

Chemical Bonds in Biology

⚡ Bond Energy Comparison

50-200

Covalent

5-10

Ionic

1-5

H-bond

<1

Van der Waals

➡ Covalent Bonds

- Strong electron sharing
- Single, double, triple bonds
- Form backbone of biomolecules
- Energy: 50-200 kcal/mol

↔ Ionic Interactions

- Electrostatic attractions
- Important in protein folding
- Salt bridges stabilize structures
- Energy: 5-10 kcal/mol

�� Hydrogen Bonds

- Weak but numerous
- Critical for DNA base pairing
- Protein secondary structure
- Energy: 1-5 kcal/mol

● Van der Waals Forces

- Weakest interactions
- Important in close packing
- Hydrophobic effect
- Energy: < 1 kcal/mol