



# Sri Chaitanya IIT Academy.,India.

☆ A.P ☆ T.S ☆ KARNATAKA ☆ TAMILNADU ☆ MAHARASTRA ☆ DELHI ☆ RANCHI

*A right Choice for the Real Aspirant***ICON Central Office - Madhapur - Hyderabad****SEC: Sr.S60\_Elite, Target & LIIT-BTs****JEE-MAIN****Date: 22-12-2024****Time: 09.00Am to 12.00Pm****GTM-08/03****Max. Marks: 300**

## KEY SHEET

### MATHEMATICS

1	3	2	2	3	2	4	1	5	2
6	1	7	1	8	4	9	3	10	2
11	3	12	4	13	2	14	2	15	3
16	1	17	2	18	1	19	3	20	4
21	7	22	6	23	3	24	7	25	3

### PHYSICS

26	2	27	2	28	2	29	2	30	3
31	2	32	3	33	4	34	4	35	4
36	1	37	2	38	1	39	4	40	3
41	4	42	4	43	4	44	2	45	3
46	67	47	208	48	8	49	1	50	1

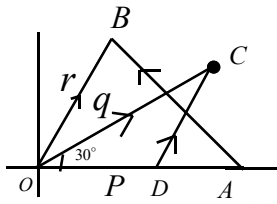
### CHEMISTRY

51	3	52	4	53	4	54	4	55	2
56	1	57	3	58	3	59	2	60	4
61	1	62	1	63	1	64	1	65	1
66	4	67	2	68	3	69	3	70	3
71	10	72	3	73	687	74	6	75	6



# SOLUTIONS

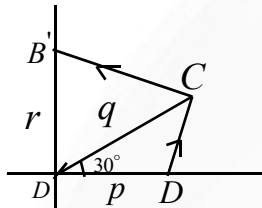
## MATHEMATICS



1.

$$\overrightarrow{DC} = \overrightarrow{a}, \overrightarrow{AB} = \overrightarrow{b}, \overrightarrow{OA} = q \hat{i}$$

Rotate  $\Delta OAB$  by  $30^\circ$  then A reaches C. B reaches  $B'$  Where  $|\overrightarrow{AB}| = |\overrightarrow{CB'}|$



$|\overrightarrow{DC}| + |\overrightarrow{CB'}|$  is min if they are collinear

$$(2, 0, -3)$$

$$\frac{x}{1} = \frac{y-2}{1} = \frac{z+4}{-1}$$

2.

3.

$\alpha - \beta$	5	4	3	2	1	0	-1	-2	-3	-4	-5
Prob	$\frac{1}{36}$	$\frac{2}{36}$	$\frac{3}{36}$	$\frac{4}{36}$	$\frac{5}{36}$	$\frac{1}{6}$	$\frac{5}{36}$	$\frac{4}{36}$	$\frac{1}{12}$	$\frac{1}{18}$	$\frac{1}{36}$

4. R be  $(2\cos\theta, 3\sin\theta)$ , OP perpendicular to OR

$$\therefore \frac{3}{2} \tan \theta \times \frac{3}{\sqrt{3}} = -1 \Rightarrow \tan \theta = \frac{-2\sqrt{3}}{9}, \text{ Required} = \frac{1}{4} \left( \frac{1}{OP^2} + \frac{1}{OR^2} \right) = \frac{13}{144}$$

5. Given  $\int \tan^{-1} x \cdot \frac{1}{x(\ln x)} + \frac{1}{1+x^2} \cdot \ln(\ln x) \therefore f(x) = (\tan^{-1} x) \ln(\ln x)$

6. By differentiating wrt x, we get  $(1+x^2)f(x) = x \rightarrow f(x)$  is odd

$$\text{Required integral} = \int_{-\pi/4}^{\pi/4} \sec^2 x \, dx = 2$$

7.  $D = p^2 - 5p$ . If  $p=9$ , then D is perfect square

$$\text{Area} = \int_0^9 (x-9)^2 \, dx = 243$$

8.  $m = {}^6C_2 \times {}^5C_4 \times 2 \times 4 \times 5!$

$$n = 5 \times 6!$$

9. Required = coefficient of  $x^{99}$  in  $x^{100} \left( 1 + 2 \left( \frac{1+x}{x} \right) + \dots + 101 \left( \frac{1+x}{x} \right)^{100} \right)$

