# Semester 2 Notes

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Spring Semester 25

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## 1 Analysis II

## 1.1 Propositional Logic

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- $\neg (p \cup q) = (\neg p) \cap (\neg q)$
- $\neg(p \cap q) = (\neg p) \cup (\neg q)$
- $\bullet \ \neg(\exists p) = \forall p$
- $\bullet \ p \Rightarrow q = (p \cap q) \cup \neg q$
- so  $\neg(p \Rightarrow q) = (\neg p \cup \neg q) \cap \neg \neg q = \neg p \cup q$

## 1.2 Continuous Functions

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Continuity

- For  $f: D \to \mathbb{R}$ , f is continuous at a point  $a \in D$  if:  $\forall \epsilon > 0, \exists \delta > 0 : \forall x \in D, |x a| < \delta \Rightarrow |f(x) f(a)| < \epsilon$
- f is uniformly coninuous if continuous  $\forall a$ , or  $\forall \epsilon > 0, \exists \delta > 0 : \forall x, y \in D, |x y| < \delta \Rightarrow |f(x) f(y)| < \epsilon$

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- A function  $f: D \to \mathbb{R}$  is continuous at  $a \in \mathbb{R}$  iff:

$$\forall (x_n)_{n \in \mathbb{N}}, x_n \in D, \lim_{n \to \infty} x_n = a, \lim_{n \to \infty} f(x_n) = f(a)$$

• An alternative statement of this is:

$$\lim f(x_n) = f(\lim(x_n))$$

if f not continuous at a:

$$\exists \epsilon>0: \forall \delta>0, \exists x\in D:$$

$$|x - a| < \delta \Rightarrow |f(x) - f(a)| \ge \epsilon$$

Theorem 1 (Intermediate Value Theorem) for  $f:D\to C\subseteq\mathbb{R}$ 

then 
$$u \in C \Rightarrow \exists x \in D : u = f(x)$$

- A function is also continuous if:
- $\delta > 0 : \forall x \in D \backslash a : |x a| > \delta$
- $\forall (x_n)_{n\in\mathbb{N}}$
- f invertable and f continuous  $\Rightarrow f^{-1}$  continuous

## 2 waves and fields

## 2.1 Periodic Motion

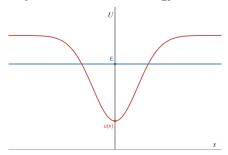
Periodic Motion In 1D

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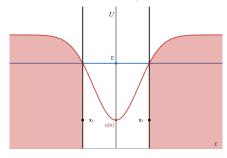
- Periodic motion is one in which the position repeats values regularly
- It can be characterised like so:

$$r(t+nT) = r(t), \forall t \in \mathbb{R}, n \in \mathbb{N}, T = period$$

• If we consider a situation in a consevative field, with a potential u(x), and a system with a total energy E:



• We can see that the system would be bound within the region U < E:



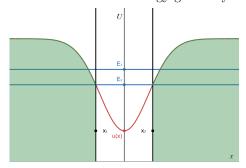
- This naturally leads to periodic motion, as at  $x = x_1, v = 0$
- At this point, the energy is not minimised, and so the system tends towards x=0, where  $v\neq 0$ , causing periodic motion.

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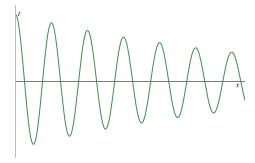
- One way to achieve this is through simple harmonic motion (SHM)
- SHM is any system where  $\ddot{x} = -x$
- We can see here that  $u(x) = -\int f dx = \int \frac{\ddot{x}}{m} dx$ , or  $f = -\frac{du}{dx} \Rightarrow m \ddot{x} = -\frac{du}{dx}$
- We can then add another term to take energy out of the system:
- $m\ddot{x} = -\frac{du}{dx} \gamma \dot{x}$  in general.

