

General Biology

SBI0202

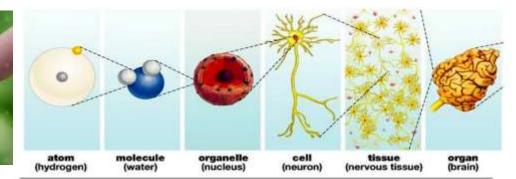
Unit 01: Structure and Function of Living Organisms

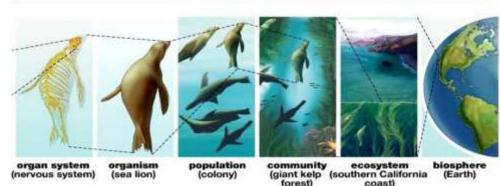
Topic 01: Major biological molecules

Key features that distinguish living organisms from non living matter

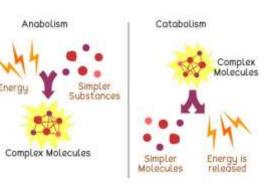
- 1. Order and Organization
- 2. Metabolism
- 3. Growth and Development
- 4. Irritability and Coordination
- 5. Adaptation
- 6. Reproduction
- 7. Heredity and Evolution



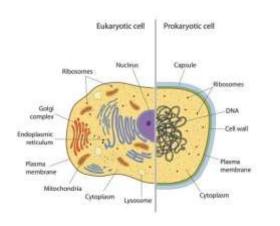


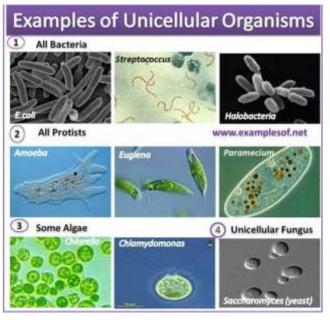






- The cell is the basic structural and functional unit of life.
- Organisms may be unicellular or multicellular.
- There are **two** distinct types of cells:
- 1. Prokaryotic cells
- 2. Eukaryotic cells





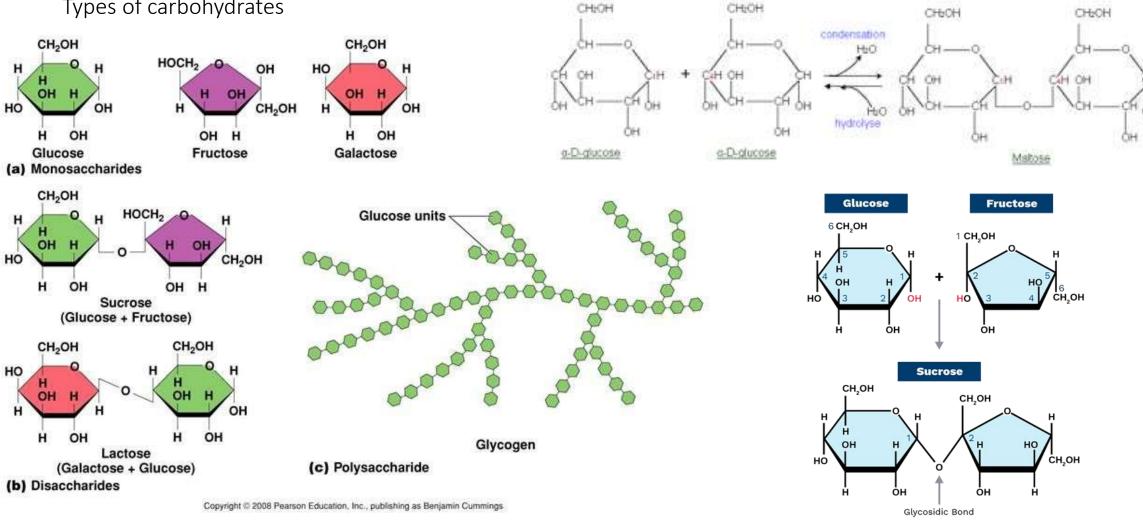
- Cells comprise of organelles which are a composed of several important biological molecules.
- An organelle is a specific structure present in the cell responsible of performing certain functions and together enable to cell to function as a unit.

