

# DISTANCE LEARNING PROGRAMME

(Academic Session : 2024 - 2025)

JEE (Main)
TEST # 14
29-12-2024

# JEE(Main+Advanced): LEADER TEST SERIES / JOINT PACKAGE COURSE

Test Type: Review (Unit Test # 06, 07, 08, 09 & 10)

### **ANSWER KEY**

### **PART-1: PHYSICS**

SECTION-I	Q.	1	2	3	4	5	6	7	8	9	10
	A.	Α	С	Α	В	D	Α	В	D	Α	Α
	Q.	11	12	13	14	15	16	17	18	19	20
	A.	С	D	С	С	С	Α	В	С	С	Α
SECTION-II	Q.	1	2	3	4	5					
SECTION-II	A.	6	5	9	1	27					

### **PART-2: CHEMISTRY**

SECTION-I	Q.	1	2	3	4	5	6	7	8	9	10
	A.	Α	D	Α	D	Α	С	D	Α	D	В
	Q.	11	12	13	14	15	16	17	18	19	20
	A.	В	С	В	В	D	D	D	D	Α	В
SECTION-II	Q.	1	2	3	4	5					
SECTION-II	A.	5	9	9	0	147					

# **PART-3: MATHEMATICS**

SECTION-I	Q.	1	2	3	4	5	6	7	8	9	10	
	SECTION I	Α.	Α	В	Α	Α	В	С	В	В	С	Α
	SECTION-I	Q.	11	12	13	14	15	16	17	18	19	20
		A.	D	С	С	В	С	Α	D	В	В	D
	SECTION-II	Q.	1	2	3	4	5					
SECTION-II	SECTION-II	Α.	1	18	1	4	99					

# (HINT – SHEET)

# **PART-1: PHYSICS**

# **SECTION-I**

1. Ans (A)

I = 2 × 0.01 × 5 = F<sub>0</sub>t  
⇒ F = 
$$\frac{2 \times 0.01 \times 5}{0.01}$$
 = 10 N

2. Ans (C)

$$f = mg$$
=  $20 \times 10^{-3} \times 10$   
=  $10^{-1} \times 2$   
=  $2 \times 10^{-1} \text{ N}$   
 $F_{photon} = fg$   

$$\frac{P}{C} = 2 \times 10^{-1}$$
  
 $P = 2 \times 10^{-1} \times 3 \times 10^{8}$   
 $P = 6 \times 10^{7} \text{ W}$ 

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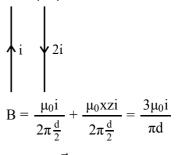
3. Ans (A)

$$I = \frac{Mr^2}{2} + \frac{mr^2}{4} = \frac{3}{4}mr^2$$

4. Ans (B)

$$m_1 v \ell = \left(\frac{m_2 \ell^2}{3} \omega\right) + m_1(\ell \omega) \ell$$

5. Ans (D)



$$\mathbf{F} = \mathbf{q}\vec{\mathbf{v}} \times \vec{\mathbf{B}} = \mathbf{0}$$

# 6. Ans (A)

The work done by the external agent = change in potential enrgy

= 
$$-(MB \cos \theta_2) - (-MB \cos \theta_1)$$
  
=  $-MB (\cos 60^\circ - \cos 0^\circ) = \frac{1}{2}MB$   
=  $\frac{1}{2} (1.0 \times 10^4 \text{J/T}) (4 \times 10^{-5} \text{T}) = 0.2 \text{J}$ 

# 7. Ans (B)

$$\begin{split} &T_S = 3mg \\ &\frac{3mg}{A_s} = y_S \times \frac{\Delta \ell_S}{\ell_S} \\ &\frac{2mg}{A_b} = y_b \times \frac{\Delta \ell_b}{\ell_b} \frac{\Delta \ell_b}{\ell_b} \\ &\frac{\Delta \ell_s}{\Delta \ell_b} = \frac{3}{2} \times \frac{\ell_s}{\ell_b} \times \frac{A_b}{A_S} \times \frac{y_b}{y_S} = \frac{3}{2} \times \frac{a}{b^2 c} \end{split}$$

# 8. Ans (D)

$$\vec{S} = \frac{\vec{E} \times \vec{B}}{\mu_0}$$

is called the Poynting vector after its discoverer, John Poynting. The average magnitude of  $\vec{S}$  is the intensity of the wave, and the direction of  $\vec{S}$  is the direction of propagation of the wave.