• This describes a frictional force, causing damped harmonic motion (DHM)

• This will lead to the energy gradually decreasing:



• Which will result in the following motion:



#### Periodic Motion In 3D

• Gravity is a conservative force so  $u=u(\vec{r})$ , and angular momentum is conserved

• Evidently, from  $\vec{L} = \vec{r} \times \vec{p}$ , the position  $\vec{r}$  is perpendicular to the constant angular momentum  $\vec{L}$ , and thus the system is confined to a single plane.

• Let  $v_{\phi}$  =the component of v perpendicular to  $\vec{r}$ , and  $v_r$  = the component of v parallel to  $\vec{r}$ 

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• Then  $L = m\vec{r} \times (r\dot{\phi})$ 

• Since  $K = \frac{1}{2}m\dot{\vec{r}} = \frac{1}{2}m(\dot{\vec{r}}x^2 + \dot{\vec{r}}x^2) = \frac{1}{2}m(\dot{r}^2 + r^2\dot{\phi}^2)$ 

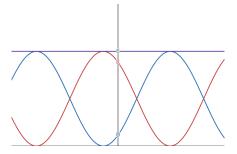
•  $\dot{\phi} = \frac{L}{mr^2} \Rightarrow K = \frac{1}{2}m(\dot{r}^2 + r^2(\frac{L}{mr^2})^2)$ 

• So  $E = K + u = \frac{1}{2}m(\dot{r}^2 + (L/mr)^2 + u(\vec{r}))$ 

## Simple Harmonic Motion

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- Simple harmonic motion is any motion of the form  $x(t) = A\cos(\omega t + \phi)$
- From this, we can see that  $\ddot{x}(t) = -A\omega^2\cos(\omega t + \varphi) = -\omega^2 x$
- This differential equation,  $\ddot{x} \propto x$ , characterises simple harmonic oscillation(SHO)
- Then,  $f = m\ddot{x} = -m\omega^2 x$ , is only dependant on position
- Then, the potential energy:  $U(x) = -\int f dx = m\omega^2 \int x dx = \frac{1}{2}m\omega^2 x^2$
- Then,  $E=K+U=\frac{1}{2}m\dot{x^2}+\frac{1}{2}m\omega^2x^2=\frac{1}{2}mA^2\omega^2$
- E, K, and U will look like this over time, with  $\overline{U} = \overline{K}$  over one period:



## Simple Pendulums and Small oscillators

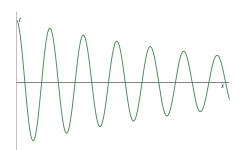
- For a simple pendulum,  $|\tau| = I\ddot{\theta} = -f_{\perp} \cdot r = -f_g \sin(\theta) \cdot l$
- Then, using the small angle approximation  $\sin(\theta) = \theta$ :  $I\ddot{\theta} = -f_g l\theta$
- Another example of simple harmonic motion is the torsion pendulum
- This is a mass on a string that can twist around its axis:  $\tau = -c\theta \Rightarrow I\ddot{\theta} = -c\theta$

#### **Damped Harmonic Motion**

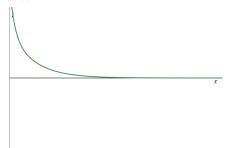
- We expect the amplitude A of a damped harmonic oscillator to decrease over time.
- For DHM, The 2nd law of motion looks like this:
- $m\ddot{x} = -kx b\dot{x}$
- Setting b=0, we recover the SHM characteristic equation
- For  $b^2 4km < 0$ , This is called underdamping

• Underdamping will have the following form of motion:

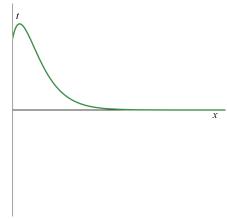
$$x(t) = Ae^{-\frac{b}{2m}t}\cos(\omega t + \varphi)$$
 (2.1)



- Then considering the quantity  $\frac{2m}{b},$  we see  $|x(t+\frac{2m}{b})|=e^{-1}|x(t)|$
- If  $b^2 > 4mk$ , we get overdamped motion:
- $\bullet \ x(t) = A_1 e^{-\lambda_1 t} + A_2 e^{-\lambda_2 t}$



- and if  $b^2 = 4mk$ , critically damped Motion:
- $x(t) = (A_1 + A_2 t)e^{-\frac{b}{2m}t}$



