

RESEARCH ARTICLE | MAY 03 2017

Vibronic exciton theory of singlet fission. I. Linear absorption and the anatomy of the correlated triplet pair state

Roel Tempelaar; David R. Reichman



J. Chem. Phys. 146, 174703 (2017)

<https://doi.org/10.1063/1.4982362>

 CHORUS



Nanotechnology &
Materials Science



Optics &
Photonics



Impedance
Analysis



Scanning Probe
Microscopy



Sensors



Failure Analysis &
Semiconductors



Unlock the Full Spectrum.
From DC to 8.5 GHz.

Your Application. Measured.

[Find out more](#)

 Zurich
Instruments

Vibronic exciton theory of singlet fission. I. Linear absorption and the anatomy of the correlated triplet pair state

Roel Tempelaar^{a)} and David R. Reichman^{b)}

Department of Chemistry, Columbia University, 3000 Broadway, New York, New York 10027, USA

(Received 28 February 2017; accepted 10 April 2017; published online 3 May 2017)

Recent time-resolved spectroscopic experiments have indicated that vibronic coupling plays a vital role in facilitating the process of singlet fission. In this work, which forms the first article of a series, we set out to unravel the mechanisms underlying singlet fission through a vibronic exciton theory. We formulate a model in which both electronic and vibrational degrees of freedom are treated microscopically and non-perturbatively. Using pentacene as a prototypical material for singlet fission, we subject our theory to comparison with measurements on polarization-resolved absorption of single crystals, and employ our model to characterize the excited states underlying the absorption band. Special attention is given to the convergence of photophysical observables with respect to the basis size employed, through which we determine the optimal basis for more expensive calculations to be presented in subsequent work. We furthermore evaluate the energetic separation between the optically prepared singlet excited state and the correlated triplet pair state, as well as provide a real-space characterization of the latter, both of which are of key importance in the discussion of fission dynamics. We discuss our results in the context of recent experimental studies. *Published by AIP Publishing.* [<http://dx.doi.org/10.1063/1.4982362>]

I. INTRODUCTION

Singlet fission, the molecular process whereby a singlet excited state is converted into two triplet excitons, has been discussed in the literature for over five decades.^{1,2} In recent years, a renewed focus on the topic has emerged, driven by both fundamental interest and the promise of improved photovoltaic devices.^{3,4} With respect to the latter, the splitting of the absorbed photon energy in two lower-energy packets enables the possibility of circumventing⁵ the Shockley-Queisser efficiency limit⁶ for single-junction photovoltaic materials. Furthermore, the triplet excitons that are produced in the process are stable against radiative losses, having no dipole-allowed optical transitions to the (singlet) ground state. This property facilitates a much longer range of energy transport toward extraction regions than would be possible with singlet excitons.^{7,8} The technological potential of singlet fission has been demonstrated in various proof-of-principle studies using polyacenes as a fission material.^{9–12} Nevertheless, beyond polyacenes, singlet fission has remained a somewhat rare and relatively exotic phenomenon, and a much-needed expansion of the spectrum of suitable materials is hampered by a lack of microscopic understanding of singlet fission through which the determining factors of a complete set of fission materials can be identified.⁴

An important question surrounding singlet fission that has remained incompletely answered is the surprisingly fast time scale of exciton multiplication in certain acene systems such as bulk pentacene. In such systems, efficient fission is

found to occur very rapidly,^{13,14} and evidence exists that a near instantaneous population of crucial fission intermediates concomitantly occurs.^{14–16} Early attempts to account for these observations relied on very strong electronic coupling between the initially excited singlet exciton and the correlated triplet pair state,¹⁴ the fission intermediate which renders the overall process spin-allowed. However such sizable coupling seems inconsistent with most electronic structure calculations.^{17–19} Recent time-resolved spectroscopic studies, seeking to find alternative explanations, have pointed to the importance of vibronic coupling.^{15,16,20} Transient absorption measurements on pentacene derivatives suggested that a conical intersection is the driving mechanism for fission.²⁰ Yet another scenario for pentacene was proposed based on two-dimensional electronic spectroscopy (2DES) in which high-frequency intramolecular vibrations generate a resonance between the singlet and correlated triplet pair states by matching their energy difference.¹⁵ Two-photon photoemission measurements on hexacene were interpreted similarly,¹⁶ despite the very different energetics of this material.

Given the recent indications of the importance of vibronic coupling to singlet fission, a theoretical study of fission materials is called for in which the involved electronic degrees of freedom and interacting intramolecular modes are treated microscopically and non-perturbatively. The aforementioned spectroscopic studies were mostly supported by calculations performed using phenomenological models, including a minimal number of electronic degrees of freedom.^{14–16} Although such modeling is extremely helpful for interpreting the spectroscopic measurements, it does not provide fully microscopic insights at the molecular level. On the other hand, available high-level theoretical studies on fission materials^{17,18,21–23} are

^{a)}Electronic mail: r.tempelaar@gmail.com

^{b)}Electronic mail: drr2103@columbia.edu

numerically constrained to identifying excited states in a time-independent framework for limited cluster sizes. A viable approach to bridge the gap between these calculations and measurements of singlet fission is through microscopic modeling. Such has been proven successful in several studies on fission materials;^{24–30} however, applications did not address the fission dynamics^{24,26,30} or were limited to the perturbative treatment of electronic couplings²⁵ or the vibrational degrees of freedom.^{27–29}

