HW 3

2.24 (a)

$$\Omega\left(N, \frac{N}{2}\right) = \frac{N!}{\left(\frac{N}{2}\right)! \left(\frac{N}{2}\right)!}$$

$$\approx \frac{N^N E^{-N} \sqrt{2\pi N}}{\left(\left(\frac{N}{2}\right)^{(N/2)} E^{-(N/2)} \sqrt{\pi N}\right) \left(\left(\frac{N}{2}\right)^{(N/2)} E^{-(N/2)} \sqrt{\pi N}\right)}$$

$$\approx \frac{N^N E^{-N} \sqrt{2\pi N}}{\left(\frac{N}{2}\right)^N E^{-N} \pi N}$$

$$\approx \frac{2^N \sqrt{2\pi N}}{\pi N}$$

$$\approx \frac{2^{N+1}}{\sqrt{2\pi N}}$$

$$\approx \frac{2^{N+\frac{1}{2}}}{\sqrt{2\pi}}$$

$$\Omega_{max} \approx 2^N$$

(b)

$$x \equiv N \uparrow -\frac{N}{2}$$

$$N \uparrow = \frac{N}{2} + 2$$

$$N = N \uparrow + N \downarrow$$

$$N = \frac{N}{2} + x + N \downarrow$$

$$\frac{N}{2} = x + N \downarrow$$

$$N \downarrow = \frac{N}{2} - x$$

$$\Omega = \binom{N \downarrow + N \uparrow + 1}{N \downarrow} = \binom{\frac{N}{2} + x + \frac{N}{2} - x + 1}{\frac{N}{2} - x}$$

$$\begin{split} \Omega &\approx \binom{N}{\frac{N}{2}-x} = \frac{N!}{\left(\frac{N}{2}-x\right)! \left(\frac{N}{2}-x\right)!} \\ &\approx \frac{N^N e^{-N} \sqrt{2\pi N}}{\left(\left(\frac{N}{2}-x\right)^{\frac{N}{2}-x}\right) e^{-\left(\frac{N}{2}-x\right)} \sqrt{2\pi \left(\frac{N}{2}-x\right)} \left(\left(\frac{N}{2}+x\right)^{\frac{N}{2}+x}\right) e^{-\left(\frac{N}{2}+x\right)} \sqrt{2\pi \left(\frac{N}{2}+x\right)} \right)} \\ &\approx \frac{N^N \sqrt{2\pi N}}{\left(\left(\frac{N}{2}-x\right)^{\frac{N}{2}-x}\right) \sqrt{2\pi \left(\frac{N}{2}-x\right)} \left(\left(\frac{N}{2}+x\right)^{\frac{N}{2}+x}\right) \sqrt{2\pi \left(\frac{N}{2}+x\right)}} \\ &\approx \frac{N^N \sqrt{2\pi N}}{2\pi \sqrt{\left(\frac{N}{2}\right)^2-x^2} \left(\frac{N}{2}-x\right)^{\frac{N}{2}} \left(\frac{N}{2}-x\right)^{-x} \left(\frac{N}{2}+x\right)^{\frac{N}{2}} \left(\frac{N}{2}+x\right)^x} \\ &\approx \frac{N^N \sqrt{2\pi N}}{2\pi \sqrt{\left(\frac{N}{2}\right)^2-x^2} \left(\frac{N}{2}-x\right)^{\frac{N}{2}} \left(\frac{N}{2}-x\right)^{-x} \left(\frac{N}{2}+x\right)^x} \\ &\approx \frac{N^N \sqrt{N}}{2\pi \sqrt{N}} \\ &\approx \frac{N^N \sqrt{N}}{\sqrt{2\pi \left(\frac{N}{2}\right)^2-x^2} \left(\left(\frac{N}{2}\right)^2-x^2\right)^{\frac{N}{2}} \left(\frac{N}{2}-x\right)^{-x} \left(\frac{N}{2}+x\right)^x} \\ &\Omega \approx \frac{N^N}{\left(\left(\frac{N}{2}\right)^2-x^2\right)^{\frac{N}{2}} \left(\frac{N}{2}-x\right)^{-x} \left(\frac{N}{2}+x\right)^x} \\ &\ln(\Omega) \approx \ln \left(\frac{N^N}{\left(\frac{N}{2}\right)^2-x^2\right)^{\frac{N}{2}} \left(\frac{N}{2}-x\right)^{-x} \left(\frac{N}{2}+x\right)^x} \\ &\approx \ln(N^N) - \ln \left(\left(\frac{N}{2}\right)^2-x^2\right)^{\frac{N}{2}} \left(\frac{N}{2}-x\right)^{-x} \left(\frac{N}{2}+x\right)^x} \\ &\approx N \ln(N) - \frac{N}{2} \ln \left(\left(\frac{N}{2}\right)^2-x^2\right) + x \ln \left(\frac{N}{2}-x\right) - x \ln \left(\frac{N}{2}+x\right) \end{aligned}$$

i.

