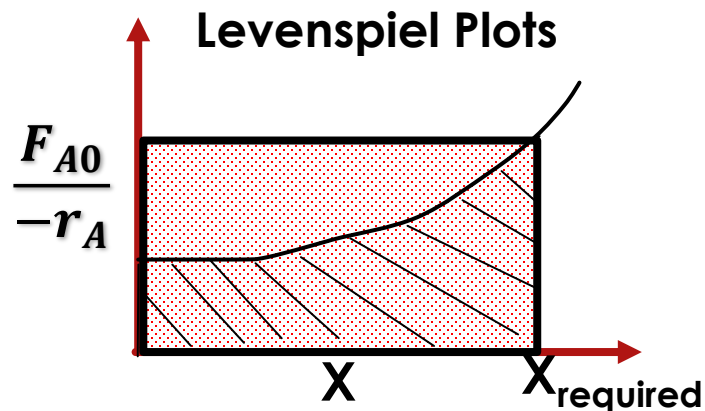


$$X = \frac{N_{A0} - N_A}{N_{A0}}$$

$$X = \frac{F_{A0} - F_A}{F_{A0}}$$

Reactor	Conversion
BR	$t = N_{A0} \int_{X \text{ at } t=0}^X \frac{dX}{-r_A \cdot V}$
CSTR	$V = \left( \frac{F_{A0}}{-r_A} \right) X$
PFR	$V = \int_{X \text{ at } V=0}^X \left( \frac{F_{A0}}{-r_A} \right) dX$
PBR	$W = \int_{X \text{ at } W=0}^X \left( \frac{F_{A0}}{-r'_A} \right) dX$



## Lecture # 3.2

- Define conversion,  $X$ , for BR and flow reactors
- Design equation in terms of  $X$
- Levenspiel plots
- Next Multiple reactors

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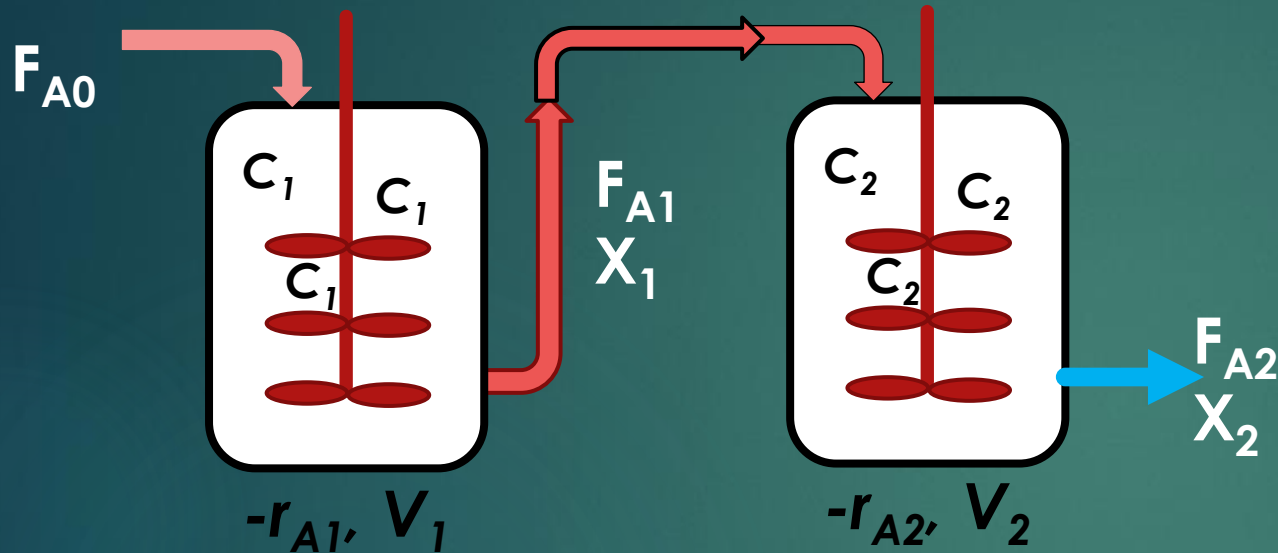
# Multiple reactors are often used in the Chemical Industry

