

SYLLABUS UPDATES FOR NEET(UG) 2024 CHEMISTRY_CLASS-11th & 12th

CLASS-11 th				
Unit Number	Units	Status		
Unit 1	Some Basic Concepts of Chemistry	No Change		
Unit 2	Structure Of Atom	Change		
Unit 3	Classification of Elements and Periodicity	Change		
Unit 4	Chemical Bonding and Molecular Structure	No Change		
Unit 5	Thermodynamics	Change		
Unit 6	Equilibrium	No Change		
Unit 7	Redox Reactions	No Change		
Unit 8	The p -Block Elements	Change		
Unit 8	Organic Chemistry- Some Basic Principles and Techniques	No Change		
CLASS-12 th				
Unit Number	Units	Status		
Unit 11	Solutions	Change		
Unit 12	Chemical Kinetics	No Change		
Unit 13	Electrochemistry			
	Electromical	Change		
Unit 14	The p-Block Elements	Change Change		
Unit 14 Unit 15	•			
	The p-Block Elements	Change		
Unit 15	The p-Block Elements The d-and f-Block Elements	Change Change		
Unit 15 Unit 16	The p-Block Elements The d-and f-Block Elements Coordination Compounds	Change Change No Change		
Unit 15 Unit 16 Unit 17	The p-Block Elements The d-and f-Block Elements Coordination Compounds Haloalkanes and Haloarenes	Change Change No Change Change		
Unit 15 Unit 16 Unit 17 Unit 18	The p-Block Elements The d-and f-Block Elements Coordination Compounds Haloalkanes and Haloarenes Alcohols, Phenols and Ethers	Change Change No Change Change Change		
Unit 15 Unit 16 Unit 17 Unit 18 Unit 19	The p-Block Elements The d-and f-Block Elements Coordination Compounds Haloalkanes and Haloarenes Alcohols, Phenols and Ethers Aldehydes, Ketones and Carboxylic Acids	Change Change No Change Change Change No Change		



CLASS-11 th				
Class-11 NCERT Chapter	Dropped Chemistry Topics/Chapters	Added Chemistry Topics/Chapters		
Some Basic Concepts of Chemistry	-	-		
Structure of Atom	-	Elementary Ideas of Quantum Mechanics, Quantum Mechanics, The Quantum Mechanical Model of the Atom, Its Important Features. Concept of Atomic Orbitals as One-Electron Wave Functions: Variation of Y and Y2 with r for Is and 2s Orbitals		
Classification of Elements and Periodicity	-	-		
Chemical Bonding and Molecular Structure	-	-		
States Of Matter	Full Chapter Deleted	-		
Thermodynamics	Third Law of Thermodynamics Enthalpy of Dilution	-		
Equilibrium	-	-		
Redox Reactions	-	-		
Hydrogen	Full Chapter Deleted	-		
The s-Block Elements	Full Chapter Deleted	-		
The p-Block Elements	Method of Preparation and Uses	-		
Organic Chemistry: Some Basic Principles and Techniques	-	-		
Hydrocarbons	-	-		
Environmental Chemistry	Full Chapter Deleted	-		



CLASS-12 th					
Class-12 NCERT Chapter	Dropped Chemistry Topics/Chapters	Added Chemistry Topics/Chapters			
The Solid State	Full Chapter Deleted	-			
Solutions	-	-			
Chemical Kinetics	-	-			
Electrochemistry	Corrosion	-			
Surface Chemistry	Full Chapter Deleted	-			
General Principles and Processes of Isolation of Elements	Full Chapter Deleted	-			
The p-Block Elements	-	-			
The d-and f-Block Elements	Chemical Reactivity of Lanthanoids	-			
Coordination Compounds	-	-			
Haloalkanes and Haloarenes	Environmental effects of dichloromethane, tetrachloromethane	-			
Alcohols, Phenols and Ethers	Ether: Preparation Physical and Chemical Properties	-			
Aldehydes, Ketones and Carboxylic Acids	-	-			
Amines	Cyanides Isocyanides and Diazonium Salt: Preparation and Physical and Chemical Properties	-			
Biomolecules	Polysaccharides (Starch Cellulose, Glycogen)	Biological Functions of Nucleic Acids			
Polymers	Full Chapter Deleted	-			
Chemistry in Everyday Life	Full Chapter Deleted	-			
Practical Chemistry	 NEWLY ADDED: Detection of extra elements (Nitrogen, Sulphur, Halogens) Inorganic Compounds; Detection of the following functional group, hydroxyl (alcoholic and phenolic), carbonyl (aldehyde and ketones) carboxyl, and amino groups in organic compounds. The chemistry involved in the preparation of the following: Inorganic compounds: Mohr's salt. potash alum' Organic compounds: Acetanilide. p-nitro acetanilide, aniline yellow, iodoform' The chemistry involved in the titrimetric exercises - Acids. bases and the use of indicators. oxalic-acid vs KMnO₄. Mohr's salt vs KMnO₄. Chemical principles involved in the qualitative salt analysis: Cations - Pb²⁺.Cu²⁺.Al³⁺, Fe³⁺.Zn₂+, Ni²⁺, Ca²⁺, Ba²⁺, Mg²⁺.NH⁴⁺ Anions - CO₂₃-, SO₃²⁻.NO₂-, Cl-, Brl- (Insoluble salts excluded). Chemical principles involved in the following experiments:				