### **Outline**

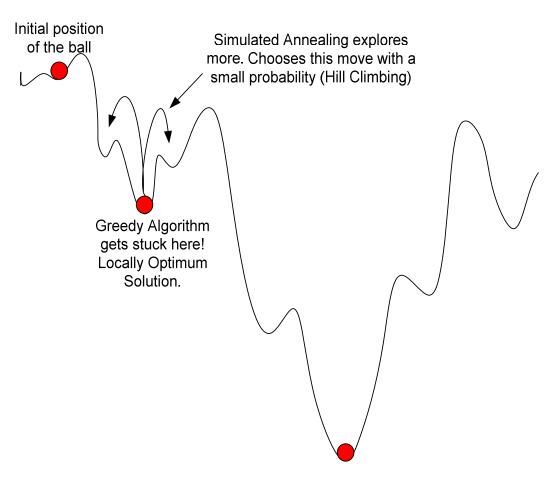
- Introduction
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# Introduction

# What Is Simulated Annealing?

- Simulated Annealing (SA)
  - SA is applied to solve optimization problems
  - SA is a stochastic algorithm
  - SA is escaping from local optima by allowing worsening moves
  - SA is a memoryless algorithm, the algorithm does not use any information gathered during the search
  - SA is applied for both combinatorial and continuous optimization problems
  - SA is simple and easy to implement.
  - SA is motivated by the physical annealing process

# SA vs Greedy Algorithms: Ball on terrain example



Upon a large no. of iterations, SA converges to this solution.

# History

• Numerical simulation of annealing, Metropolis et al. 1953.

N. Metropolis, A. Rosenbluth, M. Rosenbluth, A. Teller, and E. Teller. Equation of state calculations by fast computing machines. Journal of Chemical Physics, 21:1087–1092, 1953.

# History

- SA for combinatorial problems
  - Kirkpatrick et. al, 1986
  - Cerny, 1985
  - S. Kirkpatrick, C. D. Gelatt, and M. P. Vecchi. Optimization by simulated annealing. *Science*, 220(4598):671–680, 1983.
  - V. Cerny, Thermodynamical approach to the traveling salesman problem : an efficient simulation algorithm. J. of Optimization Theory and Applications, 45(1):41–51, 1985.

# **History**

- Originally, the use of simulated annealing in combinatorial optimization
- In the 1980s, SA had a major impact on the field of heuristic search for its simplicity and efficiency in solving combinatorial optimization problems.
- Then, it has been extended to deal with continuous optimization problems
- SA was inspired by an analogy between the physical annealing process of solids and the problem of solving large combinatorial optimization problems.

# **Applications**

## • Basic problems

- Traveling Salesman Problem
- Graph partitioning
- Matching prob.
- Quadratic Assignment
- Linear Arrangement
- Scheduling

- ....

# **Applications**

## • Engineering problem

- VLSI: Placement, routing...
- Facilities layout
- Image processing
- Code design
- Biology
- Physics
- ....

# Real Annealing and Simulated Annealing

# **Real Annealing Technique**

- Annealing Technique is known as a thermal process for obtaining low-energy state of a solid in a heat bath.
- The process consists of the following two steps:
  - Increasing temperature: Increase the temperature of the heat bath to a maximum value at which the solid melts.
  - Decreasing temperature: Decrease carefully the temperature of the heat bath until the particles arrange themselves in the ground state of the solid.

# **Real Annealing Technique**

- In the **liquid phase** all **particles** arrange themselves randomly, whereas in the ground state of the solid, the particles are arranged in a highly structured lattice, for which the corresponding energy is minimal.
- The ground state of the solid is obtained only if:
  - the maximum value of the temperature is sufficiently high and
  - the cooling is done sufficiently slow.
- Strong solid are grown from careful and slow cooling.

# **Real Annealing Technique**

### Metastable states

 If the initial temperature is not sufficiently high or a fast cooling is applied, metastable states (imperfections) are obtained.

