PHARMACEUTICS - I

(Physical Pharmacy)

1. Matter, Properties of Matter:

State of matter, change in the state of matter, latent heats and vapous pressure, sublimation - critical point, Eutectic mixtures gases, aerosols-inholers, relative humidity, liquid, complexes, liquid crystals, glassy state, colids-crystalline, amorphous and polymorphism.

2. Micrometric and Powder Rheology:

Particle size and distribution, average particle size, number and weight distribution, particle number, methods for determining particle volume, optical microscopy.

Asieving, sedimentation, measurement, barticle shape, specific surface, methods for determining surface area, permeability, adsorption, derived properties of powders, borosity, packing arrangement, densities, bulkiness, and flow properties.

3. Susface and Interfacial Phenomenon

Liquid interface, surface and interfacial tensions, surface free energy, measurement of surface and interfacial tensions, Spreading coefficient, adsorption at liquid interfaces, surface active agents, HLB classification, solubilization, defengency, adsorption at solid interfaces, solid-gas and solid-liquid interfaces, complex films and electrical properties of interface.

4. Viscosity and Rheology:

Cates Color Pages L

Newtonian systems, Law of flow, kinematic viscosity, effect of temperature, non-Newtonian systems, bseudoplastic, dilatant, plastic, thixotropy in formulation, determination of viscosity, capillary, falling ball, rotational viscometers, thixotropy,

5. Dispersion Systems:

Colloidal Dispersions:-

Definition, types, properties of colloids, protective colloids, applications of colloids in pharmacy.

	Suspencions and Emulsions:
	The thing of the transmission of the same
	Interfacial properties of suspended particles,
	settling in suspensions, theory of sedimentation,
	effect of Brownian movement, sedimentation
	of flocculated particles, sedimentation
	parameters, wetting of particles, controlled
	floculation, floculation in structured vehicles,
	sheological considerations.
	Emulsions-types, theories, physical stability.
	The state of the s
6	Complexation:
<u> </u>	- Triple Controller
	Classification of complexes, methods of
	preparation and analysis, applications.
7.	Kinetics and Doug Stability:
	General considerations and concepts, half-life
	determination, Influence of temperature,
	light, solvent, catalytic species and
	other factors, Accelerated stability study,
	expiration dating.

	Date: Page:
8.	Buffers:
	Buffer equations and buffer capacity in bhormaceutical systems,
	general, buffer in production buffered isotonic
	Solutions, measurements of tonicity, Calculations and methods of adjusting
	isotonicity.
	Consider them of the state of t
	in Comparation:
*	