#### Forced Harmonic Oscillation & Resonance

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- $\bullet \ \ddot{x} + \frac{b}{m}\dot{x} + \frac{k}{m} = \frac{1}{m}F_d$
- $\bullet\,$  If the driving force is also periodic, the motion will oscillate at the driving frequency  $f_d$

$$\omega_r = \sqrt{\frac{k}{m} - \frac{b^2}{2m^2}} \tag{2.2}$$

• Where A is maximised at  $\omega_d = \omega_r$ 

#### **Coupled Harmonic Oscillators**

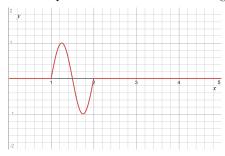
- for coupled springs with masses  $m_{1,2}$  at positions  $x_{1,2}$
- Clearly,  $m\ddot{x}_1 = -kx_1 k'(x_1 x_2)$
- And a similar result for  $x_2$
- If we then consider the case x1 = x2, the middle spring's force vanishes and we get 2 uncoupled harmonic motions with frequency  $\omega_1 = \sqrt{\frac{k}{m}}$
- This is called a normal mode of the system, as the  $x_1(t) \perp x_2(t)$ : They are independent
- Considering the alternate normal mode:  $x_1(t) = -x_2(t)$
- And the motion  $\omega_2 = \sqrt{\frac{k+2k'}{m}}$
- $\bullet\,$  In n dimensions, there are n normal modes, and all normal frequencies are normal frequencies
- All modes of a system are superpositions of the normal modes

## Waves

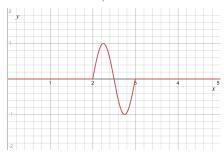
Wave Basics 13/2/25

• Waves are a form of harmonic motion

- Waves have an amplitude, wave speed, and carry energy and momentum
- There are travelling waves and standing waves
- The simplest consideration is a string moving in one dimension:



• After some time, the wave will now look like so:



- The function for a 1d wave will be y(x,t) = f(x-vt) as it moves in time but the shape doesnt move in time
- The actual particles causing the wave always return to rest once the wave has left the space
- For sinusoidal waves:
- $\bullet \ \ y(x,t) = A\cos(k[x\mp vt]+\varphi) = A\cos(kx\mp \omega t + \varphi) = A\cos(2\pi[\tfrac{x}{\lambda}\mp ft]+\varphi)$
- where k =the wave number or spacial frequency, and  $k = \frac{1}{\lambda}$
- The general wave equaiton can be written as:

$$\frac{\partial^2 y}{\partial t^2} = v^2 \frac{\partial^2 y}{\partial x^2} \tag{2.3}$$

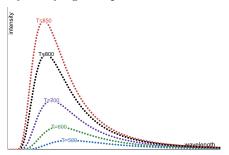
## 3 Modern Physics

## 3.1 Black Body Radiation

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#### Observations

- Black bodies satisfy 3 criteria:
  - absorbs all incident radiation
  - emits as much radiation as possible at any wavelength
  - emits radiation isotropically
- By studying the spectra of black bodies at different temperatures:



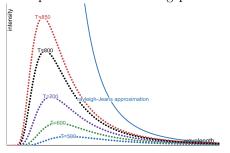
- We can observe 2 main laws:
  - Wein's displacement law:  $\lambda_{max} = \frac{b}{T}$
  - Stephans law:  $P(T) = \sigma A T^4$

#### Rayleigh-Jeans Law

- In June 1900, John Rayleigh discovered that the spectral raiance  $\propto \lambda^{-4}$
- $\bullet$  In 1905, James Jean derived the full Rayleigh-Jeans law including constants
- The Rayleigh-Jeans law is a classical approximation of the radiance of a black body
- Rayleigh-Jeans law can be expressed as so:

$$B(\lambda, T) = \frac{2ck_bT}{\lambda^4} \tag{3.1}$$

• This produced the following prediction for spectral radiance at T=800:



