## DENSITY OF MOLTEN ELEMENTS AND REPRESENTATIVE SALTS

This table lists the liquid density at the melting point,  $\rho_m$ , for elements that are solid at room temperature, as well as for some representative salts of these elements. Densities at higher temperatures (up to the  $t_{max}$  given in the last column) may be estimated from the equation

$$\rho(t) = \rho_{\rm m} - k(t-t_{\rm m})$$

where  $t_{\rm m}$  is the melting point and k is given in the fifth column of the table. If a value of  $t_{\rm max}$  is not given, the equation should not be used to extrapolate more than about 20°C beyond the melting point.

Data for the elements were selected from the primary literature; the assistance of Gernot Lang in compiling these data is gratefully acknowledged. The molten salt data were derived from Reference 1.

## REFERENCE

- 1. Janz, G. J., Thermodynamic and Transport Properties of Molten Salts: Correlation Equations for Critically Evaluated Density, Surface Tension, Electrical Conductance, and Viscosity Data, J. Phys. Chem. Ref. Data, 17, Suppl. 2, 1988.
- 2. Nasch, P. M., and Steinemann, S. G., Phys. Chem. Liq., 29, 43, 1995.

Formula	Name	$t_{ m m}/^{\circ}{ m C}$	$\rho_m/g~cm^{-3}$	$k/\mathrm{g~cm^{-3}~^{\circ}C^{-1}}$	$t_{\rm max}$
Ag	Silver	961.78	9.320	0.0009	1500
AgBr	Silver(I) bromide	432	5.577	0.001035	667
AgCl	Silver(I) chloride	455	4.83	0.00094	627
AgI	Silver(I) iodide	558	5.58	0.00101	802
$AgNO_3$	Silver(I) nitrate	212	3.970	0.001098	360
$Ag_2SO_4$	Silver(I) sulfate	652	4.84	0.001089	770
Al	Aluminum	660.32	2.375	0.000233	1340
$AlBr_3$	Aluminum bromide	97.5	2.647	0.002435	267
AlCl <sub>3</sub>	Aluminum chloride	192.6	1.302	0.002711	296
$AlI_3$	Aluminum iodide	188.32	3.223	0.0025	240
As	Arsenic	817	5.22	0.000544	
Au	Gold	1064.18	17.31	0.001343	1200
В	Boron	2075	2.08		
Ba	Barium	727	3.338	0.000299	1550
$BaBr_2$	Barium bromide	857	3.991	0.000924	900
BaCl <sub>2</sub>	Barium chloride	962	3.174	0.000681	1081
$BaF_2$	Barium fluoride	1368	4.14	0.000999	1727
$BaI_2$	Barium iodide	711	4.26	0.000977	975
Be	Beryllium	1287	1.690	0.00011	
$BeCl_2$	Beryllium chloride	415	1.54	0.0011	473
$BeF_2$	Beryllium fluoride	552	1.96	0.000015	850
Bi	Bismuth	271.40	10.05	0.00135	800
$BiBr_3$	Bismuth bromide	218	4.76	0.002637	927
BiCl <sub>3</sub>	Bismuth chloride	230	3.916	0.0023	350
Ca	Calcium	842	1.378	0.000230	1484
CaBr <sub>2</sub>	Calcium bromide	742	3.111	0.0005	791
CaCl <sub>2</sub>	Calcium chloride	775	2.085	0.000422	950
CaF <sub>2</sub>	Calcium fluoride	1418	2.52	0.000391	2027
CaI <sub>2</sub>	Calcium iodide	783	3.443	0.000751	1028
Cd	Cadmium	321.07	7.996	0.001218	500
$CdBr_2$	Cadmium bromide	568	4.075	0.00108	720
CdCl <sub>2</sub>	Cadmium chloride	564	3.392	0.00082	807
$CdI_2$	Cadmium iodide	387	4.396	0.001117	700
Ce	Cerium	799	6.55	0.000710	1460
CeCl <sub>3</sub>	Cerium(III) chloride	817	3.25	0.00092	950
CeF <sub>3</sub>	Cerium(III) fluoride	1430	4.659	0.000936	1927
Co	Cobalt	1495	7.75	0.00165	1580
Cr	Chromium	1907	6.3	0.0011	2100
Cs	Cesium	28.44	1.843	0.000556	510
CsBr	Cesium bromide	636	3.133	0.001223	860
CsCl	Cesium chloride	645	2.79	0.001065	906
CsF	Cesium fluoride	703	3.649	0.001282	912
CsI	Cesium iodide	621	3.197	0.001183	907