- ▶ Two main possibilities: Reactors in series and Reactors in parallel
  - Combinations of series and parallel also possible
- ▶ Conversion for multiple reactors are defined based on the inlet to the first reactor (no side streams!)
- ▶ Conversion after the  $i^{th}$  reactor,  $X_i$ , is: 
$$X_i = \frac{\text{total A converted after reactor } i}{\text{molar flowrate of A to first reactor}}$$
- ▶ Thus,  $X_i = \frac{F_{A0} - F_{Ai}}{F_{A0}}$  where  $F_{Ai}$  is the molar flowrate of A after reactor  $i$
- ▶ And,  $F_{Ai} = F_{A0}(1 - X_i)$



# Design equations applied to CSTR in series

- Two CSTRs in series → The outlet of the first is the feed to the second



$$X_1 = \frac{F_{A0} - F_{A1}}{F_{A0}} \text{ and } X_2 = \frac{F_{A0} - F_{A2}}{F_{A0}}$$

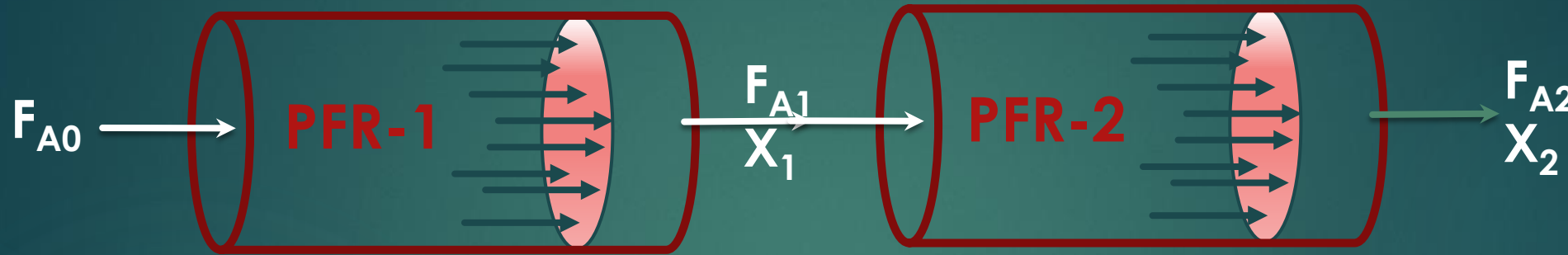
$$F_{A1} = F_{A0}(1 - X_1) \text{ and } F_{A2} = F_{A0}(1 - X_2)$$

$$F_{A1} - F_{A2} + r_{A2} \cdot V = 0$$

- $V_1 = \left( \frac{F_{A0}}{-r_{A1}} \right) X_1$  and  $V_2 = \left( \frac{F_{A1} - F_{A2}}{-r_{A2}} \right) = \frac{F_{A0}(1 - X_1) - F_{A0}(1 - X_2)}{-r_{A2}}$
- $V_2 = \left( \frac{F_{A0}}{-r_{A2}} \right) (X_2 - X_1)$  total vol. required to achieve  $X_2$ :  $V_T = V_1 + V_2$
- For a single CSTR to achieve  $X_2$ :  $V = \left( \frac{F_{A0}}{-r_A} \right) X_2$



# For PFR/PBR in series there is no saving in volume



► For 1<sup>st</sup> PFR reactor:  $V_1 = \int_0^{X_1} \left( \frac{F_{A0}}{-r_A} \right) dX$  and