• This presents many problems, mainly that it does nt match for  $\lambda<\lambda_p,$  and  $I\to\infty$  as  $\lambda\to0$ 

#### Planck's Law

- Rayleigh-Jeans law, as seen in the derivation, can be expressed as:
- $R(\lambda) \propto \rho(f) \cdot E_{avg}$
- Where  $\rho$  and E are the number of waves and average energy of each wave
- In the Rayleigh-Jeans law,  $E_{avg} = kT$
- This was due to the equipartition theorem
- Max Planck's great insight was to abandon the equipartition theorem, instead writing:
- $E = n\epsilon \Rightarrow \overline{E} = \frac{hf}{e^{\frac{hf}{kT}-1}}$
- And so:

$$R(\lambda) = \frac{2hc}{\lambda^5} \cdot \frac{1}{e^{\frac{hc}{\lambda kT} - 1}}$$
 (3.2)

#### 3.2 The Photoelectric Effect

- $\bullet\,$  In 1887, Heinrich Hertz discovered that light rays incident on a metal plate produces an emf across the material
- This is due to electrons being liberated by the light
- This works with metals due to the delocalised electron bonds
- The electrons would be liberated with a range of kinetic energies up to a maximum
- There was also an observed threshold frequency, below which a ray can't liberate electrons

#### classical Predictions

- $V_s \propto I$
- $K_{max} \propto A$
- $\frac{dN}{dt} \propto f$

#### Einstein's Explanation

- In 1905, Einstein published his explanation of the photoelectric effect
- He explained that the light interacts in a 1:1 mannor, with E = hf
- This is insightful as Max Planck had only quantised matter oscillation
- Einstein qunatised light
- $\bullet$  This explains the intensity profile, as the particle light must interact with the electrons  $1{:}1$
- It also explained the threshold frequency phenomenon, as light with insufficient energy just reflects off of the metal
- $E_{\gamma} = hf = \varphi + \frac{1}{2}m_e v_{max}^2$  or  $hf = \varphi + K_{max}$

## 3.3 Rutherford Experiment

13/2/25

- In 1814, Joseph von Fraunhofer observed line spectra
- When light shone through a gas is split through a prism, you observe a full spectrum with line gaps
- In 1885, Johann Balmer discovered the Balmer series, which described where these line gaps would occur
- Neils Bohr described these balmer seires by quantising electrons' angular momentum, meaning they can only occupy specific orbitals
- Then,  $E_{\gamma} = hf = E_i E_f$
- Bohr's model worked well only for hydrogenic atoms
- We can obtain:

$$E_{\gamma} = -(13.6 \ eV)Z^{2} \left[\frac{1}{n_{i}^{2}} - \frac{1}{n_{f}^{2}}\right]$$
(3.3)

• We call the ionisation energy  $E_i = E_{infty} - E_1$ , the energy required to leave the atom

## 4 Intro to Quantum Physics

## 4.1 A Brief History

#### Ultraviolet Catastrophe

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- In 1860, Gustav Kirchhoff discovered what is now called black body radiation
- This is where an object with internal energy radiates electromagnetic waves correlated with its thermal properties.
- Classical predictions, using namely the Rayleigh-Jeans law, predicted an infinite ammount of low-frequency light waves:
- In 1990, Max Planck produced a soloution to this problem which correctly predicted the spectral radiance of black body radiation
- This theory worked by quantising light, which he expected to disapear in the final solution
- He still did not believe that light was acually quantised
- He earned the Nobel Prize in 1918

#### Photoelectric Effect

- In 1887, Heinrich Hertz discovered the photoelectric effect
- This is when you expose a metal plate to high intensity light and observe a voltage over the plate
- The contemporary theories predicted that the energy transfered should be proportional to the frequency
- Instead, there is an observed threshold frequency, below which no energy is transfered
- In 1905, Albert Einstein explained this by employing Max Plancks quantum light
- He won the Nobel Prize in 1921

## Atomic Models

- In 1911, Ernst Rutherford discovered positively charged atomic nuclei
- The presence of the electron in the atom, then, predicted that atoms were unstable, decaying in 10<sup>-</sup>11s as the electron would fall into the nuclei
- In 1913, Neils Bohr predicted that angular momentum was quantised, restricting electrons to specific "orbits" and stopping atomic decay