DENSITY OF MOLTEN ELEMENTS AND REPRESENTATIVE SALTS (continued)

Formula	Name	$t_{ m m}/^{\circ}{ m C}$	$\rho_m/g~cm^{-3}$	$k/\mathrm{g~cm^{-3}~^{\circ}C^{-1}}$	$t_{ m max}$
CsNO <sub>3</sub>	Cesium nitrate	414	2.820	0.001166	491
$Cs_2SO_4$	Cesium sulfate	1005	3.1	0.00095	1530
Cu	Copper	1084.62	8.02	0.000609	1630
CuCl	Copper(I) chloride	430	3.692	0.00076	585
Dy	Dysprosium	1411	8.37	0.00143	1540
DyCl <sub>3</sub>	Dysprosium(III) chloride	680	3.62	0.00068	987
Er	Erbium	1529	8.86	0.00157	1700
Eu	Europium	822	5.13	0.0028	980
Fe	Iron	1538	6.98	0.000572	1680
FeCl <sub>2</sub>	Iron(II) chloride	677	2.348	0.000555	877
Ga	Gallium	29.76	6.08	0.00062	400
GaBr <sub>3</sub>	Gallium(III) bromide	121.5	3.116	0.00246	135
GaCl <sub>3</sub>	Gallium(III) chloride	77.9	2.053	0.002083	141
GaI <sub>3</sub>	Gallium(III) iodide	212	3.630	0.002377	252
Gd	Gadolinium	1314	7.4		
GdCl <sub>3</sub>	Gadolinium(III) chloride	609	3.56	0.000671	1007
GdI <sub>3</sub>	Gadolinium(III) iodide	925	4.12	0.000908	1032
Ge	Germanium	938.25	5.60	0.00055	1600
Hf	Hafnium	2233	12		
HgBr <sub>2</sub>	Mercury(II) bromide	236	5.126	0.003233	319
$HgCl_2$	Mercury(II) chloride	276	4.368	0.002862	304
$HgI_2$	Mercury(II) iodide	259	5.222	0.003235	354
Но	Holmium	1472	8.34		
In	Indium	156.60	7.02	0.000836	500
InBr <sub>3</sub>	Indium(III) bromide	420	3.121	0.0015	528
InCl <sub>3</sub>	Indium(III) chloride	583	2.140	0.0021	666
InI <sub>3</sub>	Indium(III) iodide	207	3.820	0.0015	360
Ir	Iridium	2446	19		
K	Potassium	63.38	0.828	0.000232	500
KBr	Potassium bromide	734	2.127	0.000825	930
KCl	Potassium chloride	771	1.527	0.000583	939
KF	Potassium fluoride	858	1.910	0.000651	1037
KI	Potassium iodide	681	2.448	0.000956	904
$KNO_3$	Potassium nitrate	337	1.865	0.000723	457
La	Lanthanum	920	5.94	0.00061	1600
LaBr <sub>3</sub>	Lanthanum bromide	788	4.933	0.000096	912
LaCl <sub>3</sub>	Lanthanum chloride	859	3.209	0.000777	973
LaF <sub>3</sub>	Lanthanum fluoride	1493	4.589	0.000682	2177
LaI <sub>3</sub>	Lanthanum iodide	778	4.29	0.001110	907
Li	Lithium	180.5	0.512	0.00052	285
LiBr	Lithium bromide	552	2.528	0.000652	739
LiCl	Lithium chloride	610	1.502	0.000432	781
LiF	Lithium fluoride	848.2	1.81	0.000490	1047
LiI	Lithium iodide	469	3.109	0.000917	667
LiNO <sub>3</sub>	Lithium nitrate	253	1.781	0.000546	441
Li <sub>2</sub> SO <sub>4</sub>	Lithium sulfate	859	2.003	0.000407	1214
Lu	Lutetium	1663	9.3		
Mg	Magnesium	650	1.584	0.000234	900
$MgBr_2$	Magnesium bromide	711	2.62	0.000478	935
$MgCl_2$	Magnesium chloride	714	1.68	0.000271	826
$MgI_2$	Magnesium iodide	634	3.05	0.000651	888
Mn	Manganese	1246	5.95	0.00105	1590
MnCl <sub>2</sub>	Manganese(II) chloride	650	2.353	0.000437	850
Mo	Molybdenum	2623	9.33		
Na	Sodium	97.80	0.927	0.00023	600
NaBr	Sodium bromide	747	2.342	0.000816	945
Na <sub>2</sub> CO <sub>3</sub>	Sodium carbonate	858.1	1.972	0.000448	1004
NaCl	Sodium chloride	800.7	1.556	0.000543	1027
NaF	Sodium fluoride	996	1.948	0.000636	1097
NaI	Sodium iodide	660	2.742	0.000949	912

