Implementing Molecular Hydrophobicity Potential Measurment for the Analysis of Dynamic Biomolecular Interactions

Peleg Bar Sapir¹

Under supervision of Prof. Maria Andrea Mroginski²

¹Freie Universität Berlin ²Techniche Universität Berlin

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Molecular Hydrophobicity Potential

Pelg Bar Sapir

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Hydrophobicity and log P Partition Coefficient

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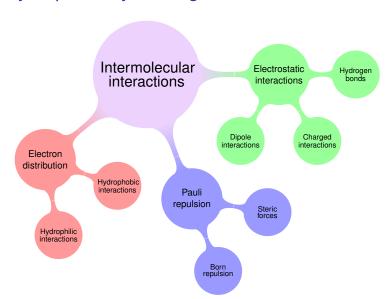
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Definition

The ratio of concentrations of a compound in a mixture of two immiscible phases at equilibrium

Molecular Hydrophobicity Potential

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Definition

The ratio of concentrations of a compound in a mixture of two immiscible phases at equilibrium

Commonly used: water and octanol

Molecular Hydrophobicity Potential

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Definition

The ratio of concentrations of a compound in a mixture of two immiscible phases at equilibrium

- Commonly used: water and octanol
- Can be measured at an ionized or unionized state

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- Can be measured at an ionized or unionized state

$$\qquad \log P_{\text{octanol/water}} = \log \left(\frac{[\text{solute}]_{\text{water}}}{[\text{solute}]_{\text{octanol}}} \right)$$

ightharpoonup Hydrophobicity increases with the (common) $\log P$

Molecular Hydrophobicity Potential

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What is Molecular Hydrophobicity Potential (MHP)?

▶ By measuring the log P of many (ca. 30,000) compounds¹, single atoms can be assigned local "force" values.

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¹Ghose et al, J. Phys. Chem. A 1998, 102, 3762-3772 () → () → ()

What is Molecular Hydrophobicity Potential (MHP)?

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- Combining these values with a distance-depended decay function, a potential can be constructed.

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- ▶ By measuring the log *P* of many (ca. 30,000) compounds¹, single atoms can be assigned local "force" values.
- Combining these values with a distance-depended decay function, a potential can be constructed.
- ▶ This potential predicts the local $\log P$ behaviour of fragments of a molecule.

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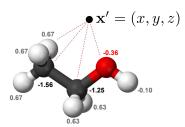
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¹Ghose et al, J. Phys. Chem. A 1998, 102, 3762-3772 () → ()



$$\mathsf{MHP}\left(\mathbf{x}'\right) = \sum_{i=1}^{k} \left[f_i \cdot D\left(\mathbf{x} - \mathbf{x}'_i\right) \right]$$

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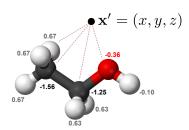
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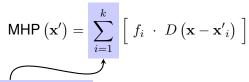
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Summing over all atoms

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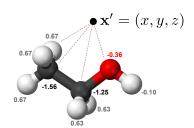
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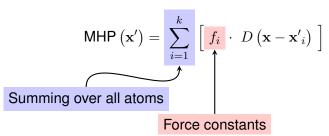
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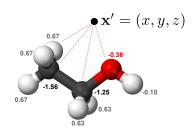
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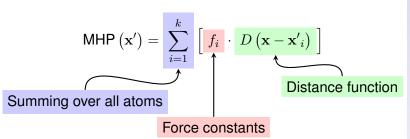
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Force Constants - Carbon

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Carbon atom contribution to hydrophobicity²

Type	Description	f_i value
	Carbon in:	
1	$\mathrm{CH_{3}R}$	-1.5603
3	CHR_3	-0.6681
7	CH_2X_2	-1.0305
13	RCX_3	0.7894
17	$=CR_2$	0.0383
24	RCHR	-0.3251
25	RCRR	0.1492
26	RCXR	0.1539

