FIITJEE ALL INDIA TEST SERIES

JEE (Advanced)-2025

FULL TEST – XI

PAPER –1

TEST DATE: 11-05-2025

ANSWERS, HINTS & SOLUTIONS

Physics

PART - I

SECTION - A

C
 Comparing with radioactive decay

$$a=a_{_0}e^{-\lambda t}\,,$$
 where $\lambda=\frac{\ell n2}{t_{_0}}$

$$\Rightarrow \frac{dv}{dt} = a_0 e^{-\lambda t} \Rightarrow \int_0^v dv = \int_0^t a_0 e^{-\lambda t} dt$$

$$\Rightarrow v = \frac{a_0}{\lambda} (1 - e^{-\lambda t})$$

Terminal velocity $v_{_T} = \frac{a_{_0}}{\lambda} = \frac{a_{_0}t_{_0}}{\ell n2}$

2. (

Sol. If R_0 be the initial activity of the sample, then $R_1 = R_0 e^{-\lambda t_1}$ and $R_2 = R_0 e^{-\lambda t_2}$

Where
$$\lambda = \frac{1}{T}$$

 ${:: Mean life T = 1/\lambda}$

$$\Rightarrow \frac{R_2}{R_1} = \frac{e^{-\lambda t_2}}{e^{-\lambda t_1}} = e^{\lambda (t_1 - t_2)}$$

$$\Rightarrow R_2 = R_1 \exp\left(\frac{t_1 - t_2}{T}\right)$$

Sol.
$$F_{87} = ma$$

$$F_{21} = 7 \text{ ma}$$

$$\frac{F_{21}}{F_{87}} = 7$$

4. E

Sol. Path difference at angular position θ is d sin θ .

- B, C, D 5.
- $\mbox{Optical path length} = \int\limits_{-\infty}^{\infty} \mu dy = 2 \mu_0 D \ , \quad \ \ t = \frac{2 \mu_0 D}{c} \label{eq:optical}$ Sol.
- 6. B, D
- Sol. $a_A = 0$

and $v_A = v_B = u$

By conservation of energy,

$$u = \sqrt{gL} = 4 \text{ m/s}$$

Acceleration of centre of mass,

$$a_C = a_B/2$$

and
$$a_B = a_C + \frac{(u/2)^2}{L/2}$$

$$\Rightarrow$$
 $a_B = \frac{u^2}{L} = g$

Thus, $T - mg = ma_B$

$$\Rightarrow$$
 T = 2 mg

and,
$$F - 2mg = 2m\left(\frac{a_B}{2}\right)$$

$$\Rightarrow$$
 F = 3mg

So,
$$F/T = 3/2$$

- 7. B, C
- $S = \pi (\ell \sin \alpha)^2 \Rightarrow \phi = Bs = \pi B (\ell \sin \alpha)^2$ Sol.

$$e_{in} = \frac{BS}{T} = \frac{BS}{2\pi/\omega} = \frac{B\omega\ell^2 \sin^2\alpha}{2}$$



Second Method

$$e_{in} = \int d\vec{x} \cdot (\vec{v} \times \vec{B})$$

$$= \int_{0}^{\ell} (\omega x \sin \alpha) (B \cdot \sin \alpha) dx$$

$$=\frac{1}{2}B\omega\ell^2\sin^2\alpha$$

Using FBD of mass m, we can write

$$mg = F \cos \alpha$$

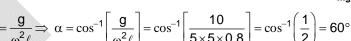
 $m\omega^2 \ell \sin \alpha = F \sin \alpha$

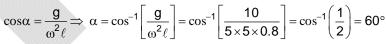
$$\Rightarrow m\omega^2 \ell = F$$

With the help of (ii) and (iii) we can write



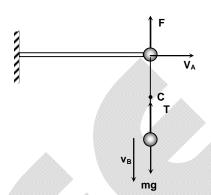






Since pendulum is rotating with uniform angular speed so no magnetic force or torque is acting on the conducting string because there is no energy loss in resistance, hence no current is flowing in the circuits.

