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OCR A Level Physics



Density & Pressure

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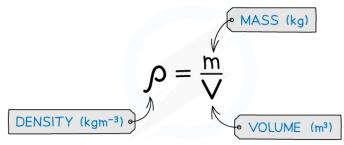
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Density

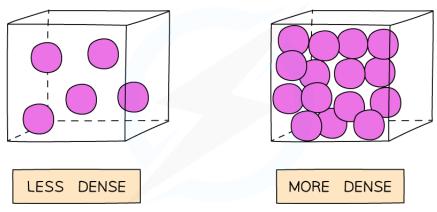
Your notes

Density

- Density is the **mass per unit volume** of an object
 - Objects made from low-density materials typically have a lower mass
 - For example, a balloon is less dense than a small bar of lead despite occupying a larger volume
- The units of density depend on the units used for mass and volume:
 - If the mass is measured in g and volume in cm³, then the density will be in g / cm³
 - If the mass is measured in kg and volume in m^3 , then the density will be in kg/m^3



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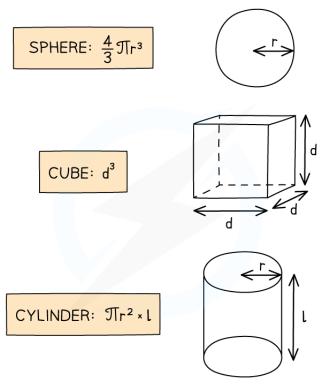
Gases are less dense than a solid



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• The volume of an object may not always be given directly, but can be calculated with the appropriate equation depending on the object's shape





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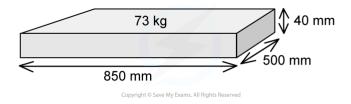
Volumes of common 3D shapes



Worked Example

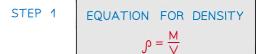
A paving slab has a mass of 73 kg and dimensions 40 mm \times 500 mm \times 850 mm.

Calculate the density, in $kg \, m^{-3}$ of the material from which the paving slab is made.



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Answer:



STEP 2 CALCULATE THE VOLUME
$$V = 40 \text{ mm} \times 500 \text{ mm} \times 850 \text{ mm} = 1.7 \times 10^7 \text{ mm}^3$$

STEP 3 CONVERT mm³ TO m³

$$1 \text{ mm} = 0.001 \text{ m} = 1 \times 10^{-3} \text{ m}$$

$$1 \text{ mm}^3 = (0.001)^3 \text{ m}^3 = (1 \times 10^{-3})^3 \text{ m}^3 = 1 \times 10^{-9} \text{ m}^3$$

$$V = 1.7 \times 10^7 \times 1 \times 10^{-9} = 1.7 \times 10^{-2} \text{ m}$$

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Examiner Tips and Tricks

- When converting a larger unit to a smaller one, you multiply (x)
 - E.g. $125 \text{ m} = 125 \times 100 = 12500 \text{ cm}$
- When you convert a **smaller** unit to a **larger** one, you **divide** (÷)
 - E.g. $5g = 5/1000 = 0.005 \text{ or } 5 \times 10^{-3} \text{ kg}$
- When dealing with squared or cubic conversions, cube or square the conversion factor too
 - E.g. $1 \text{ mm}^3 = 1/(1000)^3 = 1 \times 10^{-9} \text{ m}^3$
 - E.g. $1 \text{ cm}^3 = 1/(100)^3 = 1 \times 10^{-6} \text{ m}^3$

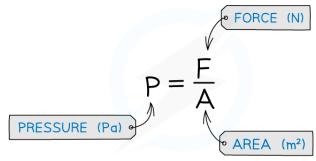


Pressure

Your notes

Pressure

• Pressure tells us how concentrated a force is, it is defined as the **force per unit area**



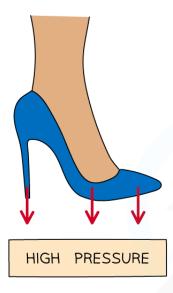
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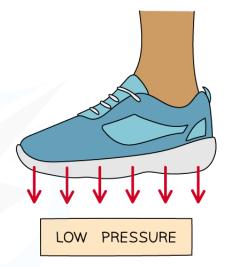
- This equation shows:
 - If a force is spread over a large area it will result in a small pressure
 - If it is spread over a small area it will result in a large pressure



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WEIGHT FROM HEELED SHOES IS SPREAD OVER A **SMALLER** AREA

THIS EXERTS A **HIGHER** PRESSURE ON THE GROUND

WEIGHT FROM FLAT SHOES IS SPREAD OVER A LARGER AREA

THIS EXERTS A **LOWER** PRESSURE ON THE GROUND

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Different pressure is exerted for the same force on different areas

