

# Entropy and the efficiency of refrigeration and air conditioning system

Group Members :Poroma,Urbi,Sudipto, Sadman,Shrewasy

Supervised By: Samiya Mahzabin, Department of Physics, Faculty of Science and Technology, American International University, Bangladesh

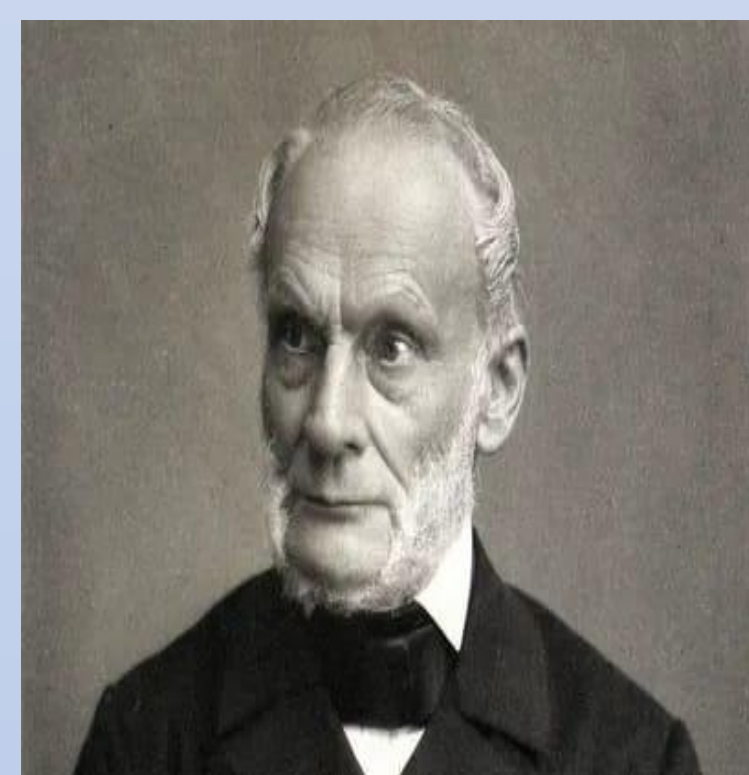


## Abstract

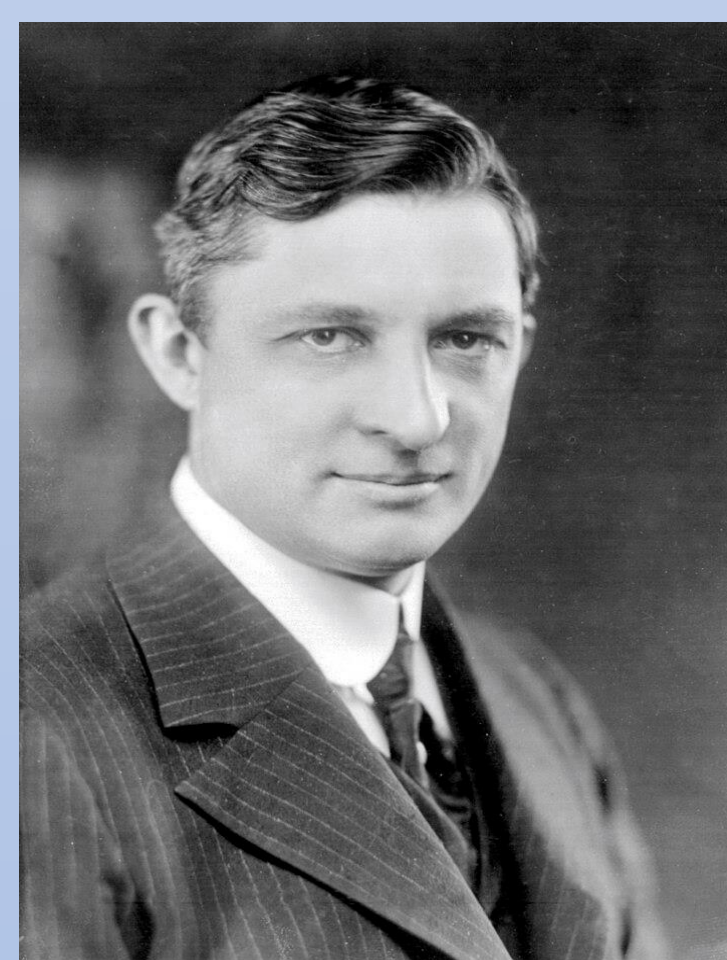
This research delves into the interplay between entropy and the efficiency of refrigeration and air conditioning systems. Through a combination of theoretical analysis and practical examinations, we elucidate the influence of entropy on system performance. Our study aims to enhance understanding and propose strategies for optimizing efficiency in these crucial thermal management systems. This abstract encapsulates the primary objectives, methodologies, and significance of our investigation.