10.  $x^2 - 16 \geq 0, x \neq 2, -2, (x+5)(x-2) \geq 0$

$$x \in (-\infty, -5) \cup (2, \infty)$$



$$11. \cos\left(2\sin^{-1}\frac{1}{\sqrt{5}}\right) = \frac{3}{5}, \sin^{-1}\left(\sin\frac{2\pi}{3}\right) = \frac{\pi}{3}$$

$$\cos^{-1}\left(\cos\frac{7\pi}{6}\right) = 2\pi - \frac{7\pi}{6}$$

$$12. ABA^T = I \Rightarrow AB^T A^T = I$$

$$13. a_1 + a_2 + a_3 = 2, b_1 + b_2 + b_3 = 2, c_1 + c_2 + c_3 = 2$$

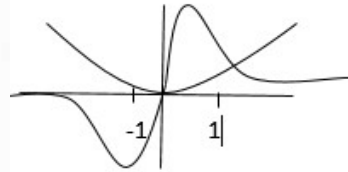
(det A) max when  $a_1 = 2, b_2 = 2, c_3 = 2$  & else are '0'.

$$14. f''(1) = 3 - 2a + b = a, f''(2) = 12 - 2a = b, f'''(3) = 6 \quad \therefore a = 3, b = 6s$$

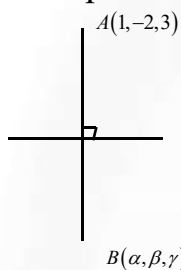
15. Conceptual

$$16. \int e^y dy = \int 2x dx = f(x) = \ln(x^2 + c)$$

$$f'(1) = 1 \Rightarrow C = 1$$

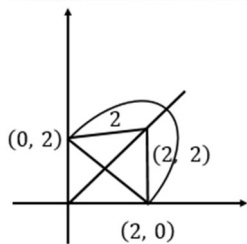


17. conceptual



$$18. AB \perp \text{line}, (\alpha - 1)2 + 4(\beta + 2) + 3(\gamma - 3) = 0 \therefore 2\alpha + 4\beta + 3\gamma = 3$$

19.

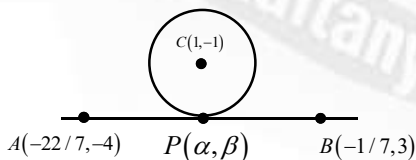


$$20. B = (1, 2) \& \text{Let } C \equiv (\alpha, 4 - 2\alpha) \therefore AB = \sqrt{2^2 + 1^2} = \sqrt{(2\alpha - 3)^2 + (3 - \alpha)^2}, 5\alpha^2 - 18\alpha + 13 = 0$$

$$21. \text{Equation of AB is } 7X - 3Y + 10 = 0 \dots (I)$$

$$\text{Equation of CP is } 3x + 7y - 4 = 0 \dots (ii)$$

$$\alpha = \frac{-41}{29} \quad \beta = \frac{1}{29}$$



$$22. |kz_1 + (1 - k)z_2 - z_3| = \text{height from vertex C.}$$

$$\therefore \text{Area} = \frac{1}{2}(AB)(\text{height}) = \frac{1}{2}|z_1 - z_3||z_2 - z_3|\sin C$$

$$\alpha = \frac{\sin c}{AB} = \frac{1}{2R} = \frac{1}{6}$$

$$23. (\sin x + 3)(\sin x - (x - 1)^2) = 0$$



$$24. \text{ Required} = \lim_{n \rightarrow \infty} \frac{3 \left( \sum_{r=1}^{n-1} (-r^3 + r^2(n+1) - nr) \right)}{n(n+1)(3n^2 - n - 2)} = \frac{1}{6}$$

$$25. \mu = \frac{40 \times 12.5 - 10 \times 15}{25} = 14 \quad \sigma^2 = 2 + \frac{15 \times 25}{40 \times 40} (4)^2 = 5.75$$

## PHYSICS

26. Gases have less viscosity.

Due to insoluble impurities like detergent surface tension decreases

$$27. \omega = \frac{2\pi}{T} \Rightarrow \omega \propto \frac{1}{T} \quad T_{\text{moon}} = 27 \text{ days}$$

$$T_{\text{earth}} = 365 \text{ days } 4 \text{ hour} \Rightarrow \omega_{\text{moon}} > \omega_{\text{earth}}$$