- Neils Bohr could not come up with a mechanism for angular momentum quantisation, however
- He won the Nobel Prize in 1913

#### DeBroglie's Electron Strings

- In 1924, Louis DeBroglie suggested wave particle duality
- This suggests that sufficient restrictions upon an electrons position force it to exhibit wave like behaviour
- When employed on an electron in orbit, the electron waves superpose, producing a standing wave
- Standing waves can only exist with an integer multiplier on the fundamental frequency
- This multiplier corresponds to the energy the electron can have, quantising orbital angular momentum
- He won the Nobel Prize in 1929

## 4.2 Rutherford And Compton Scattering

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#### **Rutherford Scattering**

- In 1987, Thompson devised the Plum Pudding model.
- $\bullet$  In 1906 in Manchester, Ernst Rutherford devised the  $\alpha$  scattering experiment
- He shot  $\alpha$  particles at a thin sheet of gold and observe the distribution of scattering angles
- The Plum Pudding model predicts the following distribution:
- It predicted little to no scattering due to the diffusity of the positive charge
- Whereas the distribution Rutherford found was the following:
- In response to this, rutherford devised the nuclear model
- This explained that the nucleus repels the  $\alpha$  particles due to the like charges, sometimes up to a full  $\pi^c$

#### Compton Scattering

- In 1923, Arthur Compton discovered Compton scattering
- This is where x-rays incident on an atom reflect at a different wavelength
- The classical model, with  $p_{\gamma} = 0$ , predicted no shift:
- But in reality, there was a shift:
- This showed that light does carry momentum, and therefore supported the photon model
- Using conservation of momentum and conservation of energy, you can arrive at the following equation:

$$\lambda - \lambda' = \frac{h}{m_0 \cdot c} [1 - \cos(\theta)] \tag{4.1}$$

## 4.3 Milikan's Oil Drop

- In 1897, Robert Milikan devised an experiment to measure the charge of the electron:
- The 2 plates are connected to a power source, producing an emf
- A nozzle containing oil is small to be sprayed
- There is a camera or microscope to observe the drops
- The drops become charged due to friction with the nozzle
- Oil drops that move up are large charge or low mass, and vice versa
- There is also the bouyant and viscous forces
- $\vec{f_b} = \frac{4}{3}\pi g r^3 \rho_{air} \hat{e}_z$  since the oil is displacing the air
- from Stoke's Law,  $\vec{f_v} = -6\pi \eta_{air} \cdot rv$
- $\vec{f_v} = -6\pi \eta_{air}$
- Since it is hard to measure the properties of the drops explicitly, we can instead formulate expressions for them:

$$r = \sqrt{\frac{6\eta_{air}v_{off}}{2(\rho_{oil} - \rho_{air})g}} \tag{4.2}$$

$$q = \frac{g\sqrt{2}(v_{on} + v_{off})d}{V_0} \cdot \sqrt{\frac{\eta_{air}^3 v_{off}}{(\rho_{oil} - \rho_{air})g}}$$
(4.3)

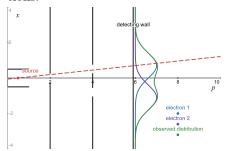
- You then repeat this multiple times and record the charge of each drop 6/2/25
- There will be a random distribution of charges, however the smallest difference in charge will be  $e=1.60\cdot 10^{-19}$

## 4.4 Quantum Behaviour

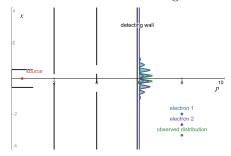
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## Young's Double Slit

 $\bullet$  If particles are sent through the slits, you expect the following distributions:



• And if waves are sent through:



- When Young's double slit experiment was performed with electrons, the distribution found was that of a wave
- However when you put a light on the other side of the slits, the distribution turns out like a particle
- This is because the light affects the motion of the electrons, spreading their wavefunction out

