	C_9H_{10}	98-83-9	118.176	11.83	21.73	3.725	-4.8214
244	Methyl <i>tert</i> -butyl ether	$C_5H_{12}O$	1634-04-4	88.148	-28.32	-11.7	3.578	-3.105
245	Methyl vinyl ether	C_3H_6O	107-25-5	58.079	-10.8	-4.73	3.08	-1.77431
246	Naphthalene	$C_{10}H_{8}$	91-20-3	128.171	15.058	22.408	3.3315	-4.9809
247	Neon	Ne	7440-01-9	20.180	0	0	1.46219	0
						-0.6125		
248	Nitroethane	$C_2H_5NO_2$	79-24-3	75.067	-10.21		3.168	-1.25
249	Nitrogen	N_2	7727-37-9	28.013	0	0	1.915	
250	Nitrogen trifluoride	F_3N	7783-54-2	71.002	-13.2089	-9.06	2.6062	
251	Nitromethane	CH_3NO_2	75-52-5	61.040	-7.47	-0.6934	2.751	-0.6432
252	Nitrous oxide	N ₂ O	10024-97-2	44.013	8.205	10.416	2.1985	-0.0820482
253	Nitric oxide	NO	10102-43-9	30.006	9.025	8.657	2.106	-0.0902489
254	Nonadecane	$C_{19}H_{40}$	629-92-5	268.521	-43.579	10.74	8.9866	-11.7812
		C191140						
255	Nonanal	$C_9H_{18}O$	124-19-6	142.239	-31.09	-7.553	5.3988	-5.35
256	Nonane	C_9H_2O	111-84-2	128.255	-22.874	2.498	5.064	-5.68455
257	Nonanoic acid	$C_9H_{18}O_2$	112-05-0	158.238	-57.73	-31.7	5.59	-5.061
258	1-Nonanol	$C_9H_{20}O$	143-08-8	144.255	-37.79	-10.86	5.579	-5.506
259	2-Nonanol	$C_9H_{20}O$	628-99-9	144.255	-39.71	-12.61	5.523	-5.506
260	1-Nonene	C_9H_{18}	124-11-8	126.239	-10.35	11.23	5.041	-5.5716
261		C ₉ H ₂₀ S	1455-21-6	160.320	-19.08	5.28	5.724	-6.006
	Nonyl mercaptan							
262	1-Nonyne	C_9H_{16}	3452-09-3	124.223	6.17	24.34	4.8699	-5.493
263	Octadécane	$C_{18}H_{38}$	593-45-3	254.494	-41.512	9.91	8.5945	-11.1715
264	Octanal	$C_8H_{16}O$	124-13-0	128.212	-29.02	-8.377	5.0063	-4.74
265	Octane	C_8H_{18}	111-65-9	114.229	-20.875	1.6	4.6723	-5.07415
266	Octanoic acid	$C_8H_{16}O_2$	124-07-2	144.211	-55.6	-32.5	5.2	-4.448
267	1-Octanol	$C_8H_{18}O$	111-87-5	130.228	-35.73	-11.7	5.187	-4.895
268	2-Octanol	$C_8H_{18}O$	123-96-6	130.228	-37.62	-13.43	5.132	-4.894
269	2-Octanone	$C_8H_{16}O$	111-13-7	128.212	-32.16	-11.38	4.962	-4.6984
270	3-Octanone	$C_8H_{16}O$	106-68-3	128.212	-33.9	-12.81	4.879	-4.711
271	1-Octene	C_8H_{16}	111-66-0	112.213	-8.194	10.57	4.637	-4.961
272	Octyl mercaptan	$C_8H_{18}S$	111-88-6	146.294	-17.01	4.457	5.331	-5.3962
273	1-Octyne	C_8H_{14}	629-05-0	110.197	8.23	23.5	4.478	-4.88145
274	Oxalic acid	$C_2H_2O_4$	144-62-7	90.035	-72.37	-66.14	3.433	-0.1989
275		O_2	7782-44-7	31.999	0	0	2.05043	0
	Oxygen							
276	Ozone	O_3	10028-15-6	47.998	14.2671	16.3164	2.38823	-0.142671
277	Pentadecane	$C_{15}H_{32}$	629-62-9	212.415	-35.311	7.426	7.4181	-9.34237
278	Pentanal	$C_5H_{10}O$	110-62-3	86.132	-22.78	-10.71	3.8289	-2.91
279	Pentane	C_5H_{12}	109-66-0	72.149	-14.676	-0.8813	3.4945	-3.24494
280					-49.13			
	Pentanoic acid	$C_5H_{10}O_2$	109-52-4	102.132		-34.7	4.02	-2.617
281	1-Pentanol	$C_5H_{12}O$	71-41-0	88.148	-29.57	-14.23	4.01	-3.064
282	2-Pentanol	$C_5H_{12}O$	6032-29-7	88.148	-31.37	-15.88	3.958	-3.058
283	2-Pentanone	$C_5H_{10}O$	107-87-9	86.132	-25.92	-13.83	3.786	-2.87956
284	3-Pentanone	$C_5H_{10}O$	96-22-0	86.132	-25.79	-13.44	3.7	-2.8804
