FIITJEE

ALL INDIA TEST SERIES

JEE (Advanced)-2025

CONCEPT RECAPITULATION TEST – II

PAPER -1

TEST DATE: 24-04-2025

ANSWERS, HINTS & SOLUTIONS

Physics

PART - I

SECTION - A

1. A

Sol. From first law of thermodynamics

$$\Delta Q = \Delta U + \Delta W$$

$$\Delta Q = nC_V (T_2 - T_1) + \frac{nR(T_1 - T_2)}{x - 1}$$

$$\frac{1}{n} \left(\frac{\Delta Q}{\Delta T} \right) = C_V - \frac{R}{x - 1}$$

$$\therefore (C-C_{\vee})(x-1) = -R$$

: graph is a rectangular, hyperbola

 \therefore co-ordinate of P₁ (O, C_P)

i.e.
$$\left(0, \frac{5}{2}R\right)$$
 and that of $P_4\left(\frac{5}{3}, 0\right)$

2. [

Sol. Impulse on ball

$$I = (1 + 0.5)\sqrt{2gh} \text{ m (vertically)}$$

Frictional impulse on block (horizontally)

$$= \mu I = 0.2 \times 1.5 \text{ m} \sqrt{2gh}$$

Decrease in velocity

$$\Delta V = \frac{\mu I}{m} = 0.3 \sqrt{2gh}$$

3.

For More J

Sol. Potential of centre of sphere =
$$\frac{Kq}{r} + V_i = \frac{Kq}{r}$$

where V_i = potential due to induced charge at centre = 0 [.:. Σq_i = 0 and all induced charges are equidistance from centre]

potential at point
$$P = \frac{Kq}{r} = \frac{Kq}{r_1} + V_i$$
 (For conductor all points are equipotential)

$$\therefore \qquad V_i = K \left(\frac{q}{r} - \frac{q}{r_1} \right)$$

4. A

$$2 \times \sin 60^{0} = (\sin 90^{0}) \times \frac{2}{(1+H^{2})}, \quad 2 \times \frac{\sqrt{3}}{2} = 1 \times \frac{2}{(1+H^{2})}$$
$$(1+H^{2}) = \frac{2}{\sqrt{3}}, \qquad H = \sqrt{(\frac{2}{\sqrt{3}}-1)}$$

Sol.
$$\Delta U = \frac{fR\Delta T}{2}$$

 $\Delta W = \frac{nR\Delta T}{1-x}$ where PV^x = constant. Here $x = -\frac{1}{2}$

6. ACD

7. ABCD

8. E

Sol. Mean value =
$$\frac{12.5 + 12.3 + 11.8 + 12.4 + 12.2 + 12.6}{6} = 12.3$$
 Mean absolute error =
$$\frac{\left|\Delta x_1\right| + \dots + \left|\Delta x_6\right|}{6}$$
 Relative error =
$$\frac{\text{Mean absolute error}}{\text{Mean value}}$$

9. C

10. C

Sol. Optical path difference at any point P on the screen,
$$\delta(P) = S_2P - S_1P - (\mu - 1)t$$
 and the intensity on the screen, at point $P = 4I_o \times \cos^2 \frac{\pi}{\lambda} \times \delta P$.

11. A

Sol. (P) A
$$\rightarrow$$
 B : V \downarrow , P constant \rightarrow T \downarrow , U \downarrow and Δ W is $-$ ve, Δ Q < 0, Δ U < 0

(Q) B
$$\rightarrow$$
 C : V is same, P \downarrow T \downarrow , U \downarrow , Δ Q < 0, Δ U < 0

No work is done.

(R) C
$$\rightarrow$$
 D : V $\uparrow \Rightarrow$ T \uparrow , $\Delta U > 0$, $\Delta Q > 0$, $\Delta W > 0$

(S) D
$$\rightarrow$$
 A : V decrease so Δ W < 0

$$T_D = T_A$$

$$\Delta U = 0$$

and then $\Delta Q < 0$.

SECTION - B

12. 6

Sol.
$$\frac{\Delta Q}{\Delta t} = K_A A \frac{dT}{dx}$$

$$\frac{dT}{dx}$$
 in conductor A = slope of graph = $\sqrt{3}$

Since both conductors are connected in series, same heat current will flow in A and B.

$$K_A (dT/dx)_A = K_B (dT/dx)_B$$

$$K_B = 6$$
.

