

CHEMICAL ENGINEERING DEPARTMENT

End-semester exam, December 2018

2-Tech. (Chemical), Fall Semester, 2018 - 19

Course Code

: CH13102

Duration

: 3 h.

Course Name

: Process Instrumentation

Max. Marks

: 60

Faculty-In-Charge : Dr Dipesh S Patle

Specific Instruction to Students

- Answer the questions sequentially
- Use of non-programmable scientific calculator is permitted ii.

Part A: Answer all questions

- Which of the following is an undesirable dynamic characteristic of an instrument? i.
 - Reproducibility₁ A.
 - Static error B.
 - Dead zone C.

 - Which of the following temperature measuring instruments need not touch the object ii. whose temperature is being measured?
 - Radiation/infrared pyrometer/ À.
 - Filled system thermometer В.
 - Mercury in glass thermometer C.
 - Thermo electric pyrometer D.
 - Which of the following is a desirable characteristic of an instrument? iii.
 - High drift Α.
 - Poor reproducibility B.
 - High fidelity C.
 - High measuring lage. D.
 - Which of the following effect is useful in thermocouples? iv.
 - Seebeck A.
 - Thomson B.
 - Pelitier C.
 - None of the above D.
 - Working principle of radiation pyrometer is based on the
 - Wiens's law A.
 - Kirchoff law B.
 - Stefan-Boltzman law œ.
 - Seebeck effect D.

2. A level sensor, having an operating range of 0.5-20 m, is used in an industry. The sensor, having a transmitter, transmits the level readings in standard voltage form (i.e. 1 VDC to 5 VDC). For this sensor, (a) find out the level when the output from the sensor is 3 VDC. (b) find out the gain (c) what is the 'ZERO' and (d) what is the SPAN. [2+1+1+1 =5 M]

3. Explain why it is important to understand the dynamics of the instrument e.g. first order or second order dynamics. Discuss their typical dynamics in terms of oscillation and response.

12+3 =5 MI

4. Write any three temperature sensors that are based on electrical effects. Describe in details how a RTD works.

[2+3=5 M]

5. Write a short note on the ionization gage instrument for pressure measurement. Also, draw a neat labeled schematic.

[3+2=5 M]

6. Explain how an Ultrasonic Probe can be used for level measurement. Discuss it in terms of accuracy and limitations.

[5 M]

Part 2: Answer the following

7. Describe how pH of a solution is measured electronically. Draw a neat labelled schematic. Explain way temperature correction is necessary for pH meter reading?

[4+2+4= 10 M]

8. Describe chromatography method for analyzing multicomponent systems? Also, explain how a mass spectroscopy works. Discuss why mass spectroscopy is often integrated with gas chromatography.

[4+4+2=10 M]

9. Explain the working principle of an orifice-flowmeter with the help of a schematic. State its major demerits in real application in the industry for flow measurement. Derive an expression for the calculation of flowrate (Q) (Hint: use Bernoulli's equation).

[3+2+5 = 10 M]

10. Explain the working principle of a ventury-flowmeter with the help of a schematic. State its major (de)merits in real application in the industry for flow measurement. Derive an expression for the calculation of flowrate (Q) (Hint: use Bernoulli's equation).

[3+2+5 = 10 M]

(Good	Luck	=
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