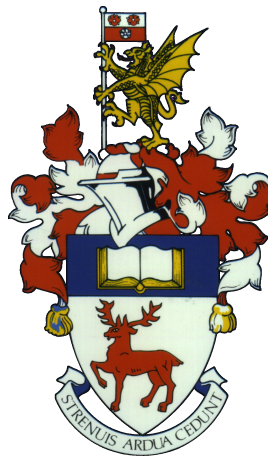


1                    Optimization of Fast Filling of Hydrogen Cylinders

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4    This report is submitted in partial fulfillment of the requirements for the MEng Mechanical  
5    Engineering, Faculty of Engineering and the Environment, University of Southampton.



6

**Declaration**

7

I, Pau Miquel Mir, declare that this thesis and the work presented in it are my own and has been generated by me as the result of my own original research. I confirm that:

8

9

1. This work was done wholly or mainly while in candidature for a degree at this University;

10

2. Where any part of this thesis has previously been submitted for any other qualification at this University or any other institution, this has been clearly stated;

11

12

3. Where I have consulted the published work of others, this is always clearly attributed;

13

4. Where I have quoted from the work of others, the source is always given. With the exception of such quotations, this thesis is entirely my own work;

14

15

5. I have acknowledged all main sources of help;

16

6. Where the thesis is based on work done by myself jointly with others, I have made clear exactly what was done by others and what I have contributed myself;

17

18

7. None of this work has been published before submission.

19

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I want to thank my advisor, Dr. Edward Richardson, for his time, dedication, patience, and support throughout this project.

21

22

Furthermore, I would like to thank Vishagen Ramasamy for his initial work upon which this paper builds.

23

24

I also want to thank my parents, for their constant support and for proofreading the paper.

25

**Abstract**

26

This is the abstract.

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60 **References**

**21**

61 **Acronyms**

62 **CFD** Computational Fluid Dynamics

63 **CNG** Compressed Natural Gas

64 **FTCS** Forward-Time, Centered-Space

65 **ODE** Ordinary Differential Equation

66 **PDE** Partial Differential Equation



## 67 **Todo list**

68	Reword . . . . .	1
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70	What the fuck are inlet profiles . . . . .	1
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86	Does this need to be cited / proved? Eplain better so you dont have to cite it . . . . .	9
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88	need to explain why? . . . . .	10
89	cite . . . . .	11
90	finish . . . . .	13
91	General stuff:	
92	- Check tense and person	
93	- Check for repetition	
94	- Check for excessive verbosity - define short and long cylinders at some point . . .	14



# 1 Introduction

## 1.1 Purpose of the investigation

Hydrogen is a very promising alternative fuel for the future, mainly due to the absence of greenhouse emissions when burning it. In this regard, it is superior to petroleum- and, more generally, carbon-based fueling systems currently used by most road vehicles. However, air pollution is a negative externality associated with this type of fuels, as it is a cost to a third party (in this case, society as a whole) that is not accounted for in the price of the good. By definition, the consequences of polluting the air are not included in the price of traditional vehicles, and therefore will have little influence on consumers' choices. Thus, if hydrogen is to succeed as an alternative to carbon-based fuels, it is essential that its use be as convenient, if not more, than traditional fuel.

One of the main aspects where hydrogen currently lags behind traditional fuel is the refueling process. Indeed, refueling hydrogen involves its compression, and thus a significant rise in temperature, which according to standards must be kept below a certain value (358 °K as per SAE J2601 ). This temperature limit in turn leads to long refueling times, potentially lasting more than five minutes, which is cumbersome for users. Therefore, it is of utmost importance to research and develop systems that enable faster refueling of hydrogen tanks. To this end, this project builds upon a model of filling a hydrogen cylinder which has already been developed by members of the Faculty of Engineering and the Environment at the Univeristy of Southampton to analyse novel methods of improving fill times.

## 1.2 Outline of the investigation

One of the current solutions to improve fill times involves cooling the hydrogen before filling the cylinder. However, this is quite an expensive process, both in energetic and economic terms. Consequently, the aim of this project is to explore several of the options available to reduce fill times while simultaneously reducing the energy consumption of the process, thus improving both convenience for users and energy efficiency of the fueling stations.

An attempt to optimise the filling profile of hydrogen cylinders will be made by building upon the existing cylinder model. The model will be developed to include transient effects by implementing a hysteresis effect. This will allow for innovative inlet profiles to be evaluated, such as sinusoidal or square waves . An attempt will be further made to completely determine the ideal filling profile by means of a constrained nonlinear optimization.

# 2 Background

## 2.1 Challenges of hydrogen fuelled for vehicles

Hydrogen is an attractive alternative fuel because it has zero carbon emissions at point of use. In order to exploit hydrogen on road vehicles it is necessary to have practical hydrogen storage and filling infrastructure. Three different technologies have been developed so far: liquid storage systems,

metal hydrides, and compressed hydrogen solutions. Liquid storage systems achieve high energy density but require cryogenic cooling solutions, and metal hydrides are an emerging solution that still needs further development to compete with compressed hydrogen. Therefore, this study focuses exclusively on compressed gas storage of hydrogen. To achieve high energy density very high pressures are required, 35 and 70 MPa being the standards. This compressed hydrogen is then used in the fuel cell of the vehicles and converted into electricity, which can be used directly by the electric motor or stored in onboard batteries. The requirements of a compressed hydrogen system are:

- High fuel capacity
- Low overall weight
- Fast fill times

Increasing fuel capacity can be achieved by either using larger tanks, which leads to higher weights, or by using higher pressures. However, fast refueling of hydrogen gas to high pressures leads to a sharp increase in gas temperature due to the Joule-Thompson effect and the quasi-adiabatic compression involved. These higher temperatures are problematic due to material constraints as explained below. This may restrict the fill rate because to avoid exceeding the temperature threshold of the material we may need to allow time for the heat to dissipate through the cylinder walls. Thus, solutions must be developed to minimize the temperature rise when fast filling to high pressures.

Currently, two types of cylinder used are: Type III cylinders, made out of an aluminum liner wrapped in a carbon-fiber/epoxy composite, and Type IV cylinders, with a plastic liner and the same composite wrap. **Of concern** is the composite material, as the polymer matrix cannot withstand high temperatures, and the material properties of the cylinder will begin to degrade. The specific temperature at which this occurs is usually around the glass transition temperature of the epoxy, where the thermosetting polymer changes from a hard “glassy” state to a more compliant “rubbery” state.

yes it gets hot, but how hot is it gonna get. Can prove this by showing temp change with adiabatic filling

cite:  
<http://www.epotek>

## 2.2 Previous work

This section begins with a summary of previous work concerning heat transfer during fast filling of hydrogen cylinders and concludes with an analysis for the need for more research.

### 2.2.1 Experimental work

Several papers describe experimental work that has been conducted regarding the fast-filling of hydrogen cylinders, in particular comparing the results to simulations. Dicken and Mérida’s work indicates that the temperature inside the cylinder is rather uniform [1]. This claim is not, however supported by the work of Zheng et al [2] nor that of Woodfield et al [3], wherein large discrepancies among gas temperatures in different regions of the cylinder are found.