The present article forms the first entry in a series of studies in which we present a microscopic vibronic exciton model aimed at unraveling the mechanistic principles underlying singlet fission, with a particular emphasis placed on the role of vibronic coupling. We provide a detailed description of the model in which both the electronic degrees of freedom and a selected intramolecular vibrational mode are treated non-perturbatively. In part, the applied methodology descends from a recent parametrization of polarization-resolved absorption of single-crystalline pentacene by Hestand *et al.*,³⁰ which we adopt and combine with a treatment of the relevant triplet excited states not previously considered.³⁰ In doing so, we base our parametrization on the first direct detection of the correlated triplet pair state through the aforementioned recent 2DES experiments on pentacene.¹⁵ We will further investigate this measurement in a follow-up companion study focused on 2DES of fission materials.⁵⁷ In order to realize (expensive) 2DES calculations, we address in the present article the convergence of the photophysical observables with respect to the crystal dimensions, from which we determine the optimal crystal size to simulate the photophysics of fission materials on a cost-effective yet accurate basis. This will also pave the way for the final piece in this series, a study of the dynamical evolution of the full singlet fission process by marrying our model with quantum dynamical methods, which forms a critical test for the functional mechanisms hypothesized to underlie this process.³¹

In the present article, we use linear absorption of crystalline pentacene as a means to determine the accuracy of the proposed model. At the same time, the model allows us to characterize the excited states underlying the absorption band, through which we shed light on the energetic separation between the singlet exciton and the correlated triplet pair state. In particular, we demonstrate how this separation is dependent on the applied crystal sizes. Another topic that we address is the spatial configuration of the correlated triplet pair state, which has come under debate with a recently reported triplet-triplet separation of several times the intermolecular distance³² contesting the conventional idea of triplets located at neighboring molecules.^{3,4} The microscopic nature of our model provides direct access to the anatomy of this elusive state, which helps to shed light on this matter.

The paper is organized as follows. In Sec. II, we present the vibronic basis set and Hamiltonian, followed by a discussion of the associated parameters for pentacene. Results for pentacene are presented and discussed in Sec. III. First, the linear absorption spectrum is analyzed, highlighting the composition of the underlying excited states. What follows is an evaluation of the excited state energies, and other photophysical properties, in relation to the applied crystal size. Finally,

we analyze the spatial composition of the correlated triplet pair state. We conclude in Sec. IV.

II. THEORY

Before presenting our theoretical framework, we begin with a few remarks on the distinction between adiabatic and diabatic excited states. Adiabatic states (eigenstates of the Hamiltonian of the fission material) belong to the quantum basis that is accessible through spectroscopy. Important examples of such states in the context of singlet fission are the optically prepared singlet excited state S_1 and the subsequently populated correlated triplet pair state TT_1 , where it is understood that both labels in reality might represent a manifold of energetically closely spaced adiabatic states. Nevertheless, we instead formulate our model in the diabatic picture (also referred to as site basis or molecular basis). The underlying approximations, as well as the utility of the diabatic representation to describe singlet fission, are extensively discussed in Ref. 27. From our diabatic calculations, the adiabatic spectroscopic observables are readily extracted. In order to clearly distinguish between these two bases, we will consistently denote diabatic states with lower case labels such as s_1 and t_1 . On the other hand, it should be understood that adiabatic states such as S_1 and TT_1 are never composed entirely of s_1 and t_1 , respectively. Rather, they consist of some admixture of the two, additionally mixed with other states such as diabatic charge transfer (CT) states.^{24,26,29,30} In this sense, the adiabatic states are labeled so as to distinguish between a predominant singlet or triplet composition.

A. Basis set

Similarly to Refs. 24–30, our diabatic basis comprises the many-body states formed by excitations within the minimal active space of all highest and lowest unoccupied molecular orbitals (HOMOs and LUMOs) of the *isolated* molecules. Explicit formulations of the symmetry-adapted linear combinations of the diabatic basis states can be found in Ref. 27. In addition to the ground state s_0 , in which the HOMO is fully occupied, we include for each molecule the aforementioned singlet excitations through the basis states $|(s_1)_m\rangle$. Here, an electron is promoted to the LUMO on a molecule labeled m , leaving a hole in the corresponding HOMO, and with all other molecules in their respective s_0 state. We also consider charge transfer (CT) states, for which a hole (cation) and an electron (anion) are found at different molecules. These states are denoted $|(c)_m, (a)_{m'}\rangle$, with m and m' labeling the molecular sites of the cation and anion, respectively. Likewise, the correlated triplet pair state is accounted for by the basis states $|(t_1)_m, (t_1)_{m'}\rangle$, where two triplets located at molecules m and m' , respectively, are entangled in an overall spin-zero state.

In addition to the electronic degrees of freedom, we explicitly include a single intramolecular vibrational mode in the basis set. Such an approach has been employed in earlier microscopic studies on fission materials,^{24,26,30} although there is only a single report in which the basis encompassed the triplet states relevant to fission.²⁶ We note that multiple modes have been explicitly included in the phenomenological

model used to support the recent 2DES study on singlet fission in pentacene.¹⁵ Although such an extension of the basis set is straightforward in principle, it becomes numerically prohibitive at the microscopic level employed in this work. We therefore limit ourselves to the explicit treatment of the spectrally most dominant mode, as was done in the aforementioned microscopic studies.^{24,26,30} We do bear in mind, however, that such an approach can easily be combined with a Redfield-type propagation scheme in which the remainder of the vibrations are treated perturbatively.