### Quenching

The process that leads to metastable states is called quenching

### • Thermal equilibrium

 If the lowering of the temperature is done sufficiently slow, the solid can reach thermal equilibrium at each temperature.

# Real Annealing and Simulated Annealing

• The analogy between the physical system and the optimization problem.

Physical System		Optimization Problem
System state		Solution
Molecular positions	<u> </u>	Decision variables
Energy		Objective function
Minimizing energy		Minimizing cost
Ground state		Global optimal solution
Metastable state		Local optimum
Quenching	<u></u>	⇒ Local search
Temperature	<u> </u>	Control parameter T
Real annealing	<u> </u>	⇒ Simulated annealing

# Real Annealing and Simulated Annealing

- The objective function of the problem is analogous to the energy state of the system.
- A solution of the optimization problem corresponds to a system state.
- The decision variables associated with a solution of the problem are analogous to the molecular positions.
- The global optimum corresponds to the ground state of the system.
- Finding a local minimum implies that a metastable state has been reached.

- In 1958 Metropolis et al. introduced a simple algorithm for simulating the evolution of a solid in a heat bath to **thermal equilibrium**.
- Their algorithm is based on Monte Carlo techniques, and generates a sequence of states of the solid in the following way.
- Given a current state i of the solid with energy  $E_i$ , a subsequent state j is generated by applying a **perturbation mechanism** that transforms the current state into a next state by a small distortion, for instance, by a displacement of a single particle.

- The energy of the next state is  $E_j$ . (Ej)
- If the energy difference,  $E_j E_i$ , is less than or equal to 0, the state j is accepted as the current state.
- If the energy difference is greater than 0, then state *j* is accepted with a probability given by

$$\exp\left(\frac{E_i - E_j}{k_B T}\right)$$

- where T denotes the temperature of the heat bath and
- k<sub>B</sub> a physical constant known as the Boltzmann constant.

- The acceptance rule described above is known as the Metropolis criterion (Metropolis rule) and the algorithm that goes with it is known as the Metropolis algorithm.
- In the Metropolis algorithm thermal equilibrium is achieved by generating a large number of transitions at a given temperature value.

- On the basis of a given, the system is subjected to an elementary,
- if this modification causes a decrease in the objective function of the system, it is accepted;
- if it causes an increase  $\Delta E$  of the objective function, it is also accepted, but with a probability

$$e^{\frac{-\Delta E}{T}}$$

- By repeatedly observing this Metropolis rule of acceptance, a sequence of configurations is generated
- With this formalism in place, it is possible to show that, when the chain is of infinite length (in practical consideration, of "sufficient" length. . . ), the system can reach (in practical consideration, can approach) thermodynamic balance (Equilibrium State) at the temperature considered

- At high temperature,  $e^{\frac{-\Delta E}{T}}$  is close to 1,
  - therefore the majority of the moves are accepted and the algorithm becomes equivalent to a simple random walk in the configuration space.
- At low temperature,  $e^{\frac{-\Delta E}{T}}$  is close to 0,
  - therefore the majority of the moves increasing energy is refused.
- At an intermediate temperature,
  - the algorithm intermittently authorizes the transformations that degrade the objective function

# **Template of SA**

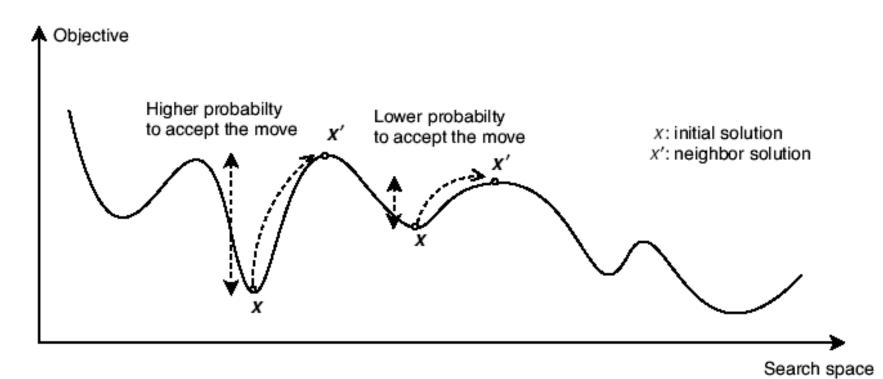
• SA can be viewed as a sequence of Metropolis algorithms, evaluated at decreasing values of the temperature.

