

McCabe & Thiele Method (Read from RE Treybal)

- McCabe & Thiele method assumes that for each section (enriching & stripping sections), straight operating lines can be used.

- Less rigorous, as detailed enthalpy data is not used. Nevertheless, this method works for most cases.

- Main assumption of McCabe & Thiele method : "Equimolar overflow & vaporization". That is, the molar flow rates of liquid & vapor streams don't change along the column for each section.

Therefore for enriching section: $\begin{cases} L_1 = L_2 = \dots = L_N = L \text{ (mole/time)} \\ G_1 = G_2 = \dots = G_N = G \end{cases}$

for stripping section: $L_m = L_{m+1} = \dots = L_{M_p} = \bar{L}$ & $G_m = G_{m+1} = \dots = G_{M_p} = \bar{G}$

Note: The molar flow rates are constant for stripping & enriching sections. However, the mass flow rates can vary because the mass of liquid & gas streams can vary along the column.

→ Average molecular mass (of liquid & gas streams) can vary along the column.

max

Background: For enriching section, we have

$$G_{int} H_{int} = L_{int} + Q_c + DHD \quad (\text{from previous lecture})$$

$$\Rightarrow G_{\text{NH}} H_{\text{GNH}} - L_{\text{NH}} = D \left(H_0 + \frac{Q_c}{D} \right) = D Q' \quad \left[Q' = \underline{H_0 + \frac{Q_c}{D}} \right]$$

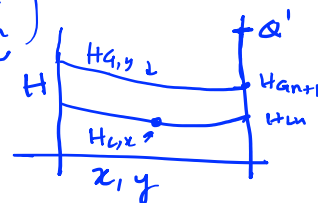
$$\Rightarrow \frac{L_n}{G_{nt}} = \frac{Q' - H_{gnt}}{Q' - H_{Ln}} = \left(1 - \frac{H_{gnt} - H_{Ln}}{Q' - H_{Ln}} \right)$$

Combine previous equation with species balance



$\frac{H_{GHH} - H_{HH}}{}$ = This term also
doesn't change much if the component
form ideal solution, heat of solution is

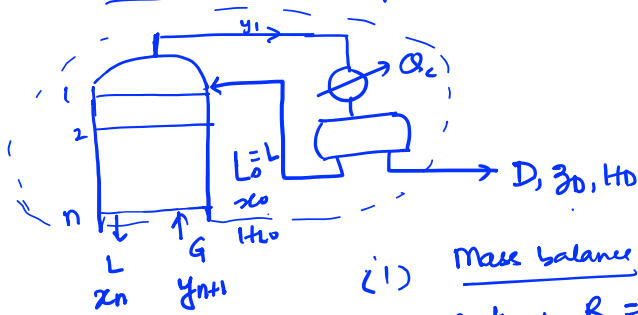
form ideal solution, heat of mixing is negligible, $H_{G,y} \approx H_{L,x}$ don't vary much with change in composition, heat losses are negligible,



Physical Significance: When L_{n+1} & G_{n+1} come into contact, they undergo equimolar counter diffusion. Therefore, their molar flow rates do not change.

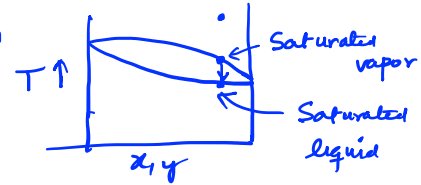


Let us obtain operating lines for the enriching section:



Assumption: Complete condensation of vapor to saturated liquid

i.e. $y_1 = z_D = x_D$



(i) Mass balance: $G = L + D$
 we know, $R = L/D \Rightarrow \boxed{G = (1+R) D}$

Reflux ratio

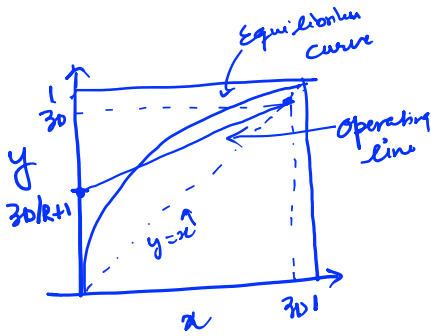
(ii) Species balance: $G y_{n+1} = L x_n + D z_D$
 Applying on the lighter component (volatile component)

$y_{n+1} = \frac{L}{G} x_n + \frac{D}{G} z_D \leftarrow \text{operating line}$

Using $R = L/D$

$\Rightarrow y_{n+1} = \left(\frac{R}{R+1} \right) x_n + \left(\frac{z_D}{R+1} \right) \leftarrow \text{operating line for enriching section}$

$\left[\begin{array}{l} @ x_n = z_D \Rightarrow y_{n+1} = z_D \\ @ x_n = 0 \Rightarrow y_{n+1} = \frac{z_D}{R+1} \text{ (intercept on the y-axis)} \end{array} \right.$



$x = x_A$

$y = y_A$

for the stripping section:

• Mass balance: $\bar{L} = \bar{G} + W \quad \text{--- (i)}$

• Species balance: $\bar{L} x_m = \bar{G} y_{m+1} + W x_w \quad \text{--- (ii)}$

(i) & (ii) $\Rightarrow \boxed{y_{m+1} = \frac{\bar{L}}{\bar{G}} x_m - \frac{W}{\bar{G}} x_w} \leftarrow \text{operating line}$

or: $y_{m+1} = \left(\frac{\bar{L}}{\bar{L} - W} \right) x_m - \left(\frac{W}{\bar{L} - W} \right) x_w \leftarrow \text{operating line}$

@ $x_m = x_w \Rightarrow y_{m+1} = x_w$

Draw a line of slope $\frac{\bar{L}}{\bar{L} - W}$ passing from (x_w, x_w) . This is the operating line for the stripping section.

