

The Fermi velocity, i.e., the velocity of electrons occupying states ("dots") on the surface of the Fermi sphere, and hence are the ones with maximum energy & momentum, is given by

$$v_F = \frac{p_F}{m} = \frac{\hbar k_F}{m} = \frac{4.20}{(r_s/a_0)} \times 10^8 \text{ cm/sec.}$$

This is about 1% of the velocity of light! At temperature $T=0$! These extremely high velocities are possible because of Pauli's exclusion principle, which disallows electrons from occupying already occupied lower energy levels.

Similarly, we can write the Fermi energy as

$$E_F = \frac{50.1 \text{ eV}}{(r_s/a_0)^2}, \text{ which implies}$$

that Fermi energies of metallic elements range from 1.5 to 15 eV.

What is the total energy of the N -electron system? we must add all e^- energies.

That is,
$$E_{\text{tot}} = \sum_{k < k_F} \frac{\hbar^2 k^2}{2m}$$

When evaluated,

$$\frac{E}{V} = \frac{1}{\pi^2} \cdot \frac{\hbar^2 k_F^5}{10m}$$

We already know,
$$\frac{N}{V} = \frac{k_F^3}{3\pi^2}$$

$\therefore \frac{E}{N} = \frac{E/V}{N/V} = \frac{3}{5} \cdot \frac{\hbar^2 k_F^2}{2m} = \frac{3}{5} E_F$

Energy per electron

$$\frac{E}{N} = \frac{3}{5} E_F = \frac{3}{5} k_B T_F$$

Fermi energy

Fermi temperature.

$\sum_{\vec{k}} F(\vec{k})$, where $F(\vec{k})$ is any function of \vec{k} , can be evaluated as follows. Since $\Delta \vec{k}$, the volume occupied by a single "dot" in k -space is $\left(\frac{2\pi}{L}\right)^3$, i.e; $\Delta \vec{k} = \left(\frac{2\pi}{L}\right)^3$,

$$\frac{V}{8\pi^3} \cdot \Delta \vec{k} = 1 \text{ \& } 80, \quad \sum_{\vec{k}} F(\vec{k}) = \frac{V}{8\pi^3} \sum_{\vec{k}} F(\vec{k}) \Delta \vec{k}$$

Taking the limit $\Delta \vec{k} \rightarrow 0, (V \rightarrow \infty)$ we get

$$\lim_{V \rightarrow \infty} \frac{1}{V} \sum_{\vec{k}} F(\vec{k}) = \int \frac{d\vec{k}}{8\pi^3} F(\vec{k}) = \int \frac{d\vec{k}}{8\pi^3} \cdot \frac{\hbar^2 k^2}{2m}$$

$$T_F = \frac{E_F}{k_B}$$

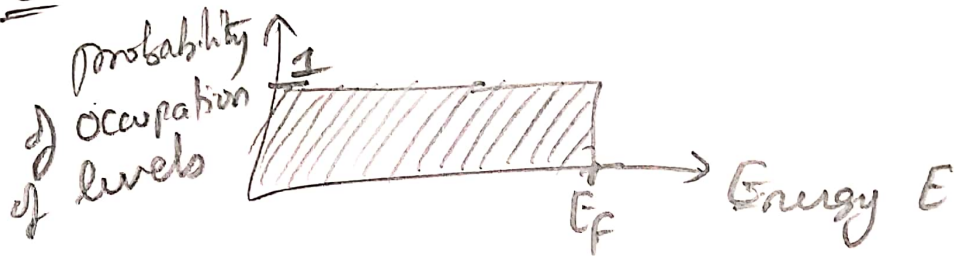
$$= \frac{58.2}{(r_s/a_0)^2} \times 10^4 \text{ K.}$$

That is, even at zero temperature, the electrons in the "Fermi gas" behave as if they are at a temperature of 10,000 K!

A corollary to this observation is that, room temperature (300 K) and zero Kelvin are nearly the same, looking from a temperature of 10,000 K. So we can treat theoretically metals at room temperature as systems at zero Kelvin, simplifying our calculations.