- From an initial solution, SA proceeds in several iterations.
- At each iteration, a random neighbor is generated.
- Moves that improve the cost function are always accepted.
- Otherwise, the neighbor is selected with a given probability that depends on the current temperature and the amount of degradation  $\Delta E$  of the objective function.
- ΔE represents the difference in the objective value (energy) between the current solution and the generated neighboring solution.

- The higher the temperature, the more significant the probability of accepting a worst move.
- At a given temperature, the lower the increase of the objective function, the more significant the probability of accepting the move.

# **Template of SA**

• As the algorithm progresses, the probability that such moves are accepted decreases.

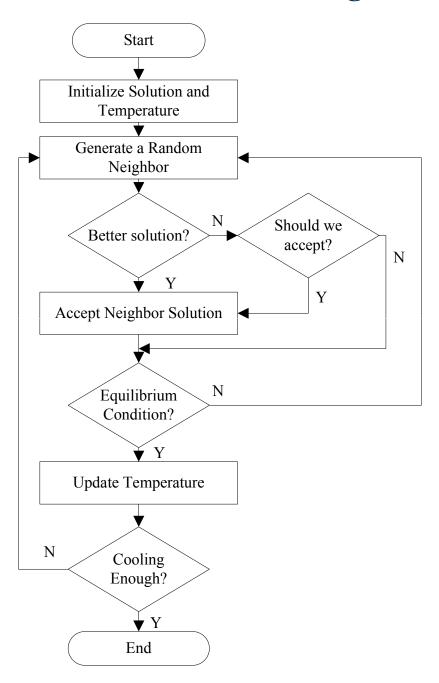


# **Template of SA**

• The acceptance probability function, in general, the Boltzmann distribution:

$$P(\Delta E, T) = e^{-\frac{f(s') - f(s)}{T}}$$

- It uses a control parameter, called temperature, to determine the probability of accepting nonimproving solutions.
- At a particular level of temperature, many trials are explored.
- Once an equilibrium state is reached, the temperature is gradually decreased according to a cooling schedule such that few nonimproving solutions are accepted at the end of the search.

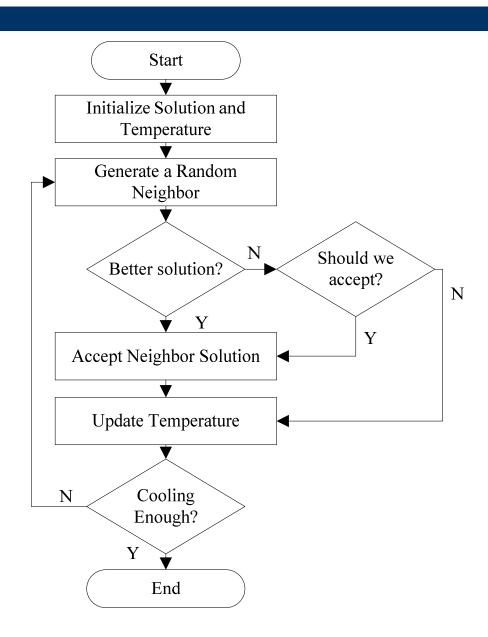


```
Input: Cooling schedule.
s = s_0; /* Generation of the initial solution */
T = T_{max}; /* Starting temperature */
Repeat
   Repeat /* At a fixed temperature */
     Generate a random neighbor s';
     \Delta E = f(s') - f(s);
     If \Delta E \leq 0 Then s = s' /* Accept the neighbor solution */
     Else Accept s' with a probability e^{\frac{-\Delta E}{T}};
   Until Equilibrium condition
   /* e.g. a given number of iterations executed at each temperature T^*/
   T = g(T); /* Temperature update */
Until Stopping criteria satisfied /* e.g. T < T_{min} */
Output: Best solution found.
```