$$\frac{N}{2}\ln\left(\left(\frac{N}{2}\right)^2 - x^2\right) = \frac{N}{2}\ln\left(\left(\frac{N}{2}\right)^2 \left(1 - \left(\frac{2x}{N}\right)^2\right)\right)$$

$$= \frac{N}{2}\ln\left(\left(\frac{N}{2}\right)^2\right) + \frac{N}{2}\ln\left(1 - \left(\frac{2x}{N}\right)^2\right)$$

$$= N\ln\left(\frac{N}{2}\right) - \frac{N}{2}\left(\frac{2x}{N}\right)^2$$

$$= N\ln(N) - N\ln(2) - \frac{2x^2}{N}$$

ii.

$$x \ln\left(\frac{N}{2} - x\right) = x \ln\left(\frac{N}{2}\left(1 - \frac{2x}{N}\right)\right)$$

$$= x \ln\left(\frac{N}{2}\right) + x \ln\left(1 - \frac{2x}{N}\right)$$

$$= x \ln\left(\frac{N}{2}\right) - x\left(\frac{2x}{N}\right)$$

$$= x \ln\left(\frac{N}{2}\right) - \frac{2x^2}{N}$$

iii.

$$x \ln\left(\frac{N}{2} + x\right) = x \ln\left(\frac{N}{2}\left(1 + \frac{2x}{N}\right)\right)$$

$$= x \ln\left(\frac{N}{2}\right) + x \ln\left(1 + \frac{2x}{N}\right)$$

$$= x \ln\left(\frac{N}{2}\right) + x\left(\frac{2x}{N}\right)$$

$$= x \ln\left(\frac{N}{2}\right) + \frac{2x^2}{N}$$

$$\begin{split} \ln(\Omega) &\approx N \ln(N) - \left(N \ln(N) - N \ln(2) - \frac{2x^2}{N} \right) + \left(x \ln\left(\frac{N}{2}\right) - \frac{2x^2}{N} \right) - \left(x \ln\left(\frac{N}{2}\right) + \frac{2x^2}{N} \right) \\ \ln(\Omega) &\approx N \ln(2) - \frac{2x^2}{N} \\ \Omega &\approx 2^N e^{-\frac{2x^2}{N}} \\ \Omega &\approx 2^N \quad (\Omega_{max}, x = 0) \end{split}$$

(c) Width measured when value falls to $\frac{1}{e}$ of max value.

$$\frac{\Omega_{max}}{e} = \Omega_{max}e^{-\frac{2x^2}{N}}$$

$$e^{-1} = e^{-\frac{2x^2}{N}}$$

$$1 = \frac{2x^2}{N}$$

$$x^2 = \frac{N}{2}$$

$$x = \sqrt{\frac{N}{2}}$$

Total width $w = 2x = 2\sqrt{\frac{N}{2}} = \sqrt{2N}$.

(d)

$$P = \frac{\Omega}{\Omega_{max}} = \frac{\Omega_{max} e^{-\frac{2x^2}{N}}}{\Omega_{max}} = e^{-\frac{2x^2}{N}}$$
$$P_{501000} = e^{-\frac{2(1000)^2}{1000000}} = 0.135$$

501000 heads is probable

$$P_{510000} = e^{-\frac{2(10000)^2}{1000000}} = 1.38 \times 10^{-87}$$

510000 heads is improbable

2.25 (a) N = Total Steps, n = Steps away from starting position

$$\Omega \approx \frac{N!}{n!(N-n)!}$$

$$\Omega_{max} \approx 2^{N}$$

$$P = \frac{\Omega}{\Omega_{max}} = \frac{N!}{n!(N-n)!} \frac{1}{2^{N}}$$

Highest probability state is when n=0, so ending up at the same place. It is also probable that you will end up in a spot that is within the width of the peak, so $\pm \sqrt{\frac{N}{2}}$ steps away from the starting position.

