



A Manual of Photographic Chemistry; Including the Practice of the Collodion Process

By T Frederick Hardwich

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ACETIC ACID. Symbol, $C_4HSO_3 + HO$. Atomic weight, 60. Acetic Acid is a product of the oxidation of Alcohol. Spirituous liquids, when perfectly pure, are not affected by exposure to air; but if a portion of yeast, or Nitrogenous organic matter of any kind, is added, it soon acts as a ferment, and causes the spirit to unite with oxygen derived from the atmosphere, and so to become sour from formation of vinegar, or Acetic Acid, as it is properly termed. The most concentrated Acetic Acid is obtained by neutralizing common vinegar with Carbonate of Soda and crystallizing out the Acetate of Soda so formed; this Acetate of Soda is then distilled with Sulphuric Acid, which removes the Soda and liberates Acetic Acid: the Acetic Acid being volatile, distils over, and may be condensed. Properties of Acetic...



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