²Source: Ghose et al, J. Phys. Chem. A 1998, 102, 3762-3772

Force Constants - Hydrogen

Hydrogen atom contribution to hydrophobicity³

Type	Description	f_i value
	Hydrogen attached to:	
46	$\overline{\mathrm{C_{sp^3}}$, no X in $lpha$	0.7341
47	$ m C_{sp}^2$	0.6301
50	X	-0.1036
52	$\mathrm{C}_{\mathrm{sp^3}}$, 1 X in $lpha$	0.6666
54	C_{sp^3} , 3 X in α	0.6338

Molecular Hydrophobicity Potential

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Force Constants

³Source: Ghose et al, J. Phys. Chem. A 1998, 102, 3762-3772



Force Constants - Oxygen

Oxygen atom contribution to hydrophobicity⁴

Type	Description	f_i value
	Oxygen in:	
56	Alcohol	-0.3567
57	Phenol, enol, carboxyl OH	-0.0127
58	Ketone	-0.0233
61	Nitro, N-oxides	1.0520
62	O_{-}	-0.7941

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⁴Source: Ghose et al, J. Phys. Chem. A 1998, 102, 3762-3772

Force Constants - Various

Various atom contribution to hydrophobicity⁵

Type	Description	f_i value
66	Primary amine	-0.5427
67	Secondary amine	-0.3168
81	F attached to $\mathrm{C}_{\mathrm{sp^3}}$	0.4797
106	S in R-SH (thiol)	1.0520
119	$P \text{ in } PR_3 \text{ (phosphine)}$	-0.7941

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⁵Source: Ghose et al, J. Phys. Chem. A 1998, 102, 3762-3772



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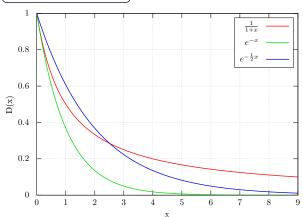
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Audry form

Exponential decay form

$$D\left(x\right) = \frac{1}{1+x}$$

$$D\left(x\right) = e^{-\alpha x}$$



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Solvent accesible surface

The surface around a molecule accesible to solvent molecules

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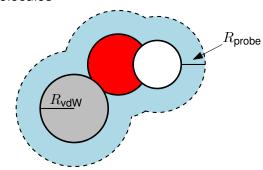
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Solvent accesible surface

The surface around a molecule accesible to solvent molecules



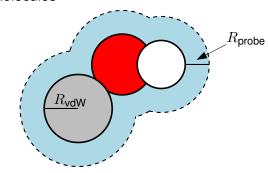
Molecular Hydrophobicity Potential

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Solvent accesible surface

Solvent accesible surface

The surface around a molecule accesible to solvent molecules



For water molecules usually r = 1.4 |A|

Molecular Hydrophobicity Potential

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Solvent accesible surface



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1. Take all atoms with their vdW-radii

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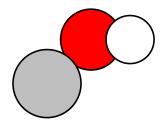
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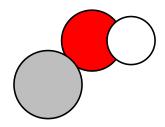
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- 1. Take all atoms with their vdW-radii
- 2. Create spheres around all atoms with $R^i = R^i_{\text{vdw}} + R_{\text{probe}}$

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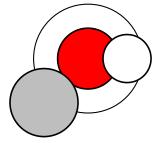
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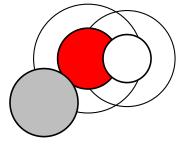
- Take all atoms with their vdW-radii
- Create spheres around all atoms with

 $R^i = R^i_{\text{vdw}} + R_{\text{probe}}$

Molecular Hydrophobicity Potential

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Solvent accesible surface



- 1. Take all atoms with their vdW-radii
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$$R^i = R^i_{\text{vdw}} + R_{\text{probe}}$$

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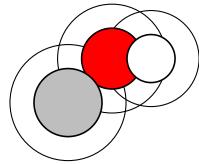
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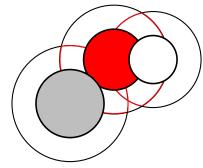
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- 1. Take all atoms with their vdW-radii
- 2. Create spheres around all atoms with $R^i = R^i_{\text{vdw}} + R_{\text{probe}}$
- 3. Delete all points that are "burried" in other extended spheres (i.e. $\Delta(p^i, c^j) \leq R^j + R_{\text{probe}}$)

