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ABSTRACT

The water molecule occurs in two nuclear-spin isomers that differ by the value of the total nuclear spin of the hydrogen atoms, i.e., $I = 0$ for *para*-H₂O and $I = 1$ for *ortho*-H₂O. Spectroscopic transitions between rovibrational states of *ortho* and *para* water are extremely weak due to the tiny hyperfine nuclear-spin-rotation interaction of only ~ 30 kHz and, so far, have not been observed. We report the first comprehensive theoretical investigation of the hyperfine effects and *ortho*-*para* transitions in H₂¹⁶O due to nuclear-spin-rotation and spin–spin interactions. We also present the details of our newly developed general variational approach to the simulation of hyperfine effects in polyatomic molecules. Our results for water suggest that the strongest *ortho*-*para* transitions with room-temperature intensities on the order of 10⁻³¹ cm/molecule are about an order of magnitude larger than previously predicted values and should be detectable in the mid-infrared v_2 and near-infrared $2v_1 + v_2$ and $v_1 + v_2 + v_3$ bands by current spectroscopy experiments.

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I. INTRODUCTION

Water is the third most abundant molecule in the universe. It is also quite unique in that it possesses a wide range of anomalous properties, some of which may be a result of nuclear spin symmetry breaking. It has two nuclear spin isomers, *ortho*, with a total nuclear spin of hydrogen atoms $I = 1$, and *para*, with a total nuclear spin of hydrogens $I = 0$. In isolated-molecule conditions, the *ortho* and *para* nuclear spin isomers show tremendously long-lasting stability to inter-conversion,^{1,2} can be spatially separated,^{3,4} and exhibit distinct physical and chemical properties.^{5,6} Thus, the nuclear spin isomers of water are frequently treated as distinct molecular species.

The concept of stable nuclear spin isomers is appealing to astrophysicists, as it allows us to deduce temperatures, below 50 K, in cometary comae and star- and planet-forming regions from the observations of relative abundance of *ortho* and *para* species.^{7–11} Some astronomical observations, however, reported anomalous *ortho*-*para* ratios (OPR), corresponding to spin temperatures that are much lower than gas kinetic temperatures in the same

region.^{12–15} These observations pose the intriguing question if the OPR values could be altered as a result of internal *ortho*-*para* conversion, which can possibly be enhanced by natural factors, such as molecular collisions,^{16–18} interaction with catalytic surfaces,¹⁹ external fields,²⁰ and radiation.²¹ Low nuclear-spin temperatures have been attributed to the photodesorption of water from colder icy grains.²² However, this theory was benchmarked and disputed in a number of recent laboratory experiments.^{23–26} Arguably, there could be another yet unknown mechanism of spin-non-destructive desorption of water molecules from ice.

The OPR values can change as a result of the interaction between the nuclear spins and an induced internal magnetic field of the rotating molecule, which is called the nuclear spin-rotation interaction. For the main water isotopologue H₂¹⁶O, considered here, the ¹⁶O has zero nuclear spin, and the hyperfine coupling between the spins of the protons is very weak, providing a fundamental rationale for neglecting the *ortho*-*para* conversion in practical applications. However, it can be significantly enhanced by accidental resonances between the *ortho* and *para* states, which

are present in vibrationally excited bands of isolated water. Their coupling can be amplified by external effects such as molecular collisions and interactions with strong external fields and field gradients. The accurate modeling of these processes may unravel previously unknown mechanisms contributing to the observed anomalous OPR of water in space. Precise knowledge of the molecular hyperfine states and corresponding transitions is mandatory for the understanding of such conversion mechanisms. This information can also be important for cold-molecule precision spectroscopy relying on controlled hyperfine transitions and hyperfine-state changing collisions.²⁷

Here, we report a complete line list of rovibrational hyperfine transitions in H₂¹⁶O at room-temperature, which we computed using an accurate variational approach^{28–31} with an empirically refined potential energy surface (PES)³² and a high-level *ab initio* spin-rotation tensor surface. The spin–spin coupling was modeled as the magnetic dipole–dipole interaction between the two hydrogen nuclei. We show that the strongest forbidden *ortho–para* transitions are on the order of 10⁻³¹ cm/molecule, which is about ten times stronger than previously reported calculations for the same lines.² We also present the details of our variational approach for computing hyperfine effects, which is general and not restricted by the numbers and specific magnitudes of the molecules' nuclear spins.

II. THEORETICAL DETAILS

A. Spin-rotation and spin–spin coupling

In this section, we describe the implementation of the nuclear spin-rotation and spin–spin coupling terms within the general variational framework of the nuclear motion approach TROVE.^{28–31} Implementation details of the hyperfine nuclear quadrupole coupling can be found in our previous studies.^{33,34}

The spin-rotation coupling is the interaction between the rotational angular momentum \mathbf{J} of the molecule and the nuclear spins \mathbf{I}_n of different nuclei,³⁵

$$H_{\text{sr}} = \sum_n^{N_I} \mathbf{I}_n \cdot \mathbf{M}_n \cdot \mathbf{J}, \quad (1)$$

where \mathbf{M}_n is the second-rank spin-rotation tensor relative to the nucleus n and the sum runs over all nuclei N_I with non-zero spin. The interaction between the nuclear spins \mathbf{I}_n of different nuclei is given by the spin–spin coupling as

$$H_{\text{ss}} = \sum_{n>n'}^{N_I} \mathbf{I}_n \cdot \mathbf{D}_{n,n'} \cdot \mathbf{I}_{n'}, \quad (2)$$

where $\mathbf{D}_{n,n'}$ is the second-rank spin–spin tensor, which is traceless and symmetric. Using the spherical-tensor representation,³⁶ the spin-rotation and spin–spin Hamiltonians can be expressed as