# 9. Ans (A)

As ratio of slit widths = Ratio of intensities

$$\therefore \frac{I_1}{I_2} = \frac{9}{4} \text{ or } \frac{a_1^2}{a_2^2} = \frac{9}{4} \Rightarrow \frac{a_1}{a_2} = \frac{3}{2}$$

$$a_{\text{max}} = a_1 + a_2 = 3 + 2 = 5 \; ; \; a_{\text{min}} = 3 - 2 = 1$$

$$\frac{I_{\text{max}}}{I_{\text{min}}} = \frac{(a_1 + a_2)^2}{(a_1 - a_2)^2} = \frac{(3 + 2)^2}{(3 - 2)^2} = \frac{25}{1}$$

### 10. Ans (A)

$$f = \frac{4}{\frac{2 \times 0.4}{2 \times 5}} \times \sqrt{\frac{90}{0.1}}$$
$$= \frac{2 \times 5}{2} \times 30 = 150 \text{Hz}$$

### 11. Ans (C)

$$\frac{A_R}{A_i} = \frac{\sqrt{\mu_1} - \sqrt{\mu_2}}{\sqrt{\mu_1} + \sqrt{\mu_2}}$$

# 12. Ans (D)

$$\omega = 2\pi \text{ f}$$
  
= 1.5 × 10<sup>3</sup>  
A =  $\frac{6}{2}$  = 3 cm = 0.03 m

# 13. Ans (C)

Loudness of sound is given by

$$dB = 10 \log \frac{I}{I_0}$$

$$120 = 10 \log \left(\frac{I}{I_0}\right)$$

$$\Rightarrow$$
 I = 1 W/m<sup>2</sup>

$$Also \quad I = \frac{P}{4\pi r^2} = \frac{2}{4\pi r^2}$$

$$r = \sqrt{\frac{2}{4\pi}} = \sqrt{\frac{1}{2\pi}} \text{ m} = 0.399 \text{ m} \approx 40 \text{ cm}$$

# 14. Ans (C)

When KE = PE then total energy is half kinetic and half potential.

So, KE = PE  

$$= \frac{1}{2}m\omega^2 (a^2 - x^2) = \frac{1}{2}m\omega^2 x^2$$

$$= \frac{1}{2}m\omega^2 a^2 = m\omega^2 x^2$$

$$x^2 = \frac{a^2}{2} \Rightarrow x = \frac{a}{\sqrt{2}}$$

# 15. Ans (C)

$$P = \left(\frac{N}{t}\right) 250 \,\text{MeV}$$

$$\left(\frac{N}{t}\right) = \frac{100 \times 10^6}{250 \times 1.6 \times 10^{-13}}$$

$$= 2.5 \times 10^{18}$$

: Number of neutron per second

$$= 2.5 \times 2.5 \times 10^{18}$$
$$= 6.25 \times 10^{18}$$

# 19. Ans (C)

Theory based

20. Ans (A)
Conceptual.



# PART-1: PHYSICS SECTION-II

# 1. Ans (6)

$$C = 2mF$$

$$\omega = \frac{1}{\sqrt{LC}} = \frac{1000}{\sqrt{1.2 \times 10^{-3}}}$$

$$q_0 = 24nC$$
,  $q = 12 mC$ 

$$i = \omega \sqrt{q_0^2 - q^2}$$

$$i = \frac{1000}{\sqrt{1.2 \times 10^{-3}}} \sqrt{144 \times 3 \times 10^{-12}} = 0.60$$

# 2. Ans (5)

$$\Delta 1 = \alpha \ell \Delta T$$

$$\frac{\Delta \ell}{\ell} = \alpha \Delta T$$

$$\sigma_b = \frac{T}{A} = \frac{y\Delta\ell}{\ell} = y\alpha\Delta T$$

$$\Rightarrow \Delta T = \frac{\sigma_b}{y\alpha} = \frac{4 \times 10^8}{2 \times 10^{11} \times 10^{-5}} = 200^{\circ} \text{C}$$

# 3. Ans (9)

Heat developed in 1 sec =  $\frac{20}{100} \times 4200$ J

$$\therefore$$
 Q = ms $\Delta$ T

$$\Rightarrow \frac{4200 \times 20}{100} = 2 \times 5 \times 4.2 \times 1.8 \times \Delta T$$