$$V = e_{in} = \frac{B\omega\ell^2}{2} \left[1 - \cos^2\alpha \right] = 0.60 \text{ Volt}$$



 $\ell \cos \alpha$

 $\ell \sin \alpha$

- 8. C
- Sol. Use basic concept
- 9. C
- Sol. efflux velocity = $\sqrt{2gH} = \sqrt{2 \times 10 \times 5}$

Time of flight

$$-11 = 6t - \frac{1}{2}(10)t^2$$

t = 2.2 sec

Horizontal range from B

$$R = 8 \times 2.2 - \left(\frac{33}{4}\right)$$

$$R = 9.35 \, \text{m}$$

$$F_x = \rho a v^2 \cos 37^\circ$$

=
$$(10^3)(0.15 \times 10^{-4})(100)\left(\frac{4}{5}\right) = 4$$

$$F_v = \rho a v^2 \sin 37^\circ = 3$$

- 10. B
- Sol. Use basic concept
- 11. D
- Sol. (P) In the process AB,

$$PT^{-1/2} = constant \Rightarrow PV^{-1} = constant$$

$$\Delta W_{AB} = \frac{nR\Delta T}{(1-x)} = \frac{2R \times 900}{2} = 900 R$$

$$\Delta W_{BC} = 0$$

$$\Delta W_{CA} = nR\Delta T = 2R(-300) = -600 R$$

$$\Delta W_{\text{cycle}} = 900 \text{ R} + 0 - 600 \text{ R} = 300 \text{ R}$$

(Q)
$$\Delta Q_{AB} = nC_P \Delta T = 2 \times \frac{5R}{2} \times 400 = 2000R$$

In the process BC, $PT^{-2} = constant \Rightarrow PV^2 = constant$

$$C = C_V + \frac{R}{(1-x)} = \frac{3R}{2} + \frac{R}{(1-2)} = \frac{R}{2}$$

$$\Delta Q_{BC} = nC\Delta T = 2 \times \frac{R}{2} \times (400 - 800) = -400R$$

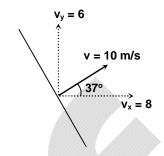
$$\Delta Q_{CA} = \Delta W_{CA} = nRT_0 \ell n \left(\frac{P_0}{4P_0}\right) = 2R \times 400 (-2\ell n2) = -1120R$$

$$\Delta W_{cycle} = \Delta Q_{cycle} = 2000R - 400R - 1120R = 480R$$

(R)
$$\Delta Q_{AB} = nC_P \Delta T = 2 \times \frac{5R}{2} \times 300 = 1500R$$

In the process BC, $VT^2 = constant \Rightarrow PV^{3/2} = constant$

$$C = C_V + \frac{R}{(1-x)} = \frac{3R}{2} + \frac{R}{\left(1-\frac{3}{2}\right)} = \frac{3R}{2} - 2R = -\frac{R}{2}$$



$$\Delta Q_{BC} = nC\Delta T = 2 \times \left(-\frac{R}{2}\right) \times (-300) = 300R$$

$$\Delta Q_{CA} = \Delta W_{CA} = nRT_0 \ell n \left(\frac{V_0}{8V_0} \right) = 2R \times 300(-3\ell n2) = -1260R$$

