

Ch/ChE 164 Winter 2024 Homework Problem

Set #3

Due Date: Thursday Feb 1, 2024 @ 11:59pm

1

1. (from Chandler 3.22)

(a) (14 pts.) Consider a region within a fluid described by the van der Waals equation

$$\beta p = \frac{\rho}{1 - b\rho} - \beta a\rho^2$$

where $\rho = \langle N \rangle / V$. The volume of the region is L^3 . Due to the spontaneous fluctuations in the system, the instantaneous value of the density in that region can differ from its average by an amount $\delta\rho$. Determine, as a function of β, ρ, a, b , and L^3 , the typical relative size of these fluctuations; that is, evaluate $\langle (\delta\rho)^2 \rangle^{1/2} / \rho$. Demonstrate that when one considers observations of a macroscopic system (i.e., the size of the region becomes macroscopic, $L^3 \rightarrow \infty$) the relative fluctuations become negligible.

1.1

We know a few things about the system. We know that the average density is given by:

$$\rho = \frac{\langle N \rangle}{V} \tag{1}$$

We also know that the fluctuations in the density are given by:

$$\langle (\delta\rho)^2 \rangle = \langle \rho^2 \rangle - \langle \rho \rangle^2 \tag{2}$$

and then the volume of the region is given by:

$$V = L^3 \quad (3)$$

We want to find the relative fluctuations, which are given by:

$$\frac{\sigma_\rho^2}{\rho^2} = \frac{\langle(\delta\rho)^2\rangle}{\rho^2} = \frac{\langle\frac{\langle N \rangle^2}{V^2}\rangle - \langle\frac{\langle N \rangle}{V}\rangle^2}{\langle\frac{\langle N \rangle}{V}\rangle^2} \quad (4)$$

We can cancel out the V^2 in the numerator and the denominator, and then we have:

$$\frac{\sigma_\rho^2}{\rho^2} = \frac{\langle N^2 \rangle - \langle N \rangle^2}{\langle N \rangle^2} = \frac{\sigma_N^2}{\langle N \rangle^2} \quad (5)$$

In class, we showed that the variance of the number of particles is:

$$\sigma_N^2 = k_B T N \rho \kappa_T \quad (6)$$

And then the isothermal compressibility is given by:

$$\kappa_T = -\frac{1}{V} \left(\frac{\partial V}{\partial p} \right)_T \quad (7)$$

SymPy gives the relative fluctuations as:

$$\frac{TVk_B(-N^2b^2 + 2NVb - V^2)}{N(2N^3ab^2 - 4N^2Vab + 2NV^2a - TV^3k_B)} \quad (8)$$

I have a V^3 in the numerator and denominator, so I am not sure what happens in the limit of $V \rightarrow \infty$.

Also, when you grade my work, I realize that the above expression is kind of unwieldy, so let me know what I could do to use the computer algebra. Probably printing out some intermediate results would help, but would it give me partial credit to give some kind of an algorithm as to how I am arriving at my results? I tried to comment my python script pretty well.

```

1 from sympy import symbols, Eq, solve, diff, simplify, latex
2
3 # Define symbols
4 beta, p, a, b, N, V, k_B, T = symbols('beta p a b N V k_B T')
5
6 # Average number density, rho = <N>/V

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```

7 rho = N/V
8
9 # Define the van der Waals equation in the given form
10 vdW_eq = Eq(beta*p, rho / (1 - b*rho) - beta*a*rho**2)
11
12 # Substitute rho = N/V in the equation and beta = 1/(k_B*T)
13 vdW_eq_sub = vdW_eq.subs(rho, N/V).subs(beta, 1/(k_B*T))
14
15 # Solve the van der Waals equation for p after substitution
16 p_solved = solve(vdW_eq_sub, p)[0]
17
18 # Differentiate the solved pressure with respect to V
19 dp_dV = diff(p_solved, V)
20
21 # Simplify the derivative to make it more manageable
22 dp_dV_simplified = simplify(dp_dV)
23
24 # Take the inverse of this to get the derivative of volume
    with respect to pressure
25 dV_dp = 1/dp_dV_simplified
26
27 # Express the isothermal compressibility using what we have
    found \kappa_T = -\frac{1}{V}\left(\frac{\partial V}{\partial p}\right)_T
28 kappa_T = -1/V * dV_dp
29
30 # Find the variance in the particle number using \sigma_{N
    }^2 = k_B T N rho \kappa_T
31 sigma_N2 = kappa_T * N * rho * k_B * T
32
33 # Divide by N^2 to get the relative variance
34 rel_var_N = sigma_N2 / N**2
35
36 # Print the result in simplified form and in latex
37 print(latex(simplify(rel_var_N)))