285			109-67-1	70.133	-2.162			
	1-Pentene	C_5H_{10}				7.837	3.462	-3.13037
286	2-Pentyl mercaptan	$C_5H_{12}S$	2084-19-7	104.214	-11.3	1.814	4.05	-3.564
287	Pentyl mercaptan	$C_5H_{12}S$	110-66-7	104.214	-10.84	1.94408	4.154	-3.5641
288	1-Pentyne	C_5H_8	627-19-0	68.117	14.44	21.03	3.298	-3.051
289	2-Pentyne	C_5H_8	627-21-4	68.117	12.89	19.45	3.3084	-3.0291
290	Phenanthrene	$C_{14}H_{10}$	85-01-8	178.229	20.12	30.219	3.945	-6.8282
291	Phenol	C_6H_6O	108-95-2	94.111	-9.6399	-3.2637	3.1481	-2.921
292	Phenyl isocyanate	C ₇ H ₅ NO	103-71-9	119.121	-1.454	4.87212	3.527	-3.298
293	Phthalic anhydride	$C_8H_4O_3$	85-44-9	148.116	-37.14	-30.7001	3.995	-3.1715
294	Propadiene '	C_3H_4	463-49-0	40.064	19.05	20.08	2.439	-1.8563
295	Propane	C_3H_8	74-98-6	44.096	-10.468	-2.439	2.702	-2.04311
296	1-Propanol	C_3H_8O	71-23-8	60.095	-25.46	-15.99	3.226	-1.844
297		C_3H_8O	67-63-0	60.095	-27.21	-17.52	3.175	-1.834
	2-Propanol							
298	Propenylcyclohexene	C_9H_{14}	13511-13-2	122.207	4.677	20.85	4.233	-5.232
299	Propionaldehyde	C_3H_6O	123-38-6	58.079	-18.63	-12.46	3.044	-1.6857
300	Propionie acid	$C_3H_6O_2$	79-09-4	74.079	-47.99	-38.5	2.949	-1.395

TABLE 2-179 Enthalpies and Gibbs Energies of Formation, Entropies, and Net Enthalpies of Combustion of Inorganic and Organic Compounds at 298.15 K (Concluded)

Cmpd.	Name	Formula	CAS no.	Mol. wt.	Ideal gas enthalpy of formation, J/kmol × 1E-07	Ideal gas Gibbs energy of formation, J/kmol × 1E-07	Ideal gas entropy, J/(kmol·K) × 1E-05	Standard net enthalpy of combustion, J/kmol × 1E-09
301	Propionitrile	C ₃ H ₅ N	107-12-0	55.079	5.18	9.74949	2.8614	-1.8007
302	Propyl acetate	$C_5H_{10}O_2$	109-60-4	102.132	-46.48	-32.04	4.023	-2.672
303	Propyl amine	C ₃ H ₉ N	107-10-8	59.110	-7.05	4.17	3.242	-2.165
304	Propylbenzene	C ₉ H ₁₂	103-65-1	120.192	0.79	13.76	4.0014	-4.95415
305	Propylene	C ₃ H ₆	115-07-1	42.080	2.023	6.264	2.67	-1.9262
306	Propyl formate	C ₄ H ₈ O ₂	110-74-7	88.105	-40.76	-29.36	3.678	-2.041
307	2-Propyl mercaptan	C ₃ H ₈ S	75-33-2	76.161	-7.59	-0.218	3.243	-2.3398
308	Propyl mercaptan	C ₃ H ₈ S	107-03-9	76.161	-6.75	0.2583	3.365	-2.3458
309	1,2-Propylene glycol	$C_3H_8O_2$	57-55-6	76.094	-42.15	-30.4	3.52	-1.6476
310	Ouinone	$C_6H_4O_2$	106-51-4	108.095	-12.29	-6.92	3.205	-2.658
311	Silicon tetrafluoride	F_4Si	7783-61-1	104.079	-161.494	-157.27	2.82651	0.7055
312	Styrene	C ₈ H ₈	100-42-5	104.149	14.74	21.39	3.451	-4.219
313	Succinic acid	$C_4H_6O_4$	110-15-6	118.088	-82.29	-69.73	4.034	-1.3591
314	Sulfur dioxide	O_2S	7446-09-5	64.064	-29.684	-30.012	2.481	1.5551
315	Sulfur hexafluoride	F ₆ S	2551-62-4	146.055	-122.047	-111.653	2.91625	0.924
316	Sulfur trioxide	O ₃ S	7446-11-9	80.063	-39.572	-37.095	2.5651	0.1422
317	Terephthalic acid	C ₈ H ₆ O ₄	100-21-0	166.131	-71.79	-59.9	4.48	-3.0576
318	o-Terphenyl	$C_{18}H_{14}$	84-15-1	230.304	27.66	42.3	5.263	-9.053
319	Tetradecane	C ₁₈ H ₁₄ C ₁₄ H ₃₀	629-59-4	198.388	-33.244	6.599	7.0259	-8.73282