$$\Delta W_{cycle} = \Delta Q_{cycle} = 1500R + 300R - 1260R = 540R$$

(S)
$$\Delta Q_{AB} = nC_P \Delta T = 2 \times \frac{5R}{2} \times 300 = 1500R$$

$$\Delta Q_{BC} = \Delta W_{BC} = nRT\ell n2 = 2R \times 600 \times 0.7 = 840R$$

In the process CA, $VT^{-2} = constant \Rightarrow PV^{1/2} = constant$

$$C = C_V + \frac{R}{(1-x)} = \frac{3R}{2} + 2R = \frac{7R}{2}$$

$$\Delta Q_{CA} = nC\Delta T = 2 \times \frac{7R}{2} \times (-300) = -2100R$$

$$\Delta W_{cycle} = \Delta Q_{cycle} = 1500R + 840R - 2100R = 240R$$

SECTION - B

Sol.
$$F - \mu N = ma$$

$$80 - \frac{1}{3} \times 6g = ma$$

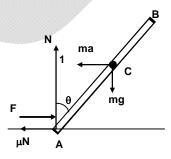
$$Ma = 60$$

$$\tau_{\text{A}}\,=0$$

$$ma\frac{\ell}{2}cos\theta-mg\frac{\ell}{2}sin\theta=0$$

$$\Rightarrow$$
 tan θ =1

$$\theta = 45^{\circ}$$



Sol. Initial potential difference between shells =
$$\frac{kq}{R} - \frac{kq}{2R} = \frac{kq}{2R} = 10 \text{ volt}$$

Let q' be the change on the outer shell after switch S_1 is closed

Then
$$k\left(\frac{q}{2R} + \frac{q'}{2R}\right) = 0 \Rightarrow q' = -q$$

If q₁ be the charge on inner shell after switch S₁ is opened and S₂ is closed

$$k\left(\frac{q_1}{R} + \frac{-q}{2R}\right) = 0 \Rightarrow q_1 = +\frac{q}{2}$$

After S₁ and S₂ are closed alternatively n time charge on the inner shell

$$q_n = \frac{q}{2^n}$$

So, potential difference between the shells

$$v = kq_n \left(\frac{1}{R} - \frac{1}{2R}\right) = \frac{k}{2R} \frac{q}{2^n} = \frac{kq}{2^{n+1}R}$$

When n = 3, V =
$$\frac{1}{8} \left(\frac{kq}{2R} \right) = \frac{5}{4} \text{ volt}$$

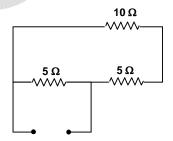
$$\left(\frac{\mathsf{T}_1}{\mathsf{T}_2}\right)^2 = \left(\frac{\ell_1}{\ell_2}\right)^3 \times \frac{\mathsf{M}_2}{\mathsf{M}_1} = \left(\frac{9}{8}\right)^3 \times \frac{4}{3}$$

$$\Rightarrow \frac{T_1}{T_2} = \frac{9}{8} \sqrt{\frac{3}{2}}$$

Sol.
$$F = 2m\eta A \left[(v_0 + v)^2 - (v_0 - v)^2 \right]$$
$$= 2(10^{-26})(10^{25})(1) \left[(5 + 2)^2 - (5 - 2)^2 \right]$$
$$= 0.2 \times 40 = 8 \text{ N}$$

Sol.
$$P = \frac{B^2}{2\mu_0} = \frac{1}{2\mu_0} \left(\frac{\mu_0 I}{2\pi R}\right)^2 = \frac{\mu_0 I^2}{8\pi^2 R^2}$$

Sol.
$$x = \frac{15 \times 5}{15 + 5} = \frac{15}{4}\Omega$$



Chemistry

PART - II

SECTION - A

18. C

Sol. Let the maximum limiting labelling of oleum is (100 + x)% In 100 gram of oleum the maximum mass of free SO_3 should be tending to 100 gram and hence, the mass of water needed x gram, should be exactly that mass which combine completely with all the free SO_3 present

$$H_2O + SO_3 \longrightarrow H_2SO_4$$

$$x = \frac{100}{80} \times 18$$

$$= 22.5$$

19. C

Sol.

$$3CH \equiv CH \xrightarrow{\text{Red hot iron tube} \atop 873 \text{ K} \atop \text{(cyclic polymerisation)}} (B) \xrightarrow{\text{Conc. HNO}_3 \atop \text{Conc. H}_2\text{SO}_4} (C) \xrightarrow{\text{Sn+HCI} \atop 6[H]} (D)$$

Catalyst 3Br₃

20. C

21. B

Sol.
$$\begin{array}{ll} \text{As } P_{ext} = 0 & W = -P_{ex} \Delta V = 0 \\ \text{As } \Delta T = 0 & \Delta E = n C_V \Delta T = 0 \\ \\ \text{But } \Delta S = n R \ell n \frac{V_2}{V_1} \\ \end{array}$$

