Competishun

52/6, Opposite Metro Mas Hospital, Shipra Path, Mansarovar

Date: 17/11/2024

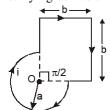
Time: 3 hours Max. Marks: 300

UTS-1 MT-2 (24-25)

Physics

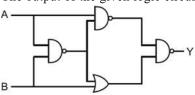
Single Choice Question

- Calculate the magnetic field amplitude produced by the radiation from a 100 W point sized bulb at a distance of 3 m from it. The efficiency of bulb is 2.5%
- **b)** $14 \times 10^{-8} \text{ T}$
- **c)** $0.7 \times 10^{-8} \text{ T}$
- d) $1 \times 10^{-8} \text{ T}$
- The magnitude of magnetic field at O (centre of the circular part) of the current carrying coil as shown is:



- **b)** $\frac{\mu_0 i}{2\pi} \left(\frac{3\pi}{2a} + \frac{\sqrt{2}}{b} \right)$ **c)** $\frac{\mu_0 i}{2\pi} \left(\frac{\pi}{3a} + \frac{3}{\sqrt{2b}} \right)$ **d)** $\frac{\mu_0 i}{4\pi} \left(\frac{3\pi}{2a} + \frac{\sqrt{2}}{b} \right)$

The output of the given logic circuit is



- $A\overline{B} + AB$
- b) $A\overline{B}$
- c) $AB + \overline{AB}$
- Particle is projected with initial velocity at an angle such that it grazes the top of two poles of same height at time 2 and 6 sec. First pole is at distance 35 m from the point of projection. Then match the list-I and List-II:

List-II List-I (P) Time of flight (in sec) (1)80(Q) Height of the pole (in m) (2)70(R) Maximum height reached by particle (in m) (3) 8(S) Distance between two poles (in m) (4) 60

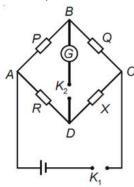
- a) $P \rightarrow 3; Q \rightarrow 2; R \rightarrow 4; S \rightarrow 1$
- **b)** $P \rightarrow 3; Q \rightarrow 4; R \rightarrow 2; S \rightarrow 1$
- c) $P \rightarrow 2; Q \rightarrow 4; R \rightarrow 1; S \rightarrow 3$
- d) $P \rightarrow 3: O \rightarrow 4: R \rightarrow 1: S \rightarrow 2$
- A metal ball of mass 0.1 kg is heated upto 500°C dropped into a vessel of heat capacity 800JK⁻¹ and containing 0.5 kg water. The initial temperature of water and vessel is 30°C. What is the approximate percentage increment in the temperature of the water? [Specific Heat Capacities of water and metal are, respectively, 4200Jkg⁻¹ K⁻¹ and $400 \text{Jkg}^{-1} K^{-1}$
 - a) 20%
- **b)** 25%
- c) 15%
- **d)** 30%

- The parallel combination of two air filled parallel plate capacitors of capacitance C and nC is connected to a battery of voltage V. When the capacitors are fully charged, the battery is removed and after that a dielectric material of dielectric constant K is placed between the plates of the first capacitor. The new potential difference of the combined system is
 - nVK + n

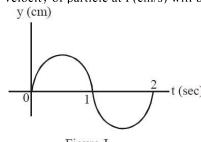
- c) $\frac{V}{K+n}$ d) $\frac{(n+1)V}{(K+n)}$
- The potential energy of a particle of mass 1 kg in motion along the x-axis is given by $U = (10 - 10\cos 6x)J$

The period of small oscillations is ($\pi^2 \approx 10$)