$$H_{\text{sr}} = \frac{1}{2} \sum_n^{N_I} \sum_{\omega=0}^2 \sqrt{2\omega+1} \left(-\frac{1}{\sqrt{3}} \right) \mathbf{I}_n^{(1)} \cdot \left((-1)^\omega [\mathbf{M}_n^{(\omega)} \otimes \mathbf{J}^{(1)}]^{(1)} + [\mathbf{J}^{(1)} \otimes \mathbf{M}_n^{(\omega)}]^{(1)} \right) \quad (3)$$

and

$$H_{\text{ss}} = \sum_{n>n'}^{N_I} \mathbf{D}_{n,n'}^{(2)} \cdot [\mathbf{I}_n^{(1)} \otimes \mathbf{I}_{n'}^{(1)}]^{(2)}, \quad (4)$$

where $\mathbf{M}_n^{(\omega)}$, $\mathbf{D}_{n,n'}^{(2)}$, $\mathbf{J}^{(1)}$, and $\mathbf{I}_n^{(1)}$ denote the spherical-tensor representations of operators in (1) and (2) and the square brackets are used to indicate the tensor product of two spherical-tensor operators. Because the spin-rotation tensor is generally not symmetric, the second term in the sum (3) is added to ensure that the Hamiltonian is Hermitian.

The nuclear-spin operator \mathbf{I}_n and the rotational-angular-momentum operator \mathbf{J} are coupled using a *nearly equal* coupling scheme, i.e., $\mathbf{I}_{1,2} = \mathbf{I}_1 + \mathbf{I}_2$, $\mathbf{I}_{1,3} = \mathbf{I}_{1,2} + \mathbf{I}_3$, ..., $\mathbf{I} \equiv \mathbf{I}_{1,N} = \mathbf{I}_{1,N-1} + \mathbf{I}_N$, and $\mathbf{F} = \mathbf{J} + \mathbf{I}$. The nuclear-spin functions $|I, m_I, \mathcal{I}\rangle$ depend on the quantum numbers I and m_I of the collective nuclear-spin operator \mathbf{I} and its projection onto the laboratory Z axis, respectively. The set of auxiliary quantum numbers $\mathcal{I} = \{I_1, I_{1,2}, I_{1,3}, \dots, I_{1,N-1}\}$ for the intermediate spin angular momentum operators provide a unique assignment of each nuclear-spin state. The total spin-rovibrational wave functions $|F, m_F, u\rangle$ are built as symmetry-adapted linear combinations of the coupled products of the rovibrational wave functions $|J, m_J, l\rangle$ and the nuclear-spin functions $|I, m_I, \mathcal{I}\rangle$. Here, J and F are the quantum numbers of \mathbf{J} and \mathbf{F} operators with m_J and m_F of their Z -axis projections. l and u denote the rovibrational and hyperfine state indices, respectively, and embrace all quantum numbers, e.g., rotational k and vibrational quantum numbers v_1, v_2, \dots , that are necessary to characterize a nuclear spin-rovibrational state.

The symmetrization postulate requires the total wavefunction of the H₂O molecule to change sign upon exchange of the protons, i.e., to transform as one of the irreducible representations B_1, B_2 of its C_{2v}(M) symmetry group. Accordingly, the *ortho* spin state $|I = 1\rangle$ of the A_1 symmetry can be coupled with the rovibrational states of B_1 and B_2 symmetries and the *para* state $|I = 0\rangle$ of the B_2 symmetry can be coupled with the rovibrational states of A_1 and A_2 symmetries.

The matrix representations of the spin-rotation and spin–spin Hamiltonians in the basis of the $|F, m_F, u\rangle$ functions are diagonal in F and m_F , with the explicit expressions given by

$$\begin{aligned} & \langle F, m_F, u' | H_{\text{sr}} | F, m_F, u \rangle \\ &= \frac{1}{2} (-1)^{I+F} \sqrt{(2J+1)(2J'+1)} \begin{Bmatrix} I' & J' & F \\ J & I & 1 \end{Bmatrix} \\ & \times \sum_n^{N_I} \sum_{\omega=0}^2 N_\omega \left[(-1)^\omega J \begin{Bmatrix} \omega & 1 & 1 \\ J & J' & J \end{Bmatrix} \begin{Bmatrix} J & 1 & J \\ -J & 0 & J \end{Bmatrix}^{-1} \right. \\ & \left. + J' \begin{Bmatrix} 1 & \omega & 1 \\ J & J' & J' \end{Bmatrix} \begin{Bmatrix} J' & 1 & J' \\ -J' & 0 & J' \end{Bmatrix}^{-1} \right] \\ & \times \mathcal{M}_{\omega, n}^{(J' J, J)} \langle I' \| \mathbf{I}_n^{(1)} \| I \rangle \end{aligned} \quad (5)$$

and

$$\begin{aligned} & \langle F, m_F, u' | H_{\text{ss}} | F, m_F, u \rangle \\ &= (-1)^{I+J'+J+F} \sqrt{(2J+1)(2J'+1)} \begin{Bmatrix} I' & J' & F \\ J & I & 2 \end{Bmatrix} \end{aligned}$$

$$\times \sum_{n>n'}^{N_l} \mathcal{D}_{n,n'}^{(J' l' J l)} \langle I' \| [\mathbf{I}_n^{(1)} \otimes \mathbf{I}_{n'}^{(1)}]^{(2)} \| I \rangle, \quad (6)$$

with the normalization constant $N_\omega = 1, -\sqrt{3}$, and $\sqrt{5}$ for $\omega = 0, 1$, and 2, respectively. The expressions for the reduced matrix elements of the nuclear-spin operators $\langle I' \| \mathbf{I}_n^{(1)} \| I \rangle$ and $\langle I' \| [\mathbf{I}_n^{(1)} \otimes \mathbf{I}_{n'}^{(1)}]^{(2)} \| I \rangle$ depend on the total number of coupled spins and can be computed using a general recursive procedure as described, for example, in Ref. 33. Here, for the two equivalent hydrogen spins $I_1 = I_2 = 1/2$, the reduced matrix elements are