22. B, D

23. A, B, D Sol. COONa $\begin{array}{c} CH_3 & CH_2Br \\ \hline \\ (A) & (B) \end{array}$ $\begin{array}{c} NBS, A \\ \hline \\ NBS, A \\ \hline \\ OH^- \\ OH^- \\ \hline \\ OH^- \\ \\ OH^- \\ \hline \\ OH^- \\ OH^- \\ \hline \\ OH^- \\ OH^- \\ \hline \\ OH^- \\ OH^$

$$\begin{array}{c|c} COONa & CH_2OH & CHO \\ \hline \\ CH_3OH + & HCOONa + & HCHO \\ \hline \\ HCOOH & (E) & CH_2Br & CH_2CN & CH_2CH_2NH_2 \\ \hline \\ CH_3CI, Anhyd. AICI_3 & CH_2Br & CH_2CN & CH_2CH_2NH_2 \\ \hline \\ (B) & CH_3CI, Anhyd. AICI_3 & CN^- & CN^- \\ \hline \end{array}$$

- 24. A, B, C, D
- 25. E
- 26. D

$$\begin{split} \text{Sol.} &\quad \text{P}_4 \text{O}_{10} + 4 \text{HNO}_3 \longrightarrow 4 \text{HPO}_3 + 2 \text{N}_2 \text{O}_5 \\ &\quad \text{Pb} \left(\text{NO}_3 \right)_2 \stackrel{\Delta}{\longrightarrow} 4 \text{NO}_2 + 2 \text{PbO} + \text{O}_2 \\ &\quad 2 \text{NaNO}_2 + 2 \text{FeSO}_4 + 3 \text{H}_2 \text{SO}_4 \longrightarrow \text{Fe}_2 \left(\text{SO}_4 \right)_3 + 2 \text{NaHSO}_4 + 2 \text{H}_2 \text{O} + 2 \text{NO} \\ &\quad \text{NH}_4 \text{NO}_3 \stackrel{\Delta}{\longrightarrow} \text{N}_2 \text{O} + 2 \text{H}_2 \text{O} \end{split}$$

Sol.
$$2Cu(NO_3)_2 + K_4[Fe(CN)_6] \longrightarrow Cu_2[Fe(CN)_6] \downarrow + 4KNO_3$$

Chocolate
brown ppt.
 $Pb(NO_3)_2 + K_2CrO_4 \longrightarrow PbCrO_4 \downarrow + 2KNO_3$
Yellow ppt.

$$\mathsf{BiCl}_3 + \mathsf{H}_2\mathsf{O} \longrightarrow \mathsf{BiOCl} \downarrow + 2\mathsf{HCl}$$

White ppt.

$$Pb(CH_3COO)_2 + H_2S \longrightarrow PbS \downarrow +2CH_3COOH$$

Black ppt.

28. B
Sol. O O
$$H_{3}C-C-CH_{3} \xrightarrow{CF_{3}CO_{3}H} H_{3}C-C-O-CH_{3}$$

$$O CH_{2}$$

$$H_{3}C-C-CH_{3} \xrightarrow{Ph_{3}\overset{+}{P}-\overset{-}{C}H_{2}} H_{3}C-C-CH_{3}$$

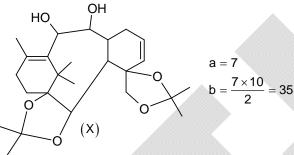
$$O H_{3}C-C-CH_{3} \xrightarrow{(i) NH_{2}-NH_{2}} H_{3}C-CH_{2}-CH_{3}$$

$$O O OH$$

$$H_{3}C-C-CH_{3} \xrightarrow{(i) NH_{2}-NH_{2}} H_{3}C-CH-CH_{3}$$

SECTION - B

Sol.
$$Pb^{2+}$$
, Bi^{3+} , Cd^{2+} , Sb^{3+} , Mn^{2+}



Methylene blue and TiO₂ are positively charged sol. Sol.