- A railway track is banked for a speed v, by making the height of the outer rail 'h' higher than that of the inner rail. The horizontal separation between the rails is d. The radius of curvature of the track is 'r': then which of the following relation is true?
- b) $\tan\left(\sin^{-1}\frac{h}{d}\right) = \frac{v^2}{rg}$ c) $\tan^{-1}\left(\frac{h}{d}\right) = \frac{v^2}{rg}$ d) $\frac{h}{r} = \frac{v^2}{dg}$
- In a Wheatstone bridge (see fig.), Resistances P and Q are approximately equal. When $R = 400 \Omega$, the bridge is balanced. On interchanging P and O, the value of R, for balance, is 405 Ω . The value of X is close to



- a) 404.5 ohm
- **b)** 401.5 ohm
- **c)** 402.5 ohm
- **d)** 403.5 ohm
- **Q10** The ratio of translational and rotational kinetic energies at 100 K temperature is 3:2. Then the internal energy of one mole gas at that temperature is [R = 8.3 J/mol-K]
 - a) 1175J
- **b)** 1037.5 J
- c) 2075 J
- Q11 A transverse sinusoidal wave moves along a string in the positive x direction. In figure (I) displacement of particle at P as a function of time is given and in figure (II) at a particular time t the snap shot of wave is shown. The wave velocity (cm/s) and velocity of particle at P(cm/s) will be:



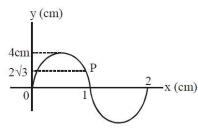
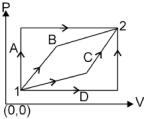


Figure-II

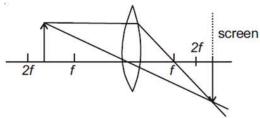
- a) $1\hat{i}$, $-3\pi\hat{j}$

- $1\hat{i}, 3\pi \hat{j}$ c) $1\hat{i}, 2\pi \hat{j}$ d) $-1\hat{i}, -2\pi \hat{j}$

Q12 An ideal gas is taken from state 1 to state 2 through optional path A, B C & D as shown in the PV diagram. Let Q, W & U represent the heat supplied, work done and change in internal energy of the gas respectively. Then,



- a) $Q_A Q_D = W_A W_D$
- **b)** $Q_B W_B > Q_C W_C$
- c) $W_A W_B < W_C W_D$
- d) $Q_A Q_B < Q_C Q_D$
- Q13 Formation of real image using a biconvex lens is shown below:



If the whole set up is immersed in water without disturbing the object and the screen positions, what will one observe on the screen?

a) Erect real image

b) No change

c) Image disappears

- d) Magnified image
- Q14 An electron of mass m and magnitude of charge |e| initially at rest gets accelerated by a constant electric field E. The rate of change of de-Broglie wavelength of this electron at time t ignoring relativistic effects is
 - |e|E√t

- Q15 Density of a planet of radius R varies with the distance r from its centre as

$$\rho = \rho_o \left(1 - \frac{r}{2R} \right)$$
 for, $0 \le r \le R$

for, r > R

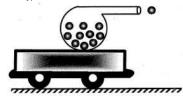
The acceleration due to gravity inside the planet at a distance r from its centre is

a) $\pi G \rho_0 r \left(\frac{4}{3} - \frac{r}{2R} \right)$

c) $\pi G \rho_0 r \left(1 - \frac{r}{2R} \right)$

- $\begin{array}{ll} ^{\text{b)}} & \pi G \rho_o r \bigg(\frac{2}{3} \frac{r}{4R} \bigg) \\ \\ ^{\text{d)}} & \pi G \rho_o r \bigg(2 \frac{r}{2R} \bigg) \end{array}$
- Q16 Two coherent sources of different intensities send waves which interfere. The ratio of the maximum intensity to the minimum intensity is 25. The intensities are in the ratio
 - a) 25:1
- **b)** 5:1
- **d)** 625:1

Q17 A cannon of mass M with n cannon balls each of mass 'm' rests on a smooth horizontal surface. The cannon fires n balls successively each with a muzzle speed v_0 relative to the cannon (in the state just before firing). The final recoil velocity of the cannon after firing the n shells will be:



- a) $\frac{nmv_0}{M}$
- **b)** $\frac{nmv_0}{M} imes \frac{1}{(n-1)} imes \frac{1}{(n-2)} imes \dots imes \frac{1}{2} imes \frac{1}{1}$
- c) $\frac{nmv_0}{M} \times \left[\frac{1}{(n-1)^2} + \frac{1}{(n-2)^2} + \dots + \frac{1}{2^2} + \frac{1}{1^2} \right]$
- d) $mv_0\left[rac{1}{M+(n-1)m}+rac{1}{M+(n-2)m}+\ldots\ldots+rac{1}{M}
 ight]$
- **Q18** Assertion: For the scattering of α -particles at a large angles, only the nucleus of the atom is responsible.

Reason: Nucleus is very heavy in comparison to electrons.

- a) If both assertion and reason are true and the reason is the correct explanation of the assertion.
- **b)** If both assertion and reason are true but reason is not the correct explanation of the assertion.
- c) If assertion is true but reason is false.
- d) If the assertion and reason both are false.
- Q19 Figure shows particles each connected rigidly by massless thin rods. Mass of each particle is m. In each case calculate moment of inertia of assembly about axis shown in figure.

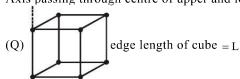
List-I (P) ed

edge length of cube = L

 $(1) 8 \mathrm{mL}^2$

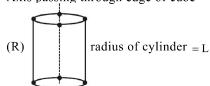
List-II

Axis passing through centre of upper and lower faces



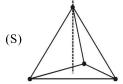
 $(2) 4mL^2$

Axis passing through edge of cube



 $(3) 3 mL^2$

Axis passing through symmetry axis of cylinder



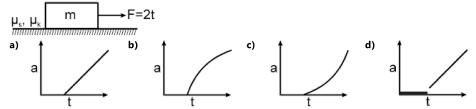
edge of tetrahedron = L

 $(4) mL^2$

Axis passing through the centroid of base

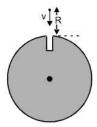
- a) $P \rightarrow 2; Q \rightarrow 4; R \rightarrow 1; S \rightarrow 3$
- **b)** $P \rightarrow 2; Q \rightarrow 3; R \rightarrow 1; S \rightarrow 4$
- c) $P \rightarrow 2; Q \rightarrow 1; R \rightarrow 1; S \rightarrow 4$
- d) $P \rightarrow 2; Q \rightarrow 1; R \rightarrow 1; S \rightarrow 3$

Q20 A force F = 2t (where t is time in seconds) is applied at t = 0 sec. to the block of mass m placed on a rough horizontal surface. The coefficient of static and kinetic friction between the block and surface are μ_s and μ_k respectively. Which of the following graphs best represents the acceleration vs time of the block. ($\mu_s > \mu_K$)



Numerical

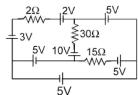
Q21 A fixed sphere of radius R and charge Q(uniformly) has a small groove of length $\frac{R}{4}$ as shown in figure. Another point charge of mass 'm' and charge 'q' is projected towards the groove with some velocity in given direction. If maximum value of v so that it does not strike to sphere is $\frac{1}{8}\sqrt{\frac{KQq}{mR\pi\varepsilon_0}}$, then find k. (Neglect gravity). (Both charges are of same nature)



Q22 In the shown figure a conducting ring of mass m=2kg and radius R=0.5 m. lies on a smooth horizontal plane with its plane vertical. The ring carries a current of $I=\frac{1}{\pi}A$. A horizontal uniform magnetic field of B=12T is switched on at t=0. The initial angular acceleration α in rad./sec² of the ring will be 4x if x is:



Q23 In the circuit shown, current through the resistance 2Ω is i_1 and current through the resistance 30Ω is i_2 . Find the ratio $\frac{i_1}{i_2}$.