$$\begin{aligned} \langle I' \| \mathbf{I}_n^{(1)} \| I \rangle &= (-1)^{I\delta_{n,1}+I'\delta_{n,2}} I_1 \\ &\times \sqrt{(2I+1)(2I'+1)} \begin{Bmatrix} I_1 & I' & I_1 \\ I & I_1 & 1 \end{Bmatrix} \\ &\times \begin{pmatrix} I_1 & 1 & I_1 \\ -I_1 & 0 & I_1 \end{pmatrix}^{-1}, \end{aligned} \quad (7)$$

with the explicit values $\langle 0 \| \mathbf{I}_n^{(1)} \| 0 \rangle = 0$, $\langle 1 \| \mathbf{I}_n^{(1)} \| 1 \rangle = \sqrt{3}/2$, $\langle 0 \| \mathbf{I}_n^{(1)} \| 1 \rangle = \pm\sqrt{3}/2$ for $n = 1$ and 2, respectively, and $\langle 1 \| \mathbf{I}_n^{(1)} \| 0 \rangle = \mp\sqrt{3}/2$.

The expressions for the $\mathcal{M}_{\omega,n}^{(J' l' J l)}$ and $\mathcal{D}_{n,n'}^{(J' l' J l)}$ tensors in Eqs. (5) and (6) depend on the chosen rovibrational wave functions $|J, m_J, l\rangle$, which are represented by the molecular rovibrational eigenfunctions calculated with the variational approach TROVE. The functions $|J, m_J, l\rangle$ are linear combinations of products of vibrational wave functions $|v\rangle = |v_1, v_2, \dots, v_M\rangle$ (M is the number of vibrational modes) and symmetric-top rotational functions,

$$|J, m_J, l\rangle = \sum_{v,k} c_{v,k}^{(J,l)} |v\rangle |J, k, m_J\rangle. \quad (8)$$

In this basis, the $\mathcal{M}_{\omega,n}^{(J' l' J l)}$ and $\mathcal{D}_{n,n'}^{(J' l' J l)}$ tensors are

$$\begin{aligned} \mathcal{M}_{\omega,n}^{(J' l' J l)} &= \sum_{v'k'v} \sum_{vk} \left[c_{v'k'}^{(J',l')} \right]^* c_{vk}^{(J,l)} (-1)^{k'} \\ &\times \sum_{\sigma=-\omega}^{\omega} \sum_{\alpha,\beta=x,y,z} \begin{pmatrix} J & \omega & J' \\ k & \sigma & -k' \end{pmatrix} \\ &\times U_{\omega\sigma,\alpha\beta}^{(2)} \langle v' | \bar{M}_{\alpha\beta,n} | v \rangle \end{aligned} \quad (9)$$

and

$$\begin{aligned} \mathcal{D}_{n,n'}^{(J' l' J l)} &= \sum_{v'k'v} \sum_{vk} \left[c_{v'k'}^{(J',l')} \right]^* c_{vk}^{(J,l)} (-1)^{k'} \\ &\times \sum_{\sigma=-2}^2 \sum_{\alpha,\beta=x,y,z} \begin{pmatrix} J & 2 & J' \\ k & \sigma & -k' \end{pmatrix} \\ &\times U_{2\sigma,\alpha\beta}^{(2)} \langle v' | \bar{D}_{\alpha\beta,n,n'} | v \rangle, \end{aligned} \quad (10)$$

where $\bar{M}_{\alpha\beta,n}$ and $\bar{D}_{\alpha\beta,n,n'}$ ($\alpha, \beta = x, y, z$) are spin-rotation and spin–spin interaction tensors in the molecule-fixed frame and the 9×9 constant matrix $U_{\omega\sigma,\alpha\beta}^{(2)}$ ($\omega = 0, \dots, 2, \sigma = -\omega, \dots, \omega$) defines the

transformation of a general second-rank Cartesian tensor operator into its spherical-tensor representation, see, e.g., (5.41)–(5.44) in Ref. 36.

The total Hamiltonian H is composed of a sum of the pure rovibrational Hamiltonian H_{rv} and hyperfine terms H_{sr} and H_{ss} . In the basis of TROVE wave functions, the rovibrational Hamiltonian H_{rv} is diagonal, and its elements are given by the rovibrational energies,

$$\begin{aligned} \langle F, m_F, u' | H | F, m_F, u \rangle &= E_u \delta_{u,u'} + \langle F, m_F, u' | H_{sr} | F, m_F, u \rangle \\ &+ \langle F, m_F, u' | H_{ss} | F, m_F, u \rangle, \end{aligned} \quad (11)$$

$$\text{where } \delta_{u,u'} = \delta_{J,J'} \delta_{l,l'} \delta_{I,I'} \delta_{\mathcal{I},\mathcal{I}'},$$

The above equations were implemented in the hyfor module of the Python software package Richmol,^{37,38} which uses rovibrational molecular states calculated in TROVE as a variational basis. Alternative approaches using Watson-type effective Hamiltonians³⁹ are also implemented in the Richmol package.

The hyperfine energies and wave functions are computed using a three step procedure. First, we solve the full rovibrational problem using TROVE and obtain the rovibrational energies and wave functions for all states with energies below a selected threshold. In the next step, the rovibrational matrix elements of the spin–rotation and spin–spin tensors are computed in the form given by Eqs. (9) and (10). These matrix elements are later used to build the spin–rotation and spin–spin interaction Hamiltonians using Eqs. (5) and (6). The total Hamiltonian is composed of the sum of a purely rovibrational part, which is diagonal and given by the rovibrational state energies, and non-diagonal spin–rotation and spin–spin parts. In the final step, the hyperfine energies and wave functions are obtained by diagonalizing the total Hamiltonian.