DENSITY OF MOLTEN ELEMENTS AND REPRESENTATIVE SALTS (continued)

Formula	Name	$t_{ m m}/^{\circ}{ m C}$	$\rho_{\text{m}}/\text{g cm}^{-3}$	k/g cm <sup>-3</sup> °C <sup>-1</sup>	$t_{ m max}$
NaNO <sub>3</sub>	Sodium nitrate	307	1.90	0.000715	370
$Na_2SO_4$	Sodium sulfate	884	2.069	0.000483	1077
Nd	Neodymium	1016	6.89	0.00076	1350
Ni	Nickel	1455	7.81	0.000726	1700
NiCl <sub>2</sub>	Nickel(II) chloride	1009	2.653	0.00066	1057
Os Pb	Osmium Lead	3033	20	0.00122	700
PbBr <sub>2</sub>	Lead(II) bromide	327.46 371	10.66 5.73	0.00122 0.00165	700 600
PbCl <sub>2</sub>	Lead(II) chloride	501	4.951	0.00165	710
PbI <sub>2</sub>	Lead(II) iodide	410	5.691	0.0015	697
Pd	Palladium	1554.9	10.38	0.001169	1700
Pr	Praseodymium	931	6.50	0.00093	1460
PrCl <sub>3</sub>	Praseodymium chloride	786	3.23	0.00074	977
Pt	Platinum	1768.4	19.77	0.0024	2200
Pu	Plutonium	640	16.63	0.001419	950
Rb	Rubidium	39.31	1.46	0.000451	800
RbBr	Rubidium bromide	682	2.715	0.001072	907
Rb <sub>2</sub> CO <sub>3</sub>	Rubidium carbonate	837	2.84 2.248	0.000640 0.000883	1007
RbCl RbF	Rubidium chloride Rubidium fluoride	715 833	2.248	0.000883	923 1067
RbI	Rubidium iodide	642	2.904	0.00102	902
RbNO <sub>3</sub>	Rubidium nitrate	305	2.519	0.001143	417
Rb <sub>2</sub> SO <sub>4</sub>	Rubidium sulfate	1050	2.56	0.000665	1545
Re	Rhenium	3186	18.9		
Rh	Rhodium	1964	10.7	0.000895	2200
Ru	Ruthenium	2334	10.65		
S	Sulfur	115.21	1.819	0.00080	160
Sb	Antimony	630.63	6.53	0.00067	745
SbCl <sub>3</sub>	Antimony(III) chloride	73.4	2.681	0.002293	77
SbCl <sub>5</sub>	Antimony(V) chloride	4	2.37	0.001869	77
SbI <sub>3</sub>	Antimony(III) iodide	168	4.171	0.002483	322
Sc Se	Scandium Selenium	1541 221	2.80 3.99		
Si	Silicon	1414	2.57	0.000936	1500
Sm	Samarium	1072	7.16	0.000750	1300
Sn	Tin	231.93	6.99	0.000601	1200
$SnCl_2$	Tin(II) chloride	247	3.36	0.001253	480