A small town with a demand of 800 kW of electric power at 220 V is situated 15 km away from an electric plant generating power at 440 V. The resistance of the two-wire line carrying power is 0.5Ω per km. The town gets power from the line through a 4,000-220 V, step-down transformer at a sub-station in the town. Estimate the line power loss in the form of heat in kW.

Two tubes of radii r_1 and r_2 and lengths I_1 and I_2 respectively, are connected in series and a liquid flows through each of them in streamline conditions. P_1 and P_2 are pressure differences across the two tubes. If $P_2=4P_1$ and $I_2=\frac{l_1}{4}$, then the radius r_2 will be equal to $\frac{r_1}{x}$. Find 'x'

Chemistry

Single Choice Question

Q26 Consider the following reaction:

$$N_2O_4(g) \rightleftharpoons 2NO_2(g); \Delta H^0 = +58 \text{ kJ}$$

For each of the following cases (a, b), the direction in which the equilibrium shifts is:

(a) Temperature is decreased

(b) Pressure is increased by adding N_2 at constant T.

a) (a) Towards product, (b) towards reactant

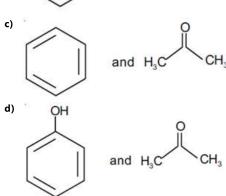
b) (a) Towards reactant, (b) no change

c) (a) Towards reactant, (b) towards product

d) (a) Towards product, (b) no change

Q27 The products formed in the reaction of cumene with O₂ followed by treatment with

dil. HCl are:



Q28 Electrolysis is carried out in three cells:

In cell (1): 1(M) CuSO₄ solution is electrolysed using Pt electrodes.

In cell (2): 1(M) CuSO₄ solution is electrolysed using Cu electrodes.

In cell (3): 1(M) KCl solution is electrolysed using Pt electrodes.

Which of the following is/are correct?

a) In cell (1): the pH of the solution is decreased

b) In cell (2): the pH of the solution remain/constant

c) In cell (3): the pH of the solution is increased

d) All of the given are correct

CHO

Q29 The tests performed on compound X and their inferences are:

Test Inference (1) 2,4-DNP test Coloured precipitate (2) Iodoform test Yellow precipitate (3) Azo-dye test No dye formation

Compound 'X' is:

Q30 The stepwise formation of $[Cu(NH_3)_4]^{2+}$ is given below

$$Cu^{2+} + NH_3 \xrightarrow{\ \ \ \ \ } \left[Cu(NH_3) \right]^{2+}$$

$$\left[Cu(NH_3)\right]^{2+} + NH_3 \xrightarrow{K_2} \left[Cu(NH_3)_2\right]^{2+}$$

$$\left[\text{Cu(NH}_3)_2\right]^{2+} + \text{NH}_3 \xrightarrow{\text{K}_3} \left[\text{Cu(NH}_3)_3\right]^{2+}$$

$$\left[\text{Cu(NH}_3)_3\right]^{2+} + \text{NH}_3 \xleftarrow{\text{K}_4} \left[\text{Cu(NH}_3)_4\right]^{2+}$$

The value of stability constants K_1 , K_2 , K_3 and K_4 are 10^4 , 1.58×10^3 , 5×10^2 and 10^2 respectively. The overall equilibrium constants for dissociation of $[Cu(NH_3)_4]^{2^+}$ is a) 1.26×10^{-12} b) 9.28×10^{-12} c) 1.26×10^{12} d) 9.28×10^{-14}

Q31 The major product obtained in the following reaction is:

 $^{\mathbf{Q32}}$ $\mathrm{CCl_4}$ is inert towards hydrolysis but $\mathrm{SiCl_4}$ is readily hydrolysed because

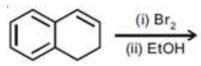
- a) carbon cannot expand its octet but silicon can expand its octet
- b) ionisation potential of carbon is higher than silicon
- c) carbon forms double and triple bonds
- d) electronegativity of carbon is higher than that of silicon

Q33 Given below are two statements: One is labeled as **Assertion A** and the other is labeled as **Reason R**

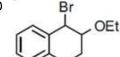
Assertion A: Zero orbital overlap is an out of phase overlap.