A full quantum treatment of the vibrational mode, in principle, entails a basis set that scales very unfavorably with increasing number of molecules. Fortunately, previous studies have demonstrated that the photophysics of fission materials is typically well accounted for when limiting the basis to so-called “two-particle” states,^{24,26,30} in which the eigenstates are decomposed into sums of pairwise (electronic and/or vibrational) excitations, an approximation that dramatically reduces the relevant Hilbert space.^{33,34} To keep computations manageable, we therefore restrict ourselves to the two-particle basis. For the ease of discussion, we proceed to formulate our theory in this basis, although noting that a general formulation in the multi-particle basis is straightforward. Accordingly, we formulate (adiabatic) eigenstates of the Hamiltonian of the material as

$$\begin{aligned}
 |\alpha\rangle = & \sum_{m,\tilde{\nu}} c_{\alpha;m,\tilde{\nu}}^{s_1} |(s_1)_m, \tilde{\nu}\rangle \\
 & + \sum_{m \neq m'} \sum_{\tilde{\nu}} \sum_{\nu' \geq 1} c_{\alpha;m,\tilde{\nu},m',\nu'}^{s_1 s_0} |(s_1)_m, \tilde{\nu}, (s_0)_{m'}, \nu'\rangle \\
 & + \sum_{m \neq m'} \sum_{\nu_+, \nu_-} c_{\alpha;m,\nu_+,m',\nu_-}^{ca} |(c)_m, \nu_+, (a)_{m'}, \nu_-\rangle \\
 & + \sum_{m > m'} \sum_{\tilde{\nu}, \tilde{\nu}'} c_{\alpha;m,\tilde{\nu},m',\tilde{\nu}'}^{t_1 t_1} |(t_1)_m, \tilde{\nu}, (t_1)_{m'}, \tilde{\nu}'\rangle. \quad (1)
 \end{aligned}$$

Here, the first summation extends over the vibronic basis states involving the electronic excitation $|(s_1)_m\rangle$ accompanied by $\tilde{\nu}$ vibrational quanta in the associated nuclear potential, while $|(s_1)_m, \tilde{\nu}\rangle$ is a short-hand notation for $|(s_1)_m\rangle \otimes |\tilde{\nu}\rangle$. In a similar fashion, the second summand contains basis states for which such a vibronic excitation is accompanied by a purely vibrational excitation involving $\nu' (\geq 1)$ quanta in the s_0 potential located at molecule m' . The subsequent term contains the CT states with ν_+ and ν_- quanta in the cationic and anionic potentials, respectively. The final summation extends over the triplet pairs, where the associated vibrational quanta $\tilde{\nu}$ and $\tilde{\nu}'$ refer to the t_1 vibrational potential. Here, the summation is restricted to $m > m'$ in order to avoid double counting.

B. Hamiltonian

The expansion coefficients in Eq. (1) are obtained upon numerically solving the eigenvalue equation, $\hat{H}|\alpha\rangle = \omega_\alpha|\alpha\rangle$, where \hat{H} is the Hamiltonian and ω_α represents the eigenenergy associated with the state α ($\hbar = 1$ is assumed throughout this paper). For clarity, we subdivide the Hamiltonian into parts,²⁶

$$\begin{aligned}
 \hat{H} = & \hat{H}_{s_1} + \hat{H}_{s_1-ca} + \hat{H}_{ca} + \hat{H}_{ca-t_1 t_1} + \hat{H}_{t_1 t_1} \\
 & + \hat{H}_{\omega_0} + \hat{H}_\lambda, \quad (2)
 \end{aligned}$$

each associated with different physical mechanisms. Note that no contribution associated with the electronic ground state s_0 is included here, implying that this state is associated with the zero point of energy.

The first part of the Hamiltonian is given by

$$\begin{aligned}
 \hat{H}_{s_1} = & E_{s_1} \sum_m |(s_1)_m\rangle\langle(s_1)_m| \\
 & + \sum_{m \neq m'} J_{m,m'} |(s_1)_m\rangle\langle(s_1)_{m'}| \quad (3)
 \end{aligned}$$

and accounts for the diagonal energies of the diabatic s_1 states, denoted as E_{s_1} and taken to be equal for all molecules. This part also describes dipole-dipole interactions between these states, denoted as $J_{m,m'}$.

The second part describes the dissociation of the Frenkel-type s_1 states into CT excitons,

$$\begin{aligned}
 \hat{H}_{s_1-ca} = & \sum_{m \neq m'} t_{m,m'}^{HH} |(s_1)_m\rangle\langle(c)_{m'}, (a)_m| \\
 & + \sum_{m \neq m'} t_{m,m'}^{LL} |(s_1)_m\rangle\langle(c)_m, (a)_{m'}| + \text{H.c.}, \quad (4)
 \end{aligned}$$

which is mediated by the HOMO-HOMO and LUMO-LUMO charge overlap integrals, $t_{m,m'}^{HH}$ and $t_{m,m'}^{LL}$, transferring electrons and holes, respectively.