{sec:experimental\_w

## 164 2.2.2 Computational Fluid Dynamics (CFD) models

{sec:cfdl}

165 A large amount of research has been conducted using multidimensional analysis, especially using  
 166 CFD. Both 2D axisymmetric models and full 3D models have been developed by many authors. 2D  
 167 models do not include the effects of gravity or buoyancy, but these can be considered negligible. On  
 168 the other hand, full 3D simulations are computationally expensive and rarely justifiable as the gain  
 169 in accuracy is very small.

170 There are many papers describing slightly different methods and setups. The models attempt to  
 171 solve the Reynolds averaged Navier Stokes equations, which are time averaged equations representing  
 172 the fluid flow. As these equations are not closed, models for the turbulent viscosity term must be  
 173 utilised.

174 Kim et al [4] propose a 3D model with a standard  $k-\epsilon$  turbulence model [5] and the Redlich-  
 175 Kwong real gas equation of state [6] for real gas properties. Dicken and Mérida [7] present an  
 176 axisymmetric model that uses a modified standard  $k-\epsilon$  turbulence model in order to reduce the jet's  
 177 spreading rate over-prediction of the standard model.

178 Other investigations use more advanced property models, such as the one presented by Zhao  
 179 et al [8], which employs REFPROP, a tool developed by the National Institute of Standards and  
 180 Technology [9], in order to have more precise real gas properties. It uses values of critical and triple  
 181 points together with equations for the thermodynamic and transport properties to calculate the state  
 182 points of fluids. Zhao et al's model is validated against experimental results by Zheng et al [2]. In  
 183 the same spirit, other authors employ more advanced turbulence modelling, such as Heitsch et al's  
 184 model [10] which uses Menter's Shear Stress Transport turbulence model [11].

185 Galassi et al [12] present a 3D model using a modified  $k-\epsilon$  turbulence model and the Redlich-  
 186 Kwong real gas equation of state similar to the work by Kim et al [4]. Melideo et al [13] compare  
 187 an axisymmetric model with this full 3D model and find a very good agreement in the temperature  
 188 profile, suggesting that the significant extra computation expense of full 3D models might not be  
 189 necessary. They also use the improved Aungier Redlich Kwang real gas equation of state [14] for  
 190 improved speed and accuracy.

191 Most models are built upon existing commercial CFD codes, mainly CFX [**cfx**], used by refer-  
 192 ences [10, 12, 13] and FLUENT [**fluent**], used by references [2, 4, 7, 8].

## 193 2.2.3 Zonal models

194 Zonal models, also referred to as a 0-dimensional or reduced order models, consider the gas inside  
 195 the cylinder as the control volume with homogenous properties. This premise allows for simpler  
 196 calculations and much lower computational time.

197 Liu et al [15] assume adiabatic filling of the cylinder. By applying the first law of thermodynamics,  
 198 they equalled the change in internal energy of the gas to the change in the product of mass and static  
 199 enthalpy. By then using real gas equations of state, they obtain a simple algebraic equation relating  
 200 the final temperature in the cylinder to the initial temperature, initial and final pressures.x

201 Hosseini et al [16] also assume adiabatic filling of the cylinder in addition to a constant mass

Deal with CFD  
shit

{sec:zonalModels}

not capital

double check if true

flow rate in order to simplify the first law of thermodynamics to an Ordinary Differential Equation (ODE). It is solved for internal energy as a function of static enthalpy of the inlet gas, initial specific internal energy, mass flow rate, initial mass, and time. The internal energy can be substituted for a real gas equation to get an expression for temperature as a function of time.

Early work regarding the filling of Compressed Natural Gas (CNG) tanks involves very similar thermodynamics to hydrogen filling. Kountz [17] presented a model which linked an energy balance for the gas to a lumped mass heat conduction model for the cylinder. It used a constant heat transfer coefficient of  $28 \text{ W/(m}^2\text{K)}$ , although it must be noted that CNG cylinders are filled to lower pressures and during longer fill times.

Woodfield et al [3] coupled a single zone model of the gas in the cylinder with a one dimensional unsteady model for the heat conduction through the cylinder walls. They used the Lee-Kesler method [18] to find the compressibility and thus the density, and from that the mass inside the cylinder. The heat transfer coefficient between the gas and the wall was assumed to  $500 \text{ W/(m}^2\text{K)}$  during filling and  $250 \text{ W/(m}^2\text{K)}$  after full.

Similarly, the work by Monde et al [19] uses the same assumption regarding heat transfer coefficient values. They achieve a reasonable fit with experimental data, shown in Fig. 1, even though the measured heat transfer coefficients were significantly lower, as shown in Table 1.

FIGURE 1: Comparison between estimated and measured temperatures by Monde et al [19].

{fig:mondeFit}

P (MPa)		Mass flow rate (g/min)	$h \text{ (W/(m}^2\text{K))}$
5	H <sub>2</sub>	168 - 276	86.4 - 97.4
	N <sub>2</sub>	456 - 1296	43.0 - 47.0
10	H <sub>2</sub>	240 - 324	143.1 - 154.5
	N <sub>2</sub>	732 - 996	38.9 - 44.7
35	H <sub>2</sub>	45-170	269.7 - 279.2

TABLE 1: Heat transfer coefficients at various conditions by Monde et al [19].

{tab:mondeHValues}

Monde et al [20] used newly published test data to validate their model presented in reference [19]. While the model matches the data reasonably well, different values of heat transfer coefficient have to be employed to achieve a good fit, with recommended values of  $150 \text{ W/(m}^2\text{K)}$  and  $200 \text{ W/(m}^2\text{K)}$  for filling to 35 MPa and 70 MPa, respectively.

Striednig et al [21] developed a whole-station model linking zero-dimensional models for the storage tank gas, the vehicle tank gas, and the vehicle cylinder. They propose using an annular flow correlation to relate the Nusselt number to the Reynolds number of the flow to dynamically calculate the heat transfer at the wall. However, as they were comparing against data with a Type I cylinder (complete steel construction) they assumed constant temperature through the thickness of the cylinder wall, an assumption which isn't valid for Type III and IV cylinders due to the low thermal conductivity of carbon fiber.

Monde and Woodfield [22] also presented an improved model with a dynamic heat transfer coefficient. The energy balance in the gas is coupled to an unsteady heat conduction equation for the wall. The Nusselt number was derived as a function of Reynolds and Rayleigh numbers by means of an empirical correlation derived from experimental results they presented in reference [23].

A similar model is presented by Johnson et al [24]. While using a proprietary software called Netflow developed by Sandia Labs, a zonal model of the cylinder is created using the same internal energy and mass balance formulation. It considers three different control volumes for the gas: each of the two half domes and the strictly cylindrical section of the tank. Furthermore, it utilizes the Reynolds number of the jet to dynamically calculate the Nusselt number, and thus the heat transfer coefficient, at the wall. The coefficients for these empirical relationships are derived from their experimental results.

Ranong et al [25] present a model with a single zone model for the gas and unsteady heat conduction through the wall. They develop a relationship between Nusselt and Reynolds numbers based upon CFD simulations also presented in that paper. Using this relationship, the heat transfer coefficient is dynamically calculated.