13. 272

Sol. Loss in
$$KE = K_f - K_i$$

$$K_f = \frac{1}{2}mv_{cm}^2 + \frac{1}{2}I_{cm}\omega^2$$

$$K_i = \frac{1}{2}(0.08)(10)^2 + \frac{1}{2}(0.08)(6)^2$$

Apply conservation of momentum and angular momentum to get V_{cm} and $\omega.$

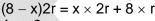
14. 2

$$\frac{R_1}{R_3} = \frac{R_2}{R_4}$$

as
$$R2 = R4$$

$$R1 = R3$$

Let the pointer at point 5 is moved to left by distance x to get nul point as shown in the figure. If resistance per unit length of wire 3 is r then that of wire 1 will be 2r.

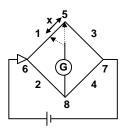


$$4x = 8$$

$$x = 2m$$

15.

Sol.
$$v_0 \le \frac{gR}{3} (7\cos\alpha - 4)$$



16. 1
Sol.
$$a = \omega^2 A$$
 $f_1 = (1)\omega^2 A \le 6$...(1)
 $f_2 = 3\omega^2 A \le 12$
 $\omega^2 A \le 4$...(2)
 $\Rightarrow \omega^2 A = 4$

$$A = \frac{4}{\left(\frac{k}{m}\right)} = \frac{4 \times 6}{24} = 1$$

17. 8
Sol.
$$\frac{4\ell_1}{3} = \frac{2\ell_2}{4}$$

Chemistry

PART – II

SECTION - A

Sol.
$$K_p = p_{NH_3} \times p_{HCI} = 6.25$$

or,
$$p_{NH_3} = p_{HCI} = \sqrt{6.25} = 2.5 \text{ atm}$$

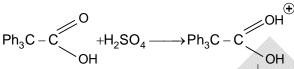
 $P_{equilibrium} = 2.5 + 2.5 = 5 atm$

Mole of HCl at equilibrium = 1

Total moles of gases at equilibrium = 1 + 1 = 2

or,
$$V = \frac{nRT}{P} = \frac{2 \times 0.0821 \times 304.55}{5} = 10L$$

19. Sol. В



$$\begin{array}{c} Ph_{3}C - C & Ph_{3}C - C \\ OH & Ph_{3}C - C \\ OH & Ph_{3}C - C \\ \end{array} \xrightarrow{P} \begin{array}{c} OH \\ OH \\ Ph_{3}C + CO & Ph_{3}C - C \\ \hline \\ CH_{3}OH \\ \end{array} \xrightarrow{P} \begin{array}{c} Ph_{3}C - C \\ OH \\ \hline \\ Ph_{3}C + CO & Ph_{3}C - C \\ \hline \\ CH_{3} & Ph_{3}COCH_{3} \\ \end{array}$$

- 20. В
- It contains three oxygen atom in the three peptide linkages and two oxygen atoms in the Sol. free COOH – group.
- 21.

For More J

Sol. Molality of benzoic acid (m) =
$$\frac{2}{122} \times \frac{1000}{25} = 0.655$$

$$2C_6H_5COOH \longrightarrow (C_6H_5COOH)_2$$

Initial 1 0

Initial 1

$$\frac{x}{2}$$

$$\therefore \text{ van't Hoff factor (i)} = \frac{1 - x + \frac{x}{2}}{1} = 1 - \frac{x}{2}$$