 $(\Delta T \text{ is in } ^{\circ}C)$ 

$$\Rightarrow \Delta T = 11.11 = \frac{100}{9} ^{\circ} C$$

$$\therefore \Delta T \text{ in } 10s = \frac{100}{9}^{\circ} C$$

# 4. Ans (1)

The change in internal energy of the gas is

$$\Delta U = 1.5 \text{ nR } (\Delta T) = 1.5 (1 \text{ mol}) (\frac{25}{3} \text{ J/mol-K}) (2\text{K})$$
  
= 25 J

The heat given to the gas = 42 J

The work done by the gas is

$$\Delta W = \Delta Q - \Delta U = 42 J - 25 J = 17 J$$

If the distance moved by the piston is x, the work done is

$$\Delta W = (100 \text{ kPa}) (8.5 \text{ cm}^2) \text{ x}$$

Thus, 
$$(10^5 \text{ N/m}^2) (8.5 \times 10^{-4} \text{ m}^2) \text{ x} = 17 \text{ J or}$$
,

$$x = 0.2m = 20 cm$$

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# 5. Ans (27)

$$Q_{absorbed} = Q_{emitted}$$

$$e(I A_{projected}) = e \sigma A T^4$$

$$1944 \times \pi R^2 = 6 \times 10^{-8} \times 4\pi R^2 T^4$$

$$T = 300 \text{ K} = 27^{\circ}\text{C}$$

# PART-2 : CHEMISTRY SECTION-I

# 1. Ans (A)

Glucose is soluble in water due to presence of alcohol functional group and extensive hydrogen bonding.

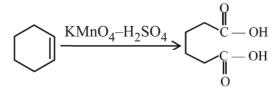
Glucose exist is open chain as well as cyclic forms in its aqueous solution.

Glucose having 6C atoms so it is hexose and having aldehyde functional group so it is aldose.

Thus, aldohexose.

Glucose is monomer unit in sucrose with fructose.

# 2. Ans (D)



# 3. Ans (A)

Stability: 
$$Cu_{(aq.)}^{2+} > Cu_{(aq.)}^+$$
.

Electrode potential is more important than electronic configuration for determining stable oxidation state.

# 4. Ans (D)

- (i) In VOSO<sub>4</sub>, 'V' is in +4 oxidation state. So it act as oxidising agent.
- (ii) Cr<sub>2</sub>O<sub>3</sub> is an amphoteric oxide.
- (iii) In RuO<sub>4</sub>, 'Ru' is in +8 oxidation state. So it act as oxidising agent.
- (iv) Red colour of ruby is due to the presence of  $Cr^{+3}$  ions in  $Al_2O_3$ .

# 5. Ans (A)

$$CO(g) + H_2O(g) \rightleftharpoons CO_2(g) + H_2(g)$$

Initial

)

Equilibrium 1-x 5-x x 1+x

Keq = 
$$\frac{1}{3}$$
 =  $\frac{x(1+x)}{(1-x)(5-x)}$ 

$$x = \frac{1}{2}$$

# 6. Ans (C)

$$M \mid M^{+2} \parallel X / X^{2-}$$

$$E_{cell}^{o} = E_{M/M^{+2}}^{o} + E_{X/X^{-2}}^{o}$$

$$=-0.46+0.34=-0.12$$
V

As  $E_{cell}^{o}$  is negative so anode becomes cathode and cathode become anode. Spontaneous reaction will be  $M^{+2} + X^{2-} \longrightarrow M + X$ 

# 7. Ans (D)

$$\stackrel{\text{O}}{\longrightarrow} \xrightarrow{\text{HCHO+NaOH}}$$

# 9. Ans (D)

$$PbS + HNO_3 \rightarrow Pb(NO_3)_2 + NO + S + H_2O$$

Nitrous oxide  $(N_2O)$  is not formed during the reaction.

# 10. Ans (B)

(A) Pb<sub>3</sub>O<sub>4</sub> is insoluble in water or do not react with water.