Finally, Khan et al [26] analyzed different tank sizes and optimised fueling characteristics by using the model presented by Monde and Woodfield in references [3, 19, 20]. They use a constant heat transfer coefficient, and make an attempt to evaluate the feasibility of step-filling. However, the analysis is hindered by the lack of unsteady modelling of the heat transfer coefficient.

#### 2.2.4 Summary and outlook

The simulation work that has been presented in this section, in broad terms, divided into complex CFD models and more simplified reduced dimensional models. It is important to analyze the difference between these two approaches.

CFD models have several advantages, as they take into consideration many details that reduced order models must assume, simplify, or neglect. Firstly, one can represent the specific effects of geometry of the cylinder more accurately using CFD. Secondly, time effects can be directly simulated without the need of timescales to be modelled. However, they are extremely computationally expensive, with run times lasting from hours to weeks. It thus follows that the main advantage of reduced dimension models is that they are much less computationally expensive. This, in turn, means they can be incorporated as a part of larger analyses, in which they must be run multiple times, such as optimization routines or probabilistic whole station models.

In compressed hydrogen fuel systems the gas is at very high pressures, meaning ideal gas approximations are inaccurate, and thus real gas properties must be employed. Several methods are employed in existing work as outlined in Section 2.2.3. Most use real gas equations of state, which although very accurate, are less accurate than property models. Property models determine any gas property from two other independent properties. The property model that will be employed throughout this analysis is REFPROP. Although the use of REFPROP is fairly established in CFD models [2, 8, 24], it is less prominent in reduced order simulations.

do I have evidence

Reduced dimension models up to date make assumptions that are not always supported with evidence, and therefore lead to less accurate results. This requires adjustments to the model to match the experimental data, such as the case of Monde et al [20], explained in Section 2.2.3. Although this produces a better fit, the model is less predictive. For this reason, it is important to improve reduced order models by devising ways to obtain quantities previously assumed. This will improve the models' predictive ability, which will increase confidence in results from simulations regarding events which haven't been tried experimentally.

An assumption typically made is that of a constant heat transfer coefficient derived from either time-averaged experimental results or CFD simulations, as explained in Section 2.2.3. A slightly better assumption employed is that of deriving the heat transfer coefficient from the inlet Reynolds number by means of empirical correlations. However, work conducted so far has used a quasi steady approach to modelling the heat transfer coefficient. This implies either assuming a constant heat transfer coefficient, or deriving it from the instantaneous Reynolds number of the inlet flow. However, for rapid transients, this approach will not work, as the changes in the inflow will inaccurately result in immediate changes to the heat transfer conditions. This report will propose a new method of deriving the unsteady heat transfer coefficient numerically to achieve better results for transient convective heat transfer.

### 3 Formulation

{sec:formulation}

Out of place, where to put it

One of the fundamental principles used throughout this paper, and indeed, throughout engineering, is that of non-dimensioning. By operating using non-dimensional parameters such as the Reynolds number  $Re = \frac{\rho u L}{\mu}$  or the Prandtl number  $Pr = \frac{c_p \mu}{k}$  solving problems involving differential equations becomes simplified. Also, the analysis becomes much more general, and can be scaled.

#### 3.1 Governing equations

The governing equation for mass and internal energy of the gas in the tank is given by:

$$\frac{dU_{gas}}{dt} = \frac{d(m_{gas}u_{gas})}{dt} = h_{in}\dot{m}_{in} - \dot{Q}_{out} \quad (3.1)$$

consistent time derivatives

where  $U_{gas}$  is the total internal energy of the gas,  $m_{gas}$  is the mass of the gas,  $u_{gas}$  is the specific internal energy of the gas,  $H_{in}$  is the enthalpy of the inlet gas,  $\dot{m}_{in}$  the mass flow into the cylinder, and  $\dot{Q}_{out}$  is the rate of heat transfer out of the gas to the cylinder.

The change in internal energy of the gas is equal to the difference in energy entering the system in the enthalpy of gas inflow and the energy leaving the system through the walls of the cylinder. The enthalpy of the inlet can be determined from real gas models as a function of the inlet pressure and temperature. Therefore, to obtain the internal energy variation one must calculate the gas mass flow and the heat transferred to the cylinder.

We also have conservation of mass in the system, which is expressed simply by:



$$\frac{dm_{gas}}{dt} = \dot{m}_{in} \quad (3.2)$$

as the change in mass of the gas in the cylinder is only influenced by the influx of gas during filling.

One of the simplest methods available to solve this system of ODEs, both conceptually and in terms of ease of implementing in code, is forward Euler time integration. It can be described as follows: given a function that can be defined by:

$$y'(t) = f(t, y(t)), \quad y(t_0) = y_0 \quad (3.3)$$

we can compute the approximate shape of the function given the initial point and finding the slope of the curve for small intervals. Indeed, from the initial point, we can find the tangent of the curve at that point, and take a small step along that tangent until arriving at the next point, where the procedure can be repeated. Denoting the step size as  $h$ , we can express forward Euler time integration as:

$$y_{n+1} = y_n + hf(t_n, y_n) \quad (3.4)$$

### 3.2 Gas mass flow into cylinder

The nozzle and corresponding gas flow will be modeled using isentropic relations, and then a discharge coefficient relationship will be used to find the approximate real values. Indeed, the isentropic Reynold's number is first calculated as follows:

$$\text{Re}_{\text{jet,ideal}} = \frac{\rho_{\text{exit}} d_{\text{inlet}} u_{\text{exit}}}{\mu_{\text{exit}}} \quad (3.5)$$

where  $d_{\text{inlet}}$  is the diameter of the inlet delivery pipe and  $\rho_{\text{exit}}$ ,  $u_{\text{exit}}$ , and  $\mu_{\text{exit}}$  are determined using real gas models as described in Section ???. More specifically:

$$\rho_{\text{exit}} = f(P_{\text{exit}}, S_{\text{in}}), \quad u_{\text{exit}} = \sqrt{2(H_{\text{in}} - H_{\text{static}})}, \quad \mu_{\text{exit}} = f(P_{\text{exit}}, S_{\text{in}}) \quad (3.6)$$

where  $P_{\text{exit}}$  is the pressure at the end of the inlet tube,  $S_{\text{in}}$  is the entropy at the inlet,  $H_{\text{in}}$  the stagnation enthalpy at the inlet,  $H_{\text{static}}$  the static enthalpy at the end of the inlet tube, and the function  $f$  represents the real gas model which computes a thermodynamic property from any two given properties. As the system is initially treated as isentropic, the inlet entropy can be used to calculate exit properties. We also have:

$$S_{\text{in}} = f(P_{\text{in}}, T_{\text{in}}), \quad H_{\text{in}} = f(P_{\text{in}}, T_{\text{in}}), \quad H_{\text{static}} = f(P_{\text{exit}}, S_{\text{in}}) \quad (3.7)$$

where  $P_{\text{in}}$  and  $T_{\text{in}}$  are the inlet pressure and temperature, respectively. The exit pressure  $P_{\text{exit}}$  is taken to be the pressure inside the gas tank. However, in the case that the inlet diameter is choked, these calculations would yield velocities higher than the speed of sound, which is impossible due to the nature of the inlet pipe. For this reason, if a velocity of Mach 1 or higher is achieved at any given time, an iterative process is used to find the value of  $P_{\text{exit}}$  that will yield a velocity equal to the speed

of sound, and the rest of properties and ultimately the Reynolds number calculated accordingly.