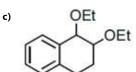
Reason R: It results due to different orientation/ direction of approach of orbitals. In the light of the above statements. Choose the correct answer from the options given below

- a) Both A and R are true and R is the correct explanation of A
- b) Both A and R are true but R is NOT the correct explanation of A
- c) A is true but R is false
- d) A is false but R is true
- Q34 The number of stereoisomers possible for 1,2- dimethyl cyclopropane is:
 - a) One
- **b)** Four
- c) Two
- d) Three
- Q35 A reaction takes place in 3 steps, the rate constant are K_1, K_2 and K_3 and energies of activation are 40,30 and 20 kJ respectively. If overall rate constant $K = \frac{K_1 K_3}{K_2}$, then overall energy of activation is
 - **a)** 10
- **b)** 15
- c) 30
- **d)** 60
- Q36 The major product of the following reaction is



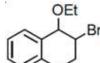
a)





b)

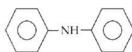
d)



- Q37 Which of the following is strongest Bronsted base
 - a)



b)



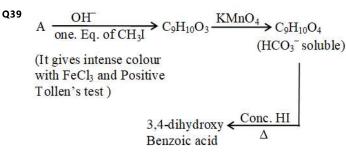
c)



d)



- **Q38** Amylopectin is composed of
 - a) β -D-glucose, C_1 — C_4 and C_2 — C_6 linkages
 - **b)** α -D-glucose, C_1 C_4 and C_2 C_6 linkages
 - c) β -D-glucose, C_1 — C_4 and C_1 — C_6 linkages
 - d) α -D-glucose, $C_1 C_4$ and $C_1 C_6$ linkages



Starting substrate 'A' is -

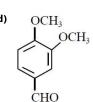


CH₃MgBr +



CHO

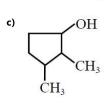


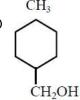


D, D is:

CH₂OH







- Q41 In the flame test of a mixture of salts, a green flame with blue centre was observed. Which one of the following cations may be present?
 - a) Cu²⁺
- **b)** Sr^{2+}
- d) Ca^{2+}
- Q42 Which of the following is not optically active?
 - a) $[Co(en)_3]^{3+}$

- b) $[Cr(ox)_3]^{3-}$
- c) $cis [CoCl_2(en)_2]^+$
- d) $trans [CoCl_2(en)_2]^+$

 CH_2OH

Q43 Given below are two statements: one is labelled as Assertion (A) and the other is labelled as **Reason (R)**. **Assertion (A)**: Cu²⁺ in water is more stable than Cu⁺.

Reason (R): Enthalpy of hydration for Cu²⁺ is much more negative than that of Cu⁺. In the light of the above statements, choose the correct answer from the options given below:

- a) Both (A) and (R) are correct and (R) is the correct explanation of (A).
- **b)** (A) is correct but (R) is not correct.
- c) (A) is not correct but (R) is correct.
- d) Both (A) and (R) are correct but (R) is not the correct explanation of (A).
- Q44 Which of the following orders are correct for the ionization energies?
 - (1) Ba < Sr < Ca
- (2) $S^{2-} < S < S^{2+}$ (3) C < O < N
- (4) Mg < Al < Si

- a) 1,2 and 4
- **b)** 1, 3 and 4
- c) 1, 2 and 3
- **d)** 1, 2, 3 and 4
- Change in internal energy (in kJ mol⁻¹)when 0.5 mole of Ar having a specific heat at constant pressure of 20.814 J g⁻¹ deg⁻¹ is heated from 27°C to 31°C at constant volume.
 - (Atomic mass of Ar = 40)
 - a) 1.65
- **b)** 3.64
- **c)** 1648
- **d)** 3.30