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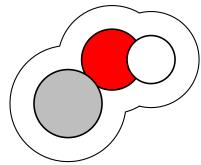
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- 2. Create spheres around all atoms with $R^i = R^i_{\text{vdw}} + R_{\text{probe}}$
- 3. Delete all points that are "burried" in other extended spheres (i.e. $\Delta(p^i, c^j) \leq R^j + R_{\text{probe}}$)
- The remaining surface is the solvent-accesible surface of the molecule

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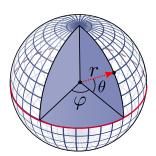
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Evenly distributed points

How to distribute N points on a surface of a sphere?



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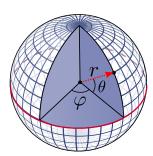
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How to distribute N points on a surface of a sphere?



$$\varphi_i = i \cdot \frac{2\pi}{N}$$

$$\theta_j = j \cdot \frac{\pi}{N}$$

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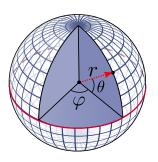
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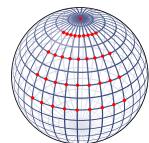
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Evenly distributed points

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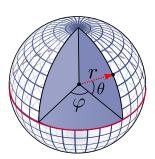
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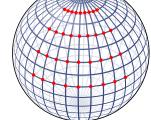
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Points are not evenly distributed Pelg Bar Sapir

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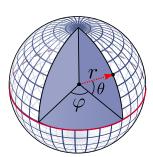
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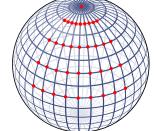
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How to distribute N points on a surface of a sphere?







$$\varphi_i = i \cdot \frac{2\pi}{N}$$

$$\theta_j = j \cdot \frac{\pi}{N}$$

$$\theta_j = j \cdot \frac{\pi}{N}$$

Points are not evenly distributed

Several points overlap at poles

Molecular Hydrophobicity Potential

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Evenly distributed points

Solution: Vogel's method

In 2 dimensions:

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Solution: Vogel's method

In 2 dimensions:

▶ Distances: $r_i = \sqrt{\frac{i}{N}}$

• Angle: $\theta_i = \varphi i$

(φ is the golden ratio!)

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Solution: Vogel's method

In 2 dimensions:

▶ Distances:
$$r_i = \sqrt{\frac{i}{N}}$$

• Angle: $\theta_i = \varphi i$ (φ is the golden ratio!)

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Evenly distributed points

Solution: Vogel's method

In 2 dimensions:

- ▶ Distances: $r_i = \sqrt{\frac{i}{N}}$
- Angle: $\theta_i = \varphi i$ (φ is the golden ratio!)

In 3 dimensions (cylindrical coordinates):

Molecular Hydrophobicity Potential

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Evenly distributed points

In 2 dimensions:

▶ Distances:
$$r_i = \sqrt{\frac{i}{N}}$$

• Angle:
$$\theta_i = \varphi i$$

(φ is the golden ratio!)

In 3 dimensions (cylindrical coordinates):

▶ Distances:
$$z_i = \left(1 - \frac{1}{N}\right) \left(1 - \frac{2i}{N-1}\right)$$

► Angles:

$$\theta_i = \varphi i, \ \rho_i = \sqrt{1 - z_i^2}$$

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Solution: Vogel's method

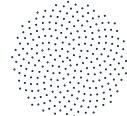
In 2 dimensions:

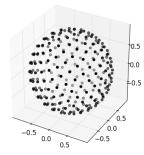
▶ Distances:
$$r_i = \sqrt{\frac{i}{N}}$$

• Angle: $\theta_i = \varphi i$ (φ is the golden ratio!)