The computation of the dipole transition intensities also proceeds in two steps. First, the rovibrational matrix elements of the dipole moment surface are computed and cast into a tensor form similar to (10),

$$\begin{aligned} \mathcal{K}_\omega^{(J' l' J l)} &= \sum_{v'k'v} \sum_{vk} \left[c_{v'k'}^{(J',l')} \right]^* c_{vk}^{(J,l)} (-1)^{k'} \\ &\times \sum_{\sigma=-\omega}^{\omega} \sum_{\alpha,\beta=x,y,z} \begin{pmatrix} J & \omega & J' \\ k & \sigma & -k' \end{pmatrix} U_{\omega\sigma,\alpha}^{(1)} \langle v' | \bar{\mu}_\alpha | v \rangle, \end{aligned} \quad (12)$$

where $\bar{\mu}_\alpha$ ($\alpha = x, y, z$) is the permanent dipole moment in the molecule-fixed frame and the 3×3 constant matrix $U_{\omega\sigma,\alpha}^{(1)}$ ($\omega = 1, \sigma = -\omega, \dots, \omega$) defines the transformation of a general first-rank Cartesian tensor operator into its spherical-tensor representation, see, e.g., (5.4) in Ref. 36. In the second step, the dipole matrix elements are transformed into the basis of hyperfine wave functions, i.e.,

$$\begin{aligned} \mathcal{K}_\omega^{(F',u',F,u)} &= \sum_{I',\mathcal{I},J',I,\mathcal{I},J,l} \sum_{I',\mathcal{I},J',I',\mathcal{I},J,l} \left[c_{I',\mathcal{I},J',I',\mathcal{I},J,l}^{(F',u')} \right]^* c_{I,\mathcal{I},J,l}^{(F,u)} (-1)^I \\ &\times \sqrt{(2J'+1)(2J+1)} \begin{Bmatrix} J' & F' & I \\ F & J & \omega \end{Bmatrix} \\ &\times \mathcal{K}_\omega^{(J',l' J,l)} \delta_{I',I} \delta_{\mathcal{I}',\mathcal{I}}, \end{aligned} \quad (13)$$

where $c_{I,\mathcal{I},J,l}^{(F,u)}$ are hyperfine wave function coefficients obtained by diagonalization of the total Hamiltonian. Finally, the line strengths

for transitions between hyperfine states $|f\rangle = |F', u'\rangle$ and $|i\rangle = |F, u\rangle$ are computed as³⁴

$$S(f \leftarrow i) = (2F' + 1)(2F + 1) \left| \mathcal{K}_1^{(F' u', F u)} \right|^2, \quad (14)$$

where we sum over all degenerate m_F and m'_F components. The expression for the integrated absorption coefficient of the dipole transition in units of cm/molecule reads

$$I(f \leftarrow i) = \frac{8\pi^3 v_{if} e^{-hcE_i/kT} (1 - e^{-hc\nu_{if}/kT})}{3hcZ(T)} S(f \leftarrow i), \quad (15)$$

where $\nu_{if} = |E_i - E_f|$ is the transition wavenumber, E_i and E_f are energy term values of the initial and final states in cm^{-1} , $Z(T)$ is the temperature dependent partition function, h (erg · s) is the Planck constant, c (cm/s) is the speed of light, and k (erg/K) is the Boltzmann constant.

B. Electronic structure calculations

The molecule-fixed frame spin-rotation tensors $\tilde{M}_{\alpha\beta,n}$ ($\alpha, \beta = x, y, z$, $n = 1, 2$) were calculated *ab initio* on a grid of 2000 different molecular geometries with electronic energies ranging up to 30 000 cm^{-1} above the equilibrium energy. We used the all-electron CCSD(T) (coupled-cluster singles, doubles, and perturbative triples) method with the augmented core-valence correlation-consistent basis sets aug-cc-pwCVTZ⁴⁰ and aug-cc-pVQZ^{41,42} for the oxygen and hydrogen atoms, respectively. The basis sets were downloaded from the Basis Set Exchange library.^{43–45} The calculations employed second-order analytical derivatives⁴⁶ together with the rotational London orbitals,^{47,48} as implemented in the quantum chemistry package CFOUR.⁴⁹

The electronic structure calculations used the principal axes of the inertia coordinate frame. For variational calculations, another frame was employed, defined such that the x axis is parallel to the bisector of the valence bond angle with the molecule lying in the xz plane at all instantaneous molecular geometries. In this frame, the z axis coincides with the molecular axis at the linear geometry. The computed spin-rotation tensors were rotated from the principal axis of inertia to the new frame. The permutation symmetry is such that the exchange of the two hydrogen atoms transforms $\tilde{M}_{\alpha\beta,1}$ into $\tilde{M}_{\alpha\beta,2}$ followed by a sign change for non-diagonal elements ($\alpha \neq \beta$).

The expression for the spin-rotation tensor, as computed in CFOUR, contains multiplication by the inverse of the tensor of inertia, see (3) and (7) in Ref. 48. For linear and closely linear geometries of the molecule, the inertial tensor becomes singular, which creates a discontinuity in the dependence of xz and zz elements of the spin-rotation tensor on the bending angle. To circumvent this problem, we have multiplied the computed spin-rotation tensors on the right side by the corresponding inertial tensors. The resulting data for the inertia-scaled spin-rotation tensor were parameterized through least-squares fitting, using a power series expansions to fourth order in terms of valence bond coordinates, with $\sigma_{\text{rms}} \leq 0.3$ kHz for all tensor components. Later, when computing the rovibrational matrix elements of the spin-rotation tensor, we have multiplied the inertia-scaled tensor with the inverse moment of inertia. The divergence of the spin-rotation tensor in the vicinity of

linear geometries is exactly canceled by the basis functions chosen to satisfy the kinetic cusp condition at the linear geometry.^{31,50}

