

Problem 1. 2-1

What target isotope must be used for forming the compound nucleus ${}_{11}^{24}\text{Na}$ when the incident projectile is:

- (a) a neutron
- (b) a proton
- (c) an alpha particle?

Solution**Part (a)**

A neutron will increase the mass number, A , by one, but leave the element number, Z , unchanged. Therefore, the answer is a lighter isotope of Neon: ${}_{10}^{23}\text{Ne}$

Part (b)

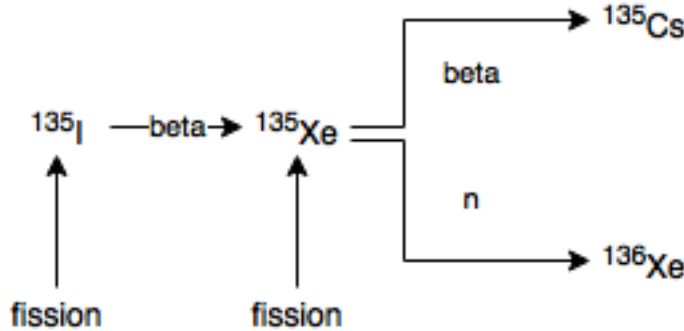
Capturing a proton increases both the mass number and element number by one: ${}_{10}^{23}\text{Ne}$

Part (c)

Capturing an α particle increases the mass number by four and the element number by two: ${}_{8}^{20}\text{O}$

Problem 2. 2-4

A fission product of very considerable importance in thermal reactor operation is ^{135}Xe , which has an enormous thermal absorption cross section of $2 * 10^6 b$. This nuclide can be produced either directly as a fission product or by beta decay of ^{135}I , as indicated by the radioactive chains below:



Write the rate equations describing the concentration of ^{135}I and ^{135}Xe in a nuclear reactor. Then assuming a constant production rate of these isotopes from fission and transmutation rate by neutron capture, determine the steady-state or saturated concentration of ^{135}Xe .

Solution

^{135}I has one production path (production as a fission daughter) and one decay path (β decay into ^{135}I). The rate equation is:

$$\dot{N}_{^{135}\text{I}} = \phi \Sigma_{fission} \gamma_{^{135}\text{I}} - \lambda_{^{135}\text{I}} N_{^{135}\text{I}}$$

^{135}Xe is also produced by fission, as well as by the β decay of ^{135}I . It is removed by β decay as well as by neutron absorption:

$$\dot{N}_{^{135}\text{Xe}} = \phi \Sigma_{fission} \gamma_{^{135}\text{Xe}} + \lambda_{^{135}\text{I}} N_{^{135}\text{I}} - \lambda_{^{135}\text{Xe}} N_{^{135}\text{Xe}} - \sigma_a \phi N_{^{135}\text{Xe}}$$

In steady state, set $\dot{N} = 0$:

Iodine:

$$\begin{aligned} \dot{N}_{^{135}\text{I}} &= 0 \rightarrow \\ \lambda_{\text{I}} N_{\text{I}} &= \phi \Sigma_f \gamma_{\text{I}} \\ N_{^{135}\text{I}} &= \frac{\phi \Sigma_f \gamma_{\text{I}}}{\lambda_{\text{I}}} \end{aligned}$$

Xenon:

$$\dot{N}_{Xe} = 0 \rightarrow$$

$$N_{Xe}(\lambda_{Xe} + \sigma_{Xe}\phi) = \phi\Sigma_f\gamma_{Xe} + \lambda_I N_I$$

$$N_{Xe} = \frac{\phi\Sigma_f\gamma_{Xe} + \lambda_I N_I}{\lambda_{Xe} + \sigma_{Xe}\phi}$$

$$= \frac{\phi\Sigma_f\gamma_{Xe} + \phi\Sigma_f\gamma_I}{\lambda_{Xe} + \sigma_{Xe}\phi}$$

$$N_{^{135}_{54}\text{Xe}} = \frac{\phi\Sigma_f(\gamma_{Xe} + \gamma_I)}{\lambda_{Xe} + \sigma_{Xe}\phi}$$

Problem 3. 2-6

Boron is a common material used to shield against thermal neutrons. Estimate the thickness of boron required to attenuate an incident thermal neutron beam to 0.1% of its intensity. (Use the thermal cross section data in Appendix A.)

Solution

From *Duderstadt* Appendix A, $\Sigma_t = 104\text{cm}^{-1}$ for Boron.

$$\begin{aligned}\left(\frac{1}{e}\right)^n &= \frac{1}{1000} \\ e^{-n} &= 1000^{-1} \\ e^n &= 1000 \\ n &= \ln(1000)\end{aligned}$$

Dividing this by the macroscopic cross section Σ_t gives:

$$\begin{aligned}\frac{n}{\Sigma_t} &= \frac{\ln(1000)}{\Sigma_t} \\ &= 0.0664\text{cm}\end{aligned}$$

Problem 4. 2-8

A free neutron is unstable against beta decay with a half-life of 11.7m. Determine the relative probability that a neutron will undergo beta-decay before being absorbed in an infinite medium. Estimate this probability for a thermal neutron in H₂O.

Solution

To determine the relative probability, we take a ratio of the mean free path for neutron absorption and divide it by the distance travelled by a thermal neutron before decaying:

$$\begin{aligned}\frac{x_{absorption}}{x_{decay}} &= \frac{\Sigma_a^{-1}}{\dot{x}t_{1/2}} \\ &= (\Sigma_a \dot{x}t_{1/2})^{-1}\end{aligned}$$

Substituting the values for a thermal ($\dot{x} = 2.2 * 10^5 cm/s$) neutron in water ($\Sigma_a = 0.022 cm^{-1}$):

$$\left((0.022 cm^{-1}) \left(2.2 * 10^5 \frac{cm}{s} \right) \left(11.7 minutes * \frac{60s}{minute} \right) \right)^{-1} = 2.94 * 10^{-7}$$

It is much more likely that the neutron will decay rather than be absorbed in water (once thermalized).

Problem 5. 2-10

How many mean free paths thick must a shield be designed in order to attenuate an incident neutron beam by a factor of 1000?

Solution

We know that for every mean free path, Σ , travelled, the incident beam attenuates by a factor of $\frac{1}{e}$. Therefore:

$$\begin{aligned}\left(\frac{1}{e}\right)^n &= \frac{1}{1000} \\ e^{-n} &= 1000^{-1} \\ e^n &= 1000 \\ n &= \ln(1000) \\ &= 6.91\end{aligned}$$