## 5 Mathematical Methods II

## 5.1 Non Cartesian Co-Ordinates

#### Polar Co-Ordinates

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- for  $x\hat{\imath} + y\hat{\jmath}$ , the equivalent polar basis vectors are:  $\hat{r} = (\cos(\theta), \sin(\theta)), \ \hat{\theta} = (-\sin(\theta), \cos(\theta))$  where  $\tan(\theta) = \frac{y}{x}$
- Then,  $\cos(\theta)\hat{r} \sin(\theta)\hat{\theta} = \hat{\imath}$  and  $\sin(\theta)\hat{r} + \cos(\theta)\hat{\theta} = \hat{\jmath}$

## 3D Polar Co-Ordinates

- There are 2 main 3D Polar Coordinate Bases:
- Spherical  $(\hat{r}, \hat{\theta}, \hat{\varphi})$
- And Cylindrical  $(\hat{\rho}, \hat{\theta}, \hat{z})$

## 6 Derivations

## 6.1 Analysis II

### 6.2 Waves and Fields

## eqn 2.1 Underdamping Motion Form

$$\bullet \ \ m\ddot{x} = -kx - b\dot{x}$$

$$\bullet \Rightarrow \ddot{x} + \frac{b}{m}\dot{x} + \frac{k}{m}x = 0$$

• This has a characteristic equation  $\lambda^2 + \frac{b}{m}\lambda + \frac{k}{m}$ 

• With soloutions 
$$\lambda = -\frac{b}{2m} \pm \frac{1}{2} \sqrt{\left(\frac{b}{m}^2\right) - 4\frac{k}{m}}$$

• For case 
$$b^2 < 4mk, \lambda = -\frac{b}{2m} \pm i\sqrt{\frac{k}{m} - \frac{b^2}{4m^2}}$$

$$\Rightarrow x(t) = Re\left[Ae^{-\frac{b}{2m}t}e^{\pm i\sqrt{\frac{k}{m} - \frac{b^2}{4m^2}}}\right]$$

• 
$$\Rightarrow x(t) = Ae^{-\frac{b}{2m}t}\cos(\omega t + \varphi)$$

#### eqn 2.2 Resonance Frequency

• For 
$$\ddot{x} + \frac{b}{m}\dot{x} + \frac{k}{m}x = \frac{1}{m}f_d, f_d = d\cos(\omega_d t + \varphi)$$

DE

• We know that for FHM,  $f = f_d$  so:

• Assume 
$$x(t) = Re[\mathcal{A}e^{i\omega_d t}], F_d = Re[\mathcal{D}e^{i\omega_d t}]$$
 where  $\mathcal{D} = De^{i\varphi_d}$ 

• Into DE: 
$$-\omega_d^2 \mathcal{A} e^{i\omega_d t} + \frac{b}{m} i\omega_d \mathcal{A} e^{i\omega_d t} + \frac{k}{m} \mathcal{A} e^{i\omega_d t} = \frac{1}{m} \mathcal{D} e^{i\omega_d t}$$

• 
$$\Rightarrow \mathcal{A}(\omega_d^2 - \frac{b}{m}i\omega_d - \frac{k}{m}) = -\frac{1}{m}\mathcal{D}$$

• 
$$\Rightarrow \mathcal{A} = \frac{\mathcal{D}/m}{\omega_d^2 - \frac{b}{m}i\omega_d - \omega_0^2}$$
 where  $\omega_0 = \sqrt{\frac{k}{m}}$ 

• Then, amplitude 
$$A=|\mathcal{A}|=\frac{D/m}{\sqrt{(\omega_d^2-\omega_0^2)^2+\frac{b^2}{m^2}\omega_d^2}}$$

 $\bullet$  It is obvious that  $A_{max}$  occurs at the minimum of the denominator

$$\bullet \ \, \frac{d\mathcal{W}}{d\omega_d} = \frac{d}{d\omega_0} \big[ \sqrt{(\omega_d^2 - \omega_0^2)^2 + \frac{b^2}{m^2} \omega_d^2} \big] = \frac{\frac{1}{2} [4(\omega_d^3 - \omega_d \omega_0^2) + 2\frac{b}{m}^2 \omega_d]}{\mathcal{W}}$$