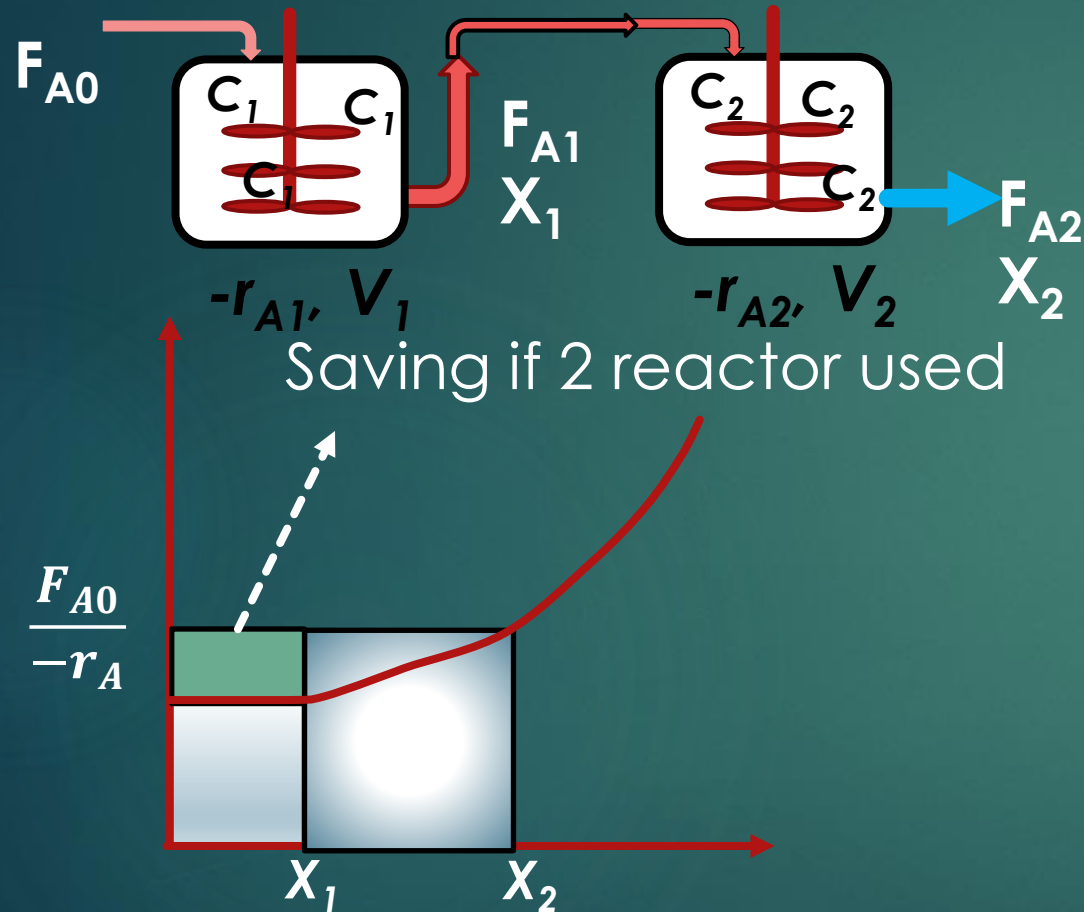
► For 2<sup>nd</sup> PFR reactor:  $V_2 = \int_{X_1}^{X_2} \left( \frac{F_{A0}}{-r_A} \right) dX$

► And,  $V = V_1 + V_2 = F_{A0} \int_0^{X_2} \frac{dX}{-r_A}$  same for a single reactor of vol.  $V$



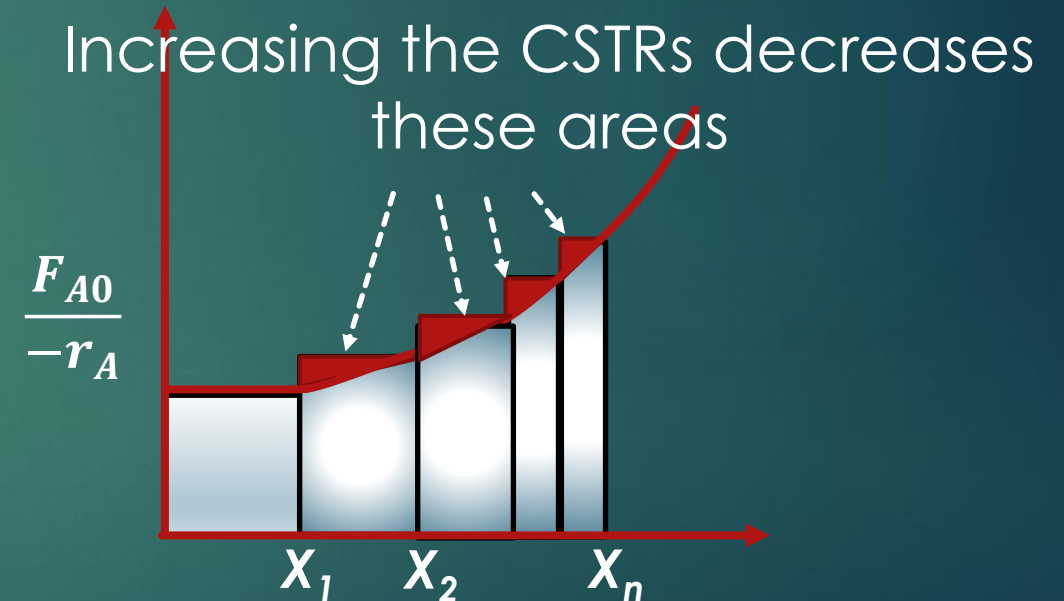
# Corresponding Levenspiel plots are useful

## Two CSTRs in series



- For achieving same conversion  $X_2$  two CSTRs are beneficial!