The spin-spin tensor elements were computed as magnetic dipole-dipole interaction between two hydrogen nuclei H_1 and H_2 ,

$$D_{\alpha\beta,12} = \frac{\mu_0}{4\pi} \frac{\mu_1 \mu_2}{I_1 I_2 r_{12}^3} (\mathbf{I} - 3\mathbf{n} \otimes \mathbf{n})_{\alpha\beta}, \quad (16)$$

where $\mu_1 = \mu_2 = 2.792\,847\,34$ are the magnetic dipole moments of H_1 and H_2 in units of the nuclear magneton, $I_1 = I_2 = 1/2$ are the corresponding hydrogen nuclear spins, r_{12} is the distance between the hydrogen nuclei, and \mathbf{n} is the unit vector directed from one hydrogen to another. The indirect spin-spin coupling constants mediated by the electronic motions were not considered here, as they are typically two orders of magnitude smaller than the direct constants.⁵¹

The magnitudes of the equilibrium *ab initio* spin-rotation and direct spin-spin diagonal tensor elements are about 30 and 60 kHz, respectively.⁵² However, the corresponding matrix elements have different selection rules. In particular, due to the traceless-tensor nature of the spin-spin interaction, it can couple only states with $|J - J'| = 2$, see (6) and (10). The spin-rotation interaction can, in principle, couple states with $|J - J'| \leq 2$, where the *ortho-para* interaction between states with $|J - J'| \leq 1$ and $|k - k'| = 1$ occurs due to the antisymmetric behavior of the off-diagonal elements of the spin-rotation tensor with respect to the proton exchange, i.e., $\tilde{M}_{xz,1} = \tilde{M}_{zx,2}$, see (5) and (9).

C. Nuclear motion calculations

We employed TROVE to calculate the rovibrational states using the exact kinetic-energy operator formalism recently developed for triatomic molecules.⁵⁰ This formalism is based on the use of associated Laguerre polynomials $L_n^l(x)$ as bending basis functions, which ensures a correct behavior of the rovibrational wave functions at linear molecular geometry.⁵⁰ The bisecting frame embedding was selected as a non-rigid reference frame, with the x axis oriented parallel to the bisector of the valence bond angle and the molecule placed in the xz plane. In this frame, the z axis coincides with the linearity axis at linear molecular geometry. An accurate empirically refined PES of H_2^{16}O was employed.³²

The primitive-stretching vibrational basis functions were generated by numerically solving the corresponding one-dimensional Schrödinger equations on a grid of 2000 points using the Numerov-Cooley approach.^{53,54} The primitive basis functions were then symmetry-adapted to the irreducible representations of the $\text{C}_{2v}(M)$ molecular symmetry group using an automated numerical procedure.³⁰ The total vibrational basis set was formed as a direct product of the symmetry-adapted stretching and bending basis functions, contracted to include states up to a polyad 48. It was used to solve the $J = 0$ eigenvalue problem for the complete vibrational Hamiltonian of H_2O . A product of the $J = 0$ eigenfunctions and symmetry-adapted rigid rotor wavefunctions formed the final rovibrational basis set. The rovibrational wavefunctions $|J, m_J, l\rangle$ for rotational excitations up to $J = 40$ and four irreducible representations A_1, A_2, B_1 , and B_2 were computed by diagonalizing the matrix representation of the total rovibrational Hamiltonian H_{rv} in the rovibrational basis set. More details about the variational approach and the basis-symmetrization procedure for the case of triatomic molecules can be found in Ref. 50.

D. Line list simulations

The line list of hyperfine rovibrational transitions for H_2^{16}O was computed with an energy cutoff at $15\,000\text{ cm}^{-1}$ and includes transitions up to $F = 39$ ($J = 40$). To further improve the accuracy of the line list, after solving the pure rovibrational problem and before entering the hyperfine calculations, the rovibrational energies E_u in (11) were replaced with the high-resolution experimental IUPAC values from Ref. 55, where available. Such empirical adjustment of the rovibrational energies has been adopted and tested, e.g., for the production of molecular line lists as a part of the ExoMol project.⁵⁶ Recently, this approach has been proven to be accurate for computing ultra-weak quadrupole transitions in water^{57,58} and carbon dioxide,^{59,60} which enabled their first laboratory (H_2O and CO_2) and astrophysical (CO_2) detection.

The final line list has been calculated at room temperature ($T = 296\text{ K}$) with the corresponding partition function $Z = 174.5813$ ⁶¹ and a threshold of $10^{-36}\text{ cm/molecule}$ for the absorption intensity based on (15). The line list stored in the ExoMol⁶² format is provided in Ref. 72.

III. RESULTS AND DISCUSSION

An overview of the calculated H_2^{16}O dipole absorption stick spectrum at $T = 296\text{ K}$ is shown in Fig. 1. The forbidden *ortho–para* transitions are plotted as red circles. Despite being, at least, 10 orders of magnitude weaker than the corresponding allowed transitions, for some of the strongest *ortho–para* transitions, the predicted absorption intensities are close to the sensitivity threshold of modern cavity ring-down spectroscopic techniques.^{64–66} All predicted *ortho–para* transitions with the line intensity larger than $10^{-31}\text{ cm/molecule}$ are listed in Table I. These transitions all occur in the fundamental v_2 bending and the overtone $2v_1 + v_2$ and $v_1 + v_2 + v_3$ bands. The off-diagonal elements of molecular-frame spin-rotation tensor $\tilde{M}_{\alpha\beta,n}$, which lead to *ortho–para* interactions, are highly dependent on the bending vibrational coordinate, indicating significance of the v_2 band in *ortho–para* transitions. The size of the off-diagonal spin-rotation matrix elements increases for bending angles close to 180° , i.e., the linear geometry of the molecule.