In order to find the real mass flow an empirical discharge coefficient is employed.

$$C_D = \frac{\dot{m}_{in}}{\dot{m}_{ideal}} = c_2 + c_3 \text{Re}_{jet,ideal} \quad (3.8)$$

A discharge coefficient must be used to account for the formation of a boundary layer inside the inlet tube. The empirical model that was used was obtained from , and uses the following values:

Citation

$$c_2 = 0.938, \quad c_3 = -2.71 \quad (3.9)$$

From the real mass flow the actual Reynold's number of the inflow can be calculated and used to find forced convection heat transfer coefficients in Section 3.3.1 and Eq. (3.14) as follows:

avoid forward references

$$\text{Re}_{jet} = \frac{4\dot{m}_{in}}{\pi\mu d} \quad (3.10)$$

### 3.3 Heat transfer from gas to cylinder

The heat transferred from the gas to the wall of the tank is given by a simple convection relationship:

$$\dot{Q} = hA(T_{gas} - T_{wall}) \quad (3.11) \quad \{\text{equ:convection}\}$$

where  $A$  is the internal surface area of the cylinder,  $h$  is the heat transfer coefficient, and  $T_{gas}$  and  $T_{wall}$  are the temperatures of the gas and the wall, respectively. The heat transfer coefficient is a result of the combination of both forced and natural convection. This is expressed as follows, as per [27] and [28]:

$$h = \sqrt[4]{h_f^4 + h_n^4} \quad (3.12)$$

where  $h_f$  is the heat transfer coefficient due to forced convection and  $h_n$  is the heat transfer coefficient due to natural convection. Each coefficient can be non-dimensionalised using Nusselt numbers, expressed as:

$$\text{Nu}_f = \frac{h_f D}{k}, \quad \text{Nu}_n = \frac{h_n D}{k} \quad (3.13)$$

where  $D$  is the characteristic length, in this case the diameter of the cylinder, and  $k$ , the thermal conductivity of the fluid. The values of the Nusselt numbers can be determined from empirical correlations, as detailed in Sections 3.3.1 and 3.3.2.

#### 3.3.1 Forced convection

{sec:forcedConvection}

As the magnitude of heat transfer is related to the flow in forced convection, the Nusselt number can be said to be a function of the Reynolds number, taking the form:

$$\text{Nu}_{f,ss} = c_4 \text{Re}_{jet}^{c_5} \quad (3.14) \quad \{\text{equ:nusseltReynold}\}$$

where  $Nu_{f,ss}$  is the steady state Nusselt number and the empirical constants  $c_4$  and  $c_5$  are given by as:

$$c_4 =, \quad c_5 = \quad (3.15)$$

The initial work this project builds upon uses the flow at the nozzle to determine the heat transfer at the wall of the cylinder. This assumes that the hydrogen flows instantaneously from the nozzle to the wall, when in reality there is of course a time delay. This assumption is acceptable for stable inflows, hence us referring to the steady state Nusselt number  $Nu_{f,ss}$ , but for more complex filling patterns and also for improved accuracy it becomes necessary to incorporate hysteresis. Indeed, the heat transfer coefficient at the wall in fact can be said to depend on the nozzle flow seconds prior, or more generally, on the history of the nozzle flow. This can be modeled as follows:

$$\frac{d}{dt}(Nu_f) = \frac{Nu_{f,ss} - Nu_f}{\tau} \quad (3.16)$$

where  $\tau$  is the time scale, which reflects the amount of time it takes for the flow to recirculate. Two time scales are considered, as two different situations may be present. The flow in the cylinder may be driven by the inflow of mass from the nozzle, in which case we have a time scale of production  $\tau_{prod}$ . However, if the inflow halts, the flow field in the cylinder will be driven by the dissipation of the existing flow and thus the time scale is denoted  $\tau_{diss}$ . Therefore the overall time scale is given by the minimum of these two:

$$\tau = \min(\tau_{prod}, \tau_{diss}) \quad (3.17)$$

**Production time scale** The main driver of flow in the cylinder, which in turn drives convective heat transfer, is the turbulent jet that develops from the end of the nozzle throughout the length of the cylinder. Experimental results can be used to derive expressions for the axial velocity of the jet. Radial profiles of mean axial velocity can be seen in Fig. 2.

FIGURE 2: Radial profiles of mean axial velocity in a turbulent round jet with  $Re = 95,000$ . Adapted from [29] with the data from [30]

By plotting the inverse of the non-dimensional speed,  $u_{exit}/u(x)$  against  $x/d$  we get the clearly linear relationship shown in Fig. 3.

FIGURE 3: Jet speed vs distance. Adapted from [29] with the data from [30]

From this experimental result the following relationship is obtained:

$$\frac{u(x)}{u_{exit}} = \frac{c_1}{(x - x_0)/d_{inlet}} \quad (3.18)$$

where  $u(x)$  is the axial velocity at a distance  $x$  from the nozzle,  $x_0$  is the position of the virtual origin, and  $c_1$  is an empirical constant.

The production time scale is given by the spatial integral of the axial speed of the jet, which is given in Eq. (3.18) in ??.

$$\begin{aligned}\tau_{prod} &= \int_0^L \frac{1}{u(x)} dx = \frac{1}{c_1 u_{exit}} \int_0^L (x - x_0) dx \\ &= \frac{\frac{L^2}{2} - x_0 L}{c_1 u_{exit} d}\end{aligned}\quad (3.19)$$

As we have  $x_0 \ll L$  we can write:

$$\tau_{prod} = \frac{L^2}{2c_1 u_{exit} d} \quad (3.20)$$

and defining  $c_6 = \frac{1}{2c_1}$ :

$$\tau_{prod} = c_6 \frac{L^2}{u_{exit} d} \quad (3.21)$$

**Dissipation time scale** The dissipation time scale can be said to be proportional to the length of the cylinder  $L$  and the recirculation velocity  $u_{recirc}$ :

$$\tau_{diss} = c_7 \frac{L}{u_{recirc}} \quad (3.22)$$

need to explain why?