SnCl <sub>4</sub>	Tin(IV) chloride	-33	2.37	0.002687	138
Sr	Strontium	777	6.980		
$SrBr_2$	Strontium bromide	657	3.70	0.000745	1004
SrCl <sub>2</sub>	Strontium chloride	874	2.727	0.000578	1037
SrF <sub>2</sub>	Strontium fluoride	1477	3.470	0.000751	1927
SrI <sub>2</sub>	Strontium iodide	538	4.085	0.000885	1026
Ta TaCl <sub>5</sub>	Tantalum Tantalum(V) chloride	3017 216	15 2.700	0.004316	457
Tb	Terbium	1359	7.65	0.004310	437
Te	Tellurium	449.51	5.70	0.00035	600
ThCl <sub>4</sub>	Thorium chloride	770	3.363	0.0014	847
$ThF_4$	Thorium fluoride	1110	6.058	0.000759	1378
Ti	Titanium	1668	4.11		
TiCl <sub>4</sub>	Titanium(IV) chloride	-25	1.807	0.001735	137
Tl	Thallium	304	11.22	0.00144	600
TlBr	Thallium(I) bromide	460	5.98	0.001755	647
TICI	Thallium(I) chloride	430	5.628	0.0018	642
TII	Thallium(I) iodide	441.8	6.15	0.001761	737
TINO <sub>3</sub>	Thallium(I) nitrate	206	4.91	0.001873	279
Tl <sub>2</sub> SO <sub>4</sub> Tm	Thallium(I) sulfate Thulium	632 1545	5.62 8.56	0.00130 0.00050	927 1675
1 m U	Uranium	1135	8.56 17.3	0.00030	10/3
UCl <sub>3</sub>	Uranium(III) chloride	837	4.84	0.007943	1057
J 2.3	(111) emonae			0.00.715	1057

DENSITY OF MOLTEN ELEMENTS AND REPRESENTATIVE SALTS (continued)

Formula	Name	$t_{\rm m}/^{\circ}{ m C}$	$\rho_{\text{m}}/\text{g cm}^{-3}$	$k/\mathrm{g~cm^{-3}~^{\circ}C^{-1}}$	$t_{\rm max}$
UCl <sub>4</sub>	Uranium(IV) chloride	590	3.572	0.001945	667
$UF_4$	Uranium(IV) fluoride	1036	6.485	0.000992	1341
V	Vanadium	1910	5.5		
W	Tungsten	3422	17.6		
Y	Yttrium	1526	4.24		
YCl <sub>3</sub>	Yttrium chloride	721	2.510	0.0005	845
Yb	Ytterbium	824	6.21		
Zn	Zinc	419.53	6.57	0.0011	700
$ZnBr_2$	Zinc bromide	394	3.47	0.000959	602
$ZnCl_2$	Zinc chloride	290	2.54	0.00053	557
$ZnI_2$	Zinc iodide	446	3.878	0.00136	588
$ZnSO_4$	Zinc sulfate	680	3.14	0.00047	987
Zr	Zirconium	1855	5.8		
$ZrCl_4$	Zirconium chloride	437	1.643	0.007464	492