

# Implementing Molecular Hydrophobicity Potential Measurement for the Analysis of Dynamic Biomolecular Interactions

Peleg Bar Sapir<sup>1</sup>

Under supervision of Prof. Maria Andrea Mroginski<sup>2</sup>

<sup>1</sup>Freie Universität Berlin

<sup>2</sup>Technische Universität Berlin

February 16, 2018

## Introduction

Hydrophobicity and log P

## Molecular Hydrophobicity Potential

### Potential

- General form

- Force constants

- Distance function

### Surface

- Solvent accesible surface

- Evenly distributed points

- Integration

### Introduction

Hydrophobicity and log P

### Molecular Hydrophobicity Potential

Potential

- General form

- Force constants

- Distance function

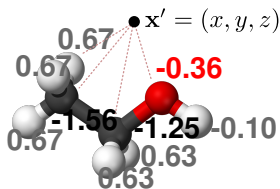
Surface

- Solvent accesible surface

- Evenly distributed points

- Integration

# The MHP Formula



$$\text{MHP}(\mathbf{x}') = \sum_{i=1}^k \left[ f_i \cdot D(\mathbf{x} - \mathbf{x}'_i) \right]$$

## Introduction

Hydrophobicity and log P

## Molecular Hydrophobicity Potential

Potential

**General form**

Force constants

Distance function

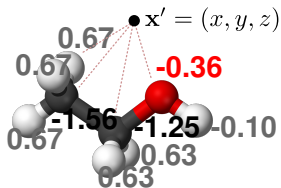
Surface

Solvent accessible surface

Evenly distributed points

Integration

# The MHP Formula



$$\text{MHP}(\mathbf{x}') = \sum_{i=1}^k \left[ f_i \cdot D(\mathbf{x} - \mathbf{x}'_i) \right]$$

Summing over all atoms

## Introduction

Hydrophobicity and log P

## Molecular Hydrophobicity Potential

Potential

General form

Force constants

Distance function

Surface

Solvent accessible surface

Evenly distributed points

Integration

# The MHP Formula

## Introduction

Hydrophobicity and log P

## Molecular Hydrophobicity Potential

Potential

General form

Force constants

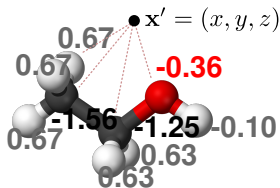
Distance function

Surface

Solvent accessible surface

Evenly distributed points

Integration

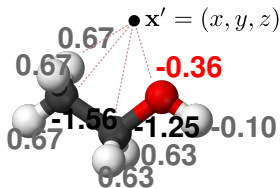


$$\text{MHP}(\mathbf{x}') = \sum_{i=1}^k \left[ f_i \cdot D(\mathbf{x} - \mathbf{x}'_i) \right]$$

Summing over all atoms

Force constants

# The MHP Formula



$$\text{MHP}(\mathbf{x}') = \sum_{i=1}^k \left[ f_i \cdot D(\mathbf{x} - \mathbf{x}'_i) \right]$$

Summing over all atoms

Force constants

Distance function

## Introduction

Hydrophobicity and log P

## Molecular Hydrophobicity Potential

Potential

General form

Force constants

Distance function

Surface

Solvent accessible surface

Evenly distributed points

Integration

# Force Constants

Carbon atom contribution to hydrophobicity<sup>1</sup>

Type	Description	$f_i$ value
	<u>C in:</u>	
1	CH <sub>3</sub> R	-1.5603
3	CHR <sub>3</sub>	-0.6681
7	CH <sub>2</sub> X <sub>2</sub>	-1.0305
13	RCX <sub>3</sub>	0.7894
17	=CR <sub>2</sub>	0.0383
24	R—CH—R	-0.3251
25	R—CR—R	0.1492
26	R—CX—R	0.1539