We can then define the Reynolds number of the recirculation flow as:

$$\text{Re}_{recirc} = \frac{u_{recirc} D}{\nu} \quad (3.23)$$

Let us assume that instantaneous Nusselt number depends on  $u_{recirc}$  as it depends on  $\text{Re}^{c_5}$  and  $c_5 \sim \mathcal{O}(1)$ . From Eq. (3.14) we know that:

$$\frac{D u_{recirc}}{\nu} = c_8 \text{Nu}_f \quad (3.24)$$

We can therefore rewrite:

$$\tau_{diss} = \frac{c_7}{c_8} \frac{LD}{\nu \text{Nu}_f} \quad (3.25)$$

Combining the two constants and absorbing the  $L/D$  coefficient:

$$\tau_{diss} = c_9 \frac{L^2}{\nu \text{Nu}_f} \quad (3.26)$$

We can then make use of the fact that in steady state we know  $\tau_{prod} = \tau_{diss}$ , which leads to:

$$c_9 \frac{L^2}{\nu \text{Nu}_f} = c_6 \frac{L^2}{u_{exit} d} \quad (3.27)$$

384 Substituting the definitions of Nusselt and Reynolds numbers from Eqs. (3.5) and (3.14):

$$c_9 = c_6 \frac{\nu \text{Nu}_f}{u_{exit} d} = c_4 c_6 \quad (3.28)$$

385 Which leads to a final definition of  $\tau_{diss}$  in terms of literature constants:

$$\tau_{diss} = \frac{c_4}{2c_1} \frac{L^2}{\nu \text{Nu}_f} \quad (3.29)$$

386 Therefore, in a pure dissipation scenario, where  $\text{Nu}_{f,ss} \rightarrow 0$ , we have:

$$\frac{d}{dt} (\text{Nu}_f) = -\text{Nu}_f^2 \frac{2c_1}{c_4} \frac{\nu}{L^2} \quad (3.30)$$

387 which can be solved by simple integration, which yields:

$$\int_{\text{Nu}_{initial}}^{\text{Nu}(t)} -\frac{d\text{Nu}_f}{\text{Nu}_f^2} = \int_0^t \frac{2c_1}{c_4} \frac{\nu}{L^2} dt \quad (3.31)$$

$$\text{Nu}_f = \frac{c_4}{2c_1} \frac{L^2}{\nu} \frac{1}{t} \quad (3.32)$$

### 3.3.2 Natural convection

388 Natural convection is caused by buoyancy driven flow, so the natural Nusselt number can be found  
 389 to be related to the Rayleigh's number (itself a product of Grashof's and Prandtl's numbers) by the  
 390 following relationship:  
 391

$$\text{Nu}_n = c_{10} \text{Ra}^{c_{11}} \quad (3.33)$$

392 where Rayleigh's number is defined as:

$$\text{Ra} = \left| \frac{g\beta (T_{wall} - T_{gas}) D^3}{\nu \alpha} \right| \quad (3.34)$$

393 where  $g$  is the acceleration due to gravity,  $\beta$  is the coefficient of thermal expansion,  $D$  is the charac-  
 394 teristic length, in this case the cylinder diameter,  $\nu$  is the kinematic viscosity, and  $\alpha$  is the thermal  
 395 diffusivity, as defined by:

$$\alpha = \frac{k}{\rho c} \quad (3.35)$$

396 where  $k$  is the material's thermal conductivity,  $\rho$  is the density of the material, and  $c$  is the specific  
 397 heat capacity of the material. The coefficients are given by :

$$c_{10} =, \quad c_{11} = \quad (3.36)$$

FIGURE 4: Diagram show composition of wall and heat fluxes into and out of wall.

ig:wall}

### 3.4 Heat transfer across cylinder

The main mode of heat transfer that occurs in the cylinder and that will be used throughout this report is heat conduction through the wall of the cylinder. A diagram showing the heat transfer through the wall is shown in Fig. 4. The heat transfer is modeled using one dimensional unsteady heat conduction:

$$\frac{\partial T}{\partial t} = \alpha \frac{\partial^2 T}{\partial x^2} \quad (3.37) \quad \{\text{equ:heatEquation}\}$$

where  $\alpha$  is the thermal diffusivity, as defined in Eq. (3.35). The boundary condition at the inner wall can be described by equalling the heat flux into the wall from convection to the conduction through the wall:

$$k \frac{dT_w}{dx} \Big|_{x=0} = hA (T_{gas} - T_{wall}) \quad (3.38) \quad \{\text{equ:innerWallBC}\}$$

For the boundary between the liner and the laminate a point precisely on the boundary is considered. At that point, the heat flux through the liner material and the heat flux through the laminate material and the respective temperatures must be equal:

$$\begin{aligned} k_{lin} \frac{\partial T_{lin}}{\partial x} &= k_{lam} \frac{\partial T_{lam}}{\partial x} \\ T_{lin} &= T_{lam} \end{aligned} \quad (3.39)$$

Solving this equation will yield the temperature distribution throughout the thickness of the cylinder wall, and more specifically, the internal wall temperature that is used to calculate the heat transferred out of the gas in Eq. (3.11).

#### 3.4.1 Outer Wall

Several cases can be considered for the outer wall, and the investigations described in Section 2.2.3 employ different boundary conditions. A constant heat flux can be applied to the outer wall, be it a constant heat loss or an adiabatic condition. The adiabatic condition represents the situation in which the cylinders are in an enclosed space which cannot transfer heat to the environment efficiently enough for it to affect the temperature of the tanks. this is the most conservative approach. Having a constant heat loss can be somewhat of an oversimplification, but a natural convection relationship could be employed to represent the heat loss to the environment in a more open environment. However, the approach used in this report will be the adiabatic condition, which can be expressed as follows:

$$\frac{\partial T_C}{\partial x} = 0 \quad (3.40)$$

where  $T_C$  is the outer wall temperature.

### 422 3.4.2 Discretisation

423 In order to solve the heat equation Partial Differential Equation (PDE) presented in Eq. (3.37) a dis-  
 424 cretisation method must be employed. The scheme employed is known as Forward-Time, Centered-  
 425 Space (FTCS), first described as such by Roache [31].

$$\frac{T_{i+1,j} - T_{i,j}}{\Delta t} = \alpha \frac{T_{i,j+1} - 2T_{i,j} + T_{i,j-1}}{\Delta x^2} \quad (3.41)$$

426 where the  $i$  and  $j$  subscripts represent the nodes of the discretisation in time and space, respectively,  
 427  $\Delta t$  is the time step, and  $\Delta x$  is the distance between grid points. This can be rearranged for  $T_{i+1,j}$  and  
 428 substituting  $r = \frac{\alpha \Delta t}{\Delta x^2}$ , we have:

$$T_{i+1,j} = T_{i,j} + r(T_{i,j+1} - 2T_{i,j} + T_{i,j-1}) \quad (3.42)$$

429 The scheme is numerically stable if and only if [32]:

$$r = \frac{\alpha \Delta t}{\Delta x^2} < \frac{1}{2} \quad (3.43)$$

430 We can also apply the discretisation scheme on the boundary conditions. For the inner wall  
 431 boundary condition expressed in Eq. (3.38) we can write:

$$T_{i+1,j} = T_{i,j} + 2r \left( T_{i,j+1} - T_{i,j} + \frac{\Delta x h (T_{gas} - T_{i,j})}{k_{lin}} \right) \quad (3.44)$$

$$T_{i+1,j} = T_{i,j} + \frac{\Delta t}{\Delta x^2} \left( \frac{k_{lam} (T_{i,j+1} - T_{i,j}) - k_{lin} (T_{i,j} - T_{i,j-1})}{0.5 (C_{lin} \rho_{lin} + C_{lam} \rho_{lam})} \right) \quad (3.45)$$

### 432 3.5 Throttling

433 Throttling is introduced in order to more accurately represent real life conditions. When a maximum  
 434 temperature is reached, the inflow of hydrogen must be stopped in order to protect the materials  
 435 of the tank, as detailed in Section 2.1. A simple method of throttling, with the flow stopping at  
 436 the designated maximum temperature, 85 °C, and the flow restarting at a chosen temperature. The  
 437 optimum temperature at which flow restarts was considered in this study, and results are presented in  
 438 ??