28. Co-efficient of depends on surface in contact So, depends on material of object.

29.: Measuring instruments Theory

30. Ampere- Maxwell law

$$\rightarrow \oint \vec{B} \cdot d\vec{l} = \mu_0 i_c + \mu_0 \epsilon_0 \frac{d\phi_E}{dt}$$

$$\text{Faraday law} \rightarrow \oint \vec{E} \cdot d\vec{l} = \frac{d\phi_B}{dt}$$

$$\text{Gauss' law for electricity} \rightarrow \oint \vec{E} \cdot d\vec{A} = \frac{Q}{\epsilon_0}$$

$$\text{Gauss 'law for magnetism} \rightarrow \oint \vec{B} \cdot d\vec{A} = 0$$

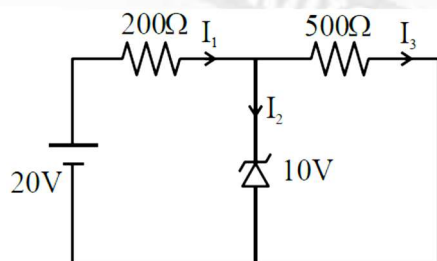
$$31. Y = AB + \bar{A}.B$$

$$= (A + \bar{A}).B$$

$$Y = 1.B$$

$$Y = B$$

32.



Zener is in breakdown region.



$$I_3 = \frac{10}{500} = \frac{1}{50}$$

$$I_1 = \frac{10}{200} = \frac{1}{20}$$

$$I_2 = I_1 - I_3$$

$$I_2 = \left( \frac{1}{20} - \frac{1}{50} \right) = \left( \frac{3}{100} \right) = 30mA$$

33. K.E. = hf -  $\phi$

$$\tan \theta = h$$

34. 
$$\mu = \frac{\sin\left(\frac{A + \delta_m}{2}\right)}{\sin \frac{A}{2}}$$

$$\frac{\cos \frac{A}{2}}{\sin \frac{A}{2}} = \frac{\sin\left(\frac{A + \delta_m}{2}\right)}{\sin \frac{A}{2}}$$

$$\sin\left(\frac{\pi}{2} - \frac{A}{2}\right) = \sin\left(\frac{A + \delta_m}{2}\right)$$

$$\frac{\pi}{2} - \frac{A}{2} = \frac{A}{2} + \frac{\delta_m}{2}$$

$$\delta_m = \pi - 2A$$

35. 
$$\vec{F} = (6t\hat{i} + 6t^2\hat{j})N$$

$$\vec{F} = m\vec{a} = (6t\hat{i} + 6t^2\hat{j})$$

$$\vec{a} = \frac{\vec{F}}{m} = (3t\hat{i} + 3t^2\hat{j})$$

$$\vec{v} = \int_0^t \vec{a} dt = \frac{3t^2}{2}\hat{i} + t^3\hat{j}$$

$$P = \vec{F} \cdot \vec{v} = (9t^3 + 6t^5)W$$

36. Since, Intensity  $\propto$  width of slit ( $\omega$ )

$$I_{\min} = (\sqrt{I_1} - \sqrt{I_2})^2 = I$$

$$\text{So, } I_1 = I, I_2 = 4I \quad I_{\max} = (\sqrt{I_1} + \sqrt{I_2})^2 = 9I$$