## Pelg Bar Sapis

0 6666



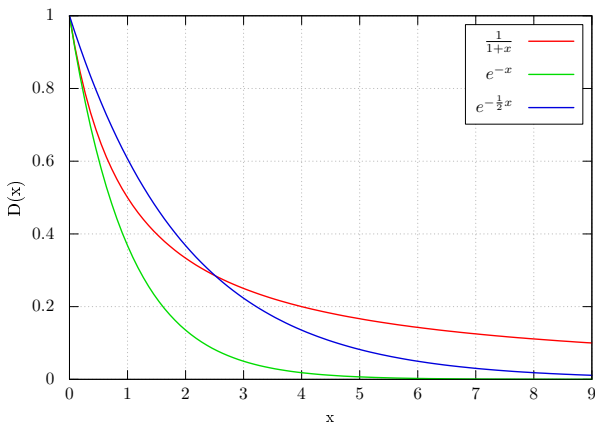
# Distance function

Audry form

$$D(x) = \frac{1}{1+x}$$

Exponential decay form

$$D(x) = e^{-\alpha x}$$



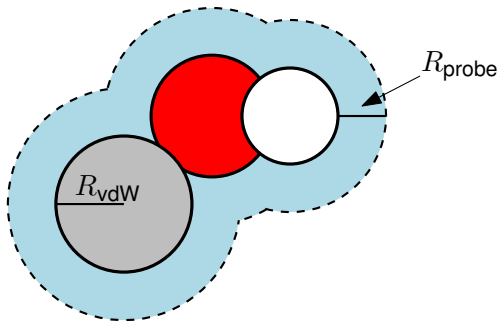
# Solvent accesible surface

- ▶ The surface around a molecule accessible to solvent molecules



# Solvent accessible surface

- ▶ The surface around a molecule accessible to solvent molecules



- ▶ For water molecules usually  $r = 1.4 \text{ [\AA]}$

# How to Create the Solvent Accessible Surface?

## Introduction

Hydrophobicity and log P

## Molecular Hydrophobicity Potential

Potential

General form

Force constants

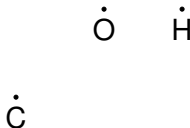
Distance function

Surface

**Solvent accessible surface**

Evenly distributed points

Integration



# How to Create the Solvent Accessible Surface?

## Introduction

Hydrophobicity and log P

## Molecular Hydrophobicity Potential

Potential

General form

Force constants

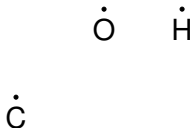
Distance function

Surface

**Solvent accessible surface**

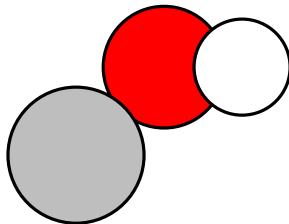
Evenly distributed points

Integration



1. Take all atoms with their vdW-radii

# How to Create the Solvent Accessible Surface?



1. Take all atoms with their vdW-radii

## Introduction

Hydrophobicity and log P

## Molecular Hydrophobicity Potential

Potential

General form

Force constants

Distance function

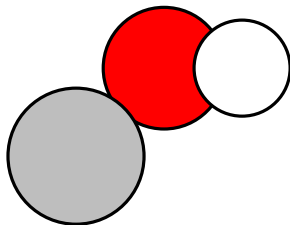
Surface

**Solvent accessible surface**

Evenly distributed points

Integration

# How to Create the Solvent Accessible Surface?

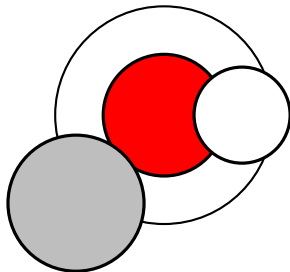


1. Take all atoms with their vdW-radii
2. Create spheres around all atoms with

$$R^i = R_{\text{vdw}}^i + R_{\text{probe}}$$



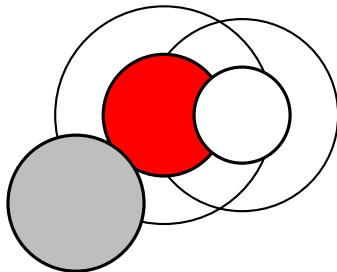
# How to Create the Solvent Accessible Surface?



1. Take all atoms with their vdW-radii
2. Create spheres around all atoms with

$$R^i = R_{\text{vdw}}^i + R_{\text{probe}}$$

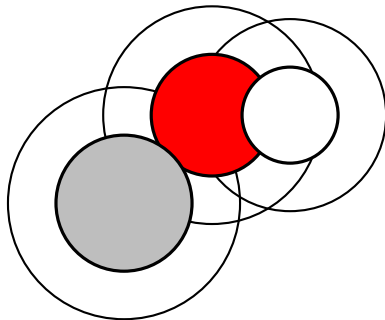
# How to Create the Solvent Accessible Surface?