Main types of biological molecules

- 1. Carbohydrates
- 2. Proteins
- 3. Lipids
- 4. Nucleic acids
- 5. Water

Carbohydrates

Types of carbohydrates



Monosaccharides are combined via glycosidic bonds to form

disaccharides and complex sugars

Functions of carbohydrates

1. Monosaccharides

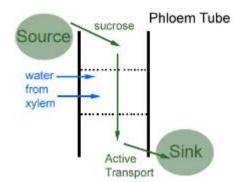
- Energy storage(glucose)
- Building blocks for complex molecules.
- Components of nucleotides (ribose, deoxyribose)

2.Disaccharides

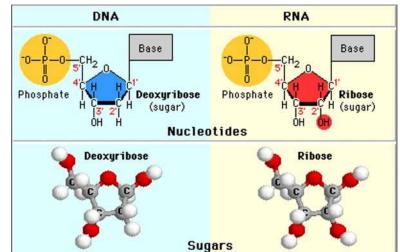
- Storage of energy
- Translocation in phloem (sucrose)
- Storage sugar in milk (Lactose)

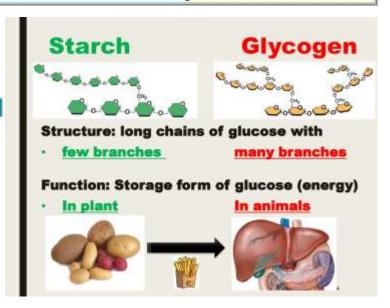
3. Polysaccharides

- Energy storage (starch, glycogen)
- Structural support (cellulose, hemicellulose, pectin in plant cell wall, chitin in fungal cell wall and exoskeletons of arthropods)



Fungal Cell Wal



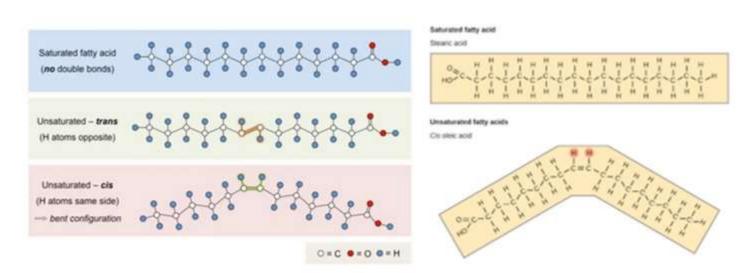


2.Lipids

- Types of lipids:
- 1. Fats and oils (triglycerides)
- 2. Phospholipids
- 3. Waxes
- 4. Steroids

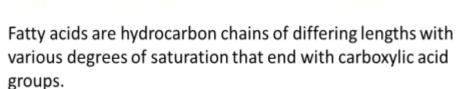
• 1.Triglycerides

- Esters of glycerol and 3 fatty acids.
- Main constituent of body fat in humans.
- Functions: Energy storage, provide insulation diversity to cells, aid in the absorption of fat-soluble vitamins.



+ 3 H₂O

+ 3 H₂O



Saturated vs. unsaturated fatty acids

2.Phospholipids

- Lipids with attached phosphate groups (hydrophilic component)
- Amphipathic
- Function: Essential component of the cell membrane (phospholipid bilayer)

3.Wax

- Esters of long chain alcohols and fatty acids
- Function: waxy cuticles of plant leaves aid in reducing water loss, cerumen (earwax) protects the ear canal.

