Cloud Chamber

Absolute Physics

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A Wilson Cloud Chamber is a particle detecting system used to visualize the path of ionizing radiation. It comprises of a sealed environment containing a supersaturated vapor of alcohol. The interaction between the vapor and ionizing radiation forms streaks in the cloud chamber which help to visualize the passage of ionizing radiation.

Aim

Build a cloud chamber to visualize the path of ionizing radiation

Resources

- 1. Glass tanks/containers
- 2. Hot bag
- 3. Black felt
- 4. Dry ice
- 5. Metal Sheet
- 6. 99% >propanol/ethanol
- 7. Light Source
- 8. Radioactive Sources

Experiment 1

- 1. Attach black felt to underside of glass tank by using plasticine as supports on sides of the tank
- 2. Spray ethanol/propanol on felt
- 3. Crush dry ice and compact it between 2 metal trays
- 4. Place glass tank upside-down on the metal tray, which should be cold due to the dry ice
- 5. Shine a light through and observe the cloud

Observations

- 1. No Cloud observed
 - Problems
 - Rate of evaporation
 - Contact of felt with top surface of glass tank
 - Felt keeps falling
- 2. Ethanol present in liquid form on metal tray obscuring observations
- 3. Lustrous surface of metal tray makes it difficult to observe the presence of a cloud

Improvements made based on observations from Experiment 1

- 1. No Cloud observed
 - Problems
 - Hot bag placed on top of the glass tank to increase the rate of evaporation
 - Use of glue to achieve contact with surface
 - Use of glue to permanently attach felt to the tank surface
- 2. Continually wipe the tray and glass to prevent condensation from obscuring observations
- 3. Painted metal tray matte black to better observe the cloud

Experiment 2

- 1. Black felt attached to underside of glass tank by using glue.
- 2. Spray ethanol/propanol on felt
- 3. Crush dry ice and compact it between 2 metal trays
- 4. Place glass tank upside-down on the metal tray, which should be cold due to the dry ice
- 5. Place hot bag on top of the tank to increase the rate of evaporation of ethanol.
- 6. Shine a light through and observe the cloud
- 7. Before repeating experiment clean surface of metal tray

Observations

- 1. Cloud Observed
- 2. Faint streaks observed