TABLE 2-153 Heat Capacities of Inorganic and Organic Liquids [J/(kmol·K)]

IADLE	2-153 Heat Capacities of I	norganic c	ina Organic	ridnias la	/(KMOI·K)]			1					
Cmpd. no.	Name	Formula	CAS no.	Mol. wt.	C1	C2	C3	C4	C5	T_{\min} , K	C_p at T_{\min} ×1E-05	T _{max} , K	C_p at T_{max} × 1E-05
1 2	Acetaldehyde Acetamide	C_2H_4O C_2H_5NO	75-07-0 60-35-5	44.053 59.067	115,100 102,300	-433 128.7	1.425			150.15 354.15	0.8221 1.4788	294.00 571.00	1.1097 1.7579
3	Acetic acid	$C_2H_4O_2$	64-19-7	60.052	139,640	-320.8	0.8985			289.81	1.2213	391.05	1.7579
4	Acetic acid Acetic anhydride	$C_{2}\Pi_{4}O_{2}$ $C_{4}H_{6}O_{3}$	108-24-7	102.089	36,600	511	0.0305			250.00	1.6435	350.00	2.1545
5	Acetone	C ₄ H ₆ O ₃ C ₃ H ₆ O	67-64-1	58.079	135,600	-177	0.2837	0.000689		178.45	1.1696	329.44	1.3271
6	Acetonie	C_2H_3N	75-05-8	41.052	97,582	-177 -122.2	0.2657	0.000009		229.32	0.8748	354.75	0.9713
7	Acetylene	C_2H_3	74-86-2	26.037	-122,020	3,082.7	-15.895	0.027732		192.40	0.8021	250.00	0.8853
8	Acrolein	C ₂ H ₂ O	107-02-8	56.063	103,090	-247.8	1.0343	0.021102		253.00	1.0660	379.50	1.5801
9	Acrylic acid	C ₃ H ₄ O ₂	79-10-7	72.063	55,300	300	1.0040			286.15	1.4114	375.00	1.6780
10	Acrylonitrile	C ₃ H ₄ O ₂ C ₃ H ₃ N	107-13-1	53.063	109,900	-109.75	0.35441			189.63	1.0183	400.00	1.2271
11	Air	Mixture	132259-10-0	28.960	-214,460	9,185.1	-106.12	0.41616		75.00	0.5307	115.00	0.7132
12	Ammonia [use Eq. (2)]	H ₃ N	7664-41-7	17.031	61.289	80,925	799.4	-2,651		203.15	0.7575	401.15	4.1847
13	Anisole	C ₇ H ₈ O	100-66-3	108.138	150,940	93.455	0.23602	2,001		298.15	1.9978	484.20	2.5153
14	Argon	Ar	7440-37-1	39.948	134,390	-1,989.4	11.043			83.78	0.4523	135.00	0.6708
15	Benzamide	C ₇ H ₇ NO	55-21-0	121.137	161,440	260.66				403.00	2.6649	563.15	3.0823
16	Benzene	C_6H_6	71-43-2	78.112	129,440	-169.5	0.64781			278.68	1.3251	353.24	1.5040
16	Benzene	C_6H_6	71-43-2	78.112	162,940	-344.94	0.85562			278.68	1.3326	500.00	2.0437
17	Benzenethiol	C_6H_6S	108-98-5	110.177	119,780	180.34				258.27	1.6636	442.29	1.9954
18	Benzoic acid	$C_7H_6O_2$	65-85-0	122.121	-5,480	647.12				395.45	2.5042	450.00	2.8572
19	Benzonitrile	C ₇ H ₅ N	100-47-0	103.121	93,383	242.61				260.40	1.5656	464.15	2.0599
20	Benzophenone	$C_{13}H_{10}O$	119-61-9	182.218	156,130	454.49				321.35	3.0218	640.00	4.4700
21	Benzyl alcohol	C_7H_8O	100-51-6	108.138	-334,997	3,644.21	-7.77514	0.00591102		257.85	1.8905	478.60	2.7617
22	Benzyl ethyl ether	$C_9H_{12}O$	539-30-0	136.191	87,500	480				275.65	2.1981	458.15	3.0741
23	Benzyl mercaptan	C_7H_8S	100-53-8	124.203	100,320	346.89				243.95	1.8494	472.03	2.6406
24	Biphenyl	$C_{12}H_{10}$	92-52-4	154.208	121,770	429.3				342.20	2.6868	533.37	3.5075
25	Bromine	Br_2	7726-95-6	159.808	179,400	-667.11	1.0701			265.90	0.7768	331.90	0.7587
26	Bromobenzene	C_6H_5Br	108-86-1	157.008	121,600	-9.45	0.358			293.15	1.4960	495.08	2.0467
27	Bromoethane	C_2H_5Br	74-96-4	108.965	94,364	-109.12	0.44032			160.00	0.8818	320.00	1.0453
28	Bromomethane	CH ₃ Br	74-83-9	94.939	129,730	-596.54	2.16	-0.0024234		184.45	0.7798	276.71	0.7870
29	1,2-Butadiene	C_4H_6	590-19-2	54.090	135,150	-311.14	0.97007	-0.0001523		136.95	1.1034	290.00	1.2279
30	1,3-Butadiene	C_4H_6	106-99-0	54.090	128,860	-323.1	1.015	0.000032		165.00	1.0333	350.00	1.4148
31	Butane	C_4H_{10}	106-97-8	58.122	191,030	-1,675	12.5	-0.03874	4.6121E-05	134.86	1.1272	400.00	2.2237
32	1,2-Butanediol [use Eq. (2)]	$C_4H_{10}O_2$	584-03-2	90.121	55.136	314,200	280.19	1,413.9		220.00	1.5590	670.00	5.2045
33	1,3-Butanediol [use Eq. (2)]	$C_4H_{10}O_2$	107-88-0	90.121	42.152	324,580	517.35	1,449.5		196.15	0.6251	670.00	5.2437
34	1-Butanol	$C_4H_{10}O$	71-36-3	74.122	191,200	-730.4	2.2998			183.85	1.3465	391.90	2.5817
35	2-Butanol	$C_4H_{10}O$	78-92-2	74.122	426,790	-3,694.6	13.828	-0.0135		158.45	1.3485	372.90	2.7190
36	1-Butene	C_4H_8	106-98-9	56.106	182,050	-1,611	11.963	-0.037454	4.5027E-05	87.80	1.1015	380.00	1.8103
37	cis-2-Butene	C_4H_8	590-18-1	56.106	126,680	-65.47	-0.64	0.002912		134.26	1.1340	350.00	1.5022
38 39	trans-2-Butene	C ₄ H ₈	624-64-6	56.106	112,760	-104.7	0.5214			167.62	1.0986	274.03	1.2322
40	Butyl acetate Butylbenzene	$C_6H_{12}O_2$ $C_{10}H_{14}$	123-86-4 104-51-8	116.158 134.218	111,850 182,470	384.52 -13.912	0.72897			298.15 185.30	2.2649 2.0492	399.26 400.00	2.6537 2.9354
40	Butyl mercaptan	$C_{10}\Pi_{14}$	109-79-5	90.187	232,190	-13.912 -804.35	2.7063	-0.0023017		157.46	1.6365	390.00	1.9359