Q46 The values of observed and calculated molecular weights of calcium nitrate are respectively 65.6 and 164. The degree of dissociation (in %) of calcium nitrate will be

Numerical

- Q47 The pH of a saturated aqueous solution of CO_2 is 5; For H_2CO_3 , $Ka1 = 10^{-7}$ and $Ka_2 = 10^{-11}$. At the given pressure the solubility of CO_2 in water is 10^{-2} (M). What is the value of $-\log [CO_2^{2-}]$ in the nearest possible integers?
- Q48 In which of the following all bond length are not equal?
 PCl₅, SF₄, ClF₃, XeF₂, [SF₅]⁺, [ClF₄]⁺, [XeF₃]⁺, O₃, P₄ (white)
- **Q49** The work function (ϕ) of some metals is listed below. The number of metals which will show photoelectric effect when light of 300 nm wavelength falls on the metal is

Metal	Li	Na	K	Mg	Cu	Ag	Fe	Pt	W
(φ eV)	2.4	2.3	2.2	3.7	4.8	4.3	4.7	6.3	4.75

Q50 Formation of polyethene from calcium carbide takes place as follows:

$$\begin{split} &\text{CaC}_2 + \text{H}_2\text{O} \rightarrow \text{Ca(OH)}_2 + \text{C}_2\text{H}_2 \; ; \\ &\text{C}_2\text{H}_2 + \text{H} \rightarrow \text{C}_2\text{H}_4 \\ &\text{n(C}_2\text{H}_4) \rightarrow (-\text{CH}_2\text{-CH}_2\text{-}) \; \text{n}. \end{split}$$

The amount of polyethylene (in kg) possibly obtainable from 64.0 kg CaC₂ can be

Mathematics

Single Choice Question

- Q51 If x + y + z = 5 and xy + yz + zx = 3, then least and largest value of x are :
- **b)** $-1, \frac{13}{2}$ **c)** $\frac{17}{2}, 7$
- d) None of these

- The sum of series $\sum_{k=1}^{n} tan^{-1} \left(\frac{2k}{2+k^2+k^4} \right)$
 - a) $tan^{-1} (n^2 + n + 1)$
- **b)** $\frac{\pi}{4}$ $\tan^{-1}(n^2 n + 1)$
- c) $\frac{\pi}{4} + \cot^{-1} (n^2 + n + 1)$
- The sum of the series $1 + \frac{1}{4.2!} + \frac{1}{16.4!} + \frac{1}{64.6!} + \dots \infty$ is -

- **Q54** The number of common terms to the two A.P's 17 + 21 + 25 + + 417 and 16 + 21 +26 ++ 466 is
 - a) 20
- **b)** 19
- **c)** 21
- **d)** 18
- Q55 No. of values of x for which the sixth term of the expansion of E=
 - $\left[3^{\log_3\sqrt{9^{|x-2|}}}+7^{\frac{1}{5}\log_7[(4).3^{|x-2|}-9]}
 ight]^7$ is 567, is/are :

- d) None
- Q56 Number of ways of selecting 5 letters from letters of word INDEPENDENT, is -
- **b)** 68

- **Q57** Let $A_0 A_1 A_2 A_3 A_4 A_5$ be a regular hexagon inscribed in a circle of unit radius. Then the product of the lengths of the line segments $A_0 A_1$, $A_0 A_2$ and $A_0 A_4$ is-**a)** 3 **b)** $3\sqrt{3}$ **c)** 3 **d)** $3\sqrt{3}$

- Q58 Out of 21 tickets marked with numbers from 1 to 21, three are drawn at random. The chance that the numbers on them are in A.P. is
- **b)** $\frac{9}{133}$
- d) None of these
- If f(y) is inverse of g(y) and g(y)=y⁵+2y³+3y+4, then the value of $28\frac{d}{dy}(g^{-1}(y))$ at
 - y = -2 is
- **c)** 1/14
- Q60 An ellipse with major axis 4 and minor axis 2 touches both the coordinate axes, then locus of its focus is

 - a) $(x^2 y^2)(1 + x^2y^2) = 16x^2y^2$ b) $(x^2 y^2)(1 x^2y^2) = 16x^2y^2$.