$$\Delta T_f = i \times K_f \times m$$

or,
$$1.62 = \left(1 - \frac{x}{2}\right) \times 4.9 \times 0.655$$

On solving, x = 0.982

- 22. AC
- Sol. The basic nature of the oxides of iron follows the following order. FeO > Fe₃O₄ > Fe₂O₃
- 23. AD
- Sol. The reaction is first order w.r.t X. So it's half-life is $\frac{0.693}{k}$. The reaction is second order with respect to Y. So its half-life is given by $\frac{1}{ka}$.
- 24. AD
- Sol. $6 \text{NaOH} + 4 \text{S} \longrightarrow 2 \text{Na}_2 \text{S} + \text{Na}_2 \text{S}_2 \text{O}_3 + 3 \text{H}_2 \text{O}$ $4 \text{P} + 3 \text{NaOH} + 3 \text{H}_2 \text{O} \longrightarrow 3 \text{NaH}_2 \text{PO}_2 + \text{PH}_3$ $\text{Si} + 2 \text{NaOH} + \text{H}_2 \text{O} \longrightarrow \text{Na}_2 \text{SiO}_3 + 2 \text{H}_2$ $3 \text{Cl}_2 + 6 \text{NaOH} \xrightarrow{\text{Heat}} 5 \text{NaCI} + \text{NaCIO}_3 + 3 \text{H}_2 \text{O}$ $\text{Cl}_2 + 2 \text{NaOH} \xrightarrow{\text{Cold}} \text{NaCI} + \text{NaOCI} + \text{H}_2 \text{O}$
- 25. D
- Sol. The orbitals which hold the electrons are:
 - (P) 4s
 - (Q) 4s and 3d
 - (R) 3p
 - (S) 4p
- 26. C
- Sol. The compound in (P) has 16 isomers(stereo)
 The compound in (Q) has 4 isomers(stereo)
 The compound in (R) has 2 isomers(stereo)
 - The compound in (S) has 4 isomers(stereo)
- 27. C
- Sol. $Cr^{2+}(3d^4 \text{ or } t_{2g}^3e_g^1)$ shows Jahn Teller distortion.

 $Co^{2+}(t_{2g}^6e_g^1)$ also shows Jahn Teller distortion.

 $Mn^{2+}(t_{2q}^3e_q^2 \text{ or } 3d^5)$ has half filled electronic configuration.

 $Ni^{2+}(t_{2q}^6e_q^2)$ has same configuration for strong field as well as weak field ligands.

- 28. B
- Sol. Azeotropic mixtures and ideal solutions can't be separated by distillation.

SECTION - B

29. 2

For More Jo

Sol. Meq of HCl = $19.8 \times 0.1 = 1.98$

Meg of 20 ml solution of $Na_2CO_3.xH_2O = 1.98$

Meq of 100 ml solution of $Na_2CO_3.xH_2O = 1.98 \times 5 = 9.9$

$$\therefore$$
 Meq of 0.7g Na₂CO₃.xH₂O = 9.9

$$\therefore \frac{0.7}{\frac{106 + 18x}{2}} \times 1000 = 9.9$$

On solving, $x = 1.98 \approx 2$

Sol.
$$Mg(OH)_2 \rightleftharpoons Mg^{2+} + 2OH^{-}$$

$$S = 4s^3 = 5 \times 10^{-10}$$

or, $S = 5 \times 10^{-4}$
[OH] = $2S = 2 \times 5 \times 10^{-4} = 10^{-3}$
 $\therefore p^{OH} = -log[OH] = 3$

Sol.

$$\mathsf{CH_3} - \mathsf{C} - \mathsf{CH} - \mathsf{C} - \mathsf{C}_2 \mathsf{H}_5 + \mathsf{CH}_3 \mathsf{Br} \longrightarrow \mathsf{CH}_3 - \mathsf{C} - \mathsf{CH} - \mathsf{C} - \mathsf{C}_2 \mathsf{H}_5$$

$$CH_{3} - C - CH - C - C_{2}H_{5} \longleftrightarrow CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \longleftrightarrow CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \to CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \to CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \to CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \to CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \to CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \to CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \to CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \to CH_{3}Br \to CH_{3} - C = CH - C - C_{2}H_{5} \to CH_{3}Br \to CH_{3} - C \to CH_{3}Br \to CH_{3} - C \to CH_{3} -$$

$$CH_{3} - C - CH - C - C_{2}H_{5} \longleftrightarrow CH_{3} - C - CH = C - C_{2}H_{5}$$

$$CH_{3} - C - CH = C - C_{2}H_{5} \longleftrightarrow CH_{3} - C - CH = C - C_{2}H_{5}$$

$$CH_{3} - C - CH = C - C_{2}H_{5} \longleftrightarrow CH_{3}Br$$

32. 7

Sol. The isomers are

 $[Co(H_2O)_4 BrCl]NO_2$, $[Co(H_2O)_4 (NO_2)Br]Cl$, $[Co(H_2O)_4 (ONO)Br]Cl$, $[Co(H_2O)_4 (NO_2)Cl]Br$, $[Co(H_2O)_4 (ONO)Cl]Br$, $[Co(H_2O)_3 (ONO)BrCl]$. $[Co(H_2O)_3 (NO_2)BrCl]$.

33. 5

$$\begin{array}{c} \mathsf{CH}_3 \\ | \\ \mathsf{H}_3\mathsf{C} - \mathsf{C} - \mathsf{CH}_3 \end{array}$$

Sol. The major product is

It forms five monochloro products.