4.Steroids

- Have a structure of 4 fused rings.
- Eg: Cholesterol
- Functions: Hormone precursor, component of animal cell membranes

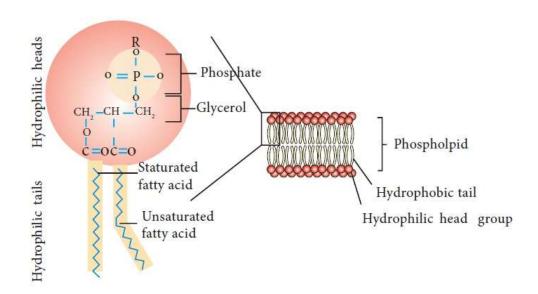
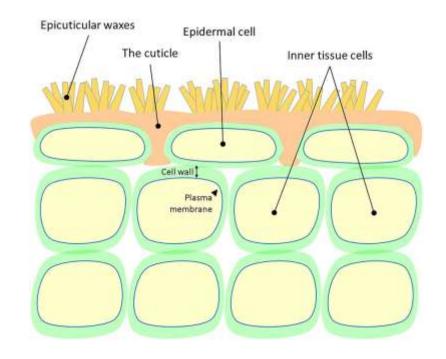


Figure 6.12 Structure of phospholipid

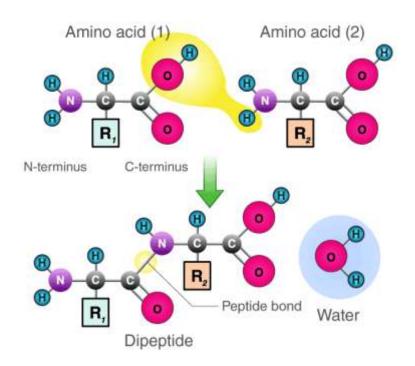


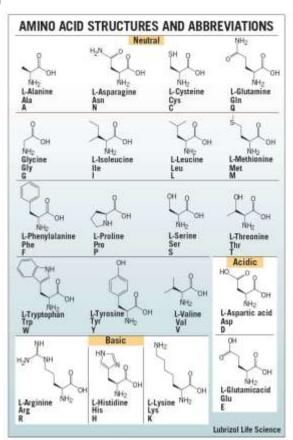
3. Proteins

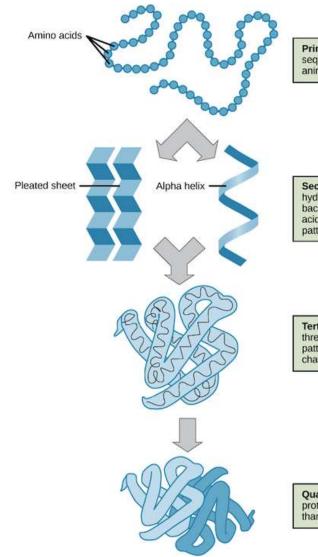
• Polymers of amino acids linked by peptide bonds.

• There are 20 different types of amino acids in

the human body.







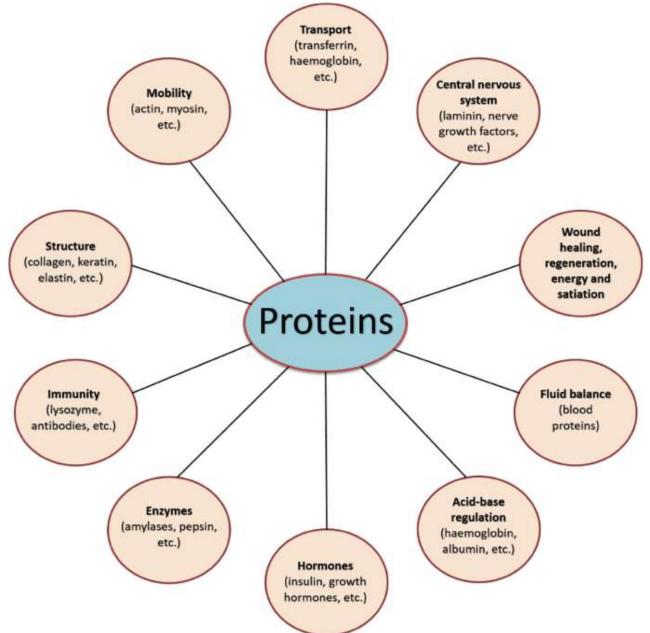
Primary protein structure sequence of a chain of animo acids