41	sec-Butyl mercaptan	$C_4H_{10}S \\ C_4H_{10}S$	513-53-1	90.187	197,890	-604.55 -491.54	1.7219	-0.0023017		133.02	1.6303	370.00	1.8844
42	1-Butyne	$C_4H_{10}S$ C_4H_6	107-00-6	54.090	136,340	-491.54 -300.4	1.7219	-0.0012499		133.02	1.1426	298.15	1.8844
44	Butyraldehyde	C_4H_8O	123-72-8	72.106	65,682	1,329.1	-7.1579	0.012755		176.75	1.4741	300.00	1.6459
45	Butyric acid	$C_4H_8O_2$	107-92-6	88.105	237,700	-746.4	1.829	0.012700		267.95	1.6902	436.42	2.6031
46	Butyronitrile	$C_4H_8O_2$ C_4H_7N	109-74-0	69.105	104,000	174	1.023			161.25	1.3206	390.75	1.7199
47	Carbon dioxide	CO_2	124-38-9	44.010	-8,304,300	104,370	-433.33	0.60052		220.00	0.7827	290.00	1.6603
48	Carbon disulfide	CS_2	75-15-0	76.141	85,600	-122	0.5605	-0.001452	2.008E-06	161.11	0.7577	552.00	1.3125
49	Carbon monoxide [use Eq. (2)]	CO	630-08-0	28.010	65.429	28,723	-847.39	1,959.6		68.15	0.5912	132.00	6.4799
50	Carbon tetrachloride	CCl ₄	56-23-5	153.823	-752,700	8,966.1	-30.394	0.034455		250.33	1.2763	388.71	1.6374
51	Carbon tetrafluoride	CF_4	75-73-0	88.004	104,600	-500.6	2.2851			89.56	0.7810	145.10	0.8007
52	Chlorine	Cl ₂	7782-50-5	70.906	63,936	46.35	-0.1623			172.12	0.6711	239.12	0.6574
53	Chlorobenzene	C ₆ H ₅ Cl	108-90-7	112.557	-1,307,500	15,338	-53.974	0.063483		227.95	1.3617	360.00	1.8101
54	Chloroethane	C ₂ H ₅ Cl	75-00-3	64.514	127,900	-345.15	0.915			134.80	0.9800	340.00	1.1632
55	Chloroform	CHCl ₃	67-66-3	119.378	124,850	-166.34	0.43209			233.15	1.0956	366.48	1.2192
56	Chloromethane	CH₃Cl	74-87-3	50.488	96,910	-207.9	0.37456	0.000488		175.43	0.7460	373.15	0.9684
57	1-Chloropropane	C ₃ H ₇ Cl	540-54-5	78.541	132,280	-153.27	0.50836			150.35	1.2073	319.67	1.3523
58	2-Chloropropane	C ₃ H ₇ Cl	75-29-6	78.541	69,362	215.01				200.00	1.1236	308.85	1.3577

TABLE 2-153 Heat Capacities of Inorganic and Organic Liquids [J/(kmol·K)] (Continued)

Cmpd.	Name	Formula	CAS no.	Mol. wt.	C1	C2	C3	C4	C5	T_{\min} , K	C_p at T_{\min} \times 1E-05	$T_{ m max}$, K	C_p at T_{max} × 1E-05
59	m-Cresol	C ₇ H ₈ O	108-39-4	108.138	-246,700	3,256.8	-7.4202	0.0060467		285.39	2.1895	400.00	2.5578
60	o-Cresol	C_7H_8O	95-48-7	108.138	-185,150	3,148	-8.0367	0.007254		304.20	2.3297	400.00	2.5243
61	p-Cresol	C ₇ H ₈ O	106-44-5	108.138	259,980	-1,112.3	4.9427	-0.0054367		307.93	2.2740	400.00	2.5794
62	Cumene	C ₉ H ₁₂	98-82-8	120.192	61,723	494.81	1.0121	0.0001001		177.14	1.4937	425.56	2.7229
63	Cyanogen	C_2N_2	460-19-5	52.035	65,516	144.7	0.063229			200.08	0.9700	300.08	1.1463
64	Cyclobutane	C_4H_8	287-23-0	56.106	101,920	-215.81	0.8103			190.00	0.9017	298.15	1.0961
65	Cyclohexane	C_6H_{12}	110-82-7	84.159	-220,600	3,118.3	-9.4216	0.010687		279.69	1.4836	400.00	2.0323
66	Cýclohexanol	$C_6H_{12}O$	108-93-0	100.159	-40,000	853				296.60	2.1300	434.00	3.3020
67	Cyclohexanone	$C_6H_{10}O$	108-94-1	98.143	6,110.4	600.94				290.00	1.8038	489.75	3.0042
68	Cyclohexene	C_6H_{10}	110-83-8	82.144	105,850	-60	0.68			169.67	1.1525	356.12	1.7072
69	Cyclopentane	C_5H_{10}	287-92-3	70.133	122,530	-403.8	1.7344	-0.0010975		179.28	0.9956	322.40	1.3584
70	Cyclopentene	C_5H_8	142-29-0	68.117	125,380	-349.7	1.143			138.13	0.9888	317.38	1.2953
71	Cyclopropane	C_3H_6	75-19-4	42.080	89,952	-196.63	0.65237			150.00	0.7514	298.15	0.8932
72	Cyclohexyl mercaptan	$C_6H_{12}S$	1569-69-3	116.224	177,560	-179.12	0.76723			189.64	1.7118	431.95	2.4334
73	Decanal	C ₁₀ H ₂₀ O	112-31-2	156.265	150,460	586.63	1.0505			267.15	3.0718	488.15	4.3682
74	Decane	$C_{10}H_{22}$	124-18-5	142.282	278,620	-197.91	1.0737			243.51	2.9409	460.00	4.1478
75 76	Decanoic acid	$C_{10}H_{20}O_2$	334-48-5	172.265	219,840	140.41	0.9968	0.27520	0.00000074	304.75	3.5521	543.15	5.9017
77	1-Decanol	C ₁₀ H ₂₂ O	112-30-1 872-05-9	158.281	4,988,500	-52,898 -1,616.5	216.35	-0.37538 -0.004348	0.00023674	280.05 206.89	3.5373 2.7541	503.00 443.75	5.0169 3.8250
78	1-Decene Decyl mercaptan	$C_{10}H_{20}$ $C_{10}H_{22}S$	143-10-2	140.266 174.347	417,440 314,570	-1,616.5 -160.93	5.3948 0.95561	-0.004345		247.56	3.3330	512.35	4.8297
79	1-Decyne	$C_{10}H_{18}$	764-93-2	138.250	276,900	-371.23	1.5774			229.15	2.7466	447.15	4.2629
80	Deuterium	D_2	7782-39-0	4.032	210,300	-571.25	1.5774			220.10	2.7400	447.10	4.2023
81	1,1-Dibromoethane	$C_2H_4Br_2$	557-91-5	187.861	149,400	-231.8	0.5946			210.15	1.2695	381.15	1.4743
82	1,2-Dibromoethane	$C_2H_4Br_2$ $C_2H_4Br_2$	106-93-4	187.861	200,560	-491.44	0.9187			282.85	1.3506	410.00	1.5350
83	Dibromomethane	CH_2Br_2	74-95-3	173.835	202,580	-726.3	1.3377			240.00	1.0532	370.10	1.1701