```

(b) (9 pts.) A fluid is at its "critical point" when

$$\left(\frac{\partial \beta p}{\partial \rho}\right)_{\beta} = \left(\frac{\partial^2 \beta p}{\partial \rho^2}\right)_{\beta} = 0$$

Determine the critical point density and temperature for a fluid obeying the van der Waals equation. That is compute β_c and ρ_c as a function of a and b .

1.2

Since the above derivatives occur at fixed β , we can factor out the β from the derivatives. We have:

$$\left(\frac{\partial \beta p}{\partial \rho}\right)_{\beta} = \beta \left(\frac{\partial p}{\partial \rho}\right)_{\beta} \quad (9)$$

and then:

$$\left(\frac{\partial^2 \beta p}{\partial \rho^2}\right)_{\beta} = \beta^2 \left(\frac{\partial^2 p}{\partial \rho^2}\right)_{\beta} \quad (10)$$

At the critical point, β and ρ are given respectively by:

$$\left[\left(\frac{27b}{8a}, \frac{1}{3b}\right)\right] \quad (11)$$

Please follow my script as it should be clearly commented

```
1 from sympy import symbols, Eq, solve, diff, simplify, latex
2
3 # Define symbols
4 beta, p, a, b, k_B, T, rho = symbols('beta p a b k_B T rho')
5     # Include rho in the symbols
6
7 # Define the van der Waals equation in the given form
8 vdw_eq = Eq(beta*p, rho / (1 - b*rho) - beta*a*rho**2)
9
10 # solve for the pressure
11 p_solved = solve(vdw_eq, p)[0]
12
13 # evaluate \left(\frac{\partial \beta p}{\partial \rho}\right)_{\beta} = \beta \left(\frac{\partial p}{\partial \rho}\right)_{\beta}
14 dp_drho = diff(p_solved, rho)
15 dp_drho_beta = beta * dp_drho
16
17 # evaluate \left(\frac{\partial^2 \beta p}{\partial \rho^2}\right)_{\beta} = \beta^2 \left(\frac{\partial^2 p}{\partial \rho^2}\right)_{\beta}
18 d2p_drho2 = diff(dp_drho, rho)
19 d2p_drho2_beta = beta**2 * d2p_drho2
20
21 # said beta equal to 0 in both equation and solve for beta
22 and rho
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23 beta_rho = solve([dp_drho_beta, d2p_drho2_beta], [beta, rho])
24 # print out the results in latex
25 print(latex(beta_rho))

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(c) (9 pts.) Focus attention on the subvolume of size L^3 in the fluid. Suppose L^3 is 100 times the space filling volume of a molecule - that is, $L^3 \approx 100b$. For this region in the fluid, compute the relative size of the density fluctuations when $\rho = \rho_c$, and the temperature is 10% above the critical temperature. Repeat this calculation for temperatures 0.1% and 0.001% from the critical temperature.

1.3

I was not able to get any numerical results for this part or the next, but this is what I tried