• 
$$\frac{d\mathcal{W}}{d\omega_d}|_{\omega_r} \Rightarrow 2\omega_r[2(\omega_r^2 - \omega_0^2) + \frac{b}{m}^2] = 0$$

• 
$$\omega_d \neq 0 \Rightarrow \omega_r^2 = \omega_0^2 - \frac{b^2}{2m^2}$$

$$\bullet \Rightarrow \boxed{\omega_r = \sqrt{\frac{k}{m} - \frac{b^2}{2m^2}}}$$

### eqn 2.3 Coupled Harmonic Oscillator

• 
$$m\ddot{x}_1 = -kx_1 - k^{prime}(x_1 - x_2)$$

• 
$$m\ddot{x}_2 = -kx_2 - k^{prime}(x_2 - x_1)$$

#### eqn 2.4 The Wave Equation

• 
$$y(x,t) = f(x-vt)$$

• Then, 
$$\frac{\partial^2 y}{\partial t^2} = \frac{\partial}{\partial t} \frac{\partial y}{\partial t} = \frac{df}{dt} \cdot \frac{\partial}{\partial t} (x - vt) = -v \frac{df}{dt}$$

• 
$$\frac{\partial^2 y}{\partial t^2} = \frac{d^2 f}{dt^2} \cdot \frac{\partial}{\partial t} (x - vt) \cdot -v = (-v)^2 \frac{d^2 f}{dt^2} = v^2 \frac{d^2 f}{dt^2}$$
 Eq1

• 
$$\frac{\partial y}{\partial x} = \frac{df}{dt} \cdot \frac{\partial (x - vt)}{\partial x} = \frac{df}{dx} \cdot 1$$

• 
$$\frac{d^2y}{dx^2} = \frac{d^2f}{dt^2} \cdot \frac{\partial(x-vt)}{\partial x} = \frac{d^2f}{dt^2}$$

• Then, using Eq1, for any travelling wave:

## 6.3 Modern Physics

#### eqn 3.1 Rayleigh-Jeans Law

• Firstly, we begin by noting that  $R(\lambda) = \rho(f) \cdot \frac{c}{4\pi}$  where  $\rho(f)$  is the energy density inside a Jeans cube

• Then, 
$$\rho(f) = N(f) \cdot E_{avg}$$

• 
$$N(f) = \frac{8\pi f^2}{c^3}$$
, and  $E_{avg} = kT$ 

• Then 
$$\rho(f) = \frac{8\pi f^2 kT}{c^3}$$

$$\bullet = \frac{8\pi kT}{\lambda^2}$$

$$\bullet \ \Rightarrow \boxed{\frac{2ckT}{\lambda^4}}$$

#### eqn 3.2 Planck's Law

#### eqn 3.3 Bohr Model

• Energies in the atom:

$$-E_k = \frac{1}{2}m_e v^2 + \frac{1}{2}m_n \cdot 0^2$$
$$-E_p = \frac{Ze^2}{4\pi\epsilon_0 r}$$

• Forces in the atom:

$$- f_c = \frac{m_e v^2}{r}$$
$$- f_e = \frac{-Ze^2}{4\pi\epsilon_0 r^2}$$

- Where Z=atomic number
- Total energy  $E_n = E_k + E_p = \frac{1}{2} m_e v^2 \frac{Ze^2}{4\pi\epsilon \alpha r}$
- Hypothesize stationary orbit:  $|f_c| = |f_e|$

$$\bullet \ \frac{m_e v^2}{r} = \frac{Ze^2}{4\pi\epsilon_0 r} \tag{Eq1}$$

• 
$$\Rightarrow m_e v^2 = \frac{Ze^2}{4\pi\epsilon_0 r}$$

• 
$$\Rightarrow E_n = \frac{1}{2} \frac{Ze^2}{4\pi\epsilon_0 r} - \frac{Ze^2}{4\pi\epsilon_0 r} = -\frac{1}{2} \frac{Ze^2}{4\pi\epsilon_0 r}$$
 (Eq2)