320	Tetrahydrofuran	C ₁₄ H ₈ O	109-99-9	72.106	-33.244 -18.418	-7.969	2.9729	-0.73262 -2.325
321	1,2,3,4-Tetrahydronaphthalene	$C_{10}H_{12}$	119-64-2	132.202	2.661	16.71	3.6964	-2.325 -5.3575
322	Tetrahydrothiophene	$C_{10}H_{12}$ C_4H_8S	119-04-2	88.171	-3.376	4.59	3.1	-3.3373 -2.76549
323	2,2,3,3-Tetramethylbutane	C ₄ H ₈ S C ₈ H ₁₈	594-82-1	114.229	-3.576 -22.56	2.239	3.893	-2.76349 -5.0639
323		C ₈ H ₁₈	110-02-1	84.140		12.67	2.784	-5.0639 -2.4352
324	Thiophene Toluene	C ₄ H ₄ S			11.544 5.017	12.67	3.2099	-2.4352 -3.734
	1,1,2-Trichloroethane	C ₇ H ₈	108-88-3	92.138				-3.734 -0.9685
326 327		C ₂ H ₃ Cl ₃	79-00-5	133.404	-14.2	-8.097 5.771	3.371	
327 328	Tridecane	C ₁₃ H ₂₈	629-50-5	184.361	-31.177		6.6337	-8.1229
	Triethyl amine	$C_6H_{15}N$	121-44-8	101.190	-9.58	11.41	4.054	-4.0405
329	Trimethyl amine	C_3H_9N	75-50-3	59.110	-2.431	9.899	2.87	-2.2449
330	1,2,3-Trimethylbenzene	C_9H_{12}	526-73-8	120.192	-0.95	12.61	3.805	-4.934
331	1,2,4-Trimethylbenzene	C_9H_{12}	95-63-6	120.192	-1.38	11.71	3.961	-4.9307
332	2,2,4-Trimethylpentane	C_8H_{18}	540-84-1	114.229	-22.401	1.394	4.2296	-5.06528
333	2,3,3-Trimethylpentane	C_8H_{18}	560-21-4	114.229	-21.845	1.828	4.2702	-5.06876
334	1,3,5-Trinitrobenzene	$C_6H_3N_3O_6$	99-35-4	213.105	6.24	26.79	4.435	-2.6867
335	2,4,6-Trinitrotoluene	$C_7H_5N_3O_6$	118-96-7	227.131	4.34	28.44	4.607	-3.2959
336	Undecane	$C_{11}H_{24}$	1120-21-4	156.308	-27.043	4.116	5.8493	-6.9036
337	1-Undecanol	$C_{11}H_{24}O$	112-42-5	172.308	-41.9	-9.177	6.363	-6.726
338	Vinyl acetate	$C_4H_6O_2$	108-05-4	86.089	-31.49	-22.79	3.28	-1.95
339	Vinyl acetylene	C_4H_4	689-97-4	52.075	30.46	30.6	2.794	-2.362
340	Vinyl chloride	C_2H_3Cl	75-01-4	62.498	2.845	4.195	2.7354	-1.178
341	Vinyl trichlorosilane	C ₂ H ₃ Cl ₃ Si	75-94-5	161.490	-48.116	-42.5514	3.73966	-1.544
342	Water	H_2O	7732-18-5	18.015	-24.1814	-22.859	1.88724	
343	m-Xylene	C_8H_{10}	108-38-3	106.165	1.732	11.876	3.5854	-4.3318
344	o-Xýlene	C_8H_{10}	95-47-6	106.165	1.908	12.2	3.5383	-4.333
345	p-Xylene	C_8H_{10}	106-42-3	106.165	1.803	12.14	3.52165	-4.333

The compounds are considered to be formed from the elements in their standard states at 298.15 K and 101,325 Pa. These include C (graphite) and S (rhombic). Enthalpy of combustion is the net value for the compound in its standard state at 298.15 K and 101,325 Pa. Products of combustion are taken to be CO_2 (gas), H_2O (gas), Cl_2 (gas), Cl_2

Values in this table were taken from the Design Institute for Physical Properties (DIPPR) of the American Institute of Chemical Engineers (AIChE), copyright 2007 AIChE and reproduced with permission of AICHE and of the DIPPR Evaluated Process Design Data Project Steering Committee. Their source should be cited as R. L. Rowley, W. V. Wilding, J. L. Oscarson, Y. Yang, N. A. Zundel, T. E. Daubert, R. P. Danner, DIPPR® Data Compilation of Pure Chemical Properties, Design Institute for Physical Properties, AIChE, New York (2007).