- The units of pressure depend on the units of area:
 - If the area is measured in cm^2 (and the force in N), then the pressure will be in N/ cm^2
 - If the area is measured in m^2 (and the force in N), then the pressure will be in N/m^2
- Pressure can also be measured in pascals, Pa where 1 Pa is the same as 1 N/m²
- Pressure, unlike force, is a **scalar** quantity
 - Therefore pressure does not have a specific direction

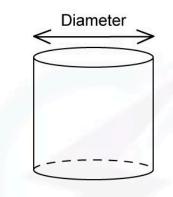


Worked Example

A cylinder is placed on a horizontal surface as shown below



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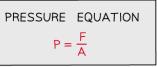




The mass of the cylinder is 4.7 kg and the diameter is 8.4 cm. Calculate the pressure produced by the cylinder on the surface in Pa.

Answer:

STEP 1



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STEP 2

CALCULATE THE FORCE PRODUCED BY THE CYLINDER

ONLY FORCE IS THE WEIGHT

$$F = mg = 4.7 \, kg \times 9.81 \, ms^{-2} = 46.1 \, N$$

STEP 3

CALCULATE THE AREA

THE FORCE IS APPLIED ON THE BASE OF THE CYLINFER WHICH IS A CIRCLE

AREA OF A CIRCLE =
$$\pi \times r^2 = \pi \left(\frac{d}{2}\right)^2 = \pi \left(\frac{8.4 \times 10^{-2} \, \text{m}}{2}\right)^2 = 5.5 \times 10^{-3} \, \text{m}^2$$

STEP 4

SUBSTITUTE BACK INTO PRESSURE EQUATION

$$P = \frac{46.1 \,\text{N}}{5.5 \times 10^{-3} \,\text{m}^2} = 8400 \,\text{Pa} \, (2 \,\text{s.f})$$

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Examiner Tips and Tricks

The area referred to is the 'cross-sectional' area of a 3D object. This is the area of the base that the force is applied on. For a cylinder, this will be a circle.





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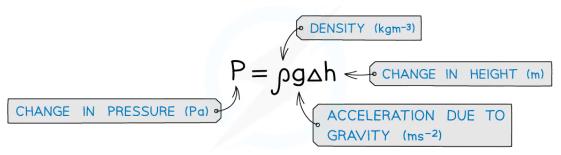
Archimedes' Principle

Your notes

Archimedes' Principle

Upthrust on an Object in a Fluid

- Pressure increases with depth in a fluid because of the force exerted by the increased weight of the fluid above
- This change in pressure can be calculated using the equation of hydrostatic pressure:



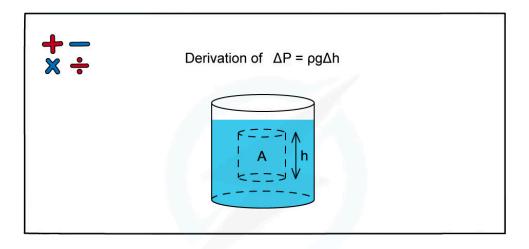
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This equation can be derived in the following way:



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THE PRESSURE ON AREA ${\bf A}$ IS DUE TO THE WEIGHT OF THE COLUMN OF WATER ${\bf h}$ ABOVE IT

$$P = \frac{F}{A}$$

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Your notes

```
FORCE IS THE WEIGHT OF THE COLUMN W = mg
MASS OF THE COLUMN IS FROM <math display="block">m = \rho * V
WHERE \qquad V = A * h
SUBSTITUTE VOLUME, MASS EQUALS
m = \rho * A * h
WEIGHT OF THE COLUMN IS NOW
W = \rho * A * h * g
```

SUBSTITUTE BACK INTO PRESSURE EQUATION
$$P = \frac{\int P \cdot A \cdot P \cdot dx}{A}$$

$$P = \int P \cdot P \cdot dx$$

MORE SPECIFICALLY, THE CHANGE IN PRESSURE P IS PROPORTIONAL TO THE CHANGE IN HEIGHT hEQUATION FOR HYDROSTATIC PRESSURE $\Delta P = \rho g \Delta h$

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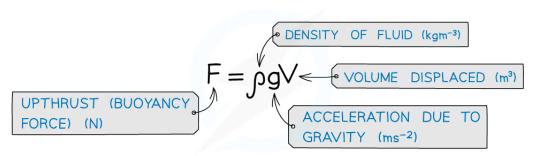
Archimedes' principle states:

An object submerged in a fluid at rest has an upward buoyancy force (upthrust) equal to the weight of the fluid displaced by the object

- The object sinks until the weight of the fluid displaced is equal to its own weight
 - Therefore the object floats when the magnitude of the upthrust equals the weight of the object
- The magnitude of upthrust can be calculated by:



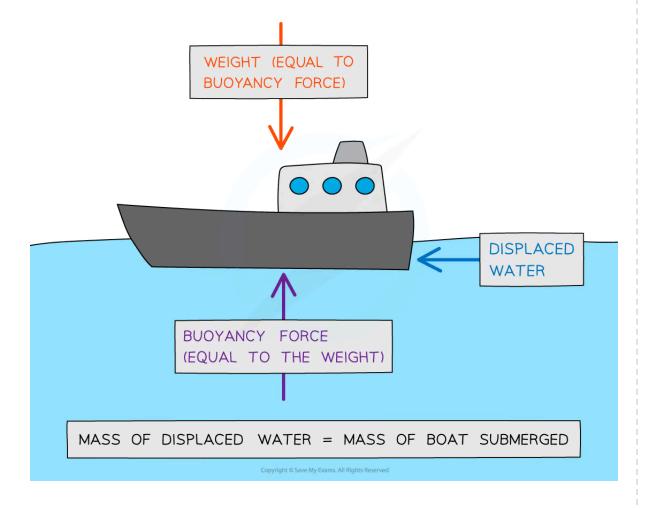
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- Since $m = \rho V$, upthrust is equal to F = mg which is the weight of the fluid displaced by the object
- Archimedes' Principle explains how ships float:



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Boats float because they displace an amount of water that is equal to their weight





Worked Example

Atmospheric pressure at sea level has a value of 100 kPa. The density of sea water is 1020 kg m $^{-3}$. At what depth in the sea would the total pressure be 250 kPa?

- **A.** 20 m
- **B.** 9.5 m
- **C.** 18 m
- **D.** 15 m

ANSWER: D

STEP 1

THE TOTAL PRESSURE = HYDROSTATIC + ATMOSPHERIC PRESSURE

$$250 \times 10^{3} = \rho g h + 100 \times 10^{3}$$

$$250 \times 10^{3} = 1020 \times 9.81 \times h + 100 \times 10^{3}$$

STEP 2 REA

REARRANGING FOR HEIGHT h
$$h = \frac{250 \times 10^{3} - 100 \times 10^{3}}{1020 \times 9.81} = 15 \text{ m}$$

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Worked Example

Icebergs typically float with a large volume of ice beneath the water. Ice has a density of 917 kg m $^{-3}$ and a volume of V_i . The density of seawater is 1020 kg m $^{-3}$. What fraction of the iceberg is above the water?

- $\textbf{A.} \ 0.10 \ V_i \qquad \textbf{B.} \ 0.90 \ V_i \qquad \textbf{C.} \ 0.97 \ V_i$
- **D.** 0.20 V_i



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ANSWER: A

STEP 1

ACCORDING TO ARCHIMEDES' PRINCIPLE, THE UPTHRUST IS EQUAL TO THE WEIGHT OF THE SEAWATER DISPLACED BY THE ICEBERG

ICEBERG WEIGHT Wi = mig

BUOYANCY FORCE IS THE

 $W_w = m_w g$

WEIGHT OF THE DISPLACED WATER

STEP 2

SINCE THE ICEBERG IS FLOATING, ITS WEIGHT IS EXACTLY EQUAL TO THE BUOYANCY FORCE

 $W_i = W_w$

 $m_i g = m_w g$

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Your notes

STEP 3

REARRANGE DENSITY EQUATION FOR MASS $m = \rho V$ $\rho_i V_i g = \rho_w V_w g$

STEP 4

CANCELLING g SHOWS THE FRACTION OF ICE UNDERWATER IS GIVEN BY RATIO OF DENSITIES

$$\frac{\bigvee_{i}}{\bigvee_{w}} = \frac{\int_{i}^{w}}{\int_{i}^{i}}$$

STEP 5

REARRANGE FOR
$$V_w$$

$$V_w = \frac{\rho_i V_i}{\rho_w}$$

$$V_w = \frac{917}{1020} V_i = 0.9 V_i$$

STEP 6

THEREFORE 90% OF THE ICEBERG'S VOLUME IS SUBMERGED UNDERWATER

THIS MEANS THAT $1-0.9 = 0.1 \, \text{V}_1$ IS ABOVE WATER

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Examiner Tips and Tricks

When asked about the **total pressure** remember to also add the atmospheric pressure

Total pressure = Hydrostatic pressure + Atmospheric pressure

Atmospheric pressure (also known as barometric pressure) is equal to **101 325 Pa** Values for pressure can vary widely and depend on metric prefixes such as kPa or MPa. When you're doing calculations make sure all the pressures are in the same units (otherwise you may be out by a factor of 1000!). To be on the safe side, you can convert them all to Pascals.