For More .

$$\frac{[B]}{[C]} = \frac{5}{4}$$

$$\frac{\begin{bmatrix} \mathsf{B} \end{bmatrix}}{\begin{bmatrix} \mathsf{C} \end{bmatrix}} = \frac{5}{4} \qquad \qquad \frac{\begin{bmatrix} \mathsf{C} \end{bmatrix}}{\begin{bmatrix} \mathsf{D} \end{bmatrix}} = \frac{4}{1} \qquad \qquad \frac{\begin{bmatrix} \mathsf{B} \end{bmatrix}}{\begin{bmatrix} \mathsf{D} \end{bmatrix}} = \frac{5}{1}$$

$$\frac{[B]}{[D]} = \frac{5}{1}$$

$$[B]:[C]:[D] = 5:4:1$$

 $k_1 = 5x k_2 = 4x k_3 = x$

$$k = \frac{0.693}{150} = 0.00462$$

$$k = k_1 + k_2 + k_3$$

$$0.00462 = 5x + 4x + x$$

$$x = \frac{0.00462}{10} = 0.000462$$

$$t_{1/2} \left(partial \ half \ life \right) = \frac{0.693}{k_3} = \frac{0.693}{0.000462}$$

= 1500 hrs.

Sol.
$$\Delta S_{lake} = \frac{q_{irr}}{T} = \frac{mS\Delta T}{T} = \frac{566 \times 90}{283}$$
$$= 180$$

Sol.
$$KCI + AgNO_3 \longrightarrow AgCI + KNO_3$$

 $M_1V_1 = M_2V_2$

$$M_1 = \frac{M_2 V_2}{V_1}$$
, $M_1 = \frac{1 \times 20}{36} = \frac{20}{36}$

$$M_1 = molality = \frac{20}{36}$$

$$i=1+\alpha\left(n-1\right)$$

$$i = 1 + \alpha$$

$$= 1 + 0.8$$

$$\Delta T_f = iK_f m$$

$$=1.8\times2\times\frac{20}{36}$$

Mathematics

PART – III

SECTION - A

Sol.
$$f^{2}(x) + xf(x) = 3 \Rightarrow f'(x) = \frac{-f(x)}{x + 2f(x)}$$
$$\Rightarrow \int \frac{3x^{3} + 6x^{2}f(x) + 2f(x)}{(x + 2f(x))(x^{3} - 2f(x))^{2}} dx = \int \frac{3x^{2}(x + 2f(x)) + 2f(x)}{(x + 2f(x))(x^{3} - 2f(x))^{2}} dx$$
$$\Rightarrow \int \frac{3x^{2} - 2f'(x)}{(x^{3} - 2f(x))^{2}} dx = \frac{1}{2f(x) - x^{3}} + c$$

Sol. Required probability =
$$\frac{1}{2} + \frac{1}{2} \left(\frac{1}{2} \cdot \frac{1}{2} + \frac{1}{2} \cdot \frac{1}{2} \cdot \frac{1}{2} + \frac{1}{2} \cdot \frac{1}{2} + \frac{1}{2} \cdot \frac{1}{2} \cdot \frac{1}{2} \right) = \frac{22}{32}$$

Sol.
$$D = (2n + 1)^2 - 4(2m + 1)(2p + 1)$$

⇒ Roots are rational if D is perfect square.

$$\Rightarrow$$
 $(2n + 1)^2 - 4(2m + 1)(2p + 1) = I^2$ (I = Integer)

$$\Rightarrow$$
 $(2n + 1)^2 - (2r + 1)^2 = 4(2m + 1)(2p + 1), I = 2r + 1, r = Integer$

$$\Rightarrow$$
 4(n + r + 1)(n - r) = 4(2m + 1)(2p + 1)

$$\Rightarrow$$
 (n + r + 1)(n - r) = odd (not possible)

Hence, roots are irrational.