The third part is given by

$$\begin{aligned}
 \hat{H}_{ca} = & \sum_{m \neq m'} (U_{m,m'} + E_{s_1}) |(c)_m, (a)_{m'}\rangle\langle(c)_m, (a)_{m'}| \\
 & + \sum_{m \neq m' \neq m''} t_{m,m'}^{HH} |(c)_m, (a)_{m''}\rangle\langle(c)_{m'}, (a)_{m''}| \\
 & + \sum_{m \neq m' \neq m''} t_{m,m'}^{LL} |(c)_{m''}, (a)_m\rangle\langle(c)_{m''}, (a)_{m'}|, \quad (5)
 \end{aligned}$$

and accounts for separated charges at molecules m and m' , whose Coulomb energy relative to the diabatic energy of s_1 is given by $U_{m,m'}$, as well as the short-ranged CT interactions mediated by the charge overlap integrals.

The next two parts are associated with the triplet pair excitations. The first one couples such states to CT states,

$$\begin{aligned}
 \hat{H}_{ca-t_1 t_1} = & \sqrt{\frac{3}{2}} \left(\sum_{m \neq m'} t_{m,m'}^{HL} |(a)_m, (c)_{m'}\rangle\langle(t_1)_m, (t_1)_{m'}| \right. \\
 & \left. + \sum_{m \neq m'} t_{m,m'}^{LH} |(c)_m, (a)_{m'}\rangle\langle(t_1)_m, (t_1)_{m'}| \right) + \text{H.c.}, \quad (6)
 \end{aligned}$$

where $t_{m,m'}^{HL}$ denotes the charge overlap integral which transfers an electron from the HOMO of molecule m to the LUMO of molecule m' and oppositely for $t_{m,m'}^{LH}$. The second part describes the diagonal energies of the triplet pairs,

$$\hat{H}_{t_1 t_1} = E_{t_1 t_1} \sum_{m > m'} |(t_1)_m, (t_1)_{m'}\rangle\langle(t_1)_m, (t_1)_{m'}|, \quad (7)$$

quantified by $E_{t_1 t_1}$ and again assumed site-independent.

The last two parts of the Hamiltonian relate to the intramolecular vibrational mode. Assuming a harmonic potential, the total vibrational energy is given by

$$\hat{H}_{\omega_0} = \omega_0 \sum_m \hat{b}_m^\dagger \hat{b}_m, \quad (8)$$

where ω_0 is the vibrational quantum ($\hbar = 1$), and $\hat{b}_m^{(\dagger)}$ is the annihilation (creation) operator pertaining to the s_0 potential

at molecule m . This mode is coupled linearly to the electronic diagonal energies (Holstein-type interaction) through

$$\hat{H}_\lambda = \omega_0 \sum_{x=s_1, c, a, t_1} \sum_m \lambda_x (\hat{b}_m^\dagger + \hat{b}_m + \lambda_x) |(x)_m\rangle \langle (x)_m|. \quad (9)$$

Here, x runs over all possible diabatic states associated with each molecule. λ_x represents the shift of the vibrational equilibrium position for each state, using the s_0 potential as a reference. The square of the vibrational displacements relative to the ground state, λ_x^2 , yields the associated Huang-Rhys (HR) factor. Since in our approach all vibrational states are represented in their respective eigenbasis, the electronic interaction terms contributing to the Hamiltonian results in a mixing between such states, dictated by vibrational overlap factors.³⁵

Extension of Eq. (2) to include two-electron couplings is straightforward (and essentially invokes a direct coupling between the singlet excited states and triplet pairs, $\hat{H}_{s_1-t_1 t_1}$). Nevertheless, their magnitudes are known to be an order of magnitude smaller than the one-electron integrals discussed above,^{4,17,22,28,36} and a significant contribution of such couplings to fission dynamics is therefore unlikely, at least in systems like pentacene.²⁸ We have performed additional calculations including typical values for the two-electron couplings and do not find noteworthy differences for the results reported in this work. We therefore disregard such couplings for the sake of simplicity. However, a brief discussion of the significance of two-electron couplings between local triplet excitations is presented in Sec. III C.

C. Parameters for pentacene

Pentacene arguably is the most widely studied fission material, attracting great interest due to its nearly unrivaled fission rate and high fission efficiency. At the same time, the rapid and efficient fission dynamics found for this material poses a challenge for theoretical modeling. These factors, as well as the wealth of information known about this material, motivate us to use pentacene as an exemplary singlet fission system in this series of articles. We expect many of our conclusions to be general for other strongly coupled acene crystals. The parameters for pentacene used in our calculations are to a large extent identical to those reported by Hestand *et al.*,³⁰ which were obtained through a combination of quantum chemical calculation and fitting to polarized absorption spectra of pentacene single crystals.^{26,30} We proceed with a detailed discussion of these parameters, in particular highlighting the adaptations made in our modeling. We note the qualitative, and in some cases semi-quantitative, agreement with the parameters obtained via a similar approach by Berkelbach *et al.* for pentacene dimers²⁸ and single crystals.²⁹ This similarity gives confidence in the robustness of the general philosophy of our work and these past studies.