1. Take all atoms with their vdW-radii
2. Create spheres around all atoms with

$$R^i = R_{\text{vdw}}^i + R_{\text{probe}}$$

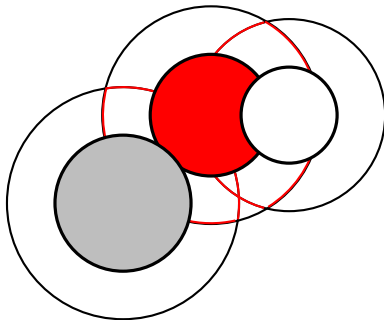
# How to Create the Solvent Accessible Surface?



1. Take all atoms with their vdW-radii
2. Create spheres around all atoms with

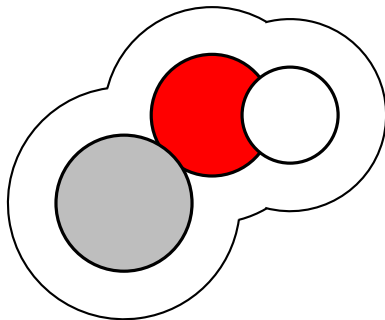
$$R^i = R_{\text{vdw}}^i + R_{\text{probe}}$$

# How to Create the Solvent Accessible Surface?



1. Take all atoms with their vdW-radii
2. Create spheres around all atoms with  
 $R^i = R_{\text{vdw}}^i + R_{\text{probe}}$
3. Delete all points that are "buried" in other extended spheres (i.e.  $\Delta(p^i, c^j) \leq R^j + R_{\text{probe}}$ )

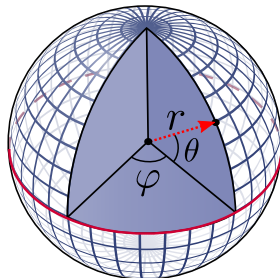
# How to Create the Solvent Accessible Surface?



1. Take all atoms with their vdW-radii
2. Create spheres around all atoms with  
 $R^i = R_{\text{vdw}}^i + R_{\text{probe}}$
3. Delete all points that are "buried" in other extended spheres (i.e.  $\Delta(p^i, c^j) \leq R^j + R_{\text{probe}}$ )
4. The remaining surface is the solvent-accessible surface of the molecule

# Evenly distributed points

How to distribute  $N$  points on a surface of a sphere?



## Introduction

Hydrophobicity and log P

## Molecular Hydrophobicity Potential

Potential

General form

Force constants

Distance function

Surface

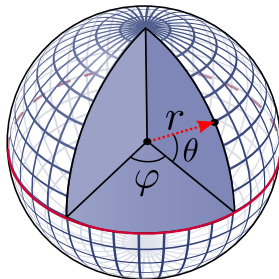
Solvent accessible surface

**Evenly distributed points**

Integration

# Evenly distributed points

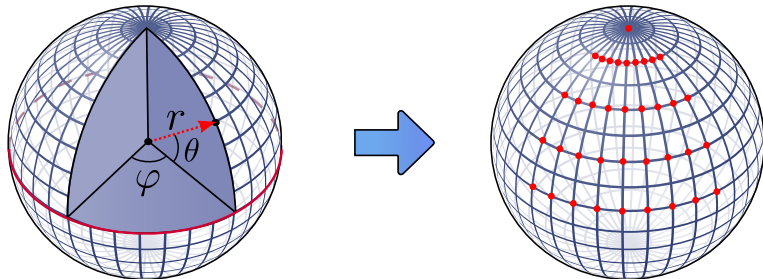
How to distribute  $N$  points on a surface of a sphere?



$$\varphi_i = i \cdot \frac{2\pi}{N}$$
$$\theta_j = j \cdot \frac{\pi}{N}$$

# Evenly distributed points

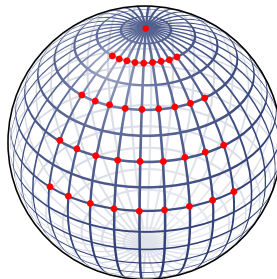
How to distribute  $N$  points on a surface of a sphere?



