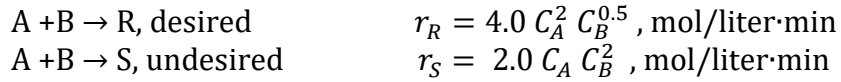


ASSIGNMENT 4
CHE331A

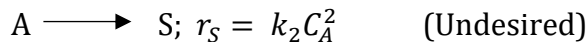
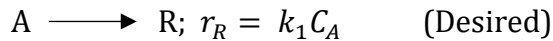
1. Consider the aqueous reactions



equal volumetric flow rates of the A and of B streams are fed to the reactor, and each stream has a concentration of 40 mol/liter of reactant.

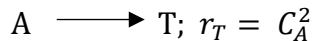
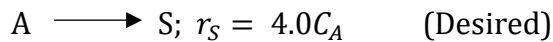
- a) Find C_R and C_S for $X_A = 0.9$ in a Plug flow reactor.
- b) Find C_R and C_S for $X_A = 0.9$ in a Mixed flow reactor.
- c) Find the maximum concentration of desired product in plug flow reactor.

2. Consider liquid phase reaction:



This reaction occurs in two CSTRs in series with $\tau_1 = 5$ min and $\tau_2 = 15$ min. The feed entering first CSTR has the compositions: $C_{A0} = 2.0$ mol/liter, $C_{R0} = 0.2$ mol/liter, $C_{S0} = 0.3$ mol/liter. Composition of stream leaving first CSTR is $C_{A1} = 1.2$ mol/liter, $C_{R1} = 0.8$ mol/liter, $C_{S1} = 0.5$ mol/liter. Find the exit composition of second reactor.

3. Consider the parallel reaction in liquid phase:



Initial concentration of A, $C_{A0} = 6$ mol/liter. For an isothermal operation, find the maximum expected concentration of S in

- a) PFR
- b) CSTR