In the crystal phase, pentacene molecules form a triclinic system with two inequivalent molecules per unit cell.^{37,38} This is illustrated in Fig. 1, showing the crystal structure taken from Ref. 37, including the crystallographic axes. As can be seen, the long axes of the pentacene molecules are oriented roughly along the c -axis, while the short axes form the characteristic herringbone structure in the ab -plane. Similarly to the work

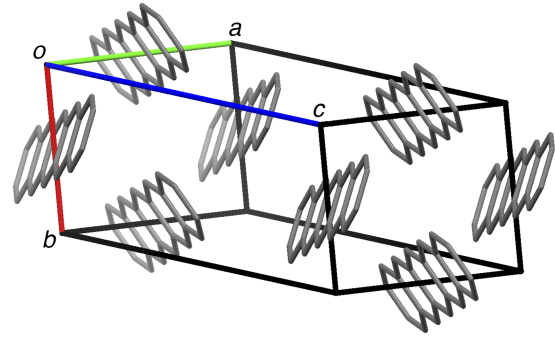


FIG. 1. Crystal structure of pentacene, taken from Ref. 37. The crystallographic a -, b -, and c -axis are indicated with green, red, and blue, respectively. The unit cell contains two inequivalent molecules with their long molecular axes oriented roughly along c , while their short axes form a characteristic herringbone structure in the ab -plane.

of Hestand *et al.*,³⁰ we consider in our calculations a herringbone lattice consisting of $M \times M$ unit cells extending in the ab -plane, which corresponds to $2M^2$ molecules (such that m runs from 1 to $2M^2$). The lengths of the a and b lattice constants are found to be 7.71 and 6.28 Å, respectively, with a mutual angle of 84.5°, while the separation between the inequivalent molecules within the same cell is given by $(a, b) = (1/2, 1/2)$.³⁷ The molecular positions are determined based on this structural data by taking for each molecule its point of inversion symmetry.

The most important parameters applied in our model are summarized in Table I, while the electronic couplings are presented in the [supplementary material](#). The dipole-dipole couplings $J_{m,m'}$ in crystalline pentacene are rather weak and have therefore been neglected by Berkelbach *et al.*^{28,29} Here, we use values reported by Hestand *et al.*,³⁰ which have been calculated through INDO/CCSD using the coordinates from Ref. 37 and screened (divided) by a directionally averaged optical dielectric constant $\epsilon = 3.5$.³⁹

The applied one-electron HOMO-HOMO and LUMO-LUMO couplings, $t_{m,m'}^{HH}$ and $t_{m,m'}^{LL}$, were originally reported in Ref. 26 based on density functional theory (using B3LYP and a double zeta basis set). Similarly to the work of Hestand *et al.*,³⁰ we apply a scaling factor of 1.1 in order to reproduce the Davydov splitting observed for pentacene single crystals. We further note that Berkelbach *et al.* found somewhat larger, yet qualitatively similar couplings through *ab initio* calculations using the Hartree-Fock molecular orbitals of isolated pentacene molecules and a 6-31G(d) basis set.²⁸ However, a

TABLE I. Parameters applied in our model. For electronic couplings, see the [supplementary material](#).

Parameter	Symbol	Value	
s_0-s_1 energy	E_{s_1}	2.09 eV	(16 890 cm ⁻¹)
$s_0-t_1 t_1$ energy	$E_{t_1 t_1}$	1.75 eV	(14 140 cm ⁻¹)
Vibr. energy	ω_0	0.17 eV	(1380 cm ⁻¹)
Huang-Rhys factors			
s_0-s_1	$\lambda_{s_1}^2$	1.1	
s_0-a	λ_a^2	0.29	
s_0-c	λ_c^2	0.39	
s_0-t_1	$\lambda_{t_1}^2$	1.1	

dramatic down-scaling of these couplings was found to be necessary in order to reproduce the Davydov components.²⁹ We point out that such is not necessary once vibronic coupling is accounted for, which by itself has a strong mitigating impact on the Davydov splitting (see, for example, Fig. 5 from Ref. 30, where a direct comparison is drawn between vibronic and purely electronic calculations).

The model applied by Hestand *et al.*³⁰ did not include triplet states, hence at this point our modeling and parameters start to differ from those used in that work. For the HOMO-LUMO couplings $t_{m,m'}^{\text{HL}}$, not reported by Hestand *et al.*, we adapt the couplings calculated by Berkelbach *et al.*²⁸ Similarly to their work on crystalline pentacene,²⁹ we apply an overall rescaling of these couplings, while keeping the signs and relative differences among them unaltered. In order to find an appropriate scaling factor, we have determined the average magnitude of the HOMO-HOMO and LUMO-LUMO couplings of Berkelbach *et al.* relative to those by Hestand *et al.* (found to be 0.68), and rescaled the HOMO-LUMO couplings accordingly.

By employing the vibronic basis set outlined in Sec. II A, we provide an exact treatment of the symmetric stretching vibration with a frequency $\omega_0 = 0.17$ eV (1380 cm^{-1}) which is the dominant progression building mode observed in absorption spectra of both dissolved and crystalline pentacene.^{24,26,30} Its HR factor associated with the diabatic singlet state is taken to be $\lambda_{s_1}^2 = 1.1$, a value derived from fitting the vibronic peak areas of dissolved pentacene to a Poisson distribution.³⁰ The ionic HR factors ($\lambda_a^2 = 0.29$ and $\lambda_c^2 = 0.39$)³⁰ are based on calculations reported in Ref. 40. We note that these values are significantly smaller than the singlet state HR factor, something that has also been observed in terrylene crystals.^{41,42} Relatively little is known about the HR factor pertaining to t_1 . In Ref. 26, the weak dependence of linear absorption on $\lambda_{t_1}^2$ was noted and an *ad hoc* value of 1.0 was applied. By contrast, a value of 0 was used in the phenomenological modeling reported in Ref. 15, based on a fitting to 2DES data. However, recent calculations on tetracene have shown the effective HR factor of triplet pair states to be competitive with $\lambda_{s_1}^2$.⁴³ Furthermore, vibrational overtones associated with ω_0 observed in S_0 - T_1 absorption of anthracene and its derivatives using the heavy-atom effect suggests this factor to be substantial.⁴⁴ Expecting the HR factors of acenes to be reasonably similar, we therefore set $\lambda_{t_1}^2 = 1.1$.