(B) 
$$2KO_2 + 2H_2O 2KOH + H_2O_2 + O_2(g) \uparrow$$

(C) 
$$Na_2O_2 + 2H_2O$$
  $2NaOH + H_2O_2$ 

(D) 
$$\text{Li}_2\text{O}_2 + 2\text{H}_2\text{O}$$
  $2\text{LiOH} + \text{H}_2\text{O}_2$ 

# 11. Ans (B)

$$AB_{2(s)} \rightleftharpoons A^{+2}_{(aq)} + 2B^{-}_{(aq)}$$

$$K_{sp} = [A^{+2}][B^{-}]^{2}$$

$$= 1.2 \times 10^{-4} \times (2.4 \times 10^{-4})^2$$

$$= 6.91 \times 10^{-12} \,\mathrm{M}^3$$

HS-4/8

# 12. Ans (C)

For endothermic reaction Keq.  $\propto \left(\frac{1}{\text{temperature}}\right)^{-1}$ 

# 13. Ans (B)

$$\begin{array}{c}
\stackrel{\text{NH}_2}{\longrightarrow} & \stackrel{\stackrel{\text{}}{\longrightarrow}}{\longrightarrow} \\
\stackrel{\text{NaNO}_2+ \text{HCl}}{\longrightarrow} & \stackrel{\stackrel{\text{}}{\longrightarrow}}{\longrightarrow} \\
\end{array}$$

$$\begin{array}{c}
\text{C1} & \text{OH} \\
& & \text{NaOH, 623K, 300 atm} \\
& & \text{H}^{+}
\end{array}$$

# 14. Ans (B)

Apart from aldehyde, Formic acid



also gives silver mirror test with ammonical silver nitrate.

# 15. Ans (D)

Hydrazine (NH<sub>2</sub>-NH<sub>2</sub>) have no carbon so does not show Lassaigne's test.

# 16. Ans (D)

$$\begin{array}{c}
O \\
O \\
\hline
Anhy.AlCl_3Cl/H^{+}
\end{array}$$

$$\begin{array}{c}
O \\
OH \\
OH \\
OH
\end{array}$$

$$\begin{array}{c}
Oi)SOCl_2 \\
Oii)Zn-Hg/HCl
\end{array}$$

$$\begin{array}{c} & & \\$$

# 17. Ans (D)

The E° value for  $Ce^{4+}/Ce^{3+}$  is +1.74 V because the most stable oxidation state of lanthanide series elements is +3. It means  $Ce^{3+}$  is more stable than  $Ce^{4+}$ .

#### 20. Ans (B)

$$E = E^{0} - \frac{2.303RT}{2F} log \frac{[Q][H^{+}]^{2}}{[H_{2}Q]}$$

$$E = 1.30 - \frac{0.06}{2} log \frac{0.01 \times (10^{-4})^{2}}{0.1}$$

$$= 1.30 + 0.27 = 1.57 \text{ V}$$

# PART-2: CHEMISTRY **SECTION-II**

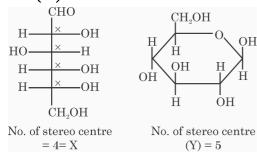
#### 1. Ans (5)

$$[Ba^{+2}] = \frac{25 \times 0.05}{50} = 0.025M$$
$$[F^{-}] = \frac{25 \times 0.02}{50} = 0.01M$$
$$[Ba^{+2}] [F^{-}]^{2} = 25 \times 10^{-7}$$

$$K_{\rm sp} = 5 \times 10^{-7} \text{ (given)}$$

Ratio = 
$$\frac{[Ba^{+2}][F^{-}]^{2}}{K_{sn}} = 5$$

#### 3. Ans (9)

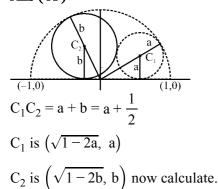


#### 4. Ans (0)

$$X \equiv Cr O_2 Cl_2$$
 "sp<sup>3</sup>"