- (b) $\max \Delta x = \pm \sqrt{\frac{10000}{2}} \approx \pm 71$. It is likely that you will end up within ± 71 steps from the starting point.
- (c) Mean free path:

$$\ell = \frac{1}{4\pi r^2} \frac{V}{N_{mol}}$$

$$\overline{v} = \sqrt{\frac{3kT}{m}}$$

Typical values for air at STP: $\ell = 1.5 \times 10^{-7} m$ and $\overline{v} = 500 m/s$.

$$N = \frac{\overline{v}t}{\ell}$$

$$\Delta x \approx x_{width} \ell = \sqrt{\frac{N}{2}} \ell = \sqrt{\frac{\overline{v}t}{2l}} \ell$$

$$\approx \sqrt{\frac{\overline{v}t\ell}{2}}$$

$$\Delta x \approx \sqrt{\frac{(500)(1)(1.5 \times 10^{-7})}{2}} = 0.0061m$$

 Δx depends on \sqrt{t} so for an increasing Δx the time it takes will grow at an exponential rate.

 Δx also has a dependence on $\sqrt[4]{T}$ so for increases in temperature Δx does not appreciably change.

2.29

$$\Omega_{most} = 6.87 \times 10^{114}
\Omega_{least} = 2.77 \times 10^{81}
S_{most} = \ln(\Omega_{most}) = \ln(6.87 \times 10^{114}) = 264.4
S_{least} = \ln(\Omega_{least}) = \ln(2.77 \times 10^{81}) = 187.5
\Omega_{total} = 9.26 \times 10^{115}
S_{total} = \ln(\Omega_{total}) = \ln(9.26 \times 10^{115}) = 267$$

2.34

$$S = k \ln(\Omega)$$

$$= Nk \left[\ln \left(\frac{V}{N} \left(\frac{4n\pi U}{3Nh^3} \right)^{\frac{3}{2}} \right) + \frac{5}{2} \right] \quad DoF = 3$$

$$Q = \Delta U + W \quad \Delta U = 0$$

$$= W = \int_{V_i}^{V_f} P dV \quad P = \frac{NkT}{V}$$

$$= NkT \int_{V_i}^{V_f} \frac{dV}{V} = NkT \ln \left(\frac{V_f}{V_i} \right)$$

$$S = Nk \left[\ln \left(\frac{V}{N} \left(\frac{4n\pi U}{3Nh^3} \right)^{\frac{3}{2}} + \frac{5}{2} \right) \right]$$

$$= Nk \left[\ln(V) \ln \left(\frac{1}{N} \left(\frac{4n\pi U}{3Nh^3} \right)^{\frac{3}{2}} \right) + \frac{5}{2} \right]$$

$$\ln\left(\frac{1}{N}\left(\frac{4n\pi U}{3Nh^3}\right)^{\frac{3}{2}}\right) + \frac{5}{2}$$
 is constant

$$\Delta S = Nk \ln(V_f) - Nk \ln(V_i) = Nk \ln\left(\frac{V_f}{V_i}\right)$$
$$= Nk \ln\left(\frac{V_f}{V_i}\right) \frac{T}{T} = \frac{NkT \ln\left(\frac{V_f}{V_i}\right)}{T}$$
$$\Delta S = \frac{Q}{T}$$

2.36

$$S \approx Nk$$

$$N_{book} = \frac{1000}{12} \times 6.022 \times 10^{22} = 5.02 \times 10^{25}$$

$$S_{book} = 5.02 \times 10^{25} \times 1.38 \times 10^{-23} = 693JK^{-1}$$

$$N_{moose} = \frac{400 \times 10^3}{18} \times 6.022 \times 10^{22} = 1.34 \times 10^{28}$$

$$S_{moose} = 1.34 \times 10^{28} \times 1.38 \times 10^{-23} = 1.84 \times 10^5JK^{-1}$$

$$N_{sun} = \frac{2 \times 10^{33}}{12} \times 6.022 \times 10^{22} = 1.2 \times 10^{57}$$

$$S_{sun} = 1.2 \times 10^{57} \times 1.38 \times 10^{-23} = 1.66 \times 10^{34}JK^{-1}$$