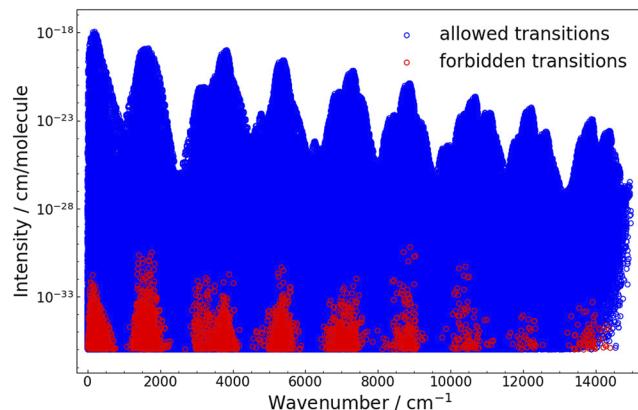


FIG. 1. Overview of the H_2^{16}O dipole absorption spectrum at $T = 296\text{ K}$. The *ortho–ortho* and *para–para* transitions are marked with blue circles, whereas the *ortho–para* transitions are given by red circles.

This leads to an increase in the *ortho–para* interaction for rovibrational energies close to the linearity barrier at $\sim 8254\text{ cm}^{-1}$ above the zero-point energy. The spin–rotation coupling in these vibrationally excited states is responsible for the *ortho–para* transitions. For example, the final transition state $F = 3$, $J_{k_a k_c} = 4_{2,3}$ (*ortho*) with energy $E = 1908.016\,319\text{ cm}^{-1}$ is mixed with the state $F = 3$, $J_{k_a k_c} = 3_{3,1}$ (*para*) with energy $E = 1907.450\,231\text{ cm}^{-1}$. The size of the rovibrational matrix element of the spin–rotation tensor $\mathcal{M}_{\omega,n}^{(J' J, II)}$ in (9) for this transition is $\pm 0.95\text{ kHz}$ and $\pm 6.3\text{ kHz}$ (\pm for $n = 1, 2$) for $\omega = 1$ and 2, respectively. Note that following (5), only the spin–rotation tensor with $\omega = 1$ contributes to the *ortho–para* coupling. Allowed transitions into these states from the ground state are quite strong, 2.07×10^{-20} and $3.52 \times 10^{-20}\text{ cm/molecule}$, respectively. Accordingly, intensity borrowing as a result of the spin–rotation interaction of excited states leads to non-zero intensities of the two corresponding forbidden transitions on the order of $10^{-31}\text{ molecule/cm}$. Similarly, for other strongest forbidden transitions listed in Table I, the enhancement occurs due to the intensity borrowing effect

TABLE I. Strongest predicted *ortho–para* transitions in H_2^{16}O at $T = 296\text{ K}$ with the $10^{-31}\text{ cm/molecule}$ intensity cut-off.

v'_1	v'_2	v'_3	F'	J'	k'_a	k'_c	I'	$E' (\text{cm}^{-1})$	v_1	v_2	v_3	F	J	k_a	k_c	I	$E (\text{cm}^{-1})$	Freq. (cm^{-1})	Int. ($\text{cm}/\text{molecule}$)
0	1	0	3	4	2	3	<i>o</i>	1908.016 319	0	0	0	4	4	4	0	<i>p</i>	488.134 170	1419.882 149	2.26×10^{-31}
0	1	0	3	3	3	1	<i>p</i>	1907.450 231	0	0	0	3	4	3	2	<i>o</i>	382.516 901	1524.933 330	1.36×10^{-31}
0	1	0	3	3	3	1	<i>p</i>	1907.450 231	0	0	0	3	4	1	4	<i>o</i>	224.838 381	1682.611 850	1.12×10^{-31}
0	1	0	3	4	2	3	<i>o</i>	1908.016 319	0	0	0	3	3	2	2	<i>p</i>	206.301 430	1701.714 889	1.02×10^{-31}
0	1	0	3	3	3	1	<i>p</i>	1907.450 231	0	0	0	2	3	1	2	<i>o</i>	173.365 811	1734.084 420	2.05×10^{-31}
0	1	0	3	4	2	3	<i>o</i>	1908.016 319	0	0	0	2	2	2	0	<i>p</i>	136.163 927	1771.852 392	3.28×10^{-31}
2	1	0	3	4	1	4	<i>o</i>	8979.657 423	0	0	0	4	4	1	3	<i>p</i>	275.497 051	8704.160 372	3.36×10^{-31}
2	1	0	3	4	1	4	<i>o</i>	8979.657 423	0	0	0	3	3	1	3	<i>p</i>	142.278 493	8837.378 930	1.01×10^{-31}
2	1	0	3	4	1	4	<i>o</i>	8979.657 423	0	0	0	2	2	1	1	<i>p</i>	95.175 936	8884.481 487	6.41×10^{-31}
1	1	1	15	14	3	11	<i>o</i>	11 067.083 574	0	0	0	14	14	0	14	<i>p</i>	2073.514 207	8993.569 367	1.92×10^{-31}
1	1	1	15	15	2	13	<i>p</i>	11 067.089 122	0	0	0	14	13	1	12	<i>o</i>	2042.309 821	9024.779 300	2.04×10^{-31}

