


Quantality parameter

In quantum mechanics, the zero-point kinetic energy, $\sim \hbar^2/Ma^2$, involved in the localization of a particle within a volume of radius a implies that the lowest energy state, the particles may be delocalized because the potential energy gain of the ^{simple} classical configuration  of fixed particles, which minimize U is overwhelmed by the quantal kinetic energy. Such delocalized quantal fluids ~~delocalized fluids~~ provide the basis for discussing the state of electrons in atoms, and metals, of the He atoms in the ground state of the He liquids (but fermionic ^3He , and bosonic ^4He), and the state of nucleons in the ground state of atomic nuclei, a (non-newtonian) quantum fluid.

The relative magnitude of the quantal kinetic energy of the localized state compared with the potential energy can be qualitatively characterized by the quantality parameter (Mottelson, 1998)

$$K = \frac{\hbar^2}{Ma^2} \frac{1}{|V_0|},$$

where M is the mass of the individual particles, while V_0 and a measure the strength of the attraction and the range corresponding to the minimum of the potential, respectively. When K is small, quantal effects are small and the lowest state of the system