In 3 dimensions (cylindrical coordinates):

- ▶ Distances: z_i = $\left(1-\frac{1}{N}\right)\left(1-\frac{2i}{N-1}\right)$
- Angles: $\theta_i = \varphi i, \ \rho_i = \sqrt{1 - z_i^2}$





Molecular Hydrophobicity Potential

Pelg Bar Sapir

Evenly distributed points

► Each atom's total surface area: $V^a = 4\pi \left(R_{\text{vdW}}^a + R_{\text{probe}}\right)^2$

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Molecular Hydrophobicity Potential

Pelg Bar Sapir

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▶ The surface is represented by *N* points

Molecular Hydrophobicity Potential

Pelg Bar Sapir

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- ▶ Meaning: each point has $V_j^a = \frac{4\pi}{N} \left(R_{\text{vdW}}^i + R_{\text{probe}} \right)^2$

Molecular Hydrophobicity Potential

Pelg Bar Sapir

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- ▶ Meaning: each point has $V_j^a = \frac{4\pi}{N} \left(R_{\text{vdW}}^i + R_{\text{probe}} \right)^2$
- ▶ In addition, each point has: MHP^a_i

Molecular Hydrophobicity Potential

Pelg Bar Sapir

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- ▶ The surface is represented by *N* points
- ▶ Meaning: each point has $V_j^a = \frac{4\pi}{N} \left(R_{\text{vdW}}^i + R_{\text{probe}} \right)^2$
- In addition, each point has: MHP^a_i

Therefore, each atom has a total MHP of:

$$\mathsf{MHP}^a = \frac{4\pi}{N} \sum_{j=0}^M \mathsf{MHP}^a_j$$

Molecular Hydrophobicity Potential

Pelg Bar Sapir

Written in Python3, utylizing ProDy

Molecular Hydrophobicity Potential

Pelg Bar Sapir

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- Written in Python3, utylizing ProDy
- Heavy calculation written in Cython

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- Written in Python3, utylizing ProDy
- Heavy calculation written in Cython
- Uses neighbor cells implementation for faster calculation

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- Written in Python3, utylizing ProDy
- Heavy calculation written in Cython
- Uses neighbor cells implementation for faster calculation
- Uses PSF, PDB and DCD files

Molecular Hydrophobicity Potential

Pelg Bar Sapir

Program Specifications

- Written in Python3, utylizing ProDy
- Heavy calculation written in Cython
- Uses neighbor cells implementation for faster calculation
- Uses PSF, PDB and DCD files
- Generates a PDB output, MHP values in beta column

Molecular Hydrophobicity Potential

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► Input: PSF + PDB or DCD

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- ► Input: PSF + PDB or DCD
- Subselection (optional): Atomic selection (like vmd)

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- ► Input: PSF + PDB or DCD
- Subselection (optional): Atomic selection (like vmd)
- ▶ Number of points per atom (default: 64)

Molecular Hydrophobicity Potential

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Validation via Known log Values

- ► Input: PSF + PDB or DCD
- Subselection (optional): Atomic selection (like vmd)
- Number of points per atom (default: 64)
- ► Solvent probe radius (defalt: 1.4Å)

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Validation via Known log p Values

- ► Input: PSF + PDB or DCD
- Subselection (optional): Atomic selection (like vmd)
- Number of points per atom (default: 64)
- ► Solvent probe radius (defalt: 1.4Å)
- ► Cutoff distance for distance function (default: 4Å)

Molecular Hydrophobicity Potential

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Validation via Known log p Values

- Input: PSF + PDB or DCD
- Subselection (optional): Atomic selection (like vmd)
- Number of points per atom (default: 64)
- Solvent probe radius (defalt: 1.4Å)
- Cutoff distance for distance function (default: 4Å)
- Frame range (if DCD)

Molecular Hydrophobicity Potential

Pelg Bar Sapir

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Validation via Known log p Values

▶ By integrating and comparing to known $\log P$ values, a correlation can be measured.