```

1 from sympy import symbols, Eq, solve, diff, simplify, latex
2
3 # Define symbols
4 a, b, k_B, T, N, V, L = symbols('a b k_B T N V L')
5
6 rho_c = 1/(3*b)
7 T_c = 8*a/(27*b*k_B) # Critical temperature
8
9 # Define the volume of the region as L^3
10 V = L**3 # Assuming L^3 is the volume
11
12 # Temperature adjustments
13 T_adjustments = [T_c * factor for factor in [1.1, 1.001,
14 1.00001]]
15
16 # Assuming you have an expression for sigma_N^2 and kappa_T
17 kappa_T = symbols('kappa_T') # Placeholder
18 sigma_N2 = k_B * T * N * rho_c * kappa_T # Placeholder
19 # formula for sigma_N^2
20
21 # Compute the relative variance sigma_rho^2 / rho^2
22 # Assuming sigma_N2 / N^2 gives sigma_rho^2 / rho^2 directly
23 rel_var_N = sigma_N2 / N**2
24
25 # Loop over each adjusted temperature
26 for T_adj in T_adjustments:
27     # Compute or substitute T_adj into your expressions as
28     # needed

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26     # This step depends on how you've structured your
    expressions for fluctuations
27     print(f'Temperature: {T_adj}, Relative Variance: {
    rel_var_N.subs(T, T_adj)}')
28
29 # Note: This script contains placeholders and conceptual
    steps. Actual computation
30 # of sigma_N2, kappa_T, and their dependencies need to be
    defined based on
31 # the van der Waals equation and fluctuation theory.

```

(d) (8 pts.) Light that we can observe with our eyes has wavelengths of order of 1000\AA . Fluctuations in density cause changes in the index of refraction, and those changes produce scattering of light. Therefore, if a region of fluid 1000\AA across contains significant density fluctuations, we will visually observe these fluctuations. On the basis of the type of calculation performed in part (b), determine how close to the critical point a system must be before critical fluctuations become optically observable (that is, when the quantity $\langle(\delta\rho)^2\rangle^{1/2}/\rho$ is of order one). The phenomenon of long wavelength density fluctuations in a fluid approaching the critical point is known as opalescence. (Note: You will need to estimate the size of b , and to do this you should note that the typical diameter of a small molecule is around 5\AA).

2

2. (i) (15 points) Consider a perfect crystal originally comprised of N molecules on an equal number of lattice sites. The creation of vacancies in the lattice (or equivalently adding "surface" sites) is a thermally activated process with Boltzmann factor, $e^{-w/kT}$ where w is the energy required to bring each of the molecules to the surface.

If n of these N molecules move to the surface and leave n vacant lattice sites, show that the partition function is well approximated by

$$Q(N, T) = \sum_{n=0}^{\infty} \frac{(N+n)!}{n!N!} e^{-nw/kT}$$

Considering the thermodynamic limit and that w is several kT , find the maximal term in the above sum to show that

$$\frac{n}{N+n} = e^{-w/kT}$$

2.1

Each vacancy might be accompanied by an energy ω (the energy of a surface site) and the total energy of the microstate would, therefore, be $n \times \omega$. As usual the Boltzmann factor is $e^{-\beta E}$, where $\beta = 1/kT$ and $E = n\omega$. Considering how canonical ensemble partition function is defined, we have:

$$Q(N, T) = \sum_{n=0}^{n_{max}} \Omega(N, n) e^{-\beta E} \quad (12)$$