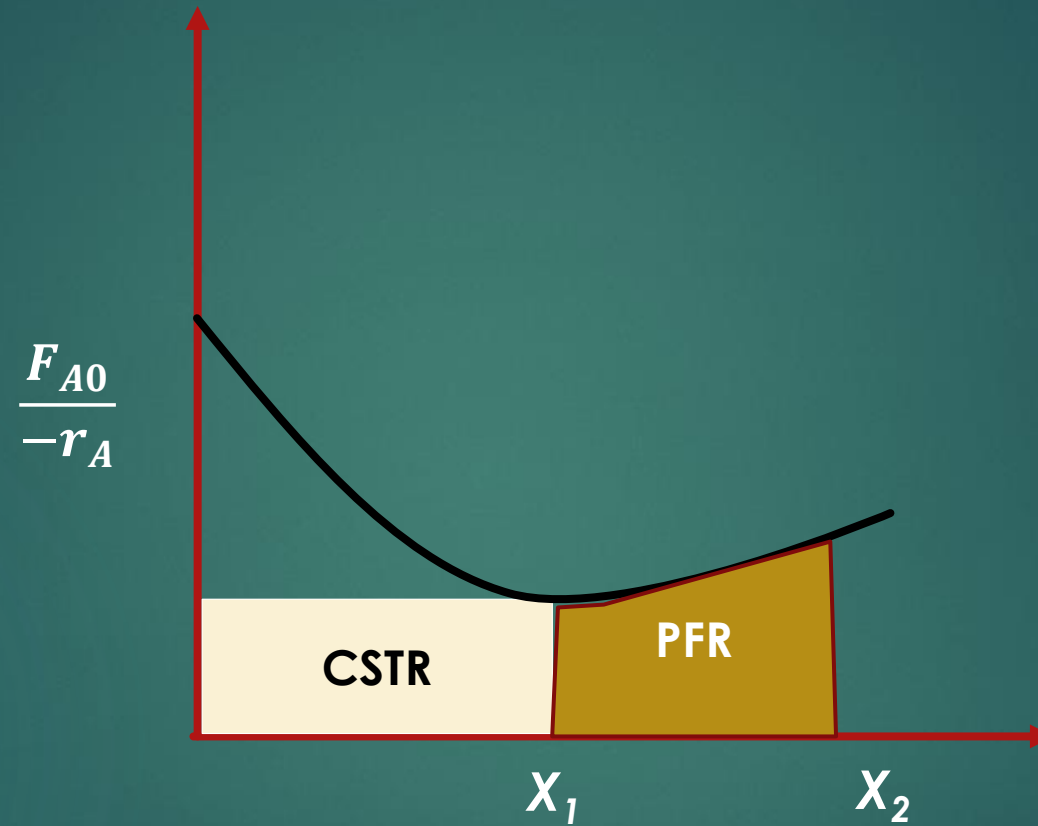
## Multiple CSTRs in series



- A PFR/PBR can be modelled as a series of CSTRs



# Combination of CSTR and PFR/PBR reactors



- A CSTR + PFR is better than (2 PFRs) or (2CSTRs) or (PFR+CSTR)





# Miscellaneous information about flow reactors

- ▶ Space time:  $\tau = \frac{V}{v_0} (\equiv \frac{m^3}{m^3/s})$  which has the unit of time
- ▶ It is the time required to process one reactor volume of fluid based on the inlet conditions
- ▶ For ideal flow reactors the space time is the mean residence time, which is the average time the molecules spend in the reactor
- ▶ Related to space time is the space velocity (SV) =  $\frac{v_0}{V}$
- ▶ SV is referred to as LHSV or GHSV. For LHSV  $v_0$  is measured as that of a liquid at 60°F or 75°F, even though the actual feed may be a vapor at some high temperature. For GHSV  $v_0$  is measured at STP.



# Some typical space time for Industrial Reactors

Reactor Type	Mean Residence Time Range	Production Capacity
Batch	15 min to 20 h	Few kg/day to 1,00,000 tons/day
CSTR	10 min to 4 h	10 to 30,00,000 tons/day
Tubular (PFR/PBR)	0.5 s to 1 h	50 to 50,00,000 tons /day

Trambouze, Landeghem and Wauquier, Chemical Reactors, p. 154  
(Paris: Editions Technip, 1988; Houston: Gulf Publishing Company, 1988)