84	Dibutyl ether	C ₈ H ₁₈ O	142-96-1	130.228	270,720	-259.83	0.95427			175.30	2.5450	450.00	3.4704
85	m-Dichlorobenzene	$C_6H_4Cl_2$	541-73-1	147.002	114,880	187.25				248.39	1.6139	400.00	1.8978
86	o-Dichlorobenzene	C ₆ H ₄ Cl ₂	95-50-1	147.002	93,093	183.97	0.2314			273.15	1.6061	528.75	2.5506
87	p-Dichlorobenzene	$C_6H_4Cl_2$	106-46-7	147.002	133,950	-24.84	0.48191			326.14	1.7711	513.56	2.4829
88	1,1-Dichloroethane	C ₂ H ₄ Cl ₂	75-34-3	98.959	126,340	-94.63	0.32			176.19	1.1960	330.45	1.3001
89	1,2-Dichloroethane	$C_2H_4Cl_2$	107-06-2	98.959	179,170	-444.74	0.93009			237.49	1.2601	356.59	1.3885
90	Dichloromethane	CH_2Cl_2	75-09-2	84.933	98,968	-62.941	0.23265			180.00	0.9518	320.00	1.0265
91	1,1-Dichloropropane	$C_3H_6Cl_2$	78-99-9	112.986	144,560	-53.605	0.30617			180.00	1.4483	361.25	1.6515
92	1,2-Dichloropropane	$C_3H_6Cl_2$	78-87-5	112.986	111,560	149.44				275.00	1.5266	369.52	1.6678
93	Diethanol amine	$C_4H_{11}NO_2$	111-42-2	105.136	184,200	286				301.15	2.7033	541.54	3.3908
94 95	Diethyl amine	$C_4H_{11}N$	109-89-7	73.137	101,330	243.18		0.000700		223.35	1.5564	328.60	1.8124
95 96	Diethyl ether Diethyl sulfide	$C_4H_{10}O$ $C_4H_{10}S$	60-29-7 352-93-2	74.122 90.187	44,400 238,520	1,301 -1,038.4	-5.5 4.0587	0.008763 -0.0044691		156.92 181.95	1.4698 1.5703	460.00 322.08	3.3202 1.7579
96 97	1,1-Difluoroethane [use Eq. (2)]	$C_4H_{10}S$ $C_2H_4F_2$	75-37-6	66.050	67.155	-1,038.4 105,580	310.21	-0.0044691 -490.54		154.56	0.9915	359.98	1.6874
98	1,2-Difluoroethane [use Eq. (2)]	$C_{2}H_{4}F_{2}$ $C_{2}H_{4}F_{2}$	624-72-6	66.050	82,577	105,580	310.21	-490.34		215.00	1.0619	283.65	1.1374
99	Difluoromethane	CH ₂ F ₂	75-10-5	52.023	263,980	-1,791.1	4.3666			200.00	0.8042	250.00	0.8912
100	Di-isopropyl amine	$C_6H_{15}N$	108-18-9	101.190	98,434	429.04	1.5000			275.00	2.1642	357.05	2.5162
101	Di-isopropyl ether	$C_6H_{14}O$	108-20-3	102.175	163,000	-4.5	0.62			187.65	1.8399	341.45	2.3375
102	Di-isopropyl ketone	C ₇ H ₁₄ O	565-80-0	114.185	179,270	28.37	0.5375			204.81	2.0763	410.00	2.8126
103	1,1-Dimethoxyethane	$C_4H_{10}O_2$	534-15-6	90.121	187,790	-313.41	1.1023			159.95	1.6586	337.45	2.0755
104	1,2-Dimethoxypropane	$C_5H_{12}O_2$	7778-85-0	104.148	199,930	-191.5	0.87664			226.10	2.0145	366.15	2.4734
105	Dimethyl acetylene	C_4H_6	503-17-3	54.090	88,153	124.16				240.91	1.1806	300.13	1.2542
106	Dimethyl amine	C_2H_7N	124-40-3	45.084	-214,870	3,787.2	-13.781	0.016924		180.96	1.1947	298.15	1.3779
107	2,3-Dimethylbutane	C_6H_{14}	79-29-8	86.175	129,450	18.5	0.608			145.19	1.4495	331.13	2.0224
108	1,1-Dimethylcyclohexane	C_8H_{16}	590-66-9	112.213	134,500	8.765	0.81151			239.66	1.8321	392.70	2.6309
109	cis-1,2-Dimethylcyclohexane	C_8H_{16}	2207-01-4	112.213	150,130	-62.38	0.8851			223.16	1.8029	402.94	2.6870
110	trans-1,2-Dimethylcyclohexane	C_8H_{16}	6876-23-9	112.213	155,560	-145.26	1.0932			184.99	1.6610	396.58	2.6989
111	Dimethyl disulfide	$C_2H_6S_2$	624-92-0	94.199	171,580	-256.67	0.5727			188.44	1.4355	360.00	1.5340
112	Dimethyl ether	C ₂ H ₆ O	115-10-6	46.068	110,100	-157.47	0.51853			131.65	0.9836	250.00	1.0314
113 114	N,N-Dimethyl formamide	C ₃ H ₇ NO	68-12-2	73.094	147,900	-106	0.384 0.604			273.82	1.4767	466.44 380.00	1.8200
114	2,3-Dimethylpentane Dimethyl phthalate	$C_7H_{16} \\ C_{10}H_{10}O_4$	565-59-3 131-11-3	100.202 194.184	146,420 206,560	59.2 325.75	0.004			90.00 274.16	1.5664 2.9587	360.00	2.5613 3.2383
116	Dimethyl phthalate Dimethylsilane	$C_{10}\Pi_{10}O_4$ C_2H_8Si	1111-74-6	60.170	131,810	3∠3.13				298.15	1.3181	298.15	1.3181
110	Dimeniyishane	C211851	1111-14-0	00.170	101,010	'	. '	'	'	200.10	1.0101	200.10	1.0101

117	Dimethyl sulfide	C_2H_6S	75-18-3	62.134	146,950	-380.06	1.2035	-0.00084787		174.88	1.1276	310.48	1.1959
118	Dimethyl sulfoxide	C ₂ H ₆ OS	67-68-5	78.133	240,300	-595	1.013			291.67	1.5293	422.15	1.6965
119	Dimethyl terephthalate	$C_{10}H_{10}O_4$	120-61-6	194.184	190,020	431.04				423.40	3.7252	466.35	3.9104
120	1,4-Dioxane	$C_4H_8O_2$	123-91-1	88.105	956,860	-5,559.9	9.6124			284.95	1.5306	374.47	2.2277
121	Diphenyl ether	$C_{12}H_{10}O$	101-84-8	170.207	134,160	447.67				300.03	2.6847	570.00	3.8933
122	Dipropyl amine	$C_6H_{15}N$	142-84-7	101.190	49,120	562.24				277.90	2.0537	407.90	2.7846
123	Dodecane	$C_{12}H_{26}$	112-40-3	170.335	508,210	-1,368.7	3.1015			263.57	3.6292	330.00	3.9429
124	Eicosane	$C_{20}H_{42}$	112-95-8	282.547	352,720	807.32	0.2122			309.58	6.2299	616.93	9.3154
125	Ethane [use Eq. (2)]	$C_{20}H_{6}$	74-84-0	30.069	44.009	89,718	918.77	-1,886		92.00	0.6855	290.00	1.2444
126	Ethane [use Eq. (2)] Ethanol		64-17-5	46.068						159.05	0.0333	390.00	1.6450
		C ₂ H ₆ O			102,640	-139.63	-0.030341	0.0020386					
127	Ethyl acetate	$C_4H_8O_2$	141-78-6	88.105	226,230	-624.8	1.472			189.60	1.6068	350.21	1.8796
128	Ethyl amine	C_2H_7N	75-04-7	45.084	121,700	38.993				192.15	1.2919	289.73	1.3300
129	Ethylbenzene	C_8H_{10}	100-41-4	106.165	154,040	-142.29	0.80539			178.20	1.5426	409.35	2.3075
130	Ethyl benzoate	$C_9H_{10}O_2$	93-89-0	150.175	124,500	370.6				238.45	2.1287	486.55	3.0482
131	2-Ethyl butanoic acid	$C_6H_{12}O_2$	88-09-5	116.158	56,359	603.02				258.15	2.1203	466.95	3.3794
132	Ethyl butyrate	$C_6H_{12}O_2$	105-54-4	116.158	82,434	422.45	0.20992			285.50	2.2015	428.25	3.0185
133	Ethylcyclohexane	C_8H_{16}	1678-91-7	112.213	132,360	72.74	0.64738			161.84	1.6109	404.95	2.6798
134	Ethylcyclopentane	C_7H_{14}	1640-89-7	98.186	178,520	-518.35	2.3255	-0.0016818		134.71	1.4678	301.82	1.8767
135	Ethylene	C_2H_4	74-85-1	28.053	247,390	-4,428	40.936	-0.1697	0.00026816	104.00	0.7012	252.70	0.9758
136	Ethylenediamine	C_2H_4 $C_2H_8N_2$	107-15-3	60.098	184,440	-150.2	0.37044	0.1001	0.00020010	284.29	1.7168	390.41	1.8226
137	Ethylene glycol	$C_2H_6O_2$	107-21-1	62.068	35,540	436.78	-0.18486			260.15	1.3666	493.15	2.0598
138	Ethylene glycol Ethyleneimine	C_2H_5N	151-56-4	43.068	46,848	205.35	-0.10400			250.00	0.9819	329.00	1.1441
139			75-21-8	44.053	144,710		2 0201	0.000004		160.65	0.8303	283.85	0.8693
	Ethylene oxide	C ₂ H ₄ O				-758.87	2.8261	-0.003064					
140	Ethyl formate	$C_3H_6O_2$	109-94-4	74.079	80,000	223.6				254.20	1.3684	374.20	1.6367
141	2-Ethyl hexanoic acid	$C_8H_{16}O_2$	149-57-5	144.211	207,670	-17.907	1.0493			235.00	2.6141	510.10	4.7157
142	Ethylhexyl ether	$C_8H_{18}O$	5756-43-4	130.228	146,040	458.22				298.15	2.8266	417.15	3.3719
143	Ethylisopropyl ether	$C_5H_{12}O$	625-54-7	88.148	106,250	292.15				298.15	1.9335	326.15	2.0153
144	Ethylisopropyl ketone	$C_6H_{12}O$	565-69-5	100.159	229,250	-404.54	1.1382			204.15	1.9410	386.55	2.4295
145	Ethyl mercaptan	C_2H_6S	75-08-1	62.134	134,670	-234.39	0.59656			125.26	1.1467	315.25	1.2007
146	Ethyl propionate	$C_5H_{10}O_2$	105-37-3	102.132	76,330	400.1				298.15	1.9562	410.00	2.4037
147	Ethylpropyl ether	$C_5H_{12}O$	628-32-0	88.148	103,680	726.3	-2.6047	0.0040957		145.65	1.6686	320.00	2.0358
148	Ethyltrichlorosilane	C ₂ H ₅ Cl ₃ Si	115-21-9	163.506	105,150	85.318	0.46693			167.55	1.3255	371.05	2.0109
149	Fluorine	F ₂	7782-41-4	37.997	-94,585	7,529.9	-139.6	1.1301	-0.0033241	58.00	0.5541	98.00	0.5966
149	Fluorine	F ₂	7782-41-4	37.997	1724,400	-59,924	537.85	1.1501	0.0000211	53.48	0.5798	56.00	0.5535
150	Fluorobenzene	C_6H_5F	462-06-6	96.102	-991,200	11,734	-40.669	0.047333		239.99	1.3675	319.99	1.5018
151	Fluoroethane	C_2H_5F	353-36-6	48.060	85,663	-118.56	0.55459	0.011000		140.00	0.7994	240.00	0.8915
152	Fluoromethane	CH ₃ F	593-53-3	34.033	74,746	-132.32	0.53772			140.00	0.6676	220.00	0.7166
153	Formaldehyde	CH ₂ O	50-00-0	30.026	61,900	28.3	0.00112			204.00	0.6767	234.00	0.6852
154	Formanide		75-12-7	45.041	63,400	150.6				292.00	1.0738	493.00	1.3765
155		CH ₃ NO	64-18-6	46.026	78,060					281.45	0.9820	380.00	
	Formic acid	CH ₂ O ₂				71.54	0.50001						1.0525
156 157	Furan	C_4H_4O He	110-00-9 7440-59-7	68.074	114,370 387,220	-215.69	0.72691	12.404	3212.9	187.55 2.20	0.9949 0.1087	304.50	1.1609 0.2965
	Helium-4			4.003		-465,570	211,800	-42,494	3212.9			4.60	
157	Helium-4	He	7440-59-7	4.003	410,430	-464,890	135,100			1.80	0.1135	2.10	0.2995
158	Heptadecane	C ₁₇ H ₃₆	629-78-7	240.468	376,970	347.82	0.57895			295.13	5.3005	575.30	7.6869
159	Heptanal	$C_7H_{14}O$	111-71-7	114.185	222,360	-105.17	0.65074			229.80	2.3256	381.25	2.7685
160	Heptane [use Eq. (2)]	C_7H_{16}	142-82-5	100.202	61.26	314,410	1,824.6	-2,547.9		182.57	1.9989	520.00	4.0657
161	Heptanoic acid	$C_7H_{14}O_2$	111-14-8	130.185	194,570	-23.206	0.88395			265.83	2.5087	496.15	4.0065
162	1-Ĥeptanol	$C_7H_{16}O$	111-70-6	116.201	2,416,800	-26,105	110.03	-0.19172	0.00011968	239.15	2.3590	448.60	3.8766
163	2-Heptanol	$C_7H_{16}O$	543-49-7	116.201	283,127	-1,037.63	3.44064			230.00	2.2649	432.90	4.7873
164	3-Heptanone	$C_7H_{14}O$	106-35-4	114.185	270,730	-399.89	1.0601			234.15	2.3522	480.00	3.2303
165	2-Heptanone	$C_7H_{14}O$	110-43-0	114.185	265,040	-375.68	1.0024			238.15	2.3242	490.00	3.2163