For temperatures $T \neq 0$, if we want to have precise, quantitative results, we must have an idea of the probability of occupation ^{by electrons} of states with energy $E > E_F$. Since electrons can gain thermal energy & can move above their Fermi energy states.

At low temperatures, particularly at $T=0$, all the low energy levels are occupied upto the Fermi energy E_F & the levels above are empty. That is the probability of occupation of energy levels with energy $E < E_F$ is 1 & the probability for levels with energy $E > E_F$ is 0.



What is the probability of occupation of energy levels when $T \neq 0$? The answer to this question is given by the Fermi-Dirac distribution function:

$$f(E) = \frac{1}{\exp((E - \mu)/k_B T) + 1}$$

Probability of occupation $\leftarrow f(E)$
 energy of the energy level whose occupation probability we want $\leftarrow E$
 Boltzmann const $\leftarrow k_B$
 temperature $\leftarrow T$
 Chemical potential $\leftarrow \mu$
 $\approx E_F$, Fermi energy near $T=0$.

At Temperature $T \neq 0$, the F-D distribution looks like below:



Since $f(E)$ gives the probability of occupation of a given level of energy E , the total number of electrons in a given system at a temp T & chemical potential μ is

$$N = \sum_{\substack{i \\ \text{levels}}} f_i = \sum \frac{1}{\exp\left(\frac{E - \mu}{k_B T}\right) + 1}$$

This connects T , N & μ .

The total energy of a system is

$$U = 2 \sum_{\vec{k}} E(\vec{k}) f(E(\vec{k}))$$

\vec{k} ← wave vectors of electrons
 $E(\vec{k})$ ← energy
 $f(E(\vec{k}))$ ← F-D distribution

This can be written as an integral
 if we allow $V \rightarrow \infty$:

$$u = \frac{U}{V} = \int \frac{d\vec{k}}{4\pi^3} \mathcal{E}(\vec{k}) f(\mathcal{E}(\vec{k}))$$

energy density.

and

$$n = \frac{N}{V} = \int \frac{d\vec{k}}{4\pi^3} f(\mathcal{E}(\vec{k}))$$

It is important to note that these \int 's are over k -space or the momentum space, whereas the integrands enter as functions of energy. It is useful to do the integration over energy, instead of momentum. That is, we go from $d\vec{k}$ to $d\mathcal{E}$. Since $d\vec{k} = dk_x dk_y dk_z$ & in spherical coordinates, it becomes $d\vec{k} = 4\pi k^2 dk$, we get

$$\int \frac{d\vec{k}}{4\pi^3} F(\mathcal{E}(\vec{k})) = \int_0^\infty \frac{4\pi k^2}{4\pi^3} F(\mathcal{E}) |dk|$$

any F

$$= \int \frac{k^2 dk}{\pi^2} F(\mathcal{E})$$

using $\mathcal{E} = \frac{\hbar^2 k^2}{2m}$,

$$d\mathcal{E} = \frac{2\hbar^2 k dk}{2m}$$

$$\text{or } d\mathcal{E} = \frac{2\hbar^2}{2m} \cdot \sqrt{\frac{2m\mathcal{E}}{\hbar^2}} dk$$

Substituting,

$$\int_0^{\infty} \frac{k^2 dk}{\pi^2} F(\epsilon) = \int_{-\infty}^{\infty} \frac{2m\epsilon}{\hbar^2} \cdot \frac{m}{\hbar^2} \cdot \sqrt{\frac{\hbar^2}{2m\epsilon}} \frac{d\epsilon}{\pi^2} F(\epsilon)$$

$$= \int_{-\infty}^{\infty} \left[\frac{m}{\hbar^2 \pi^2} \cdot \sqrt{\frac{2m\epsilon}{\hbar^2}} \right] F(\epsilon) d\epsilon$$

$g(\epsilon)$ = Density of states [calculated by going from dk to $d\epsilon$, momentum to energy].

$$= \int_{-\infty}^{\infty} \underset{\substack{\uparrow \\ \text{DOS per unit volume}}}{g(\epsilon)} F(\epsilon) d\epsilon$$