3.1

$$\frac{1}{T} = \frac{\partial S}{\partial U}$$

$$T = \frac{\partial U}{\partial S} = \frac{\Delta U}{\Delta S} = \frac{\epsilon \Delta q}{\Delta S}$$

$$\epsilon = 0.1 \text{ eV} = 1.6 \times 10^{-20} \text{ J}$$

(a) $q_a = 1$

$$T_A = \frac{\epsilon(2-0)}{S_A(2) - S_A(0)} = \frac{2\epsilon}{k(10.7-0)} = 0.187 \frac{\epsilon}{k}$$
$$= 0.187 \left(\frac{1.6 \times 10^{-20}}{1.38 \times 10^{-23}}\right) = 216.7K$$

$$T_B = \frac{\epsilon(100 - 98)}{S_B(100) - S_B(98)} = \frac{2\epsilon}{k(187.53 - 185.33)} = 0.909 \frac{\epsilon}{k}$$
$$= 0.909 \left(\frac{1.6 \times 10^{-20}}{1.38 \times 10^{-23}}\right) = 1054K$$

(b)
$$q_a = 60$$

$$T_A = \frac{\epsilon(61 - 59)}{S_A(61) - S_A(59)} = \frac{2\epsilon}{k(160.9 - 157.35)} = 0.56 \frac{\epsilon}{k}$$
$$= 0.56 \left(\frac{1.6 \times 10^{-20}}{1.38 \times 10^{-23}}\right) = 649K$$

$$T_B = \frac{\epsilon(41 - 39)}{S_B(41) - S_B(39)} = \frac{2\epsilon}{k(107.0 - 103.5)} = 0.56 \frac{\epsilon}{k}$$
$$= 0.56 \left(\frac{1.6 \times 10^{-20}}{1.38 \times 10^{-23}}\right) = 649K$$
$$T_A = T_B$$

$$\frac{1}{T} = \frac{\partial S}{\partial U} \Rightarrow T = \frac{\partial U}{\partial S}$$

$$T_A = \frac{\partial U_A}{\partial S_A} \quad T_B = \frac{\partial U_B}{\partial S_B} \quad T_C = \frac{\partial U_C}{\partial S_C}$$

$$\begin{split} T_A &= T_B \\ \frac{\partial U_A}{\partial S_A} &= \frac{\partial U_B}{\partial S_B} \\ T_B &= T_C \\ \frac{\partial U_B}{\partial S_B} &= \frac{\partial U_C}{\partial S_C} \Rightarrow \frac{\partial U_A}{\partial S_A} = \frac{\partial U_C}{\partial S_C} \Rightarrow T_A = T_C \end{split}$$

3.3 The two objects will exchange thermal energy until the slopes of their entropy vs energy graphs are equal.

$$\begin{split} \frac{1}{T} &= \frac{\partial S}{\partial U} \quad \Omega = \left(\frac{Ne}{q}\right)^q \\ U &= q\epsilon \Rightarrow q = \frac{U}{\epsilon} \\ \omega &= \left(\frac{Ne\epsilon}{U}\right)^{\frac{U}{\epsilon}} \end{split}$$

$$S = k \ln(\Omega) = k \ln\left[\left(\frac{Ne\epsilon}{U}\right)^{\frac{U}{\epsilon}}\right]$$

$$= \frac{Uk}{\epsilon} \ln\left(\frac{Ne\epsilon}{U}\right)$$

$$= \frac{Uk}{\epsilon} (\ln(N\epsilon) + \ln(e) - \ln(U))$$

$$\frac{1}{T} = \frac{\partial S}{\partial U} = \frac{\partial}{\partial U} \left[\frac{Uk}{\epsilon} (\ln(N\epsilon) + 1 - \ln(U))\right]$$

$$= \frac{\partial}{\partial U} \left[\frac{Uk}{\epsilon} \ln(N\epsilon) + \frac{Uk}{\epsilon} - \frac{Uk}{\epsilon} \ln(U)\right]$$

$$= \frac{k}{\epsilon} \ln(N\epsilon) + \frac{k}{\epsilon} - \left(\frac{k}{\epsilon} \ln(U) + \frac{Uk}{U\epsilon}\right)$$

$$\frac{1}{T} = \frac{k}{\epsilon} (\ln(N\epsilon) - \ln(U))$$

$$\frac{\epsilon}{kT} = \ln(N\epsilon) - \ln(U)$$

$$e^{\ln(U)} = e^{\left(\ln(N\epsilon) - \frac{\epsilon}{kT}\right)}$$

$$U = N\epsilon e^{-\frac{\epsilon}{kT}}$$

3.10 (a) Heat of formation for water $L_{H_20(\ell)} = 334 Jg^{-1}$

$$Q = mL = (30)(334) = 10020J$$

$$\Delta S_A = \frac{Q}{T} = \frac{10020}{273} = 36.7JK^{-1}$$

(b)