# **PART-3: MATHEMATICS SECTION-I**

### 1. Ans (A)



### 2. Ans (B)

Point of intersection of normals 3x - 4y - 7 = 0 and

$$2x - 3y - 5 = 0$$
 is  $(1, -1)$ 

Radius is 
$$\sqrt{(2-1)^2 + (0+1)^2} = \sqrt{2}$$

Equation of circle is  $(x-1)^2 + (y+1)^2 = 2$ 

#### 3. Ans (A)

Centre = 
$$(1, -2)$$
 radius = 3

Image of centre (1, -2) about line

$$x - y + 5 = 0$$
 is  $(-7, 6)$ 

∴ Equation of circle

$$(x+7)^2 + (y-6)^2 = (3)^2$$

$$x^2 + y^2 + 14x - 12y + 76 = 0$$

# Ans (A)



Area = 
$$\int_{0}^{1} (x^2 + 1) dx + \frac{1}{2} = \frac{11}{6}$$

#### 5. Ans (B)

$$\frac{dP(t)}{dt} = \frac{1}{2}P(t) - 450$$

integrate

$$\int \frac{dP}{P - 900} = \int \frac{1}{2} dt$$

$$\ln |(P - 900)| = \frac{1}{2} t + C \qquad \dots (1)$$

given 
$$t = 0 \rightarrow P = 850$$

$$\therefore$$
 C =  $\ell$ n 50

from (1)

$$\ln|(P - 900)| = \frac{1}{2}t + \ln 50$$

$$\begin{aligned} &\frac{1}{2}t = \ell n \left| \left( \frac{P - 900}{50} \right) \right|; & t = 2\ell n \left| \left( \frac{P - 900}{50} \right) \right| \\ & t = 2\ell n \frac{900}{50}; & t = 2\ell n 18 \end{aligned}$$

# 6. Ans (C)

$$\left(\frac{2+\sin x}{1+y}\right)\frac{dy}{dx} = \cos x$$

$$\frac{dy}{dx} = \frac{(\cos x)(1+y)}{(2+\sin x)}$$

$$\frac{dy}{dx} \sim \left(\frac{\cos x}{2+\sin x}\right)y = \left(\frac{\cos x}{2+\sin x}\right)$$

$$IF = e^{\int \left(\frac{-\cos x}{2+\sin x}\right)dx}$$

$$2 + \sin x = t$$

$$\cos x dx = dt$$

$$=e^{-\int \frac{dt}{t}}$$

$$= e-ln t$$

$$= \frac{1}{t} = \frac{1}{2 + \sin x}$$

$$\frac{y}{2 + \sin x} = \int \frac{\cos x}{(2 + \sin x)^2} dx$$

$$= \int \frac{\cot x \csc x}{(2 \csc x + 1)^2} dx + C$$

$$2 \csc x + 1 = t$$

$$-2 \csc x \cot x dx = dt$$

$$= -\int \frac{dt}{2t^2} + C = \frac{1}{2t} + C$$

$$\frac{y}{2 + \sin x} = \frac{1}{2(2\csc x + 1)} + C$$

$$= \frac{\sin x}{2(2 + \sin x)} + C$$

$$y = \frac{\sin x}{2} + C(2 + \sin x)$$

$$y(0) = 2$$

$$\Rightarrow$$
 c = 1

$$y = \frac{3\sin x}{2} + 2$$

$$y\left(\frac{x}{2}\right) = \frac{7}{2}$$

# 7. Ans (B)

Centre = 
$$(3, 4)$$

$$PS + PS' = 2a$$

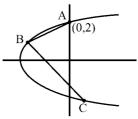
Here 
$$P = (1, 1), S(4, 5), S'(2, 3)$$

$$2a = 5 + \sqrt{5} \Rightarrow a = \frac{5 + \sqrt{5}}{2}$$

$$(x-3)^2 + (y-4)^2 = \left(\frac{5+\sqrt{5}}{2}\right)^2$$

HS-6/8

8. Ans (B)



$$\mathbf{B} \equiv (\mathbf{t}_1^2 - \mathbf{4}, \mathbf{t}_1)$$

$$C \equiv (t_2^2 - 4, t_2)$$

$$\left(\frac{t_2 - t_1}{t_2^2 - t_1^2}\right) \times \left(\frac{t_1 - 2}{t_1^2 - 4}\right) = -1$$

$$\Rightarrow t_1^2 + t_1t_2 + 2t_1 + 2t_2 + 1 = 0$$

$$\Rightarrow$$
  $(t^2 + 2)^2 - 4(2t_2 + 1) \ge 0$  (:  $t_1$  is real)

$$\Rightarrow$$
  $t_2 \in (-\infty, 0] \cup [4, \infty)$ 

least positive value of t<sub>2</sub> is 4.