### 439 3.6 Optimization

440 The optimization was realized using a constrained nonlinear multivariable minimiser. It accepts  
 441 problems in the following form:

$$\min f(x) \text{ such that } \begin{cases} c(x) & \leq 0 \\ ceq(x) & = 0 \\ A \cdot x & \leq b \\ Aeq \cdot x & = beq \\ lb \leq x & \leq ub \end{cases} \quad (3.46)$$

where  $x$  is the input variable;  $A$  and  $Aeq$ , and  $b$  and  $beq$  are matrices and vectors respectively which describe the linear constraints,  $c(x)$  and  $ceq(x)$  are nonlinear functions constraining  $x$ , and  $lb$  and  $ub$  are vectors which describe the upper and lower bounds of the problem [**fmincon**].

The optimization problem for the hydrogen filling can be expressed as follows:

$$\min f(x) \text{ such that } c(x) \leq 85 \quad (3.47)$$

where  $f(x)$  returns the fill time given inlet pressure profile parameters  $x$ , and  $c(x)$  returns the maximum temperature reached by the gas inside the cylinder during the duration of the fill. The parameters  $x$  described can create several profiles:

- $x = m$ ,  $P(t) = mt + P_0$
- $x = [A, B, C]$ ,  $P(t) = A \cdot \sin\left(\frac{2\pi}{B}t\right) + Ct$
- $x = [A, B, C, D]$ ,  $P(t) = A \cdot g\left(\frac{2\pi}{B}t, C\right) + Dt$ , where  $g(x, d)$  creates a square wave with duty cycle  $d$ .
- $x = [P_1, P_2, \dots, P_n]$ ,  $P(t) = h(P_1, P_2, \dots, P_n)$  where  $h([P])$  is a spline interpolating function.

By specifying which type profile will be used, the optimiser will take the initial parameters, find the fill time and maximum temperature, and use finite differences to find the partial derivatives  $\frac{\partial f}{\partial x_i}$  and  $\frac{\partial c}{\partial x_i}$ . From there, the interior point algorithm [**interiorPoint**] will choose a new set of parameters, and repeat the process. This will continue until the first order optimality, that is, a measure of how close to a local minimum the objective function is [**firstOrderOpt**], reaches a given threshold.

General stuff:  
 - Check tense and person  
 - Check for repetition  
 - Check for excessive verbosity - define short and long cylinders at some point



## <sup>460</sup> **4 Results and Discussion**

{sec:results}

### <sup>461</sup> **4.1 Fitting to CFD data**

### <sup>462</sup> **4.2 Square wave inlet profile**

### <sup>463</sup> **4.3 Random inlet profile**

## <sup>464</sup> **5 Conclusion**

{sec:conclusion}

<sup>465</sup> This is the conclusion.

466 **Appendices**467 **A Filling Code**

{app:filling\_code}

```

1  function [x,f,eflag,output] = provaOpt(x0, profileFunction, lb, ub)
2
3  options = optimoptions('fmincon','Display','iter-detailed','Algorithm',...
4                      'interior-point','UseParallel',true,'Diagnostics',...
5                      'on', 'OutputFcn', @outfun);
6
7
8  % Set up shared variables with OUTFUN
9  history.x = [];
10 history.fval = [];
11 searchdir = [];
12
13
14 xLast = []; % Last place computeall was called
15 myf = []; % Use for objective at xLast
16 myc = []; % Use for nonlinear inequality constraint
17 myceq = []; % Use for nonlinear equality constraint
18
19 fun = @objfun; % the objective function, nested below
20 cfun = @constr; % the constraint function, nested below
21
22 A = [];
23 b = [];
24 Aeq = [];
25 beq = [];
26
27 % Call fmincon
28 [x,f,eflag,output] = fmincon(fun,x0,A,b,Aeq,beq,lb,ub,cfun,options);
29
30     function y = objfun(x)
31         sprintf('X = %2.16e ', x)
32         if ~isequal(x,xLast) % Check if computation is necessary
33             disp("Evaluating objective function")
34             [myf,myc,myceq] = MainRoutine(x, profileFunction);
35             xLast = x;
36         else
37             disp("Obtaining evaluated objective function")
38         end
39         % Now compute objective function
40         y = myf;
41     end
42
43     function [c,ceq] = constr(x)
44         sprintf('X = %2.16e ', x)
45         if ~isequal(x,xLast) % Check if computation is necessary
46             disp("Evaluating nonlinear constraints")
47             [myf,myc,myceq] = MainRoutine(x, profileFunction);
48             xLast = x;
49         else
50             disp("Obtaining evaluated nonlinear constraints")
51         end
52         % Now compute constraint functions
53         c = myc; % In this case, the computation is trivial
54         ceq = myceq;
55     end
56
57     function stop = outfun(x,optimValues,state)
58     stop = false;
59
60     switch state
61         case 'init'

```

```
62         hold on
63     case 'iter'
64         % Concatenate current point and objective function
65         % value with history. x must be a row vector.
66         history.fval = [history.fval; optimValues.fval];
67         history.x = [history.x; x];
68         % Concatenate current search direction with
69         % searchdir.
70         searchdir = [searchdir;...
71                     optimValues.searchdirection'];
72
73         plot(x(1),x(2),'o');
74         % Label points with iteration number.
75         % Add .15 to x(1) to separate label from plotted 'o'
76         text(x(1)+.15,x(2),num2str(optimValues.iteration));
77
78         plot(profileFunction(x, 2000),)
79
80     case 'done'
81         hold off
82     otherwise
83     end
84 end
85
86 end
```