## Background

In the early 1850s, Rudolf Clausius set forth the concept of the thermodynamic system and posited the argument that in any irreversible process a small amount of heat energy  $\delta Q$  is incrementally dissipated across the system boundary. Clausius continued to develop his ideas of lost energy, and coined the term entropy.



On July 17, 1902, Willis Haviland Carrier designed the first modern air-conditioning system, launching an industry that would fundamentally improve the way we live, work and play.

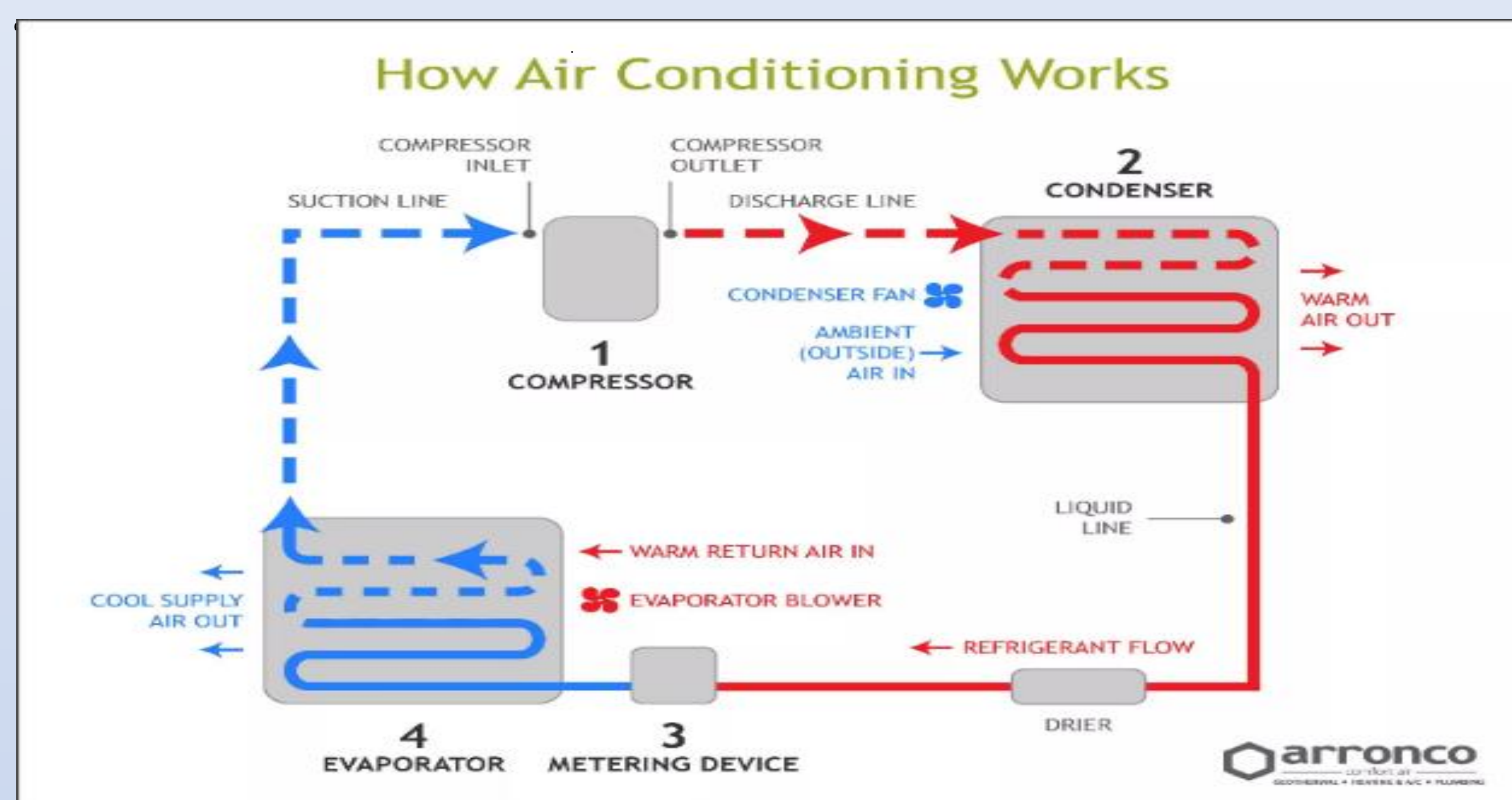
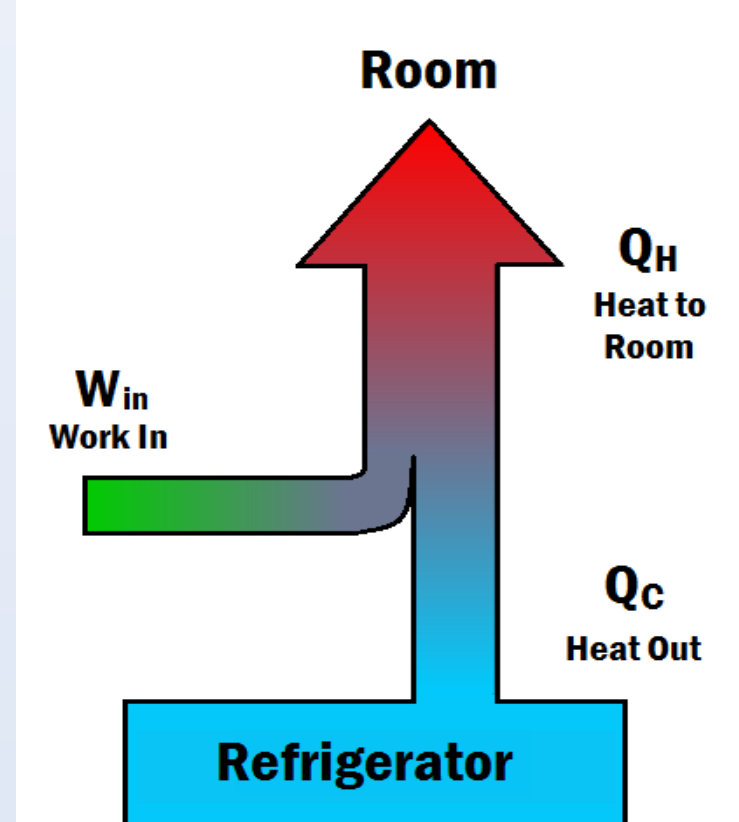