Secondary protein structure hydrogen bonding of the peptide backbone causes the amino acids to fold into a repeating pattern

Tertiary protein structure three-dimensional folding pattern of a protein due to side chain interactions

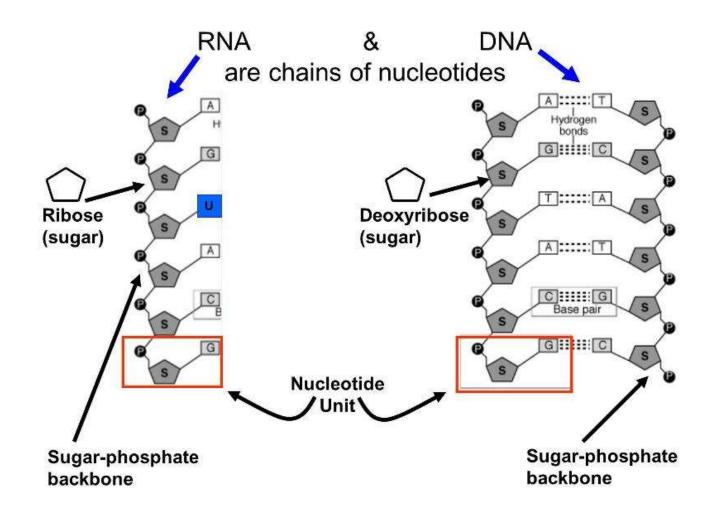
Quaternary protein structure protein consisting of more than one amino acid chain

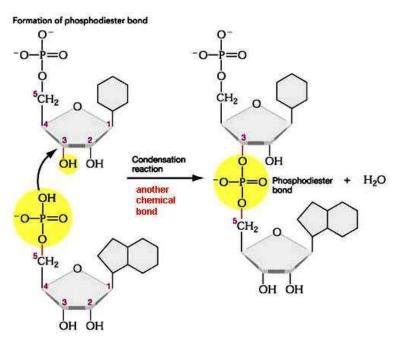
Functions of proteins



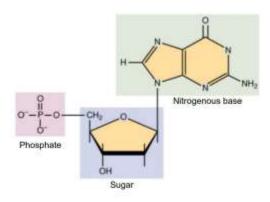
4. Nucleic acids

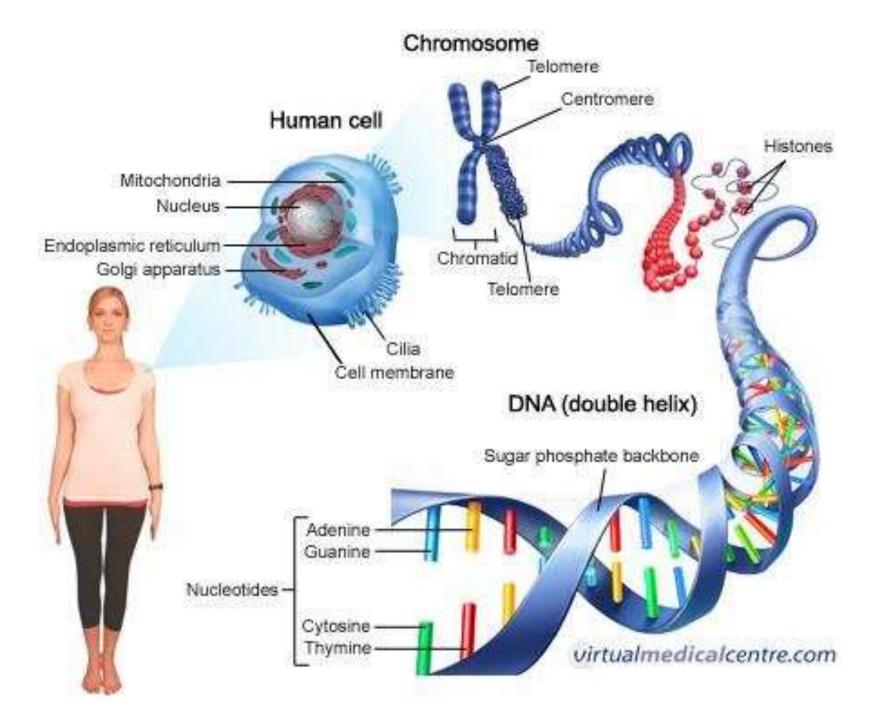
Deoxyribonucleic acid and ribonucleic acid