## References

- [1] C. Dicken and W. Mérida, “Measured effects of filling time and initial mass on the temperature distribution within a hydrogen cylinder during refuelling”, *Journal of power sources*, vol. 165, no. 1, pp. 324–336, Feb. 2007, ISSN: 03787753. DOI: 10.1016/j.jpowsour.2006.11.077. [Online]. Available: [https://www.engineeringvillage.com/share/document.url?mid=cpx%7B%5C\\_%7D30c221110dfeaa71cM69f92061377553%7B%5C\\_%7Ddatabase=cpx](https://www.engineeringvillage.com/share/document.url?mid=cpx%7B%5C_%7D30c221110dfeaa71cM69f92061377553%7B%5C_%7Ddatabase=cpx).
- [2] J. Zheng, J. Guo, J. Yang, Y. Zhao, L. Zhao, X. Pan, J. Ma, and L. Zhang, “Experimental and numerical study on temperature rise within a 70 MPa type III cylinder during fast refueling”, *International journal of hydrogen energy*, vol. 38, no. 25, pp. 10956–10962, Aug. 2013, ISSN: 0360-3199. DOI: 10.1016/J.IJHYDENE.2013.02.053. [Online]. Available: [https://www.sciencedirect.com/science/article/pii/S0360319913004199?via%7B%5C\\_%7D3Dihub](https://www.sciencedirect.com/science/article/pii/S0360319913004199?via%7B%5C_%7D3Dihub).
- [3] P. L. Woodfield, M. Monde, and T. Takano, “Heat Transfer Characteristics for Practical Hydrogen Pressure Vessels Being Filled at High Pressure”, *Journal of thermal science and technology*, vol. 3, no. 2, pp. 241–253, 2008. DOI: 10.1299/jtst.3.241.
- [4] S. C. Kim, S. H. Lee, and K. B. Yoon, “Thermal characteristics during hydrogen fueling process of type IV cylinder”, *International journal of hydrogen energy*, vol. 35, no. 13, pp. 6830–6835, Jul. 2010, ISSN: 0360-3199. DOI: 10.1016/J.IJHYDENE.2010.03.130. [Online]. Available: [https://www.sciencedirect.com/science/article/pii/S0360319910006476?%7B%5C\\_%7Drdoc=1%7B%5C\\_%7D%7B%5C\\_%7Dfmt=high%7B%5C\\_%7D%7B%5C\\_%7Dorigin=gateway%7B%5C\\_%7D%7B%5C\\_%7Ddocanchor=%7B%5C\\_%7Dmd5=b8429449ccfc9c30159a5f9aeaa92ffb%7B%5C\\_%7Dccp=y](https://www.sciencedirect.com/science/article/pii/S0360319910006476?%7B%5C_%7Drdoc=1%7B%5C_%7D%7B%5C_%7Dfmt=high%7B%5C_%7D%7B%5C_%7Dorigin=gateway%7B%5C_%7D%7B%5C_%7Ddocanchor=%7B%5C_%7Dmd5=b8429449ccfc9c30159a5f9aeaa92ffb%7B%5C_%7Dccp=y).
- [5] B. Launder and D. Spalding, “The numerical computation of turbulent flows”, *Computer methods in applied mechanics and engineering*, vol. 3, no. 2, pp. 269–289, Mar. 1974, ISSN: 0045-7825. DOI: 10.1016/0045-7825(74)90029-2. [Online]. Available: [https://www.sciencedirect.com/science/article/pii/0045782574900292?via%7B%5C\\_%7D3Dihub](https://www.sciencedirect.com/science/article/pii/0045782574900292?via%7B%5C_%7D3Dihub).
- [6] O. Redlich and J. N. S. Kwong, “On the Thermodynamics of Solutions. V. An Equation of State. Fugacities of Gaseous Solutions.”, *Chemical reviews*, vol. 44, no. 1, pp. 233–244, Feb. 1949, ISSN: 0009-2665. DOI: 10.1021/cr60137a013. [Online]. Available: <https://pubs.acs.org/doi/abs/10.1021/cr60137a013>.
- [7] C. J. B. Dicken and W. Mérida, “Modeling the Transient Temperature Distribution within a Hydrogen Cylinder during Refueling”, *Numerical heat transfer, part a: Applications*, vol. 53, no. 7, pp. 685–708, Nov. 2007, ISSN: 1040-7782. DOI: 10.1080/10407780701634383. [Online]. Available: <http://www.tandfonline.com/doi/abs/10.1080/10407780701634383>

- [8] Y. Zhao, G. Liu, Y. Liu, J. Zheng, Y. Chen, L. Zhao, J. Guo, and Y. He, “Numerical study on fast filling of 70 MPa type III cylinder for hydrogen vehicle”, *International journal of hydrogen energy*, vol. 37, no. 22, pp. 17 517–17 522, 2012, ISSN: 0360-3199. DOI: <https://doi.org/10.1016/j.ijhydene.2012.03.046>. [Online]. Available: <http://www.sciencedirect.com/science/article/pii/S0360319912006477>.
- [9] E. W. Lemmon, M. L. Huber, and M. O. McLinden, *NIST Standard Reference Database 23: Reference Fluid Thermodynamic and Transport Properties-REFPROP, Version 9.1*, National Institute of Standards and Technology, 2013. DOI: <http://dx.doi.org/10.18434/T4JS3C>. [Online]. Available: <https://www.nist.gov/srd/refprop>.
- [10] M. Heitsch, D. Baraldi, and P. Moretto, “Numerical investigations on the fast filling of hydrogen tanks”, *International journal of hydrogen energy*, vol. 36, no. 3, pp. 2606–2612, Feb. 2011, ISSN: 0360-3199. DOI: 10.1016/J.IJHYDENE.2010.04.134. [Online]. Available: <https://www.sciencedirect.com/science/article/pii/S0360319910008360>.
- [11] F. Menter, “Two-equation eddy-viscosity turbulence models for engineering applications”, *Aiaa journal*, vol. 32, pp. 1598–1605, Aug. 1994. DOI: 10.2514/3.12149.
- [12] M. C. Galassi, D. Baraldi, B. Acosta Iborra, and P. Moretto, “CFD analysis of fast filling scenarios for 70 MPa hydrogen type IV tanks”, *International journal of hydrogen energy*, vol. 37, no. 8, pp. 6886–6892, Apr. 2012, ISSN: 0360-3199. DOI: 10.1016/J.IJHYDENE.2012.01.041. [Online]. Available: <https://www.sciencedirect.com/science/article/pii/S0360319912000912%7B%5C%7Dfig4>.
- [13] D. Melideo, D. Baraldi, M. C. Galassi, R. Ortiz Cebolla, B. Acosta Iborra, and P. Moretto, “CFD model performance benchmark of fast filling simulations of hydrogen tanks with pre-cooling”, *International journal of hydrogen energy*, vol. 39, no. 9, pp. 4389–4395, Mar. 2014, ISSN: 0360-3199. DOI: 10.1016/J.IJHYDENE.2013.12.196. [Online]. Available: <https://www.sciencedirect.com/science/article/pii/S0360319913031881>.
- [14] R. H. Aungier, “A Fast, Accurate Real Gas Equation of State for Fluid Dynamic Analysis Applications”, *Journal of fluids engineering*, vol. 117, no. 2, pp. 277–281, Jun. 1995, ISSN: 0098-2202. [Online]. Available: <http://dx.doi.org/10.1115/1.2817141>.
- [15] Y.-L. Liu, Y.-Z. Zhao, L. Zhao, X. Li, H.-g. Chen, L.-F. Zhang, H. Zhao, R.-H. Sheng, T. Xie, D.-H. Hu, and J.-Y. Zheng, “Experimental studies on temperature rise within a hydrogen cylinder during refueling”, *International journal of hydrogen energy*, vol. 35, no. 7, pp. 2627–2632, Apr. 2010, ISSN: 03603199. DOI: 10.1016/j.ijhydene.2009.04.042. [Online]. Available: [https://www.engineeringvillage.com/share/document.url?mid=cpx%7B%5C\\_%7D6e3d601283fcb129fM7f742061377553%7B%5C%7Ddatabase=cpx](https://www.engineeringvillage.com/share/document.url?mid=cpx%7B%5C_%7D6e3d601283fcb129fM7f742061377553%7B%5C%7Ddatabase=cpx).
- [16] M. Hosseini, I. Dincer, G. Naterer, and M. Rosen, “Thermodynamic analysis of filling compressed gaseous hydrogen storage tanks”, *International journal of hydrogen energy*, vol. 37, no. 6, pp. 5063–5071, Mar. 2012, ISSN: 0360-3199. DOI: 10.1016/J.IJHYDENE.2011.