where $\Omega(N, n)$ is the number of microstates with N molecules and n vacancies and n_{max} is the maximum number of vacancies that can be created. Considering that we have N molecules and n vacancies, there are n ways to arrange the $N + n$ spaces, so:

$$\Omega_1(N, n) = \binom{N+n}{n} = \frac{(N+n)!}{n!N!} \quad (13)$$

We might also consider $\Omega_2(N, n)$ as the number of ways to place N particles in n vacancies, but in the limit of large N with $\Omega_2(N, n) = \frac{N!}{n!(N-n)!}$ and $\Omega_1(N, n) = \frac{(N+n)!}{n!N!}$, we have:

$$\lim_{N \rightarrow \infty} \frac{\Omega_2(N, n)}{\Omega_1(N, n)} = 1 \quad (14)$$

so it doesn't make a difference. Continuing on, combining our results we have, for the partition function:

$$Q(N, T) = \sum_{n=0}^{n_{max}} \frac{(N+n)!}{n!N!} e^{-\beta n\omega} \quad (15)$$

Now, we wish to show that the sum can well be extended to infinity, by considering where the argument of the sum is maximized. We can do this by taking the derivative of the argument of the sum with respect to n , or alternatively, by taking the logarithm of the argument and then differentiating. We want to set this equal to zero, so we have:

$$\frac{d}{dn} \ln \left(\frac{(N+n)!}{n!N!} e^{-\beta n\omega} \right) = 0 \quad (16)$$

Applying Stirling's approximation to the logarithm of the factorials, we have:

$$\ln(t_n) \approx (N+n) \ln(N+n) - (N+n) - n \ln n - N \ln N + N - \beta n \omega \quad (17)$$

Taking the derivative of the logarithm of the argument of the sum with respect to n and setting it equal to zero, we have:

$$\frac{d \ln(t_n)}{dn} \approx \ln(N+n) - \ln n - \beta \omega = 0 \quad (18)$$

Exponentiating both sides of the equation, we have:

$$\frac{N+n}{n} = e^{\beta \omega} \quad (19)$$

or flipping the equation, we have:

$$\frac{n}{N+n} = e^{-\beta \omega} \quad (20)$$

Now, we want to show the equivalnence of $\sum_{n=0}^{n_{max}}$ and $\sum_{n=0}^{\infty}$, when the argument is $t_n = \Omega(N, n) e^{-\beta n \omega}$. Now, we want to Taylor expand the logarithm of the argument of the sum around the maximum term, which we will call n^* . We have:

$$\ln(t_n) \approx \ln(t_{n^*}) - \left. \frac{d \ln(t_n)}{dn} \right|_{n=n^*} (n - n^*) + \frac{1}{2} \left. \frac{d^2 \ln(t_n)}{dn^2} \right|_{n=n^*} (n - n^*)^2 \quad (21)$$

We know that the first derivative of the logarithm of the argument of the sum is zero at the maximum term, so the first term of the Taylor expansion is zero:

$$\ln(t_n) \approx \ln(t_{n^*}) + \frac{1}{2} \left. \frac{d^2 \ln(t_n)}{dn^2} \right|_{n=n^*} (n - n^*)^2 \quad (22)$$

We can now take the exponential of both sides of the equation, and we have:

$$t_n \approx t_{n^*} \times \exp \left(\frac{1}{2} \left. \frac{d^2 \ln(t_n)}{dn^2} \right|_{n=n^*} (n - n^*)^2 \right) \quad (23)$$

We want to evaluate the second derivative of the logarithm of the argument of the sum at the maximum term. The term from the Boltzmann factor vanishes in the second derivate with respect to n , so we have:

$$\left. \frac{d^2 \ln(t_n)}{dn^2} \right|_{n=n^*} = \left. \frac{d^2 \ln(\Omega(N, n))}{dn^2} \right|_{n=n^*} = \frac{1}{(N+n^*)} - \frac{1}{n^*} = \frac{1}{\sigma_n^2} \rightarrow \sigma_n^2 = \frac{N}{n^*(N+n^*)} \quad (24)$$

So, our expression for the maximal term becomes:

$$t_n \approx t_{n^*} \times \exp\left(\frac{1}{2\sigma_n^2}(n - n^*)^2\right) \quad (25)$$

We want to consider the square of the variance divided by the square of the maximal n , n^* in the limit of large N . We have:

$$\frac{\sigma_n^2}{n^{*2}} = \frac{N}{n^*(N + n^*)n^*} = \frac{1}{n^*(1 + \frac{N}{n^*})} \quad (26)$$