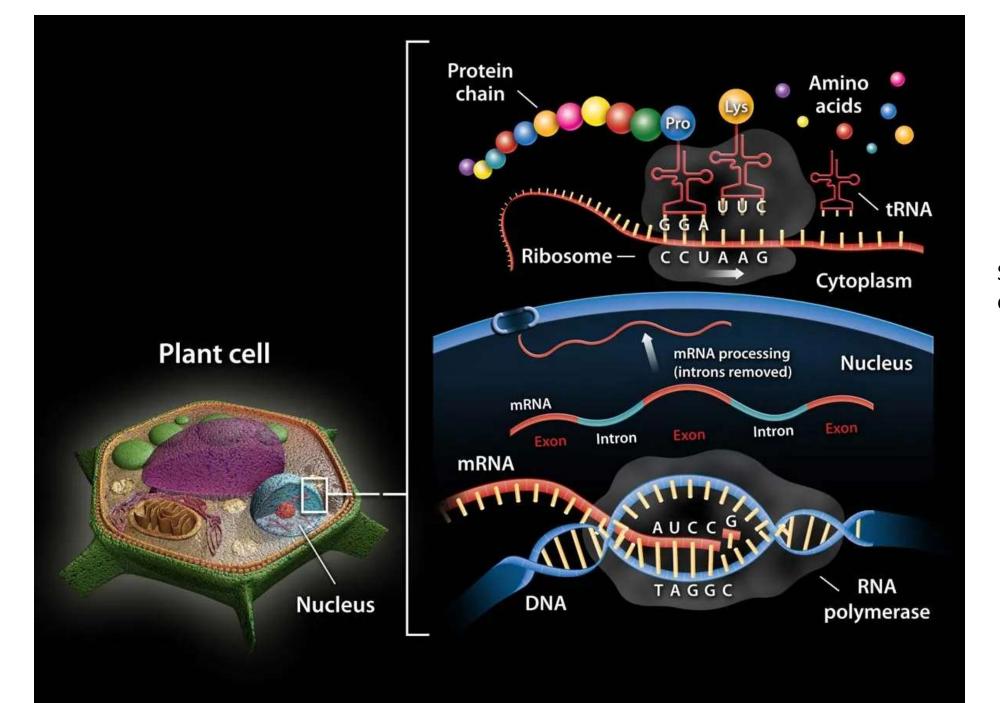


Nucleotide





Structural arrangement of DNA inside the cell



Structure and function of RNA inside the cell

Functions of DNA

- Heredity
- Crossing over results in recombination
- Mutations give rise to variation thereby contributing to the evolutionary process.

Functions of RNA

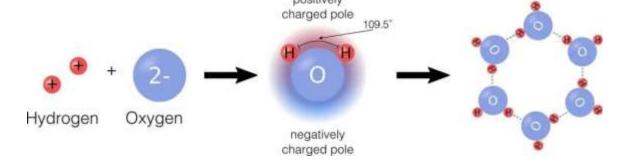
- Main function is protein synthesis
- mRNA (messenger RNA): Act as a template containing transcripted genetic information required for translation of a particular protein.
- tRNA: Delivers the amino acids complementary to the mRNA codons.
- rRNA: Formation of the ribosomes

Biological Role of Water

- molecular structure- Asymmetrical (angular) polar molecule
- Form strong intermolecular H bonds

Properties and functions

- 1. High cohesion- Enable water uptake by plant roots, Maintain body temperature.
- 2.Universal solvent-Enable transport of oxygen, nutrients etc
- 3. Support cellular structure and shape
- 4. Facilitate protein folding
- 5. Reactant eg: Photosythesis
- 6.Buffer-pH regulation in the body



Thank You