## Inhomogeneous vs. Homogeneous Algorithm

- SA has two variants:
  - Homogeneous variant
    - Previous algorithm is the homogeneous variant
    - T is kept constant in the inner loop and is only decreased in the outer loop
  - Inhomogeneous variant
    - There is only one loop
    - T is decreased each time in the loop, but only very slightly

# **Inhomogeneous variant**



# Inhomogeneous variant

```
Input: Cooling schedule. s = s_0; /* Generation of the initial solution */ T = T_{max}; /* Starting temperature */
Repeat
Generate a random neighbor s'; \Delta E = f(s') - f(s);
If \Delta E \leq 0 Then s = s' /* Accept the neighbor solution */
Else Accept s' with a probability e^{\frac{-\Delta E}{T}}; T = g(T); /* Temperature update */
Until Stopping criteria satisfied /* e.g. T < T_{min} */
Output: Best solution found.
```

# **Cooling Schedule**

- The cooling schedule defines for each step of the algorithm i the temperature  $T_i$ .
- It has a great impact on the success of the SA optimization algorithm.
- The parameters to consider in defining a cooling schedule are:
  - the starting temperature,
  - the equilibrium state,
  - a cooling function, and
  - the final temperature that defines the stopping criteria

- Main components of SA:
  - Acceptance Function
  - Initial Temperature
  - Equilibrium State
  - Cooling Function
  - Stopping Condition

# A Simple Example

# A Simple Example

- Let us maximize the continuous function  $f(x) = x^3 60x^2 + 900x + 100$ .
- A solution x is represented as a string of 5 bits.
- The neighborhood consists in flipping randomly a bit.
- The initial solution is 10011 (x = 19, f(x) = 2399)
- Testing two sceneries:
  - First scenario: initial temperature  $T_0$  equal to 500.
  - Second scenario: initial temperature  $T_0$  equal to 100.
- Cooling: T = 0.9 . T

# A Simple Example

- In addition to the current solution, the best solution found since the beginning of the search is stored.
- Few parameters control the progress of the search, which are:
  - The temperature
  - The number of iterations performed at each temperature

# A Simple Example

• First Scenario T = 500 and Initial Solution (10011)

T	Move	Solution	f	$\Delta f$	Move?	New Neighbor Solution
500	1	00011	2287	112	Yes	00011
450	3	00111	3803	<0	Yes	00111
405	5	00110	3556	247	Yes	00110
364.5	2	01110	3684	<0	Yes	01110
328	4	01100	3998	<0	Yes	01100
295.2	3	01000	3972	16	Yes	01000
265.7	4	01010	4100	<0	Yes	01010
239.1	5	01011	4071	29	Yes	01011
215.2	1	11011	343	3728	No	01011

# A Simple Example

• Second Scenario: T = 100 and Initial Solution (10011).

$\overline{T}$	Move	Solution	f	$\Delta f$	Move?	New Neighbor Solution
100	1	00011	2287	112	No	10011
90	3	10111	1227	1172	No	10011
81	5	10010	2692	< 0	Yes	10010
72.9	2	11010	516	2176	No	10010
65.6	4	10000	3236	< 0	Yes	10000
59	3	10100	2100	1136	Yes	10000

• When Temperature is not High Enough, Algorithm Gets Stuck

# References

### References

- El-Ghazali Talbi, Metaheuristics: From Design to Implementation, John Wiley & Sons, 2009.
- J. Dreo A. Petrowski, P. Siarry E. Taillard, **Metaheuristics** for Hard Optimization, Springer-Verlag, 2006.

# The End