12.047. [Online]. Available: <https://www.sciencedirect.com/science/article/pii/S0360319911026991?via%7B%5C%7D3Dihub>.

[17] K. J. Kountz, "Modeling the fast fill process in natural gas vehicle storage cylinders", *207th american chemical society meeting*, 1994.

[18] B. I. Lee and M. G. Kesler, "A generalized thermodynamic correlation based on three-parameter corresponding states", *Aiche journal*, vol. 21, no. 3, pp. 510–527, May 1975, ISSN: 0001-1541. DOI: 10.1002/aic.690210313. [Online]. Available: <http://doi.wiley.com/10.1002/aic.690210313>.

[19] M. Monde, Y. Mitsutake, P. L. Woodfield, and S. Maruyama, "Characteristics of heat transfer and temperature rise of hydrogen during rapid hydrogen filling at high pressure", *Heat transfer - asian research*, vol. 36, no. 1, pp. 13–27, Jan. 2007, ISSN: 10992871. DOI: 10.1002/htj.20140. [Online]. Available: <http://doi.wiley.com/10.1002/htj.20140>.

[20] M. Monde, P. Woodfield, T. Takano, and M. Kosaka, "Estimation of temperature change in practical hydrogen pressure tanks being filled at high pressures of 35 and 70 MPa", *International journal of hydrogen energy*, vol. 37, no. 7, pp. 5723–5734, 2012, ISSN: 0360-3199. DOI: <https://doi.org/10.1016/j.ijhydene.2011.12.136>. [Online]. Available: <http://www.sciencedirect.com/science/article/pii/S0360319911028709>.

[21] M. Striednig, S. Brandstätter, M. Sartory, and M. Klell, "Thermodynamic real gas analysis of a tank filling process", *International journal of hydrogen energy*, vol. 39, no. 16, pp. 8495–8509, May 2014, ISSN: 0360-3199. DOI: 10.1016/J.IJHYDENE.2014.03.028. [Online]. Available: <https://www.sciencedirect.com/science/article/pii/S036031991400682X?%7B%5C%7Drdoc=1%7B%5C%7D%7B%5C%7Dfmt=high%7B%5C%7D%7B%5C%7Dorigin=gateway%7B%5C%7D%7B%5C%7Ddocanchor=%7B%5C%7Dmd5=b8429449ccfc9c30159a5f9aeaa92ffb>.

[22] P. L. Woodfield and M. Monde, "A Thermodynamic Model for a High-Pressure Hydrogen Gas Filling System Comprised of Carbon-Fibre Reinforced Composite Pressure Vessels", 2010. [Online]. Available: <http://people.eng.unimelb.edu.au/imarusic/proceedings/17/198%7B%5C%7DPaper.pdf>.

[23] P. L. Woodfield, M. Monde, and Y. Mitsutake, "Measurement of Averaged Heat Transfer Coefficients in High-Pressure Vessel during Charging with Hydrogen, Nitrogen or Argon Gas", *Journal of thermal science and technology*, vol. 2, no. 2, pp. 180–191, 2007, ISSN: 1880-5566. DOI: 10.1299/jtst.2.180. [Online]. Available: <http://joit.jlc.jst.go.jp/JST.JSTAGE/jtst/2.180?from=CrossRef>.

[24] T. Johnson, R. Bozinoski, J. Ye, G. Sartor, J. Zheng, and J. Yang, "Thermal model development and validation for rapid filling of high pressure hydrogen tanks", *International journal of hydrogen energy*, vol. 40, no. 31, pp. 9803–9814, Aug. 2015, ISSN: 0360-3199. DOI: 10.1016/J.IJHYDENE.2015.05.157. [Online]. Available: <https://www.sciencedirect.com/science/article/pii/S0360319915013981>.

- [25] C. N. Ranong, S. Maus, J. Hapke, G. Fieg, D. Wenger, and C. N. Ranong, "Approach for the Determination of Heat Transfer Coefficients for Filling Processes of Pressure Vessels With Compressed Gaseous Media", *Heat transfer engineering*, vol. 32, no. 2, pp. 127–132, 2011, ISSN: 1521-0537. DOI: 10.1080/01457631003769187. [Online]. Available: <http://www.tandfonline.com/action/journalInformation?journalCode=uhte20%20https://doi.org/10.1080/01457631003769187>.
- [26] M. T. I. Khan, M. Monde, and T. Setoguchi, "Hydrogen gas filling into an actual tank at high pressure and optimization of its thermal characteristics", *Journal of thermal science*, vol. 18, no. 3, pp. 235–240, Sep. 2009, ISSN: 1993-033X. DOI: 10.1007/s11630-009-0235-x. [Online]. Available: <https://doi.org/10.1007/s11630-009-0235-x>.
- [27] F. P. Incropera, *Fundamentals of heat and mass transfer*, ser. Fundamentals of Heat and Mass Transfer v. 1. John Wiley, 2007, ISBN: 9780471457282. [Online]. Available: [https://books.google.co.uk/books?id=%7B%5C\\_%7DP9QAAAAMAAJ](https://books.google.co.uk/books?id=%7B%5C_%7DP9QAAAAMAAJ).
- [28] S. Kakaç, R. K. Shah, and W. Aung, *Handbook of single-phase convective heat transfer*, ser. A Wiley Interscience publication. Wiley, 1987, ISBN: 9780471817024. [Online]. Available: <https://books.google.co.uk/books?id=FCJRAAAAMAAJ>.
- [29] S. B. Pope, *Turbulent Flows*. Cambridge University Press, 2000, ISBN: 9780521598866. [Online]. Available: <https://books.google.co.uk/books?id=HZsTw9SMx-0C>.
- [30] H. J. Hussein, S. P. Capp, and W. K. George, "Velocity measurements in a high-Reynolds-number, momentum-conserving, axisymmetric, turbulent jet", *Journal of fluid mechanics*, vol. 258, no. -1, p. 31, Jan. 1994, ISSN: 0022-1120. DOI: 10.1017/S002211209400323X. [Online]. Available: [http://www.journals.cambridge.org/abstract%7B%5C\\_%7DS002211209400323X](http://www.journals.cambridge.org/abstract%7B%5C_%7DS002211209400323X).
- [31] P. J. Roache, *Computational fluid dynamics*, ser. IAHR Monograph. Hermosa Publishers, 1976. [Online]. Available: [https://books.google.co.uk/books?id=5%7B%5C\\_%7DNQAAAAMAAJ](https://books.google.co.uk/books?id=5%7B%5C_%7DNQAAAAMAAJ).
- [32] G. D. Smith, *Numerical solution of partial differential equations: with exercises and worked solutions*, ser. Oxford mathematical handbooks. Oxford University Press, 1965. [Online]. Available: <https://books.google.co.uk/books?id=wnKNq67GYrEC>.