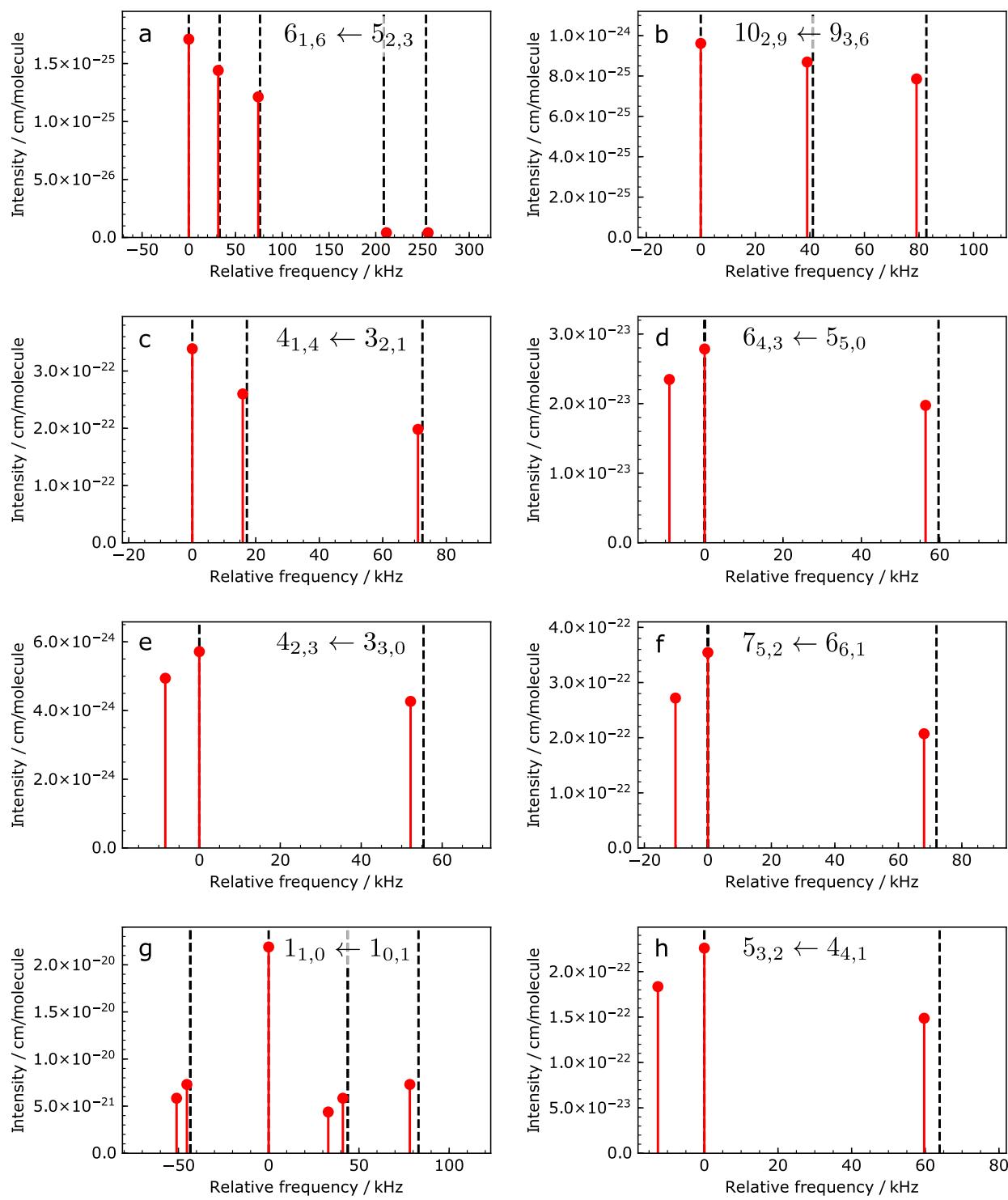


FIG. 2. Comparison of calculated hyperfine transitions (red stems) with experimental data (dashed lines) from (a) Ref. 63 and [(b)–(h)] Ref. 52. Different panels show hyperfine transitions for different rotational bands $J'_{k'_a k'_c} \leftarrow J_{k_a k_c}$. The measured (calculated) zero-crossing frequencies, in MHz, are 22 235.0447 (22 235.0322), 321 225.6363 (321 225.6311), 380 197.3303 (380 197.3361), 439 150.7746 (439 150.7857), 443 018.3358 (443 018.4016), 448 001.0538 (448 001.0359), 556 935.9776 (556 935.9849), and 620 700.9334 (620 700.8889) for panels (a)–(h), respectively.

from strongly allowed transitions with coincidence near resonance between the excited states, accompanied by a relatively large value of the spin-rotation matrix element $\mathcal{M}_{\omega=1,n}^{(J'J,J)}$.

Though *ortho-para* transitions are yet to be observed in H₂O, there are several spectroscopic studies of the allowed hyperfine transitions in the pure rotational spectrum of H₂¹⁶O.^{52,63,67,68} We used these data to validate the accuracy of our predictions. In Fig. 2, the calculated transitions (stems) are compared with the available experimental data (dashed lines), demonstrating an excellent agreement, within 1–4 kHz, for the hyperfine splittings. For example, the root-mean square (rms) deviation of the predicted hyperfine splittings from the experiment is 2.1 kHz in Fig. 2(a), while for the absolute line positions, it is 12.3 kHz. The latter can be explained by the discrepancies in predictions of the pure rotational transitions. The errors in predictions of the hyperfine splittings can be attributed to the level of electronic structure theory, in particular the basis set, employed in the calculations of the spin-rotation tensor surface. The basis set convergence of the equilibrium spin-rotation constants of H₂O was investigated elsewhere.⁵² According to the

results, the employed aug-cc-pwCVTZ basis set produces an average error of 1.3 kHz with a maximum of 1.8 kHz for one of the off-diagonal elements, when compared with the results obtained with the aug-cc-pwCV6Z basis set. There are several predicted splittings in Figs. 2(d)–2(h) that are less than 12 kHz and were not resolved in the experiment.⁵² Indeed, by visual inspection of the Lamb-dip spectrum plotted in Fig. 1 of Ref. 52, which was provided as an example of the experimental resolution achieved in that study, the transition profiles' full width at half maximum is about 13 kHz.