34.

For More Join:

Sol.
$$N_2 + 3F_2 \Longrightarrow 2NF_3$$

$$\begin{array}{ccc} & N_2 + 3F_2 \Longrightarrow 2NF_3 \\ \text{Initial} & a & 3a & 0 \\ \text{At Equi} & \left(a - \frac{a}{4}\right) \left(3a - \frac{3a}{4}\right) \left(\frac{2a}{4}\right) \end{array}$$

Total moles at equilibrium = $\frac{7a}{2}$

$$p_{NF_3} = \frac{2a/4}{7a/2} \times 28 = 4 \text{ atm}$$

Mathematics

PART – III

SECTION - A

Sol. We have,
$$(\cos\theta + i\sin\theta)(\cos 2\theta + i\sin 2\theta)..... \times (\cos n\theta + i\sin n\theta) = 1$$

$$\Rightarrow \cos(\theta + 2\theta + 3\theta + + n\theta) + i\sin(\theta + 2\theta + + n\theta) = 1$$

$$\Rightarrow \cos\left(\frac{n(n+1)}{2}\theta\right) + i\sin\left(\frac{n(n+1)}{2}\theta\right) = 1$$

$$\Rightarrow \cos\left(\frac{n(n+1)}{2}\theta\right) = 1 \text{ and } \sin\left(\frac{n(n+1)}{2}\theta\right) = 0$$

$$\Rightarrow \frac{n(n+1)}{2}\theta = 2m\pi$$

$$\Rightarrow \theta = \frac{4m\pi}{n(n+1)}, \text{ where } m \in Z.$$

Sol. We have
$$f'(x) = f(x) \Rightarrow \frac{f'(x)}{f(x)} = 1 \Rightarrow \ln(f(x)) = x + c$$

As
$$x = 0$$
, $f(0) = 1 \implies c = 0$

Now,
$$g(x) = e^{x}(x+1)^{2} - e^{x} = e^{x}(x^{2} + 2x)$$

So,
$$\int_{0}^{1} f(x)g(x)dx = \int_{0}^{1} e^{2x}(x^{2} + 2x)dx$$
,

Put
$$2x = t, dx = \frac{1}{2}dt$$

$$=\frac{1}{2}\int_{0}^{2}e^{t}\left(\frac{t^{2}}{4}+t\right)dt$$

$$=\left(\frac{3}{4}\right)e^{2}+\frac{1}{4}=ae^{2}+b$$

$$\therefore (a+b) = \frac{3}{4} + \frac{1}{4} = 1$$

37. B
Sol. Here,
$$\begin{vmatrix} -1 & a & a \\ b & -1 & b \\ c & c & -1 \end{vmatrix} = 0$$

Applying,
$$C_2 \to C_2 - C_1$$
; $C_3 \to C_3 - C_1$, we get $\begin{vmatrix} -1 & a+1 & a+1 \\ b & -(b+1) & 0 \\ c & 0 & -(1+c) \end{vmatrix} = 0$

Applying
$$R_1 \rightarrow \frac{R_1}{a+1}, R_2 \rightarrow \frac{R_2}{b+1}, R_3 \rightarrow \frac{R_3}{c+1}$$

$$\begin{vmatrix} -\frac{1}{a+1} & 1 & 1 \\ \frac{b}{b+1} & -1 & 0 \\ \frac{c}{c+1} & 0 & -1 \end{vmatrix} = 0$$

$$\Rightarrow -\frac{1}{a+1} + \frac{b}{b+1} + \frac{c}{c+1} = 0$$

$$\therefore -\frac{1}{a+1} + 1 - \frac{1}{b+1} + 1 - \frac{1}{c+1} = 0$$

$$\Rightarrow \frac{1}{a+1} + \frac{1}{b+1} + \frac{1}{c+1} = 2$$

Sol.
$$|\vec{u} \times \vec{v}| = 2|\vec{a} \times \vec{b}|$$

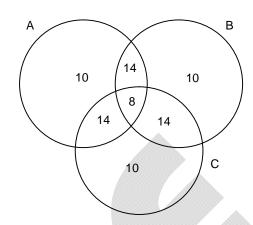
$$\Rightarrow$$
 Result

⁹C₂, number of ways of selecting two non – zero digits

$${}^{9}C_{2} \times (4+6+4) = 36 \times 14$$

$$1 3 3$$
 $9 \times 7 = 63$

Sol.
$$P\left(\frac{A \cap B \cap C}{A \cap B}\right) = \frac{P(A \cap B \cap C)}{P(A \cap B)} = \frac{x}{x+14}$$
$$\frac{x}{x+14} = \frac{1}{3} \Rightarrow x = 7$$
$$P(A \cup B \cup C) = 79$$