$$c_{H_2O(\ell)} = 4.181 J g^{-1} K^{-1}$$

 $\Delta S_B = C_V \int_{T_i}^{T_f} \frac{1}{T} dT$

$$C_V = mc$$

$$\Delta S_B = mc \int_{T_i}^{T_f} \frac{1}{T} dT$$

$$= mc \ln \left(\frac{T_f}{T_i}\right)$$

$$= (30)(4.181) \ln \left(\frac{298}{273} = 11JK^{-1}\right)$$

$$\Delta S_{water} = \Delta S_A + \Delta S_B = 36.7 + 11 = 47.7 J K^{-1}$$

$$Q = Q_{ice} + Q_{water} = Q_{ice} + mc\Delta T$$
$$= 10020 + (30)(4.181)(25) = 13156J$$
$$\Delta S_{room} = \frac{-13156}{298} = -44.7JK^{-1}$$

(d) $\Delta S_{univ} = \Delta S_{room} + \Delta S_{water} = -44.1 + 47.7 = 3.6 J K^{-1}$

A positive entropy is expected since the reaction occurs spontaneously

3.11

$$T_1 = 55C = 328K \quad T_2 = 10C = 283K \quad V_1 = 50L \quad V_2 = 25L$$

$$T_f = \frac{T_1V_1 + T_2V_2}{V_1 + V_1}$$

$$= \frac{(328)(50) + (283)(25)}{50 + 25} = 313K$$

$$\Delta S = C_V \int_{T_i}^{T_f} \frac{1}{T} dT = C_v \ln\left(\frac{T_f}{T_i}\right) = mc \ln\left(\frac{T_f}{T_i}\right)$$

$$\Delta S_{hot} = (50000)(4.181) \ln\left(\frac{313}{328}\right) = -9786JK^{-1}$$

$$\Delta S_{cold} = (25000)(4.181) \ln\left(\frac{313}{283}\right) = 10532JK^{-1}$$

$$\Delta S_{univ} = \Delta S_{hot} + \Delta S_{cold} = -9786 + 10532 = 746JK^{-1}$$

3.13 (a)

$$\Delta S = \frac{Q}{T} \quad Q = F_S T$$

$$F_s = 1000W m^{-2} \quad t = 365 \times 8 \times 60 \times 60 = 10512000s$$

$$Q = (1000)(10512000) = 1.0512 \times 10^{10}$$

$$\Delta S_{\bigoplus} = \frac{1.0512 \times 10^{10}}{300} = 3.5 \times 10^7 J K^{-1}$$

$$\Delta S_{\bigodot} = \frac{-1.0512 \times 10^{10}}{6000} = -1.75 \times 10^6 J K^{-1}$$

$$\Delta S_{univ} = \Delta S_{\bigoplus} + \Delta S_{\bigodot} = 3.5 \times 10^7 - 1.75 \times 10^6 = 3.325 J K^{-1}$$

(b) In order for the grass to grow it needs sunlight. The entropy that is gained in the 1 square meter of grass is many magnitudes higher than the entropy lost in the formation of complex hydrocarbons.

3.14

$$C_V = aT + bT^3 \quad a = 0.00135 \quad b = 2.48 \times 10^{-5}$$

$$S(T_f) - S(T_i) = \int_{T_i}^{T_f} \frac{C_V(T)}{T} dT$$

$$S(T_f) - S(0) = \int_0^{T_f} (a + bT^2) dT$$

$$S(T_f) = \left(aT + \frac{bT^3}{3}\right)\Big|_{T=0}^{T=T_f}$$

$$S(T_f) = aT_f + \frac{bT_f^2}{3}$$

$$S(1) = (0.00135)(1) + \frac{2.48 \times 10^{-5}}{3}(1)^3 = 1.36 \times 10^{-3} J K^{-1}$$
$$\frac{S(1)}{k} = \frac{1.36 \times 10^{-3}}{1.38 \times 10^{-23}} = 9.86 \times 10^{19}$$

$$S(10) = (0.00135)(10) + \frac{2.48 \times 10^{-5}}{3}(10)^3 = 2.18 \times 10^{-2} J K^{-1}$$

$$\frac{S(10)}{k} = \frac{2.18 \times 10^{-2}}{1.38 \times 10^{-23}} = 1.58 \times 10^{21}$$