The diagonal energies appearing in the Hamiltonian are notoriously difficult to quantitatively determine based on unaltered first-principles quantum chemical methods (see also the discussion in Ref. 27). Most microscopic models for fission materials have instead used values taken from experiments, mostly through fitting to linear absorption.^{24,26,29,30} For example, such fitting was applied in Ref. 26 to determine the Coulomb energies $U_{m,m'}$ of crystalline pentacene. Here we use the slightly modified values proposed by Hestand *et al.*³⁰ (which are summarized in the [supplementary material](#)). Regarding the diabatic $s_0 \rightarrow s_1$ transition energy, our value is slightly red-shifted relative to the one reported in that work,³⁰ namely, $E_{s_1} = 2.09$ eV (16890 cm^{-1}). Still, this value is higher than the energy used by Berkelbach *et al.*²⁹ by about 100 meV. The origin of this discrepancy is discussed in

Sec. III B. Lastly, the energy of the diabatic triplet pair excitation is taken to be $E_{t_1 t_1} = 1.75$ eV (14140 cm^{-1}), which yields adiabatic TT_1 state energies in agreement with 2DES measurements¹⁵ as discussed in Sec. III B.

In order to keep calculations computationally manageable, the basis set presented in Sec. II A is minimized as much as possible while retaining convergence for the photophysics of interest. Similarly to Hestand *et al.*,³⁰ the maximum charge separation is set to 20 Å (where the separation is based on the molecular positions). Instead of also adapting this value for the maximum separation for non-ionic two-particle states,³⁰ we have found that a smaller truncation radius of 7 Å suffices. The physical origin of such a truncation lies in the finite extent of the vibrational distortion surrounding the electronic excitation.⁴⁵ We furthermore reduced the total number of vibrational quanta per basis state to 2. The maximum separation for the triplet pair excitations was limited to 6.2 Å such that only nearest-neighbors in the $(a, b) = (1/2, 1/2)$ and $(a, b) = (1/2, -1/2)$ directions contribute. The case of a basis with a more extended set of pair states is explored in Sec. III C. Lastly, periodic boundary conditions are imposed. Accordingly, the coupling between two molecules is determined based on the minimal real-space separation of these molecules resulting from any translation of one molecule by multiples of M lattice spacings in the a - and b -directions.

III. RESULTS AND DISCUSSION

A. Linear absorption and excited state admixtures

Earlier studies of the theory of singlet fission have employed the comparison to experimental linear absorption as a means of obtaining parameters and as a critical test of the modeling itself.^{24,26,29,30} Such an approach has proven to be particularly fruitful when applied to polarization-resolved spectra of single crystals, which provides detailed insight into the involved excited states.^{27,30} The theory presented here is partly derived from earlier work on pentacene which centered on such a comparison.³⁰ In a follow-up article, we apply an additional stringent test via the comparison to 2DES measurements. Nonetheless, we will start by revisiting linear absorption to evaluate the modifications made with respect to the earlier model,³⁰ in particular focusing on the effect of added triplet states, and to characterize in detail the excited states underlying the absorption band.

Within the framework of Fermi's Golden Rule, the j -polarized linear absorption spectrum is expressed as

$$A_j(\omega) = \sum_{\alpha} |\langle S_0 | \hat{M}_j | \alpha \rangle|^2 W_j(\omega_{\alpha} - \omega). \quad (10)$$

Here, \hat{M}_j is the j -component of the transition dipole moment operator $\hat{\mathbf{M}}$ of the fission material given by

$$\hat{\mathbf{M}} = \sum_m \boldsymbol{\mu}_m | (s_0)_m \rangle \langle (s_1)_m | + \text{H.c.}, \quad (11)$$

where $\boldsymbol{\mu}_m$ is the transition dipole moment associated with the $s_0 \rightarrow s_1$ optical transition at molecule m . Note that molecular transition dipole moments are three-dimensional vector objects (and thus, so is the transition dipole moment operator) which are indicated using bold characters to distinguish

them from scalar objects. For pentacene, the molecular transition dipole moments are oriented along the short molecular axes,^{3,46} and the applied vector components of μ_m are accordingly extracted from the crystal structure.³⁷ The transition dipole moment operator \hat{M} acts exclusively on the electronic subspace, effectively inducing a “vertical” transition, and as such mixes vibrational states through vibrational overlap factors.⁴⁷ Also appearing in Eq. (10) are the adiabatic excited states α and associated energies ω_α . These states are coupled through the transition dipole moment operator to the ground state S_0 for which all molecules reside in the electronically and vibrationally unexcited state $|(s_0)_m, \nu = 0\rangle$. And lastly, $W_j(\omega)$ represents the j -polarized line shape function.