 - c) $(x^2 + y^2)(1 + x^2y^2) = 16x^2y^2$ d) $(x^2 + y^2)(1 x^2y^2) = 16x^2y^2$.
- **Q61** The area enclosed between the curves $y=x^2$ and $y=\sqrt{|x|}$ is
 - a) 1/3
- **b)** 2/3

$$(\alpha = \pi/2)$$

; $\alpha \neq \pi/2$, where {.} represents fractional part function,

then

- a) $f(\alpha)$ is continuous at $\alpha = \pi/2$
- b) $\lim_{\alpha \to \pi/2} f(\alpha)$ exists but not continuous at $\alpha = \pi/2$
- c) $\lim_{\alpha \to \pi/2} f(\alpha)$ does not exist
- d) $\lim_{\alpha \to \pi/2} f(\alpha) = 1$
- Q63 Match the following

	Column - I	Column - II		
(A)	$g(x) = 2 - x^{1/3}$ and $f(g(x)) = -x + 5x^{1/3} - x^{2/3}$, the local maximum value of $f(x)$ is	(P)	0	
(B)	No. of points of intersection of the curves $\arg\left(\frac{z-3}{z-1}\right) = \frac{\pi}{4}$ and $z(1-i) + \bar{z}(1+i) - 4 = 0$	(Q)	1	
(C)	If $f(x) = ax^3 + bx^2 + cx + d$, $(a, b, c, d \in Q)$ and two roots of $f(x) = 0$ are eccentricities of a parabola and a rectangular hyperbola, then $a + b + c + d =$	(R)	2	
(D)	Number of solution of equation $1^x + 2^x + 3^x \dots + n^x = (n+1)^x$ are	(S)	3	

- a) $A \rightarrow Q$; $B \rightarrow S$; $C \rightarrow R$; $D \rightarrow P$
- **b)** $A \rightarrow S$; $B \rightarrow Q$; $C \rightarrow P$; $D \rightarrow Q$
- c) $A \rightarrow S$; $B \rightarrow Q$; $C \rightarrow R$; $D \rightarrow Q$
- d) $A \rightarrow S$; $B \rightarrow Q$; $C \rightarrow P$; $D \rightarrow R$

Q64

The value of [100(k-1)] where [x] represents the G.I.F. and $k=rac{\sum_{r=1}^{44^0}\cos r^o}{\sum_{r=1}^{44^0}\sin r^o}$ is

- a) 144

Q65 Solve $xdy + ydx + x^2y^3(xdy - ydx) = 0$

- a) $rac{1}{(xy)^{5/2}}=5\sqrt{rac{y}{x}}+c$ c) $rac{-5}{(xy)^{5/2}}=-2\sqrt{y/x}+c$
- b) $rac{5}{(xy)^{5/2}}=2\sqrt{y/x}+c$
- d) None of these

Q66 Let x = 2 be a local minima of the function $f(x) = 2x^4 - 18x^2 + 8x + 12$, $x \in (-4, 4)$. If M is local maximum value of the function $f(x) = 2x^4 - 18x^2 + 8x + 12$, $x \in (-4, 4)$. If M is local maximum value of the function $f(x) = 2x^4 - 18x^2 + 8x + 12$, $x \in (-4, 4)$. If M is local maximum value of the function $f(x) = 2x^4 - 18x^2 + 8x + 12$, $x \in (-4, 4)$. If M is local maximum value of the function $f(x) = 2x^4 - 18x^2 + 8x + 12$, $x \in (-4, 4)$. If M is local maximum value of the function $f(x) = 2x^4 - 18x^2 + 8x + 12$, $x \in (-4, 4)$.