I got mixed up in the algebra here, but the point is to show that the distribution is sharply peaked about n^* , so that the sum can be extended from n_{max} to infinity with little error.

2.2

(ii) (15 pts.) Evaluate the above full partition function and then obtain $\langle n \rangle$ by suitable differentiation of $\log Q$. (Hint: Think of $Q(N, V, T)$ as a power series $Q(N, V, T) = \sum_{n=0}^{\infty} a_n z^n$.) Are the results for $\langle n \rangle$ from the maximum term and the full partition function the same?

We have the partition function:

$$Q(N, T) = \sum_{n=0}^{\infty} \frac{(N + n)!}{n!N!} e^{-n\omega/kT} \quad (27)$$

We can consider $z = e^{-\omega/kT}$ and $a_n = \frac{(N+n)!}{n!N!}$, so that we have:

$$Q(N, T) = \sum_{n=0}^{\infty} a_n z^n \quad (28)$$

We can recreate a_n by considering:

$$a_n = \frac{1}{n!} \left. \frac{d^n f}{dz^n} \right|_{z=0} \quad (29)$$

with $f(z) = \frac{1}{(1-z)^{(N+1)}}$. It was showed in OH how differentiation of this function is able to recreate $\frac{(N+n)!}{N!}$. We can then write the partition function as:

$$Q(N, T) = \sum_{n=0}^{\infty} \frac{1}{n!} \left. \frac{d^n f}{dz^n} \right|_{z=0} z^n = \frac{1}{(1-z)^{(N+1)}} \quad (30)$$

We can then write the average number of vacancies as:

$$\langle n \rangle = -\frac{d \ln Q}{d\beta\omega} = -\frac{\partial}{\partial\beta\omega} \left(\ln \frac{1}{(1-z)^{(N+1)}} \right) = -\frac{\partial}{\partial\beta\omega} (-(N+1) \ln(1-z)) \quad (31)$$

Inserting the definition of z and taking the derivative, we have:

$$\langle n \rangle = (N+1) \frac{e^{-\beta\omega}}{1 - e^{-\beta\omega}} \quad (32)$$

I am not sure how to go further here.

3

3. (30 pts.) Show that the entropy for the Bose-Einstein and Fermi-Dirac gas can be written in the form

$$S = -k \sum_{\alpha=1}^{\infty} [\langle n_{\alpha} \rangle \ln \langle n_{\alpha} \rangle \mp (1 \pm \langle n_{\alpha} \rangle) \ln (1 \pm \langle n_{\alpha} \rangle)]$$

where $\langle n_{\alpha} \rangle$ is the average occupation number for state α .

3.1

The Bose-Einstein and Fermi-Dirac distributions are given by:

$$\langle n_{\alpha} \rangle = \frac{1}{e^{\beta(\epsilon_{\alpha} - \mu)} \mp 1} \quad (33)$$

where the upper sign is for Bose-Einstein and the lower sign is for Fermi-Dirac. The grand potential is given by:

$$W = -PV = -kT \ln \Xi \quad (34)$$

The differential of the grand potential is given by:

$$dW = -SdT - pdV - Nd\mu \quad (35)$$

We can then write the entropy as:

$$S = -\left(\frac{\partial W}{\partial T} \right)_{V,\mu} = k \ln \Xi + kT \left(\frac{\partial \ln \Xi}{\partial T} \right)_{V,\mu} \quad (36)$$

Now, the grand partition function can be written as as a product over the single particle partition functions:

$$\Xi = \prod_{\alpha} \zeta_{\alpha}^{\mp} = \prod_{\alpha} (1 \mp e^{-\beta(\epsilon_{\alpha}-\mu)})^{\mp} \quad (37)$$

So that the logarithm of the grand partition function is:

$$\ln \Xi = \mp \sum_{\alpha} \ln (1 \mp e^{-\beta(\epsilon_{\alpha}-\mu)}) \quad (38)$$

Now, we can consider the derivative of the logarithm of the grand partition function with respect to temperature. We have:

$$\frac{\partial \ln \Xi}{\partial T} = -\frac{\partial \ln \Xi}{\partial \beta} \frac{\partial \beta}{\partial T} = \mp \sum_{\alpha} \frac{e^{-\beta(\epsilon_{\alpha}-\mu)}}{1 \mp e^{-\beta(\epsilon_{\alpha}-\mu)}} \left(-\frac{\epsilon_{\alpha}-\mu}{kT^2} \right) \quad (39)$$