# 9. Ans (C)

$$3x^2 - 10x - 5y - 20 = 0$$

$$3\left(x^2 - \frac{10x}{3}\right) = 5(y+4)$$

$$\left(x - \frac{5}{3}\right)^2 = \frac{5}{3}\left(y + \frac{17}{3}\right)$$

$$L(LR) = \frac{5}{3}$$

# 10. Ans (A)

$$M^2 = I - M$$
,  $M^3 = M(I - M) = M - M^2 = 2M - I$ 

$$M^4 = M^3M = 2I - 3M$$

$$\therefore$$
 n = 4

# 11. Ans (D)

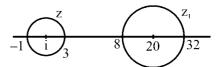
$$C = adj(adj(A)) = |A|A$$

$$\Rightarrow$$
 tr(C) = |A| tr(A) = 60

# 12. Ans (C)

$$|4w - 20| = 12 \text{ let } z_1 = 4w$$

$$\Rightarrow |z_1 - 20| = 12$$



so max value of  $|z - z_1|$  is 33.

# 13. Ans (C)

$$x^3 + y^2 = 5 \& x^3y^2 = -6$$

$$\Rightarrow$$
  $t^2 - 5t - 6 = 0$  has roots given by  $x^3 \& y^2$ .

or 
$$t = 6, -1$$

$$\Rightarrow$$
  $x^3 = -1 \& y^2 = 6$ 

$$\Rightarrow$$
 arg  $(x + iy) = \tan^{-1}(\pm\sqrt{6})$  or  $\tan^2\theta = 6$ 

# 14. Ans (B)

$$(x, y, z) = (9\lambda + 1, -\lambda + 2, -3\lambda - 3)$$

$$\frac{x - (9\lambda + 1)}{3} = \frac{y - (-\lambda + 2)}{-3} = \frac{z - (-3\lambda - 3)}{10}$$

$$=-2\left[\frac{3(9\lambda+1)-3(-\lambda+2)+10(-3\lambda-3)}{3^2+3^2+10^2}\right]$$

$$\frac{x - (9\lambda + 1)}{3} = \frac{y - (-\lambda + 2)}{-3} = \frac{z + (3\lambda + 3)}{10}$$

$$=-2\left[\frac{-59}{118}\right]=1$$

$$(x, y, z) = (9\lambda + 4, -\lambda - 1, 7 - 3\lambda)$$

$$\frac{x-4}{9} = \frac{y+1}{-1} = \frac{z-7}{-3}$$

# 15. Ans (C)

$$dr of line = \begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ 1 & -1 & 2 \\ 1 & 2 & 2 \end{vmatrix}$$

$$=-6\hat{i}+\hat{k}(3)=3(-2\hat{i}+\hat{k})$$

Hence vector equation of line is

$$\vec{r} = (-\hat{i} + \hat{j} + 3\hat{k}) + \lambda(2\hat{i} - \hat{k})$$

# 16. Ans (A)

Here solution R is reflexive since  $1 + a \times a > 0 \ \forall$  real numbers a. It is symmetric since 1 + ab > 0  $\Rightarrow 1 + ba > 0$ . However R is not transitive:

consider three real numbers  $2, -\frac{1}{6}$  and -2.

We have  $1 + 2 \times \left(-\frac{1}{6}\right) = \frac{2}{3} > 0$  and

$$1 + \left(-\frac{1}{6}\right)(-2) = \frac{4}{3} > 0$$
 Hence,

$$2R\left(-\frac{1}{6}\right)$$
 and  $\left(-\frac{1}{6}\right)R(-2)$  But  $2R-2$ 

since 
$$1 + 2(-2) = -3 \ge 0$$

# 17. Ans (D)

$$\frac{\sum_{i=1}^{n} a_i + 2i}{n} = \frac{n.\overline{a} + 2.\frac{n(n+1)}{2}}{n} = \overline{a} + (n+1)$$

# 18. Ans (B)

$$\bar{x} = \frac{8 + 12 + 13 + 15 + 22}{5} = \frac{70}{5} = 14$$

$$\sigma_x^2 = \frac{\sum (x_i - \bar{x})^2}{n}$$

$$= \frac{(-6)^2 + (-2)^2 + (-1)^2 + (1)^2 + (8)^2}{5}$$

$$= \frac{36 + 4 + 1 + 1 + 64}{5} = \frac{106}{5} = 21.2$$

# 19. Ans (B)

$${}^{5}C_{2}\left(\frac{1}{4}\right)^{2}\left(\frac{3}{4}\right)^{3}\frac{1}{4} = \frac{10.27}{4^{6}} = \frac{135}{2^{11}}$$