166	1-Heptene	C_7H_{14}	592-76-7	98.186	267,950	-1,315.9	6.5242	-0.011994	9.3808E-06	154.12	1.8150	366.79	2.4096
167	Heptyl mercaptan	$C_7H_{16}S$	1639-09-4	132.267	236,870	-158.01	0.78982			229.92	2.4229	460.00	3.3131
168	1-Heptyne	C_7H_{12}	628-71-7	96.170	46,798	761.13	-0.62882			200.00	1.7387	372.93	2.4319
169	Hexadecane	$C_{16}H_{34}$	544-76-3	226.441	370,350	231.47	0.68632			291.31	4.9602	560.01	7.1521
170	Hexanal	C ₆ H ₁₂ O	66-25-1	100.159	117,700	329.52				217.15	1.8926	401.45	2.4999
171	Hexane	C_6H_{14}	110-54-3	86.175	172,120	-183.78	0.88734			177.83	1.6750	460.00	2.7534
172	Hexanoic acid	$C_6H_{12}O_2$	142-62-1	116.158	161,980	44.116	0.709			269.25	2.2526	478.85	3.4568
173	1-Hexanol	$C_6H_{14}O$	111-27-3	102.175	1,638,600	-17,261	71.721	-0.12026	0.000071087	228.55	1.9821	460.00	3.5197
173	2-Hexanol	$C_6H_{14}O$ $C_6H_{14}O$	626-93-7	102.175	267,628	-1,201 -1,033.06	3.35185	-0.12020	0.000071007	223.00	2.0394	585.30	8.1124
174	2-Hexanone		591-78-6	102.173		-1,033.00 -107.47	0.2062	0.00070293		217.35	2.0394	460.00	2.7087
176	3-Hexanone	$C_6H_{12}O$ $C_6H_{12}O$	589-38-8	100.159	208,250 235,960	-345.94	0.2002	0.00070293		217.50	2.0133	460.00	2.7632
177	1-Hexene	C_6H_{12}	592-41-6	84.159	164,640	-200.37	0.8784	I	1	133.39	1.5354	336.63	1.9673

TABLE 2-153 Heat Capacities of Inorganic and Organic Liquids [J/(kmol·K)] (Continued)

Cmpd.			0.10		0.4						C_p at T_{\min}		C_p at T_{\max}
no.	Name	Formula	CAS no.	Mol. wt.	C1	C2	C3	C4	C5	T_{\min} , K	× 1E-05	$T_{\rm max}$, K	× 1E-05
178 179	3-Hexyne Hexyl mercaptan	C_6H_{10} $C_6H_{14}S$	928-49-4 111-31-9	82.144 118.240	82,795 303,320	283.4 -1,009	3.3885	-0.002762		300.00 192.62	1.6781 2.1495	354.35 430.00	1.8322 2.7639
180	1-Hexyne	$C_{6}H_{10}$	693-02-7	82.144	93,000	326	3.3365	-0.002702		200.00	1.5820	344.48	2.0539
181	2-Hexyne	C_6H_{10}	764-35-2	82.144	94,860	254.15				300.00	1.7110	357.67	1.8576
182	Hydrazine	H_4N_2	302-01-2	32.045	79,815	50.929	0.043379			274.69	0.9708	653.15	1.3158
183	Hydrogen [use Eq. (2)]	H ₂	1333-74-0	2.016	66.653	6,765.9	-123.63	478.27		13.95	0.1262	32.00	1.3122
184	Hydrogen bromide	HBr	10035-10-6	80.912	57,720	9.9	120.00	410.21		185.15	0.5955	206.45	0.5976
185	Hydrogen chloride	HCl	7647-01-0	36.461	47,300	90				165.00	0.6215	185.00	0.6395
186	Hydrogen cyanide	CHN	74-90-8	27.025	95,398	-197.52	0.3883			259.83	0.7029	298.85	0.7105
187	Hydrogen fluoride	HF	7664-39-3	20.006	62,520	-223.02	0.6297			189.79	0.4288	292.67	0.5119
188	Hydrogen sulfide [use Eq. (2)]	H ₂ S	7783-06-4	34.081	64.666	49,354	22.493	-1,623		187.68	0.6733	370.00	4.9183
189	Isobutyric acid	$C_4H_8O_2$	79-31-2	88.105	127,540	-65.35	0.82867			270.00	1.7031	427.65	2.5114
190	Isopropyl amine	C ₃ H ₉ N	75-31-0	59.110	-32,469	1,977.1	-7.0145	0.0086913		177.95	1.4621	320.00	1.6671
191	Malonic acid	$C_3H_4O_4$	141-82-2	104.061	157,850	-41.619	0.42817			407.95	2.1213	603.75	2.8880
192	Methacrylic acid	$C_4H_6O_2$	79-41-4	86.089	146,290	-58.59	0.3582			288.15	1.5915	434.15	1.8837
193	Methane [use Eq. (2)]	CH_4	74-82-8	16.042	65.708	38,883	-257.95	614.07		90.69	0.5361	190.00	14.9780
194	Methanol	CH ₄ O	67-56-1	32.042	105,800	-362.23	0.9379			175.47	0.7112	400.00	1.1097
195	N-Methyl acetamide	C ₃ H ₇ NO	79-16-3	73.094	62,600	243.4				359.00	1.4998	538.50	1.9367
196	Methyl acetate	$C_3H_6O_2$	79-20-9	74.079	61,260	270.9				253.40	1.2991	373.40	1.6241
197	Methyl acetylene	C_3H_4	74-99-7	40.064	79,791	89.49				200.00	0.9769	249.94	1.0216
198	Methyl acrylate	$C_4H_6O_2$	96-33-3	86.089	275,500	-1,147	2.568			196.32	1.4930	353.35	1.9084
199	Methyl amine	CH ₅ N	74-89-5	31.057	92,520	37.45				179.69	0.9925	266.82	1.0251
200 201	Methyl benzoate 3-Methyl-1,2-butadiene	$C_8H_8O_2$ C_5H_8	93-58-3 598-25-4	136.148 68.117	125,630 135,370	279.75 -133.34	0.63868			260.75 159.53	1.9857 1.3035	472.65 314.56	2.5785 1.5662
201	2-Methylbutane	C_5H_8 C_5H_{12}	78-78-4	72.149	108,300	-133.34 146	-0.292	0.00151		113.25	1.2328	310.00	1.7048
202	2-Methylbutanoic acid	$C_5H_{10}O_2$	116-53-0	102.132	74,200	417.4	-0.292	0.00151		321.50	2.0839	481.50	2.7518
204	3-Methyl-1-butanol	$C_5H_{10}O_2$ $C_5H_{12}O$	123-51-3	88.148	247,870	-1,145	3.4223			155.95	1.5254	404.15	3.4411
205	2-Methyl-1-butene	C_5H_{10}	563-46-2	70.133	149,510	-247.63	0.91849			135.58	1.3282	304.31	1.5921
206	2-Methyl-2-butene	C ₅ H ₁₀	513-35-9	70.133	151,600	-266.72	0.90847			139.39	1.3207	311.71	1.5673
207	2-Methyl-1-butene-3-yne	C_5H_6	78-80-8	66.101	81,919	181.01				298.15	1.3589	305.40	1.3720