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Results

Validation via Known log p Values

- ▶ By integrating and comparing to known $\log P$ values, a correlation can be measured.
- A groups of amino acids of varying hydrophobicity where simulated and their MHP calculated

Molecular Hydrophobicity Potential

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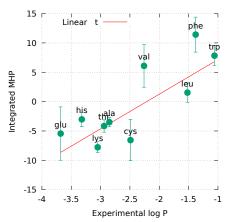


Molecular Hydrophobicity Potential

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Validation via Known log p Values

Validation in vacuum (5 frames per molecule)⁶, $R^2 = 0.668$



⁶MD simulation using NAMD, performed by Dr. Tillmann Utesch 990

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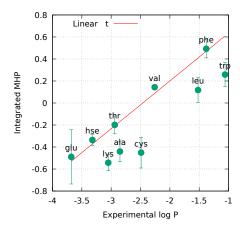
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Validation via Known log p Values

An Evennele Cueto

Validation in water + structural optimization (10 frames per molecule), $R^2 = 0.748$



Molecular Hydrophobicity Potential

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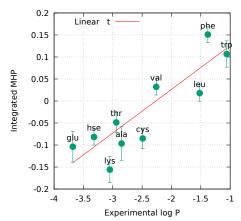
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Validation via Known log p Values

An Evample Custo

Validation in water + structural optimization + SAS normalization (10 frames per molecule), $R^2 = 0.760$



An Example System

An existing trajectory (100 frames) of a protein-membrane system⁷ was analazyed.

The peptide: OP-145, a Cathelicidin derivative with improved properties.

Molecular Hydrophobicity Potential

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⁷Trajectory provided by Dr. Alejandra de Miguel Catalina ≥ ✓ ९ ०

An Example System

An existing trajectory (100 frames) of a protein-membrane system⁷ was analazyed.

- The peptide: OP-145, a Cathelicidin derivative with improved properties.
- The interaction mechanism pathway was studied by means of all-atom simulation.

Molecular Hydrophobicity Potential

Pelg Bar Sapir

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⁷Trajectory provided by Dr. Alejandra de Miguel Catalina = → ≥ → < ○

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Validation via Known log p

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An Example System

An existing trajectory (100 frames) of a protein-membrane system⁷ was analazyed.

- The peptide: OP-145, a Cathelicidin derivative with improved properties.
- The interaction mechanism pathway was studied by means of all-atom simulation.
- The membrane used for the study consists of a mixture of two lipids, PG and PE, in agreement with experimental measurements.

⁷Trajectory provided by Dr. Alejandra de Miguel Catalina → → → → →

A picture of the system

Molecular Hydrophobicity Potential

Pelg Bar Sapir

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Hydrophobicity and log P Partition Coefficient

Molecular Hydrophobicity Potential

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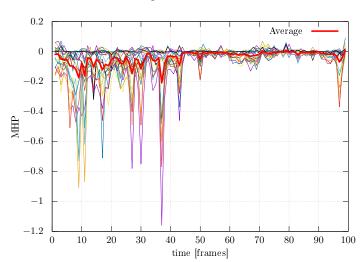
What are we interested in: Program Specifications

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MHP Change Over Time

MHP change over time for ARG-7



Molecular Hydrophobicity Potential

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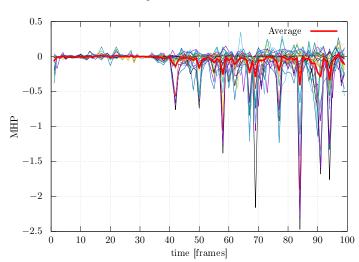
Result

Validation via Known log p



MHP Change Over Time

MHP change over time for ARG-24



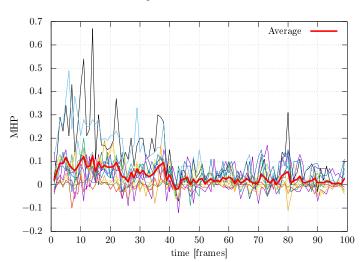
Molecular Hydrophobicity Potential

Pelg Bar Sapir



MHP Change Over Time

MHP change over time for PRO-22



Molecular Hydrophobicity Potential

Pelg Bar Sapir

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