$$\varphi_i = i \cdot \frac{2\pi}{N}$$
$$\theta_j = j \cdot \frac{\pi}{N}$$



## Molecular Hydrophobicity Potential



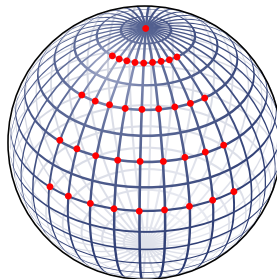
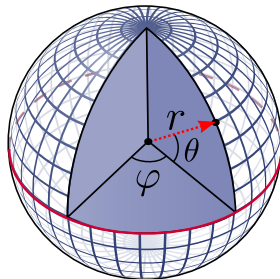
- ▶ Points are not evenly distributed

$$\begin{aligned}\varphi_i &= i \cdot \frac{2\pi}{N} \\ \theta_j &= j \cdot \frac{\pi}{N}\end{aligned}$$

$$\theta_j = j \cdot \frac{\pi}{N}$$

# Evenly distributed points

How to distribute  $N$  points on a surface of a sphere?



$$\varphi_i = i \cdot \frac{2\pi}{N}$$
$$\theta_j = j \cdot \frac{\pi}{N}$$

- Points are not evenly distributed
- Several points overlap at poles

# Evenly distributed points

Solution: **Vogel's method**

In 2 dimensions:

## Introduction

Hydrophobicity and log P

## Molecular Hydrophobicity Potential

Potential

General form

Force constants

Distance function

Surface

Solvent accessible surface

**Evenly distributed points**

Integration

# Evenly distributed points

Solution: **Vogel's method**

In 2 dimensions:

- ▶ Distances:  $r_i = \sqrt{\frac{i}{N}}$
- ▶ Angle:  $\theta_i = \varphi i$   
( $\varphi$  is the golden ratio!)

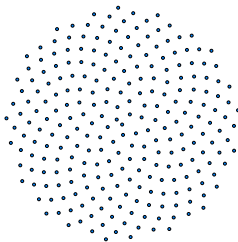


# Evenly distributed points

Solution: **Vogel's method**

In 2 dimensions:

- ▶ Distances:  $r_i = \sqrt{\frac{i}{N}}$
- ▶ Angle:  $\theta_i = \varphi i$   
( $\varphi$  is the golden ratio!)



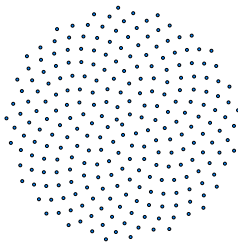
In 3 dimensions (cylindrical coordinates):

# Evenly distributed points

Solution: **Vogel's method**

In 2 dimensions:

- ▶ Distances:  $r_i = \sqrt{\frac{i}{N}}$
- ▶ Angle:  $\theta_i = \varphi i$   
( $\varphi$  is the golden ratio!)



In 3 dimensions (cylindrical coordinates):

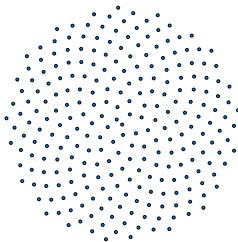
- ▶ Distances:  $z_i = \left(1 - \frac{1}{N}\right) \left(1 - \frac{2i}{N-1}\right)$
- ▶ Angles:  
 $\theta_i = \varphi i, \rho_i = \sqrt{1 - z_i^2}$

# Evenly distributed points

Solution: **Vogel's method**

In 2 dimensions:

- ▶ Distances:  $r_i = \sqrt{\frac{i}{N}}$
- ▶ Angle:  $\theta_i = \varphi i$   
( $\varphi$  is the golden ratio!)



In 3 dimensions (cylindrical coordinates):

- ▶ Distances:  $z_i = \left(1 - \frac{1}{N}\right) \left(1 - \frac{2i}{N-1}\right)$
- ▶ Angles:  
 $\theta_i = \varphi i$ ,  $\rho_i = \sqrt{1 - z_i^2}$

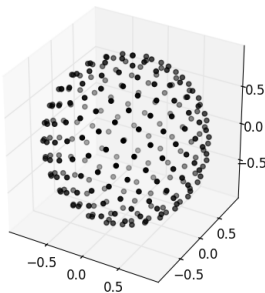


Image source: Marmakoide's Blog



# Integration

## Molecular Hydrophobicity Potential

Pelg Bar Sapir

### Introduction

Hydrophobicity and log P

### Molecular Hydrophobicity Potential

Potential

General form

Force constants

Distance function

Surface

Solvent accessible surface

Evenly distributed points

**Integration**