208	Methylbutyl ether	$C_5H_{12}O$	628-28-4	88.148	177,850	-171.57	0.74379			157.48	1.6928	343.31	2.0661
209	Methylbutyl sulfide	$C_5H_{12}S$	628-29-5	104.214	198,390	-220.35	0.76096			175.30	1.8315	510.00	2.8394
210	3-Methyl-1-butyne	C_5H_8	598-23-2	68.117	105,200	191.1				200.00	1.4342	299.49	1.6243
211	Methyl butyrate	$C_5H_{10}O_2$	623-42-7	102.132	102,930	129.1	0.62516			277.25	1.8678	415.87	2.6474
212	Methylchlorosilane	CH₅ClSi	993-00-0	80.589	47,726	338.4				250.00	1.3233	325.00	1.5771
213	Methylcyclohexane	C_7H_{14}	108-87-2	98.186	131,340	-63.1	0.8125			146.58	1.3955	320.00	1.9435
214	1-Methylcyclohexanol	C ₇ H ₁₄ O	590-67-0	114.185	50,578	508.59				300.00	2.0315	441.15	2.7494
215 216	cis-2-Methylcyclohexanol	C ₇ H ₁₄ O C ₇ H ₁₄ O	7443-70-1 7443-52-9	114.185 114.185	118,600 118,170	447.07 447.99				300.00 300.00	2.5272 2.5257	438.15 440.15	3.1448 3.1535
217	trans-2-Methylcyclohexanol Methylcyclopentane	$C_{7}H_{14}O$ $C_{6}H_{12}$	96-37-7	84.159	155,920	-490	2.1383	-0.0015585		130.73	1.2492	366.48	1.8682
218	1-Methylcyclopentene	C_6H_{10}	693-89-0	82.144	53,271	327.92	2.1303	-0.0010000		200.00	1.1885	348.64	1.6760
219	3-Methylcyclopentene	C_6H_{10}	1120-62-3	82.144	46,457	346.93				200.00	1.1584	338.05	1.6374
220	Methyldichlorosilane	CH ₄ Cl ₂ Si	75-54-7	115.034	27,030	413				250.00	1.3028	350.00	1.7158
221	Methylethyl ether	C_3H_8O	540-67-0	60.095	85,383	199.08	-0.061547			160.00	1.1566	280.50	1.3638
222	Methylethyl ketone	C_4H_8O	78-93-3	72.106	132,300	200.87	-0.9597	0.0019533		186.48	1.4905	373.15	1.7511
223	Methylethyl sulfide	C ₃ H ₈ S	624-89-5	76.161	161,240	-288.61	0.78179			167.23	1.3484	339.80	1.5344
224	Methyl formate	$C_2H_4O_2$	107-31-3	60.052	130,200	-396	1.21			174.15	0.9793	304.90	1.2195
225	Methylisobutyl ether	$C_5H_{12}O$	625-44-5	88.148	92,919	324.43				298.15	1.8965	350.00	2.0647
226	Methylisobutyl ketone	$C_6H_{12}O$	108-10-1	100.159	183,650	-79.862	0.60769			189.15	1.9029	389.15	2.4460
227	Methyl Isocyanate	C ₂ H ₃ NO	624-83-9	57.051	149,770	-529.82	1.3499			256.15	1.0263	366.00	1.3668
228	Methylisopropyl ether	$C_4H_{10}O$	598-53-8	74.122	143,440	-154.07	0.7255			127.93	1.3560	310.00	1.6540
229 230	Methylisopropyl ketone Methylisopropyl sulfide	$C_5H_{10}O$	563-80-4 1551-21-9	86.132	191,170	-331.04	0.98445	0.0001000		180.15	1.6348	440.00 357.91	2.3610
230	Methylisopropyl sulfide Methyl mercaptan	$C_4H_{10}S$ CH_4S	74-93-1	90.187 48.107	211,170 115,300	-661.97 -263.23	2.4216 0.60412	-0.0021383		171.64 150.18	1.5808 0.8939	357.91 298.15	1.8641 0.9052
231	Methyl mercaptan Methyl methacrylate	$C_5H_8O_2$	80-62-6	100.116	255,100	-263.23 -938.4	2.413			224.95	1.6611	373.45	2.4118
232	2-Methyloctanoic acid	$C_{5}H_{8}O_{2}$ $C_{9}H_{18}O_{2}$	3004-93-1	158.238	226,650	15.421	1.0578			240.00	2.9128	518.15	5.1864
234	2-Methylpentane	$C_{6}H_{14}$	107-83-5	86.175	142,220	-47.83	0.739			119.55	1.4706	333.41	2.0842
235	Methyl pentyl ether	$C_6H_{14}O$	628-80-8	102.175	251,890	-468.32	1.2209			176.00	2.0728	372.00	2.4663
236	2-Methylpropane	C_4H_{10}	75-28-5	58.122	172,370	-1,783.9	14.759	-0.047909	0.00005805	113.54	0.9961	380.00	2.0725
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238 2-Methyl-proponel C,H ₁₀ 75-65-0 74.122 -925.460 7.594-9 -17.661 0.013617 299.96 2.2016 460.00	2.9455
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246 Naphthalene C_nH_1	1.2507
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250 Nitrogen trifluoride	0.7960
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$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	2.9890
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	5.2498
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	4.6494
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	11.7608
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	3.3583
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	4.3491
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	3.6566
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	8.2276
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	3.3795
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	3.4189
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	4.6358
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	4.1566
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	5.7113
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	3.6660
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	3.4335