# 20. Ans (D)

Given 
$$P(A) = \frac{3}{5}$$
,  $P(B) = P(B) = \frac{1}{2}$ ,  $P(A \cap B) = \frac{1}{5}$ 

$$P(A \cap B) = P(A) - P(A \cap B)$$

$$\frac{1}{5} = \frac{3}{5} - P(A \cap B) \implies P(A \cap B) = \frac{2}{5}$$

$$\therefore P(\text{only A}) = \frac{1}{5}, P(\text{only B}) = \frac{1}{10}.$$

∴ P (A U B) = 
$$\frac{1}{5} + \frac{1}{10} + \frac{2}{5} = \frac{7}{10}$$

$$\Rightarrow P(\bar{A} \cap \bar{B}) = \frac{3}{10}$$

$$Now, P\left(\frac{\bar{A}}{A \cup \bar{B}}\right) = \frac{P\left(\bar{A} \cap \left(A \cup \bar{B}\right)\right)}{P\left(A \cup \bar{B}\right)} = \frac{P\left(\bar{A} \cap \bar{B}\right)}{P\left(A \cup \bar{B}\right)}$$

$$=\frac{\frac{3}{10}}{1-\frac{1}{10}}=\frac{1}{3}$$

# PART-3: MATHEMATICS SECTION-II

# 1. Ans (1)

$$y = mx + \frac{4}{m}$$

$$y = mx \pm 2\sqrt{2}\sqrt{1 + m^2}$$

$$\left(\frac{4}{m}\right)^2 = 8(1 + m^2)$$

$$\frac{2}{m^2} = 1 + m^2$$

$$m^2 = t$$

$$2 = t + t^2$$

$$t^2 + t - 2 = 0$$

$$t = -2, 1$$

# 2. Ans (18)

 $m^2 = 1$ 

$$det \{adj(A \ adj(2A))\} = |A \ adj(2A)|^2$$
$$= |A|^2 |2A|^4$$
$$= |A|^6 2^{12} = 2^{12} \times 3^6$$

$$\Rightarrow a + b = 18$$

# 3. Ans (1)

$$|\mathbf{a}| = 1 \Rightarrow \mathbf{a}\mathbf{a} = 1$$
  
 $|\mathbf{b}| = 1 \Rightarrow \mathbf{b}\mathbf{b} = 1$   
 $|\mathbf{c}| = 1 \Rightarrow \mathbf{c}\mathbf{c} = 1$ 

$$a + b + c = abc$$

$$1 \quad 1 \quad 1$$

$$\Rightarrow \frac{1}{ab} + \frac{1}{bc} + \frac{1}{ca} = 1$$
$$-\frac{1}{ab} + \frac{1}{bc} + \frac{1}{ca} = 1$$

$$\Rightarrow$$
 ab + bc + ca = 1

$$|ab + bc + ca| = 1$$

# 4. Ans (4)

$$L_{1}: \vec{r} = (\hat{i} + \hat{j}) + \lambda (\hat{i} - 2\hat{j} + \hat{k})$$

$$L_{2}: \vec{r} = 2\mu \hat{i} + (1 - \mu)\hat{j} + (1 + \mu)\hat{k}$$
S.D. between  $\vec{r} = \vec{a} + \lambda \vec{b} \& \vec{r} = \vec{c} + \mu \vec{d}$  is
$$S.D. = \frac{(\vec{a} - \vec{c}) \cdot (\vec{b} \times \vec{d})}{|\vec{b} \times \vec{d}|}$$

$$d = \frac{4}{\sqrt{11}} \Rightarrow (\frac{11}{4}d^{2}) = 4$$

# 5. Ans (99)

That total number of ways of choosing 4 shoes (in order) out of 10 pairs (or 20 shoes) is  $20 \times 19 \times 18 \times 17$ . The number of ways in which no pair is selected is  $20 \times 18 \times 16 \times 14$ . Thus, the probability of not getting a pair is:

$$\frac{20 \times 18 \times 16 \times 14}{20 \times 19 \times 18 \times 17} = \frac{224}{323}$$

Hence, the probability of getting at least one pair is: