

Bioactivity Prediction

Advanced Computational Approaches in Drug Discovery

Overview

Bioactivity prediction is a critical component of modern drug discovery, combining computational methods with experimental validation to identify promising therapeutic compounds efficiently.

Activity Cliffs

Small structural changes, large activity differences

Matched Pairs

Systematic SAR analysis

Free Energy Perturbation

Physics-based predictions

Active Learning

Iterative experiment design

Experimental Validation

Wet-lab confirmation

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Activity Cliffs

Definition & Importance

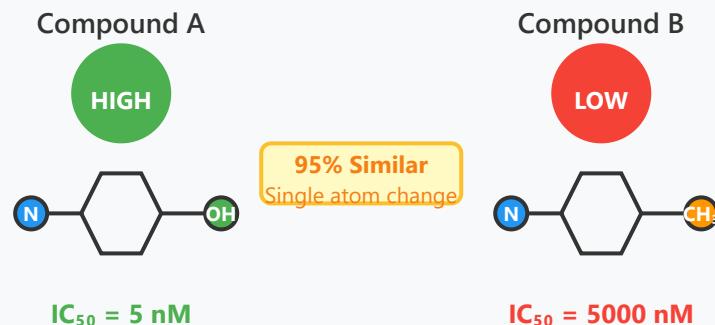
Activity cliffs represent pairs or groups of structurally similar compounds that exhibit dramatically different biological activities. These are among the most challenging and informative features in structure-activity relationship (SAR) studies. Understanding activity cliffs is crucial because they reveal sensitive regions in molecular structure where minor modifications can lead to substantial changes in potency, selectivity, or other pharmacological properties.

Activity cliffs challenge traditional medicinal chemistry assumptions and provide valuable insights into binding mechanisms. They often indicate critical interaction points with the target protein, such as key hydrogen bonds, hydrophobic contacts, or specific conformational requirements that are highly sensitive to structural perturbations.

Key Characteristics

- ▶ High structural similarity (>85% Tanimoto coefficient)
- ▶ Large activity difference (>100-fold potency change)

Activity Cliff Example: Minor Structural Change, Major Activity Impact



1000-fold Activity Loss
 $\text{OH} \rightarrow \text{CH}_3$ substitution

Activity Cliff Interpretation:

The hydroxyl group (OH) forms critical hydrogen bond with target protein. Replacing with methyl (CH_3) eliminates this interaction, drastically reducing potency.

- ▶ Reveal critical SAR features
- ▶ Guide optimization strategies
- ▶ Indicate binding mode sensitivities

Applications

- ▶ Lead optimization prioritization
- ▶ Identification of "hot spots" in molecules
- ▶ Understanding mechanism of action
- ▶ Improving predictive models

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Matched Molecular Pairs (MMP)

Definition & Methodology

Matched Molecular Pair (MMP) analysis is a systematic approach to understanding structure-activity relationships by

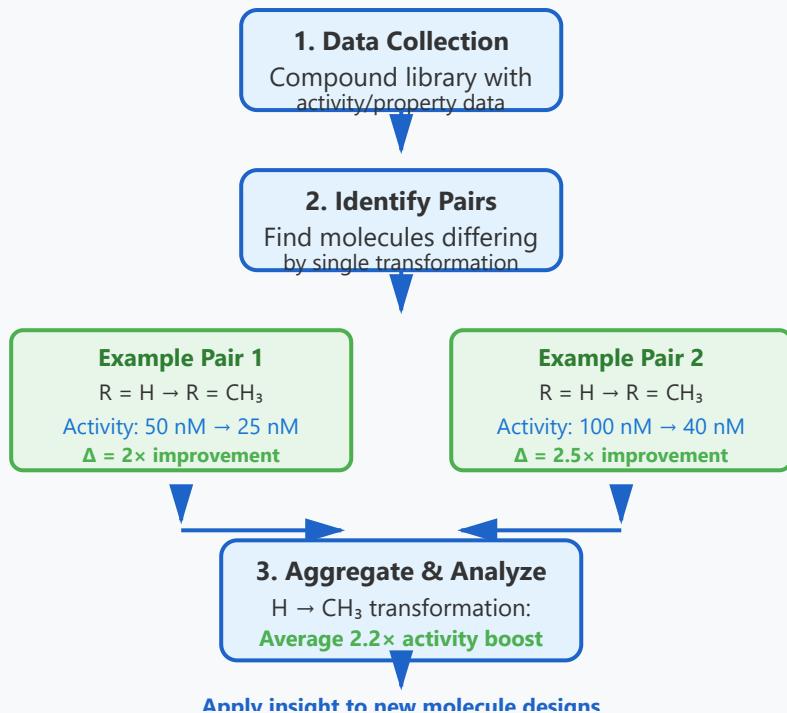
Matched Molecular Pair Analysis Workflow

examining pairs of molecules that differ by a single well-defined structural transformation. This method provides a rigorous framework for quantifying the impact of specific chemical modifications on biological activity, ADME properties, or other molecular characteristics.

The power of MMP analysis lies in its ability to isolate the effect of individual structural changes while holding the rest of the molecule constant. By aggregating data from multiple matched pairs across different chemical series, researchers can derive general rules about the effects of specific transformations, such as the addition of a fluorine atom, methylation of a nitrogen, or replacement of a benzene ring with a pyridine.

Core Principles

- ▶ Single structural transformation between pairs
- ▶ Systematic SAR knowledge extraction
- ▶ Context-dependent transformation effects
- ▶ Statistical aggregation across datasets
- ▶ Transferable medicinal chemistry insights



Applications

- ▶ Property prediction (logP, solubility, permeability)

- ▶ Activity optimization strategies
 - ▶ ADME property improvements
 - ▶ Building design rules for medicinal chemistry
 - ▶ Virtual screening prioritization
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Free Energy Perturbation (FEP)

Physics-Based Approach

Free Energy Perturbation (FEP) is a rigorous computational method rooted in statistical mechanics that predicts binding free energies of molecules to their target proteins. Unlike empirical scoring functions, FEP calculations explicitly account for entropic and enthalpic contributions to binding, providing quantitative predictions of relative binding affinities with chemical accuracy (typically within 1 kcal/mol).

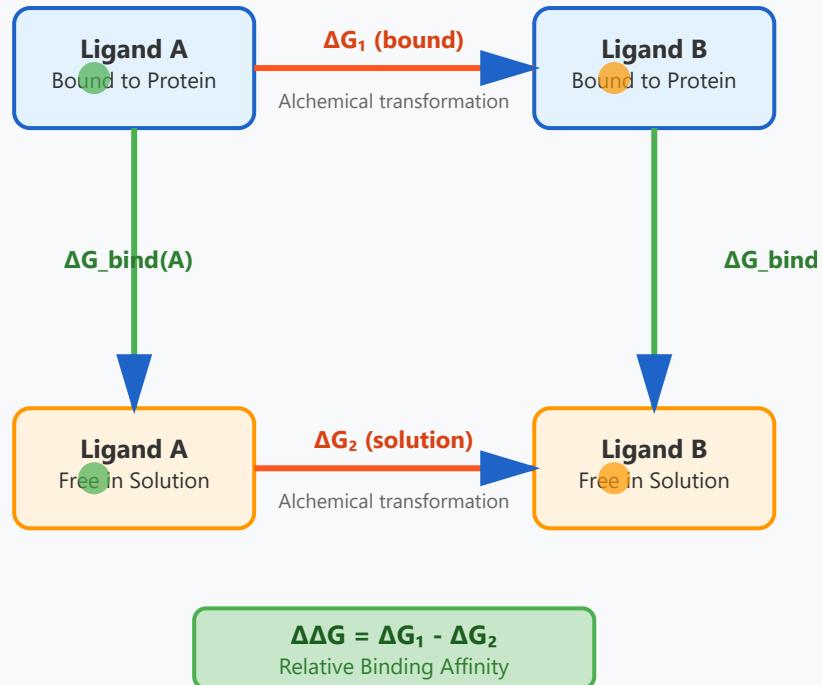
FEP works by computationally "alchemically" transforming one molecule into another while bound to the protein and in solution, calculating the free energy difference between these