Sol. (A) Let
$$\sin \frac{1}{x} = 0 \Rightarrow \lim_{x \to 0} g(f(x)) = 0$$

$$Let \ sin\frac{1}{x}\neq 0 \Rightarrow \underset{x\rightarrow 0}{Lim}\Big(f\Big(x\Big)\Big) = \underset{x\rightarrow 0}{Lim}cos\Bigg(x^2 sin\frac{1}{x}\Bigg) = 1 \, .$$

(B) Let
$$h(x) = [x] \rightarrow GIF$$

$$g(x) = \cos x$$
; $f(x) = x$

f is constant at $x = \pi$, g is continuous at $x = f(\pi) = \pi$

h is discontinuous at $x = g(f(\pi)) = -1$

but
$$\phi = h(g(f(x))) = [\cos x] \rightarrow 41F$$

$$\phi(\pi) = -1; \lim_{x \to \pi} [\cos x] = -1 = \lim_{x \to \pi^+} [\cos x]$$

(C) Check left LHL + RHL

(D)
$$f(x) = x, g(x) = \begin{cases} \frac{1}{x^2} : & x \neq 0 \\ 0 : & x = 0 \end{cases}$$

$$f(x).g(x) = \begin{cases} \frac{1}{x}: & x \neq 0 \\ 0: & x = 0 \end{cases}$$
 Not Continuous

Sol. L:
$$3x - 2y - 4 + \lambda(x - 2y + 4) = 0$$

$$P(a,b) \equiv (4, 4)$$

$$S: x^2 + y^2 = 8$$

(P)
$$a + b = 8$$

(Q)
$$L_T = \sqrt{S_1} = \sqrt{16 + 18 - 8} = 2\sqrt{6}$$

(R) Least distance =
$$OP - r = 4\sqrt{2} - 2\sqrt{2} = 2\sqrt{2}$$

(S) Least distance of the circle containing the given circle is
$$= OP + r = 4\sqrt{2} + 2\sqrt{2} = 6\sqrt{2}$$

43. A Sol.
$$3x + y - z = 0$$
(1)

$$x - \frac{py}{4} + z = 0 \qquad \qquad \dots (2)$$

$$2x - y + 2z = q$$
(3)

Equation $(2) \times (2)$ - equation (3)

$$\Rightarrow \qquad \left(1 - \frac{p}{2}\right) y = 4 - q$$

For unique solution, $p \neq 2$, $q \in N \Rightarrow$ Number of ordered pairs (p, q) in [1, 10] are 90.

For infinite solution, p = 2 and $q = 4 \Rightarrow$ exactly one ordered pair.

For no solution, p = 2 and $q \ne 4 \Rightarrow$ Number of ordered pairs (p, q) in [1, 10] are 9.

44. C

Sol. (P) L.H.S. =
$$(\cos 30^{\circ} + 3\cos 10^{\circ}) + (3\sin 20^{\circ} - \sin 60^{\circ})$$

= $\frac{\sqrt{3}}{2} + 3\cos 10^{\circ} + 3\sin 20^{\circ} - \frac{\sqrt{3}}{2} = 3(\cos 10^{\circ} + \sin 20^{\circ})$

(Q) L.H.S. =
$$\frac{\cos\theta\cos3\theta}{\sin4\theta} - \frac{\sin3\theta\sin\theta}{\sin4\theta} = \frac{\cos4\theta}{\sin4\theta} = \cot4\theta$$

$$(R) \quad \frac{1-\cos 2\theta + \sin 2\theta}{1+\cos 2\theta + \sin 2\theta} = \frac{2\sin^2\theta + 2\sin\theta\cos\theta}{2\cos^2\theta + 2\sin\theta\cos\theta} = \tan\theta$$

(S)
$$\frac{\cot \theta - 1}{\cot \theta + 1} = \frac{1 - \sin 2\theta}{\cos 2\theta}$$

45. B

Sol. (P) Let
$$z = (a + b\omega + c\omega^2)$$
, then $\overline{z} = (a + b\omega^2 + c\omega)$