$$\frac{I_{\max}}{I_{\min}} = \frac{9I}{I} = \frac{9}{1}$$

37.  $a \times 10^b$

If  $a > 5$  order is b  $a > 5$  order is  $b + 1$



38.  $f = \rho A v^2$
39.  $n = 4$
40. NCERT Theory
41. A pressure node is a displacement anti-node and displacement node is a pressure anti node.  $\Delta P = -B \frac{\partial y}{\partial x}$  At displacement nodes,  $\frac{\partial y}{\partial x}$  is both maximum and minimum
42.  $I_1 = 3A$ ,  $I_2 = 4\sqrt{2} \sin wt$   $I_{\text{net}} = I_1 + I_2 = (3 + 4\sqrt{2} \sin wt)$   
 $I_{\text{rms}} = \sqrt{3^2 + (4\sqrt{2})^2} \times \frac{1}{2} = 5A$
43. Rubber experiences larger lateral strain than steel
44. According to Gauss's theorem in electrostatics,  $\oint E \cdot ds = \frac{q}{\epsilon_0}$ . Here, E is due to all the charges  $q_1, q_2, q_3, q_4$  and  $q_5$ . As q is charge enclosed by the Gaussian surface, therefore,  $q = q_2 + q_4$
45. Given, radius of earth  $R = 6.37 \times 10^6 \text{ m}$   
 Negative surface charge density  $\sigma = 10^{-9} \text{ C/m}^2$   
 Potential  $V = 400 \text{ kV} = 400 \times 10^3 \text{ V}$  Current on the globe  $I = 1800 \text{ A}$   
 Surface area of earth  $A = 4\pi R^2 = 4 \times 3.14 \times (6.37 \times 10^6)^2 = 509.64 \times 10^{12} \text{ m}^2$   
 Charge on earth surface  $Q = \text{Area of earth surface} \times \text{Surface charge}$   
 $Q = A\sigma = 509.64 \times 10^{12} \times 10^{-9} = 509.64 \times 10^3 \text{ C}$   
 We know that  $Q = It \therefore$  Time required to neutralize earth's surface  
 $t = \frac{Q}{I} = \frac{509.64 \times 10^3}{1800} \text{ s} = 283.1 \text{ s}$  or  $t = 4 \text{ min } 43 \text{ s}$   
 Thus, the time required to neutralize the earth's surface is 283.1s.
46.  $I_{\text{sphere}} = \frac{2}{3} MR^2 = Mk_1^2$   
 $I_{\text{cylinder}} = \frac{1}{12} M(4R^2) + \frac{1}{4} MR^2 + M(2R)^2 = \frac{67}{12} MR^2 = Mk_2^2, \frac{k_1}{k_2} = \sqrt{\frac{2}{3} \cdot \frac{12}{67}} = \sqrt{\frac{8}{67}}$
47.  $^{235}\text{U} \rightarrow ^{140}\text{Ce} + ^{94}\text{Zr} + n$   
 Disintegration energy  $Q = (m_R - m_P) \cdot c^2, m_R = 235.0439 \text{ u}$   
 $m_P = 139.9054 \text{ u} + 93.9063 \text{ u} + 1.0086 \text{ u} = 234.8203 \text{ u}$   
 $Q = (235.0439 \text{ u} - 234.8203 \text{ u}) c^2 = 0.2236 \text{ u} \cdot c^2 = 0.2236 \times 931 \text{ MeV}, Q = 208.1716 \text{ MeV}$
48.  $dF = \frac{\mu_0}{2\pi a} \left( \frac{i}{2\pi R} dx \right)^2 l = \frac{\mu_0}{2\pi \times 2R} \cdot \frac{i^2}{4\pi^2 R^2} \cdot dx \cdot dx \cdot l$  Pressure on the element  
 $= \frac{\mu_0}{2\pi \cdot 2R} \cdot \frac{i^2 dx}{4\pi^2 R^2} \cdot \frac{dx \cdot l}{dx \cdot l}$
49. Potential  $V = xy$





50. Redraw the circuit in series and parallel combination

## CHEMISTRY

51.  $A = PhCH_2OH$ ,  $B = HCOOH$

52.  $A = PhOH$ ,  $B = PhH$ ,  $C = PhCH_3$

53. due to lanthanide contraction size of  $Mo = W$

54.  $O_2$  has 2 upe-

55. only  $I^-$  is oxidized by  $O_2$  based on SRP values

56.  $K_2SO_4 \rightarrow 2K^+ + SO_4^{2-}$   $i = 3$

57. Tollen's Test – Aldehydes

Barfoed Test – Monosaccharides

Molisch Test – Carbohydrates

Biuret Test – Proteins and peptides

58.  $d^8 SFL \Rightarrow 2upe^-$

59. NCERT

60. III and IV are correct

61. Both correct & correct explanation

62. both correct and correct explanation

63.  $\left(t_{\frac{1}{2}}\right)_o = \frac{[A]_o}{2K}$

64. Based on  $n+1$  value

65.  $0.02 \times 100 \times 10^{-23} \times 6.022 \times 10^{23} = 12.044 \times 10^{20}$

66.  $2NO_2 + 2OH^- \rightarrow NO_2^- + NO_3^- + H_2O$

67.  $\frac{3+6+1+0}{5} = \frac{10}{5} = 2$

68.  $\frac{1}{\lambda} \propto R_n z^2 \Rightarrow \lambda_1 : \lambda_2 : \lambda_3 = \frac{1}{1^2} : \frac{1}{2^2} : \frac{1}{3^2}$   $1 : \frac{1}{4} : \frac{1}{9} = 36 : 9 : 4$

69.  $Ni(OH)_2 \rightarrow Ni^{2+} + 2OH^-$   $2 \times 10^{-15} = x = 2 \times 10^{-15}$

71.  $t_{99.9} = \frac{2.303}{K} \log \frac{100}{0.1} = \frac{6.909}{K} = \frac{6.909}{\frac{t_1}{2}} = 10t_{\frac{1}{2}}$

72. if  $P = 1 \Rightarrow q = 2$ ,  $\frac{2+1}{2-1} = 3$

73.  $N_2O_4(g) \rightarrow 2NO(g)$   
 $\begin{array}{cc} | & | \\ 1-0.5 & 2 \times 0.5 \\ =0.5 & =1 \end{array}$

74.  $Sr^{2+}, Ba^{2+}, Ca^{2+}, Cu^{2+}, Na^+, K^+$

75. Except sucrose remaining given sugars are reducing