Free Energy Perturbation: Thermodynamic Cycle

This approach leverages molecular dynamics simulations to sample conformational space and evaluate ensemble averages, making it particularly powerful for comparing closely related analogs where small structural changes need to be accurately assessed.

Technical Features

- ▶ Alchemical transformation methodology
- ▶ Explicit solvent molecular dynamics
- ▶ Thermodynamic cycle calculations
- ▶ Chemical accuracy (± 1 kcal/mol)
- ▶ Accounts for protein flexibility
- ▶ Considers entropic contributions



Applications & Advantages

- ▶ Lead optimization prioritization
- ▶ Rank-ordering compound synthesis
- ▶ Understanding binding mechanisms
- ▶ Reducing experimental synthesis burden

- ▶ Complementary to experimental assays

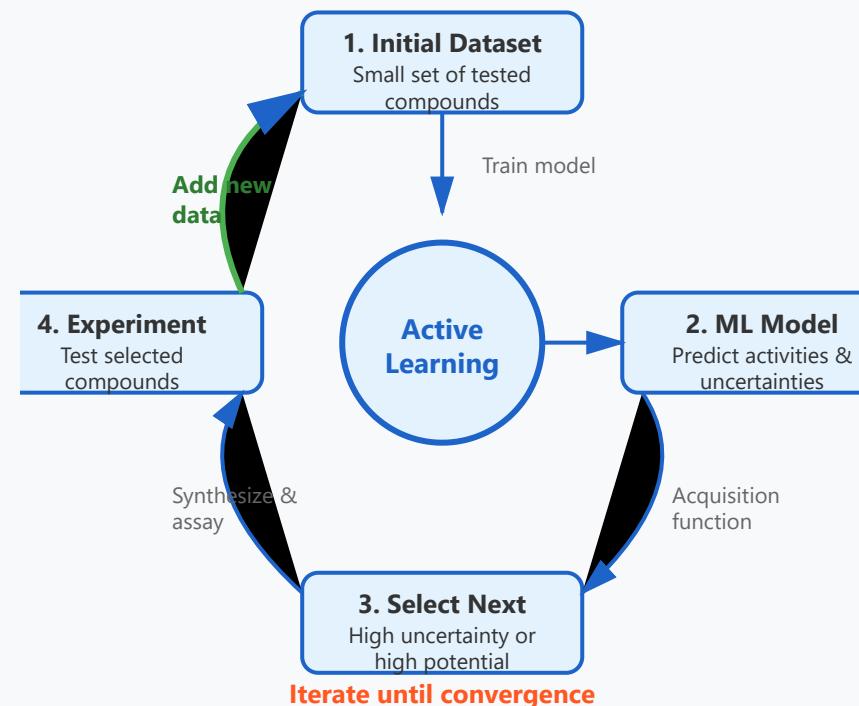
4 Active Learning

Iterative Optimization Strategy

Active learning represents a paradigm shift in drug discovery by intelligently selecting which compounds to test next based on their potential to maximize information gain. Rather than randomly screening large libraries or relying solely on chemist intuition, active learning algorithms identify molecules that are most likely to improve our understanding of the structure-activity landscape, reduce model uncertainty, or explore promising but undersampled chemical space.