Polarized linear absorption of single-crystalline pentacene is calculated through Eq. (10) using a Gaussian lineshape $W_j(\omega) \propto \exp(-\omega^2/2\sigma_j^2)$. In Ref. 30, a polarization-dependent line width σ_j was applied to match the measurements, although the physical origin of this property was left undiscussed. Here, we adjust the line widths accordingly in order to unambiguously evaluate the modifications made in our modeling, setting $\sigma_j = 46$ meV (374 cm^{-1}) for polarization j along the crystallographic b -axis, and $\sigma_j = 27$ meV (214 cm^{-1}) otherwise.³⁰ Shown in Fig. 2(a) are spectra calculated for a 6×6 crystal (72 molecules, 24 408 basis states), polarized parallel to b with its perpendicular component in the ab -plane, alongside the experimental equivalents recorded for single-crystalline pentacene.³⁰ When compared to the results reported in Ref. 30, we observe that an excellent agreement is retained upon the modifications made to the modeling. In particular, the Davydov-splitting (energy difference between the lowest-energy peaks in the $\parallel b$ and $\perp b$ spectra) is reproduced, and most of the band features are accounted for. We note that a damping of transition above 2.3 eV was applied in Ref. 30, representing rapid relaxation resulting from an enhanced density of states at higher energies, which further improves the agreement.³⁰ We will nevertheless omit such empirical refinements for now, in order not to further complicate the model, although we plan to address the effects of relaxation in a future study. Finally, we have performed comparative calculations excluding the triplet pair excitations from the basis set (see [supplementary material](#)), which demonstrate that the impact of these states on linear absorption is generally negligible, with a weak imprint only observable for the second peak polarized perpendicular to b .

Since pentacene single-crystals are difficult to grow, most time-resolved spectroscopic measurements on pentacene,^{13,48–50} including the recent 2DES measurements,¹⁵ have been performed on polycrystalline samples instead. In what follows, we mimic polycrystallinity by considering unpolarized linear absorption, consisting of the sum of the aforementioned $\parallel b$ and $\perp b$ components plus the component perpendicular to the ab -plane, which corresponds to a hypothetical sample of isotropic crystal fragments. Actual polycrystalline pentacene is isotropic only in the ab -plane.^{51,52} However, by showing that unpolarized calculations form a good proxy for this situation we simplify the 2DES simulations presented in a follow-up article. The calculated unpolarized linear absorption spectrum is shown in Fig. 2(b). The spectrum

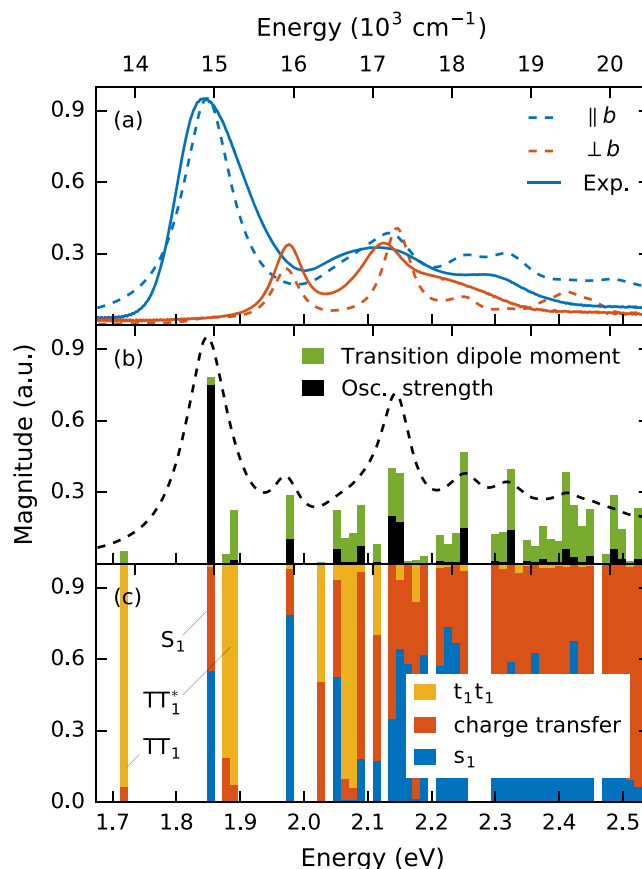


FIG. 2. (a) Polarization-resolved linear absorption measured for pentacene single crystals (solid curves),³⁰ shown together with the numerical results from our model for a 6×6 pentacene crystal (dashed). (b) Calculated unpolarized linear absorption (dashed). Also shown is a stick spectrum representing the transition dipole moments (green) and oscillator strengths (black) associated with the underlying optical transitions. (c) Dissection of the stick spectrum into diabatic singlet (blue), charge transfer (red), and triplet pair contributions (yellow).

closely resembles reported measurements of polycrystalline pentacene,^{53,54} indicating that our procedure for polycrystallinity indeed appears to be valid for this material. The reason for this is simply that the molecular transition dipole moments of pentacene, being oriented along the short molecular axes, lie largely in the ab -plane (see Fig. 1) so that in our approach the isotropy is always projected on this plane, just as is the case for polycrystals.