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	2.8235
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	3.7573
1 2 2 4	2.8619
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	1.8052
	0.9066
276 Ozone O ₃ 10028-15-6 47.998 60,046 281.16 90.00 0.8535 150.00	1.0222
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	6.6042
278 Pentanal C ₅ H ₁₀ O 110-62-3 86.132 112,050 257.78 20.0000 1.6361 376.15	2.0901
=10 -0.0000 -0.000	2.0498
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	2.9228
281 1-Pentanol $C_5H_{12}O$ 71-41-0 88.148 201,200 -651.3 2.275 200.14 1.6198 389.15	2.9227
282 2-Pentanol $C_5H_{12}O$ $6032-29-7$ 88.148 $251,596$ $-1,028.49$ 3.26306 200.00 1.7642 561.00	7.0158
283 2-Pentanone $C_5H_{10}O$ $107-87-9$ 86.132 194.590 -263.86 0.76808 196.29 1.7239 375.46	2.0380
284 3-Pentanone $C_5H_{10}O$ 96-22-0 86.132 193,020 -176.43 0.5669 234.18 1.8279 375.14	2.0661
285 1-Pentene C_5H_{10} 109-67-1 70.133 156,100 -456.94 2.255 -0.003163 0.00000238 108.02 1.2939 350.00	1.7251
	2.2827
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	2.3546
288 1-Pentyne C_5H_8 627-19-0 68.117 86,200 256.6 200.00 1.3752 313.33	1.6660
289 2-Pentyne C_5H_8 627-21-4 68.117 68.671 246.66	1.4989
290 Phenanthrene C ₁₄ H ₁₀ 85-01-8 178.229 103,370 527.03 372.39 2.9963 500.00	3.6688
291 Phenol C ₆ H ₆ O 108-95-2 94.111 101,720 317.61 314.06 2.0147 425.00	2.3670
291 Phenyl isocyanate $C_{6}H_{6}O$ $103-39-2$ 34.111 $104,720$ 317.01 9.29552 11.01 1.3080	2.3745
293 Phthálic anhydride C _s H ₂ O ₃ 85-44-9 148.116 145,400 252.4 404.15 2.4741 557.65	2.8615
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	0.8968
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	2.6079
296 1-Propanol C_3H_8O 71-23-8 60.095 158,760 -635 1.969 146.95 1.0797 400.00	2.1980
297 2-Propanol C ₃ H ₃ O 67-63-0 60.095 471,710 -4,172.1 14.745 -0.0144 185.26 1.1329 355.30	2.0487
298 Propenylcyclohexene C_9H_{14} 13511-13-2 122.207 201,400 -450.6 1.7053 199.00 1.7926 431.65	3.2463
299 Propionaldehyde C ₃ H ₄ 1317-13-8-6 58,079 99,306 115.73 1.705 200.00 1.2245 328.75	1.3735
200 Hopomatachyte C3160 125-50-0 50.010 30,000 115.10 200.00 1.22±3 525.10	1.0100

TABLE 2-153 Heat Capacities of Inorganic and Organic Liquids [J/(kmol·K)] (Concluded)

	- 100 110a: tapatimos oi iii	9	J. 3	olas [s/ (ici	101 11/1 (00)	10.0000		1					
Cmpd. no.	Name	Formula	CAS no.	Mol. wt.	C1	C2	C3	C4	C5	T_{\min} , K	C_p at T_{\min} × 1E-05	$T_{ m max}$, K	C_p at $T_{\rm max}$ ×1E-05
300	Propionic acid	$C_3H_6O_2$	79-09-4	74.079	213,660	-702.7	1.6605			252.45	1.4209	414.32	2.0756
301	Propionitrile	$C_3H_6O_2$ C_3H_5N	107-12-0	55.079	118,190	-102.7 -120.98	0.42075			180.26	1.4209	370.50	1.3112
302	Propyl acetate	$C_{5}H_{10}O_{2}$	107-12-0	102.132	83,400	384.1	0.42075			274.70	1.8891	404.70	2.3885
303	Propyl amine	C ₃ H ₉ N	107-10-8	59.110	139,530	78				188.36	1.5422	340.00	1.6605
304	Propylbenzene	C ₉ H ₁₂	103-65-1	120.192	174,380	-101.8	0.79			173.55	1.8051	432.39	2.7806
305	Propylene	C ₃ H ₆	115-07-1	42.080	114,140	-343.72	1.0905			87.89	0.9235	225.45	0.9208
306	Propyl formate	C ₄ H ₈ O ₂	110-74-7	88.105	75,700	326.1	1.0000			298.15	1.7293	398.15	2.0554
307	2-Propyl mercaptan	C ₃ H ₂ S	75-33-2	76.161	138,390	-117.11	0.47059			142.61	1.3126	350.00	1.5505
308	Propyl mercaptan	C ₃ H ₂ S	107-03-9	76.161	167,330	-319.1	0.8127			159.95	1.3708	340.87	1.5299
309	1,2-Propylene glycol	C ₃ H ₈ O ₂	57-55-6	76.094	58,080	445.2	0.0121			213.15	1.5297	460.75	2.6321
310	Quinone	$C_6H_4O_2$	106-51-4	108.095	45,810	368.33				388.85	1.8904	683.00	2.9738
311	Silicon tetrafluoride	F ₄ Si	7783-61-1	104.079	829,380	-7,331.5	19.203			186.35	1.3000	253.15	2.0403
312	Styrene	C_8H_8	100-42-5	104.149	113,340	290.2	-0.6051	0.0013567		242.54	1.6749	418.31	2.2816
313	Succinic acid	$C_4H_6O_4$	110-15-6	118.088	244,770	-236.96	0.63148			460.65	2.6961	604.50	3.3228
314	Sulfur dioxide	O ₂ S	7446-09-5	64.064	85,743	5.7443				197.67	0.8688	350.00	0.8775
315	Sulfur hexafluoride	F ₆ S	2551-62-4	146.055	119,500					230.15	1.1950	230.15	1.1950
316	Sulfur trioxide	O ₃ S	7446-11-9	80.063	258,090					303.15	2.5809	303.15	2.5809
317	Terephthalic acid	$C_8H_6O_4$	100-21-0	166.131	,								
318	o-Terphenyl	$C_{18}H_{14}$	84-15-1	230.304	182,900	635.09				329.35	3.9207	609.15	5.6977
319	Tetradecane	$C_{14}H_{30}$	629-59-4	198.388	353,140	29.13	0.86116			279.01	4.2831	526.73	6.0741
320	Tetrahydrofuran	C_4H_8O	109-99-9	72.106	171,730	-800.47	2.8934	-0.0025015		164.65	1.0721	339.12	1.3546