The approach combines machine learning models with strategic experimental design. After each round of testing, the model is retrained with new data, and its predictions become more accurate. The algorithm then identifies the next batch of compounds to synthesize and test, focusing on areas where the model is most uncertain or where potential for high activity is greatest. This closed-loop approach dramatically

Active Learning Cycle in Drug Discovery



reduces the number of experiments needed to identify optimal compounds.

Core Concepts

- ▶ Exploitation vs. Exploration balance
- ▶ Uncertainty-based compound selection
- ▶ Model-driven experimental design
- ▶ Iterative model refinement
- ▶ Efficient chemical space navigation

Benefits

- ▶ Reduced experimental costs and time
- ▶ Faster convergence to optimal compounds
- ▶ Better exploration of chemical space
- ▶ Data-efficient optimization
- ▶ Adaptable to changing objectives

5 Experimental Validation

Wet-Lab Confirmation

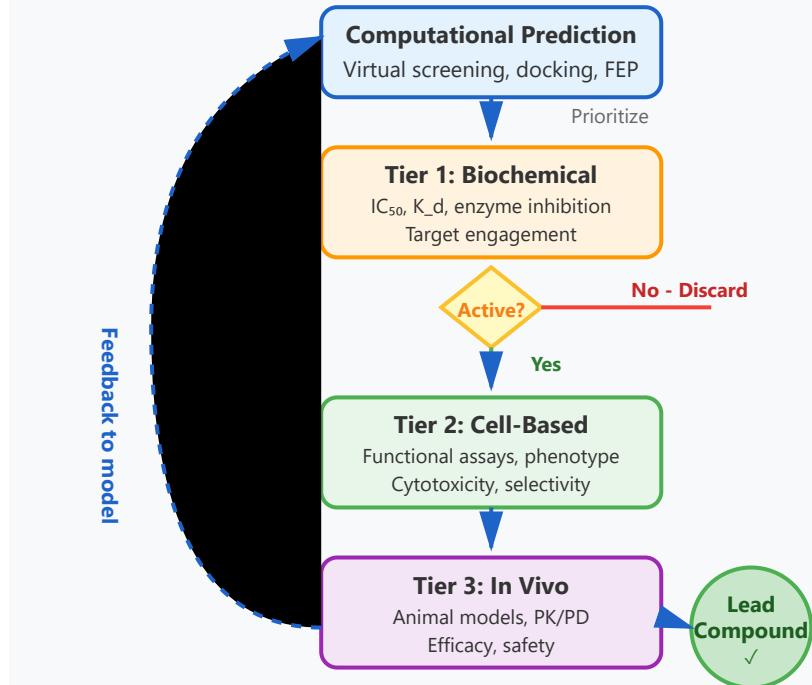
Experimental validation is the critical final step that bridges computational predictions with biological reality. No matter how sophisticated the computational methods, drug discovery ultimately depends on empirical confirmation through well-designed biochemical and cellular assays. Experimental validation not only confirms predictions but also provides essential feedback to refine computational models and improve future predictions.

A comprehensive validation strategy typically involves multiple tiers of assays, starting with target-based biochemical assays (IC_{50} , K_i), progressing through cell-based phenotypic screens, and ultimately advancing to *in vivo* studies in animal models. Each tier provides increasingly complex and biologically relevant information, helping to identify not just active compounds but those with the right balance of potency, selectivity, and drug-like properties necessary for therapeutic development.

Validation Hierarchy

- ▶ Biochemical assays (binding, inhibition)
- ▶ Cell-based functional assays

Experimental Validation Pipeline



- ▶ ADME profiling (absorption, metabolism, etc.)
- ▶ Safety and toxicity screening
- ▶ In vivo efficacy studies

Key Considerations

- ▶ Assay reproducibility and robustness
- ▶ Appropriate positive/negative controls
- ▶ Statistical significance assessment
- ▶ Orthogonal validation methods
- ▶ Translation to physiological relevance
- ▶ Feedback loop to computational models

These methodologies work synergistically to identify and optimize therapeutic candidates efficiently