Let A(0,1), B(1,1) and C(1,0) be the mid – points of the sides of a triangle with in Centre at the point D. If the focus of the parabola $y^2 = 4ax$ passing through D is

 $(\alpha + \beta\sqrt{2}, 0)$, where α and β are rational numbers, then $\frac{\alpha}{\beta^2}$ is equal to

Q68 Let $P(a_1, b_1)$ and $Q(a_2, b_2)$ be two distinct points on a circle with center $C(\sqrt{2}, \sqrt{3})$. Let O be the origin and OC be perpendicular to both CP and CQ. If the area of the triangle OCP is $\frac{\sqrt{35}}{2}$, then $a_1^2 + a_2^2 + b_1^2 + b_2^2$ is equal to_____. **a)** 26 **b)** 28 **c)** 12

Let $A = \begin{bmatrix} 1 & \frac{1}{51} \\ 0 & 1 \end{bmatrix}$. If $B = \begin{bmatrix} 1 & 2 \\ -1 & -1 \end{bmatrix} A \begin{bmatrix} -1 & -2 \\ 1 & 1 \end{bmatrix}$, then the sum of all the elements of the matrix

 $\sum_{n=0}^{\infty} B^n$ is equal to -

- a) 100
- **b)** 50
- c) 75
- **d)** 125

Q70 For the system of linear equations

$$2x - y + 3z = 5$$

$$3x + 2y - z = 7$$

$$4x + 5y + \alpha z = \beta$$

Which of the following is NOT correct?

- a) The system has infinitely many solutions for $\alpha = -5$ and $\beta = 9$
- **b)** The system has a unique solution for $\alpha \neq -5$ and $\beta = 8$
- c) The system has infinitely many solutions for $\alpha = -6$ and $\beta = 9$
- d) The system is inconsistent for $\alpha = -5$ and $\beta = 8$

Numerical

An equilateral triangle has its centroid at the origin and one side is x+y=1, then the sum of the slopes of the other two sides is

If
$$\int \frac{dt}{(t + \sqrt{t(1+t)})^2} = -2\log\left(1 + \sqrt{1 + \frac{1}{t}}\right) - f(t) + c \text{ then } 3\left(\lim_{t \to \infty} f(t)\right) =$$

- Let A={1,2,3,4,5}. The number of unordered pairs of subsets P and Q of A such that $P \cap Q = \phi$ is 'n'. Then the sum of digits of n is
- Q74 If (α,β,γ) is the foot of perpendicular drawn from the point P(1,0,3) to the line joining the points A(4,7,1) and B(3,5,3). Then the value of $3\gamma+6\beta+9\alpha=$
- Q75 The mean and standard deviation of 10 observations are 20 and 8 respectively. Later on, it was observed that one observation was recorded as 50 instead of 40. Then the correct variance is:

Answer Key

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Que.	1	2	3	4	5	6	7	8	9	10
Ans.	Α	D	В	D	А	D	Α	Α	С	С
Que.	11	12	13	14	15	16	17	18	19	20
Ans.	С	А	С	В	А	U	D	Α	С	D
Que.	21	22	23	24	25	26	27	28	29	30
Ans.	23	3	9	600	2	В	D	D	А	Α
Que.	31	32	33	34	35	36	37	38	39	40
Ans.	D	Α	А	D	С	D	D	D	В	А
Que.	41	42	43	44	45	46	47	48	49	50
Ans.	Α	D	А	С	А	75	10	6	4	28
Que.	51	52	53	54	55	56	57	58	59	60
Ans.	В	D	А	А	В	А	С	С	D	С
Que.	61	62	63	64	65	66	67	68	69	70
Ans.	В	С	В	С	А	А	В	D	А	С
Que.	71	72	73	74	75		·	·		
Ans.	4	3	5	46	13					