Sol. Let
$$I(t) = \int_0^1 \frac{\sin(\ln x t)}{\ln x} dx$$

$$I'(t) = \int_0^1 \frac{\ln x}{\ln x} \cos(t \ln x) dx = \int_0^1 Re(e^{it \ln |x|}) dx$$

$$I'(t) = Re\left(\frac{1-it}{1+t^2}\right) = \frac{1}{1+t^2}$$

$$\int_{0}^{1} I'(t) dt = \frac{\pi}{4}$$

$$Sol. \quad \ \ Consider \ \Delta A'B'C', \ where \ \ \angle A' = \frac{2\pi}{3} - \angle A \ , \ \ \angle B' = \frac{2\pi}{3} - \angle B \ , \ \ \angle C' = \frac{2\pi}{3} - \angle C \ .$$

Now apply triangle inequality to get $S > -\frac{1}{\sqrt{2}}$

Sol. Take base as the x - y plane with one vertex as origin. A, B be vertices along the x, y-axes

$$\overrightarrow{OA}=2\hat{i}$$
 , $\overrightarrow{OB}=2\hat{j}$, $\overrightarrow{OC}=\hat{i}+\hat{j}+2\hat{k}$, where C is the apex

$$\therefore \overrightarrow{OA} \times \overrightarrow{OB} = 4\hat{k}; \overrightarrow{OA} \times \overrightarrow{OC} = 2\hat{k} - 4\hat{j}; \overrightarrow{OB} \times \overrightarrow{OC} = -2\hat{k} + 4\hat{i}$$

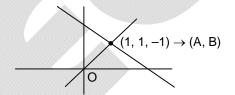
Volume is
$$\frac{1}{3} \times \text{parallelopiped} = \frac{8}{3}$$

Tetrahedron with half of pyramid's base and the pyramid's apex has volume = $\frac{4}{3}$

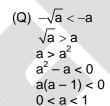
- 41. C, D
- Sol. $F(x) = \int_1^n \frac{x^u}{u} du \; ; \; u \in [1, \, n] \Rightarrow \frac{x^u}{u} \; \text{is strictly monotonic in } x$
 - \Rightarrow F(x) is strictly monotonic

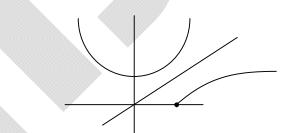
$$x \int_{1}^{n} \frac{du}{u} \le \int_{1}^{n} \frac{x^{u}}{u} du \le x^{n} \int_{1}^{n} \frac{du}{u} \text{ as } x \to 1^{+} \text{ and } x \int_{1}^{n} \frac{du}{u} \ge \int_{1}^{n} \frac{x^{u}}{u} du \ge x^{n} \int_{1}^{n} \frac{du}{u} \text{ as } x \to 1^{-}$$

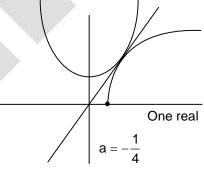
- $\Rightarrow \lim_{x \to 1} F(x) = \ln n$
- 42. E
- Sol. (P) $d = 2\sqrt{3}$
 - (Q) These lines are skew and O lies on shortest distance
 - (R) Lines are parallel and O lies mid way between them
 - (S) Lines are coplanar and perpendicular

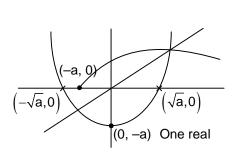


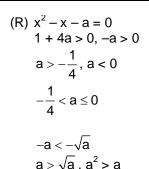
- 43. D
- Sol. We can consider 9 cells as 9 different boxes and we have to fill these boxes by 3 identical balls (2 written on them), 4 identical balls (3 written on them) and 7 identical balls (5 written on them) as per given conditions
- 44. *F*
- Sol. (P) $x^2 a = x$ $x^2 - x - a = 0$ 0 < 0 1 + 4a < 0 $a < -\frac{1}{4}$

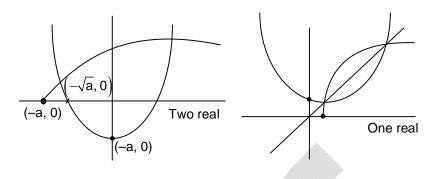












- 45. B
- Sol. (P) Use ${}^{n}C_{r} + {}^{n}C_{r-1} = {}^{n+1}C_{r}$

 $a(a - 1) > 0, a \ge 1$

SECTION - B

Sol. Let
$$a = 2^x$$
, $b = -3^{x-1}$ and $c = -1$
So, $2^x = 3^{x-1}$
 $\Rightarrow 3^{x-1} - 2^{x-1} = 2^{x-1} - 1$
Let $f(t) = t^{x-1}$
So, $f'(t_1) = f'(t_2)$ from LMVT on (1, 2) and (2, 3) and $t_1 \in (1, 2)$, $t_2 \in (2, 3)$
 $(x-1)t_1^{x-2} = (x-1)t_2^{x-2}$
Clearly, there are only 2 solutions $x = 1$ and $x = 2$