- Hypothesize quantized angular momentum:  $L_n = m_e v r_n = n\hbar$
- $\bullet \Rightarrow v = \frac{n\hbar}{m_e r_n}$
- From Eq1,  $m_e(\frac{n\hbar}{m_e r_n})^2 = \frac{Ze^2}{4\pi\epsilon_0 r_n} \Rightarrow r_n = \frac{n^2}{Z} \cdot \frac{4\pi\epsilon_0 \hbar^2}{m_e e^2}$
- Then, setting n=1, Z=1, we find the ground state in a hydrogen atom has  $r_b=5.29\cdot 10^{-11}$
- Then, from  $r_n$ , Eq2,
- $2(4\pi\epsilon_0\hbar)^2 \cdot \frac{Z^2}{n}$
- Then, for Z=1, n=1, or the case of the ground state of the hydrogen atom:
- $E_0 = -13.6 \ eV$

• 
$$E_{\gamma} = hf = E_0(\frac{1}{n_i^2} - \frac{1}{n_f^2})$$

## 6.4 Intro to Quantum Physics

### eqn 4.1 Compton Wavlength

#### Milikan's Experiment

For plates with p.d.  $V_0$ , separation d, and droplet of charge q and radius r moving at velocity v:

• 
$$\vec{f_e} = q \cdot \vec{E} = q \cdot \frac{V_0}{q} \cdot \hat{e}_z$$

• 
$$\vec{f}_g = mg \cdot \hat{e}_z$$
 and  $m = V \cdot \rho_{oil} = \frac{4}{3}\pi r^3 \rho_{oil} \Rightarrow \vec{f}_g = \frac{4}{3}\pi r^3 \rho_{oil} g \hat{e}_z$ 

- oil displacing air  $\Rightarrow \vec{f_b} = w_{displaced} = v_{air} \rho_{air}$
- $v_{air} = v_{oil} \Rightarrow \vec{f_b} = \frac{4}{3}\pi g r^3 \rho_{air} \hat{e}_z$
- from Stoke's Law,  $\vec{f_v} = -6\pi \eta_{air} \cdot rv$

### eqn 4.2 Milikan's Oil Drop Radius

If we set  $V_0 = 0$  and consider the terminal velocity of the drop  $v_{off}$ 

- $\bullet \ \vec{f_a} + \vec{f_b} + \vec{f_v} = 0$
- Note,  $\vec{f_g} + \vec{f_b} + \vec{f_v} = 0 \Rightarrow f_v = f_g f_b$
- $\frac{4}{3}\pi\rho_{oil}gr^3\hat{e}_z \frac{4}{3}\pi\rho_{air}gr^3\hat{e}_z 6\pi\eta_{air}v_{off}\hat{e}_z = 0$
- $\frac{4}{3}\pi(\rho_{oil} \rho_{air}g)r^3 = 6\eta_{air}v_{off}r$

• 
$$r = \sqrt{\frac{6\eta_{air}v_{off}}{2(\rho_{oil} - \rho_{air})g}}$$

## eqn 4.3 Milikan's Oil Drop Charge

apply voltage  $V_0$  such that drop moves upwards to a terminal velocity  $v_{on}$ 

- $\vec{f_e} + \vec{f'_v} + \vec{f_b} + \vec{f_e} = 0$
- $f_g \hat{e}_z + f'_v \hat{e}_z f_b \hat{e}_z f_e \hat{e}_z = 0$
- $f_e = f'_v + (f_g f_b) = f'_v + f_v$ , where  $f_v = \text{viscous force from off case}$
- $\frac{qV_0}{d} = 6\pi\eta_{air}rV_{on} + 6\pi\eta_{air}v_{off} = 6\pi\eta_{air}r(V_{on} + V_{off})$
- Leading to:

$$\bullet \quad q = \frac{g\sqrt{2}(V_{on} + V_{off}d)}{V_0} \cdot \sqrt{\frac{\eta_{air}^3 V_{off}}{(\rho_{oil} - \rho_{air})g}}$$

## 6.5 Maths Methods II