${\bf 1.1 \ Exercise \ 3 - electronic \ configuration}$

Write the electronic configuration of the following using the arrow and box method:

1. C
2. Cu
$3. Mg^+$
Write the electronic configuration of the following using the orbital method:
4. N^{3-}
5. Ar
6. Sc^{3+}
7. Mn ²⁺
8. Fe^{3+}
9. V^{3+}
Write the electronic configuration of the following using the shorthand arrow and bo method:
10. Cl ⁻
11. Fe
12. Br
Write the electronic configuration of the following using the shorthand orbital method:
13. Cr
14. Ga ³⁺
15. Pb ²⁺