Clearly $z\overline{z} = (a + b\omega + c\omega^2)(a + b\omega^2 + c\omega)$

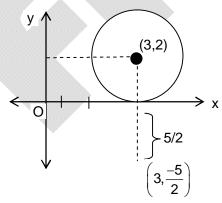
$$= (a^2+b^2+c^2-ab-bc-ca)$$

$$now \mid a + b\omega + c\omega^2 \mid + (a + b\omega^2 + c\omega \mid = \mid z \mid + \mid \overline{z} \mid = 2 \mid z \mid = 2 \cdot \frac{1}{\sqrt{2}} \sqrt{(a - b)^2 + (b - c)^2 + (c - a)^2}$$

$$\geq \sqrt{2}\sqrt{1^2+1^2+2^2} = \sqrt{12} = 144^{1/4}$$

 \therefore minimum value $144^{1/4} \Rightarrow n = 144$.

(Q)
$$|z-3-2i| \le 2$$



Represents interior of a circle with centre (3, 2) and radius 2.

$$\therefore \left| 2z - 6 + 5i \right| = 2 \left| z - 3 + \frac{5i}{2} \right|$$

= 2 × distance of z from
$$\left(3, \frac{-5}{2}\right)$$

$$\Rightarrow$$
 Minimum value is $2 \times \left(\frac{5}{2}\right) = 5$.

SECTION - B

Sol.
$$3\sin x + 4\cos x = 5$$
 and $\cos y = 1 \Rightarrow 2$ ordered (or) $3\sin x + 4\cos x = -5$ and $\cos y = -1 \Rightarrow 1$ ordered

Sol.
$$z_1 + z_2 = 1$$
, $|z_1 - z_2| = 1$
 $|(z_1 + z_2)^2 - 2z_1z_2| = 1$
 $|1 - 2ai| = 1$
 $|1^2 + 4a^2 = 1$

Focal segment subtends 90° at point on y – axis but we get angle is acute from circle Sol.

Sol.
$$I_{n} = 2 \int_{0}^{\pi} \frac{\cos n\theta}{\cos \theta} d\theta$$

$$I_{n} + I_{n-2} = 2 \int_{0}^{\pi} \frac{\cos \theta + \cos(n-2)\theta}{\cos \theta} d\theta$$

$$= 2 \left[2\cos(n-1)\theta \cos \theta d\theta \right]$$

$$= 4 \left[\frac{\sin(n-1)\theta}{n-1} \right]_0^{\pi} = 0$$

$$I_1 = 2 \int_0^{\pi} d\theta = 2\pi, I_2 = 2 \int_0^{\pi} \frac{2\cos^2\theta - 1}{\cos\theta} d\theta = 0$$

$$\Rightarrow I_1 = 2\pi, I_3 = -2\pi$$

For More J

Sol.
$$a_1 + a_2 + \dots + a_n + 1 + 3 + 5 \dots + 2n - 1$$

= 835
 $714 + n^2 = 835$
 $n^2 = 121 \Rightarrow n = 11$

$$11a + 55d$$

= $715a + 5d$
= 34

51. 4

Sol.
$$\vec{d} \cdot \vec{c} = \lambda (\vec{a} \times \vec{b}) \cdot \vec{c} + \mu (\vec{b} + \vec{c}) \cdot \vec{c} + \nu (\vec{c} \times \vec{a}) \cdot \vec{c}$$

$$= \lambda [\vec{a} \quad \vec{b} \quad \vec{c}] + 0 + 0 = \lambda [\vec{a} \quad \vec{b} \quad \vec{c}] = \frac{\lambda}{8}$$
Hence $\hat{\lambda} = 0 (\vec{d} \cdot \vec{c}) = 0$ Similarly, $\hat{\beta} = 0 (\vec{d} \cdot \vec{c}) = 0$

Hence,
$$\lambda=8\left(\vec{d}.\vec{c}\right)$$
 Similarly, $\mu=8\left(\vec{d}.\vec{a}\right)$ and $\nu=8\left(\vec{d}.\vec{b}\right)$

$$\therefore \quad \lambda + \mu + v = 8\vec{d}.\vec{c} + 8\vec{d}.\vec{a} + 8\vec{d}.\vec{b}$$
$$= 8\vec{d}.(\vec{a} + \vec{b} + \vec{c}) = 64$$