321	1,2,3,4-Tetrahydronaphthalene	$C_{10}H_{12}$	119-64-2	132.202	81,760	455.38				237.38	1.8986	480.77	3.0069
322	Tetrahydrothiophene	C_4H_8S	110-01-0	88.171	123,300	-130.1	0.6229			176.98	1.1979	394.27	1.6883
323	2,2,3,3-Tetramethylbutane	C_8H_{18}	594-82-1	114.229	43,326	630.73				375.41	2.8011	426.00	3.1202
324	Thiophene	C_4H_4S	110-02-1	84.140	84,864	91.725	0.13243			234.94	1.1372	357.31	1.3455
325	Toluene	C_7H_8	108-88-3	92.138	140,140	-152.3	0.695			178.18	1.3507	500.00	2.3774
326	1,1,2-Trichloroethane	$C_2H_3Cl_3$	79-00-5	133.404	103,350	159.3				236.50	1.4102	300.00	1.5114
327	Tridecane	$C_{13}H_{28}$	629-50-5	184.361	350,180	-104.7	1.0022			267.76	3.9400	508.62	5.5619
328	Triethyl amine	$C_6H_{15}N$	121-44-8	101.190	111,480	368.13				200.00	1.8511	361.92	2.4471
329	Trimethyl amine	C_3H_9N	75-50-3	59.110	136,050	-288	0.9913			156.08	1.1525	276.02	1.3208
330	1,2,3-Trimethylbenzene	C_9H_{12}	526-73-8	120.192	119,450	324.54				247.79	1.9987	449.27	2.6526
331	1,2,4-Trimethylbenzene	C_9H_{12}	95-63-6	120.192	178,800	-128.47	0.83741			229.33	1.9338	350.00	2.3642
332	2,2,4-Trimethylpentane 2,3,3-Trimethylpentane	C_8H_{18}	540-84-1	114.229	95,275	696.7	-1.3765	0.0021734		165.78	1.8285	520.00	3.9095
333	2,3,3-Trimethylpentane	C_8H_{18}	560-21-4	114.229	388,620	-1,439.5	3.2187			280.00	2.3791	320.00	2.5757
334	1,3,5-Trinitrobenzene	$C_6H_3N_3O_6$	99-35-4	213.105	40,364	664.46				398.40	3.0508	475.47	3.5629
335	2,4,6-Trinitrotoluene	$C_7H_5N_3O_6$	118-96-7	227.131	133,530	514.64				354.00	3.1571	475.00	3.7798
336	Undecane	$C_{11}H_{24}$	1120-21-4	156.308	293,980	-114.98	0.96936			247.57	3.2493	433.42	4.2624
337	1-Undecanol	$C_{11}H_{24}O$	112-42-5	172.308	129,450	-3,039.5	27.927	-0.061847	4.3042E-05	289.05	3.9103	520.30	5.5127
338	Vinyl acetate	$C_4H_6O_2$	108-05-4	86.089	136,300	-106.17	0.75175			259.56	1.5939	389.35	2.0892
339	Vinyl acetylene	C ₄ H ₄	689-97-4	52.075	68,720	135				200.00	0.9572	278.25	1.0628
340	Vinyl chloride	C ₂ H ₃ Cl	75-01-4	62.498	-10,320	322.8				200.00	0.5424	400.00	1.1880
341 342	Vinyl trichlorosilane	C ₂ H ₃ Cl ₃ Si	75-94-5	161.490	49,516	420.35	0.105	0.014110	0.0701E.00	178.35	1.2449	363.85	2.0246
	Water	H ₂ O	7732-18-5	18.015	276,370	-2,090.1	8.125	-0.014116	9.3701E-06	273.16	0.7615	533.15	0.8939
343	m-Xylene	C_8H_{10}	108-38-3	106.165	133,860	7.8754	0.52265	0.00202		217.00	1.6018	540.15	2.9060
344	o-Xylene	C_8H_{10}	95-47-6	106.165	36,500	1,017.5	-2.63	0.00302		247.98	1.7314	417.58	2.2269
345	p-Xylene	C_8H_{10}	106-42-3	106.165	-35,500	1,287.2	-2.599	0.002426		286.41	1.7697	600.00	3.2520

For the 11 substances, ammonia, 1,2-butanediol, 1,3-butanediol, carbon monoxide, 1,1-difluoroethane, ethane, heptane, hydrogen, hydrogen sulfide, methane, and propane, the liquid heat capacity $C_{\rm pL}$ is calculated with Eq. (2) below. For all other compounds, Eq. (1) is used. For benzene, fluorine, and helium, two sets of constants are given for Eq. (1) that cover different temperature ranges, as shown in the table.

(1)
$$C_{pL} = C1 + C2T + C3T^2 + C4T^3 + C4T^4$$

(2)
$$C_{\rm pL} = \frac{\text{C1}^2}{t} + \text{C2} - 2\text{C1C3}t - \text{C1C4}t^2 - \frac{\text{C3}^2t^3}{3} - \frac{\text{C3C4}t^4}{2} - \frac{\text{C4}^2t^5}{5}$$

where $t = 1 - T_r$, $T_r = T/T_c$, T_c is the critical temperature from Table 2-141, $C_{\rm pL}$ is in J/(kmol·K) and T is in K. All substances are listed by chemical family in Table 2-6 and by formula in Table 2-7. For temperatures less than the normal boiling point, the pressure is 1 atm. Above the normal boiling point, the pressure.

Values in this table were taken from the Design Institute for Physical Properties (DIPPR) of the American Institute of Chemical Engineers (AIChE), copyright 2007 AIChE and reproduced with permission of AICHE and of the DIPPR Evaluated Process Design Data Project Steering Committee. Their source should be cited as R. L. Rowley, W. V. Wilding, J. L. Oscarson, Y. Yang, N. A. Zundel, T. E. Daubert, R. P. Danner, DIPPR® Data Compilation of Pure Chemical Properties, Design Institute for Physical Properties, AIChE, New York (2007).

The number of digits provided for values at T_{\min} and T_{\max} was chosen for uniformity of appearance and formatting; these do not represent the uncertainties of the physical quantities, but are the result of calculations from the standard thermophysical property formulations within a fixed format.