The sensitivity and resolution required to observe the *ortho-para* transitions in a prospective experiment can be estimated from the simulated absorption spectrum, shown in Fig. 3 for selected wavenumber ranges with strong *ortho-para* transitions. Since the Doppler linewidth would be around 0.01 cm⁻¹ at room temperature and even much higher-resolution spectroscopy was demonstrated,⁶⁹ we used simple Gaussian line profiles with half-width at half-maximum (HWHM) fixed at 0.01, 0.005, and 0.001 cm⁻¹ and computed absorption cross-sections at $T = 296$ K using ExoCross⁷⁰ to predict the experimental spectra. The *ortho-para* transitions (red)

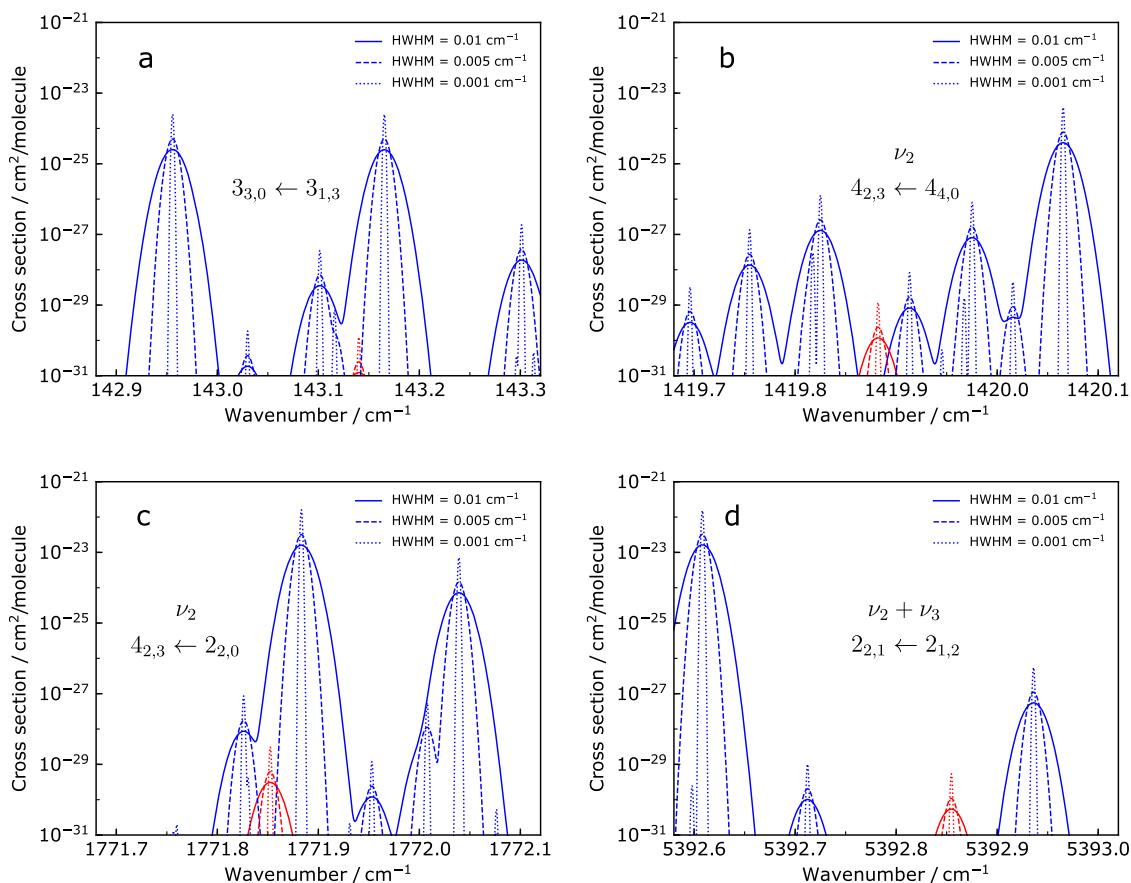


FIG. 3. Absorption cross-sections computed at $T = 296$ K for selected rotational bands in (a) ground vibrational state, [(b) and (c)] ν_2 , and (d) $\nu_2 + \nu_3$ vibrational bands, using Gaussian lineshapes with HWHMs of 0.01 cm^{-1} (solid lines), 0.005 cm^{-1} (dashed lines), and 0.001 cm^{-1} (dotted lines). The cross-sections for allowed *ortho-ortho* and *para-para* transitions are plotted with blue color lines and cross-sections for forbidden *ortho-para* transitions are plotted with red color lines.

in Figs. 3(a) and 3(c) show considerable overlap with the allowed transitions (blue) for purely rotational transitions and in the fundamental ν_2 excitation band and could only be detected with an experimental HWHM below 0.005 cm^{-1} at an experimental sensitivity of 10^{-30} and $10^{-29}\text{ cm}^2/\text{molecule}$, respectively. In Figs. 3(b) and 3(d), showing parts of the ν_2 and $\nu_2 + \nu_3$ bands, the predicted *ortho–para* transitions are better separated from the allowed transitions and should already be detectable at lower resolution, i.e., at a HWHM of 0.01 cm^{-1} , but demand a greater sensitivity of 10^{-30} and $10^{-31}\text{ cm}^2/\text{molecule}$, respectively. Such high-sensitivity measurements of intensities on the scale of $10^{-30}\text{ cm}^2/\text{molecule}$ are currently within reach, for example, using continuous wave laser cavity ring down spectroscopy.^{58,71}

IV. CONCLUSIONS

We developed and performed comprehensive variational calculations of the room temperature line list of H_2O with hyperfine resolution, including forbidden *ortho–para* transitions. The calculations were based on accurate rovibrational energy levels and wavefunctions produced using the variational approach TROVE. The nuclear hyperfine effects were modeled as spin–rotation and direct spin–spin interactions, with the spin–rotation coupling surface calculated at a high level of the electronic-structure theory. We found excellent agreement between the calculated transition frequencies and available hyperfine-resolved spectroscopic data of allowed transitions.

The predicted *ortho–para* transitions are useful for guiding future experimental spectroscopic studies in search of these forbidden transitions in the laboratory as well as in astrophysical environments. Our accurate predictions of hyperfine effects complement the spectroscopic data for water.

The variational approach we developed for computing these hyperfine effects is general. It includes nuclear quadrupole,^{33,34} spin–rotation, and spin–spin interactions and can be applied to other molecular systems without restrictions on the number and values of nuclear spins.

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AUTHOR DECLARATIONS

Conflict of Interest

The authors have no conflicts to disclose.

DATA AVAILABILITY

The data that support the findings of this study are available on Zenodo.⁷² The codes used in the presented calculations are openly available at <https://github.com/Trovemaster/TROVE> and <https://github.com/CFEL-CMI/richmol>.

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