## Methodology

2<sup>nd</sup> law of Thermodynamics how works in these systems: 1. According to the Second Law of Thermodynamics, heat will be somewhat visualized with help from Figure 1 always flow spontaneously from hot to cold, and never the other way around. A refrigerator causes heat to flow from cold to hot by inputting work, which cools the space inside the refrigerator. It does this by following the steps below, which can :  
2. Work is inputted ( ) which compresses a coolant, increasing it's temperature above the room's temperature. 3. Heat flows from this this coolant to the air in the room ( ) , reducing the

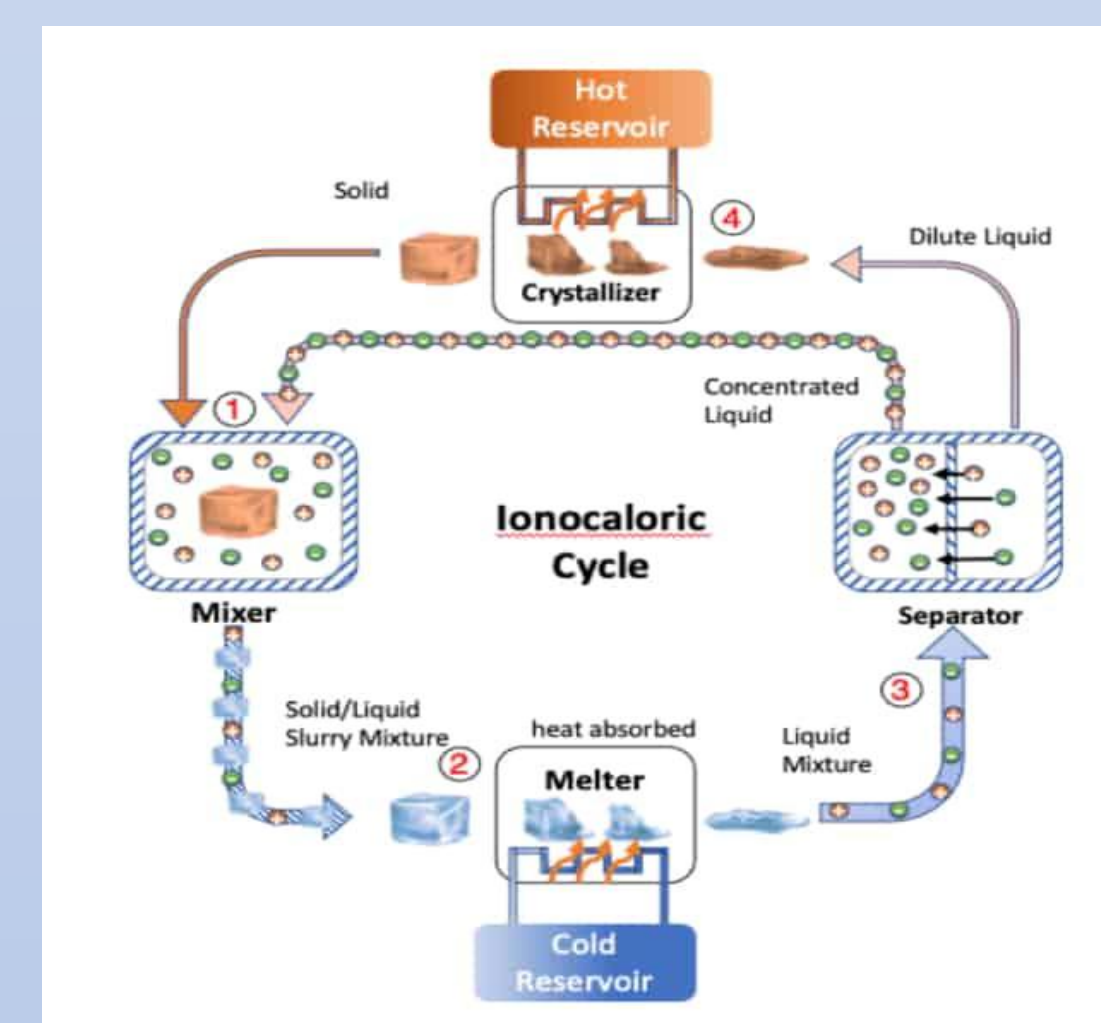
temperature of the coolant.

4. The coolant expands, and it cools down below the temperature inside the refrigerator. 5. Heat flows from the refrigerator to the coolant ( ), decreasing the temperature inside.



## Current Status and Future Scope

The newest air conditioning technology uses renewable energy sources. This type of HVAC technology is not limited to solar energy and natural gas, but also thermally driven heats and colds. The hottest new



HVAC technology uses natural gas to cool buildings, which reduces electricity costs. Over the next three decades, the use of ACs is set to soar, becoming one of the top drivers of global electricity demand. A new analysis by the International Energy Agency shows how new standards can help the world avoid facing such a "cold crunch" by helping improve efficiency while also staying cool.

## Conclusion

Entropy is pivotal for refrigeration and air conditioning efficiency. Managing entropy through effective heat transfer, insulation, and compressor performance is essential for optimizing these systems.

## Reference

1. [https://en.m.wikipedia.org/wiki/History\\_of\\_entropy](https://en.m.wikipedia.org/wiki/History_of_entropy)
2. <https://chasechillers.com/brief-history-of-refrigeration/>
3. <https://www.coynecollege.edu/a-brief-history-of-hvac-air-conditioning/>