Sol.
$$|\operatorname{adj} \lambda A| = |\lambda A|^2 = \lambda^6 |A|^2 = \lambda^6 \text{ (as A is orthogonal)}$$

 $|\operatorname{Adj}(\operatorname{adj} \lambda A)| = |\operatorname{adj} \lambda A|^2 = \lambda^{12}$

$$\therefore \frac{\lambda^{12}}{\lambda^6} = \lambda^6 = 16 \Rightarrow [\lambda^2] = \left[\sqrt[3]{16}\right] = 2$$

$$Sol. \qquad \int\limits_{0}^{1} \lim\limits_{n \to \infty} \sum\limits_{k=0}^{n} \frac{x^{k+1} 3^{k}}{k!} dx = \int\limits_{0}^{1} x \lim\limits_{n \to \infty} \sum\limits_{k=0}^{n} \frac{x^{k} 3^{k}}{k!} dx = \int\limits_{0}^{1} x e^{3x} dx = \frac{x e^{3x}}{3} \bigg|_{0}^{1} - \frac{e^{3x}}{9} \bigg|_{0}^{1} = \frac{2 e^{3} + 1}{9}$$

Sol.
$$(56x + 33y)i = \frac{-iy}{x^2 + y^2}$$
, $33x - 56y = \frac{x}{x^2 + y^2}$ $(z = x + iy)$

$$56iz + 33z = \frac{x - iy}{x^2 + y^2}$$

$$56iz + 33z = \frac{1}{z} \implies z^2 = \frac{1}{33 + 56i} \implies z = \pm \frac{1}{7 + 4i} = \pm \frac{7 - 4i}{65}$$

$$|x| + |y| = \frac{11}{65} = \frac{p}{q} \implies 6p - q = 66 - 65 = 1$$

Sol.
$$\Rightarrow x^2 + \frac{1}{x^2} + b + a\left(x + \frac{1}{x}\right) = 0 \Rightarrow x + \frac{1}{x} = t$$

$$t^2 + at + b - 2 = 0 \Rightarrow at + b + t^2 - 2 = 0, t^2 \in [4, \infty)$$

 t^2 + at + b - 2 = 0 \Rightarrow at + b + t^2 - 2 = 0, t^2 \in [4, ∞) This represents equation of line in a-b plane and a^2 + b^2 represent square of distance of a point on this line from O (origin)

$$d = \frac{t^2 - 2}{\sqrt{1 + t^2}} \implies t^2 \in [4, \infty), \ d_{min} = \frac{2}{\sqrt{5}} \ at \ t^2 = 4$$

$$d_{min}^2 = \frac{4}{5} = \frac{p}{q} \implies p^2 + q^2 = 41$$

Sol. Let
$$\vec{a} = a_1 \hat{i} + a_2 \hat{j} + a_3 \hat{k}$$

$$\vec{b} = b_1 \hat{i} + b_2 \hat{j} + b_3 \hat{k}$$

$$\vec{c} = c_1 \hat{i} + c_2 \hat{j} + c_3 \hat{k}$$

$$\vec{a} \times \vec{b} = D_1 \hat{i} + D_2 \hat{j} + D_3 \hat{k}$$

$$|\vec{a} \times \vec{b}| = \sqrt{D_1^2 + D_2^2 + D_3^2} = 4$$

$$|c| = \sqrt{c_1^2 + c_2^2 + c_3^2} = 2$$

Maximum value of $\begin{bmatrix} \vec{a} & \vec{b} & \vec{c} \end{bmatrix} = |\vec{a} \times \vec{b}| |\vec{c}| = 4 \times 2 = 8$