$g(\epsilon) d\epsilon = \frac{1}{\text{Volume}} \times \text{Number of one electron energy levels in the energy range } \epsilon \text{ to } \epsilon + d\epsilon.$

Since we know that the electronic density

$$n = \frac{k_F^3}{3\pi^2} \quad \& \quad \text{Fermi energy}$$

$$\epsilon_F = \frac{\hbar^2 k_F^2}{2m}$$

We can use $n \propto E_f$ to simplify the "appearance" of $g(E)dE$:

$$g(E) = \frac{m}{\hbar^2 \pi^2} \sqrt{\frac{2mE}{\hbar^2}} = \left[\frac{m}{\hbar^2 \pi^2} \sqrt{\frac{2m}{\hbar^2}} \right] \sqrt{E}.$$

$$E_f^{3/2} = \frac{\hbar^3 K_f^3}{(2m)^{3/2}} \quad \& \quad \frac{n}{E_f^{3/2}} = \frac{K_f^3 / 3\pi^2}{\hbar^3 K_f^3 / (2m)^{3/2}}$$

$$= \frac{(2m)^{3/2}}{3\hbar^3 \pi^2} = \frac{1}{3} \frac{2m}{\hbar^2 \pi^2} \sqrt{\frac{2m}{\hbar^2}} = \frac{2}{3} \left[\frac{m}{\hbar^2 \pi^2} \sqrt{\frac{2m}{\hbar^2}} \right]$$

$$\therefore \boxed{g(E) = \frac{2}{3} \frac{n}{E_f^{3/2}} \sqrt{E} = \frac{2}{3} \left(\frac{n}{E_f} \right) \sqrt{\frac{E}{E_f}}}$$

dimension \downarrow $\frac{e^- \text{ density}}{\text{energy}}$

We can now get back to evaluating the energy density & number density of electrons when $T \neq 0$:

$$u = \int \frac{d\vec{k}}{4\pi^3} E(\vec{k}) f(E(\vec{k}))$$

$$\& n = \int \frac{d\vec{k}}{4\pi^3} f(E(\vec{k}))$$

where the $\int k$ is over k -space.

We can write this $\int k$ as one over energy by introducing the density of states:

$$U = \int \frac{d\vec{k}}{4\pi^3} \underbrace{E(\vec{k})}_{\text{energy}} \underbrace{f(E(\vec{k}))}_{\text{FD dist. fn}} = \int \boxed{dE g(E)} \underbrace{E \cdot f(E)}_{\text{for energy density}}$$

and

$$n = \int \frac{d\vec{k}}{4\pi^3} f(E(\vec{k})) = \int \boxed{dE g(E)} f(E)$$

This form of equations for evaluating U & n is important because only $g(E)$ depends on the geometry of the substance studied (1-D, 2-D or 3-D) and can be modified accordingly.

Our goal, as in Drude's model, is to evaluate the specific heat of electron gas. This involves evaluating $C_V = \left(\frac{\partial U}{\partial T} \right)_V$.

Since the details of calculation are quite involved, we can just write down the answer for u , the energy density:

$$u = u_0 + \frac{\pi^2}{6} (k_B T)^2 g(E_F)$$

ground state
density at $T=0$

DOS at Energy E_F .

Therefore, C_V is

$$C_V = \left(\frac{\partial u}{\partial T} \right)_{n,V} = \frac{\pi^2}{3} k_B^2 T \cdot g(E_F)$$

for free electrons, $g(E_F)$ is $\frac{3}{2} \cdot \frac{n}{E_F}$ (see above)

$\therefore C_V$ for free electrons is ≈ 0.01

$$C_V \approx \frac{\pi^2}{2} \left(\frac{k_B T}{E_F} \right) \cdot n k_B$$

Compare this to $\frac{3}{2} n k_B$ from classical kinetic theory of gases. This explains why C_V of e^- s is $1/100^{\text{th}}$ of $\frac{3}{2} n k_B$ from classical physics.