Also shown in Fig. 2 is a characterization of the excited states underlying the absorption band. Panel (b) presents a stick spectrum, representing a histogram of the oscillator strengths and transition dipole moments underlying the absorption band, which are defined as $\sum_j |\langle S_0 | \hat{M}_j | \alpha \rangle|^2$ and its square root, respectively. In an accompanying histogram shown in Fig. 2(c), each underlying state $|\alpha\rangle$ is dissected into contributions from diabatic singlet states (s_1), CT states, and triplet pair states ($t_1 t_1$). One aspect that these data reveal is that the direct mixing between s_1 and $t_1 t_1$ is minimal, albeit non-zero. As a consequence, a direct excitation of triplets through strong mixing into S_1 , as proposed in Ref. 14, seems unlikely based on our data. CT states clearly act as a facilitator of the interaction between singlet and triplets, having significant admixtures in both TT_1 and S_1 . (No significant additional mixing results from the direct two-electron diabatic coupling,

the effect of which we have found to be minimal and is therefore ignored here.) In the low-energy region, most of the oscillator strength is concentrated in two Davydov components. Consistent with earlier reports,^{24,26,29,30} the lower component at 1.85 eV (14 900 cm⁻¹) is found to consist of an almost 50%-50% admixture of s_1 and CT states. The upper component at 1.97 eV (15 900 cm⁻¹), on the other hand, has a more pronounced s_1 contribution, whereas CT states dominate overall at higher energies.³⁰ By contrast, we find most of t_1t_1 to be concentrated in a limited number of states with a small admixture of CT contributions and very small oscillator strengths owing to the intensity borrowing from s_1 . By adjusting the diabatic triplet pair energy $E_{t_1t_1}$, we have located the lowest-energy candidates for such triplet-dominated states at 1.72 eV (13 860 cm⁻¹), which is consistent with the direct observation of the fission product states through 2DES,¹⁵ as is further discussed in Sec. III B.

In order to connect these results to experimental singlet fission studies, we have labeled the involved adiabatic states in Fig. 2(c). The lower Davydov component is obviously the strongest absorbing state and can therefore be regarded as the initially populated exciton.³ This state is thus labeled as S_1 , which is also consistent with its substantial s_1 admixture. The lowest-energy triplet-dominated stick is found to represent a handful of states, which in a corresponding fashion are identified with TT_1 . Interestingly, an analogous set of states is found roughly 0.17 eV (1380 cm⁻¹) higher in energy, slightly above S_1 , which are found to consist largely of t_1t_1 contributions dressed with a single vibration ω_0 . Using an asterisk to loosely refer to this vibrational sublevel, these states are denoted as TT_1^* . A similar vibrationally dressed state was found in the phenomenological modeling supporting the 2DES measurements on pentacene in Ref. 15, where it was hypothesized that its quasi-resonance with S_1 is a key factor in the fission dynamics of this material. TT_1^* also plays an important role in the direct detection of TT_1 , which we further investigate in a companion article.⁵⁷

Although the low-energy region of the stick spectrum looks quite similar to the one obtained through the phenomenological model reported in Ref. 15, the oscillator strength of TT_1 is found to be about a factor of 5 smaller. This discrepancy may be explained by noting that the phenomenological model does not include CT states, but instead considers two states, s_1 and t_1t_1 , coupled directly with an interaction strength of 31 meV (248 cm⁻¹). This value is generous compared to the effective interaction strength of 10 meV (81 cm⁻¹) derived using perturbation theory including CT intermediates,²⁸ and as such expectedly overestimates the intensity borrowing of TT_1 from singlet excitons. Even so, it should be noted that the quantitative accuracy of the perturbative approximation is questionable,²⁸ and that a more reliable measure of the degree of intensity borrowing is obtained through non-perturbative calculations, such as presented here. In order to provide further evidence that our model provides reasonable quantitative estimates for the TT_1 oscillator strength, we have performed additional calculations while conforming the electronic basis as well as the parameters to those employed in Ref. 15 (resulting linear absorption and stick spectra are included in the [supplementary material](#)). The stick spectra obtained upon

these modifications are consistent with the results reported in that work. However, in the corresponding linear absorption spectrum, a distinct feature associated with TT_1 shows up as a shoulder of the main S_1 absorption peak, which is not observed experimentally.^{30,53,54} Furthermore, considering the difficulty encountered when experimentally probing the TT_1 state, we conclude that the oscillator strength found here is most likely more accurate.

B. Diabatic vs. adiabatic energies and size convergence

The Davydov splitting observed for pentacene is a manifestation of the general principle that adiabatic energies ω_α of coupled molecular systems deviate from the diabatic energies such as E_{s_1} . This deviation is crystal size dependent and converges with increasing dimension. However, with increasing dimension the vibronic basis employed in our model grows very rapidly. It is therefore crucial to determine the minimal crystal size sufficient to obtain reliable results. This is particularly important in the (expensive) modeling of 2DES.

In Fig. 3, we have plotted the adiabatic S_1 and TT_1 energies calculated for pentacene while varying the crystal size $M \times M$. Also shown are the (size-independent) diabatic energies E_{s_1} and $E_{t_1t_1}$. This figure demonstrates that the adiabatic S_1 energy undergoes a dramatic red-shift relative to the diabatic analogue, eventually stabilizing at a value roughly 250 meV (2015 cm⁻¹) below E_{s_1} . In marked contrast, the energy of TT_1 is apparently fairly insensitive to M and does not change much apart from an initial redshift of about 30 meV (250 cm⁻¹). From these data, the minimal crystal size is dictated by the point of convergence of S_1 , which is found to be at $M = 3$. A further investigation (presented in the [supplementary material](#)) demonstrates that all of the photophysics in the range up to 2.2 eV is well accounted for by limiting the crystal to this size. In our companion article on 2DES,⁵⁷ we therefore adjust M accordingly.

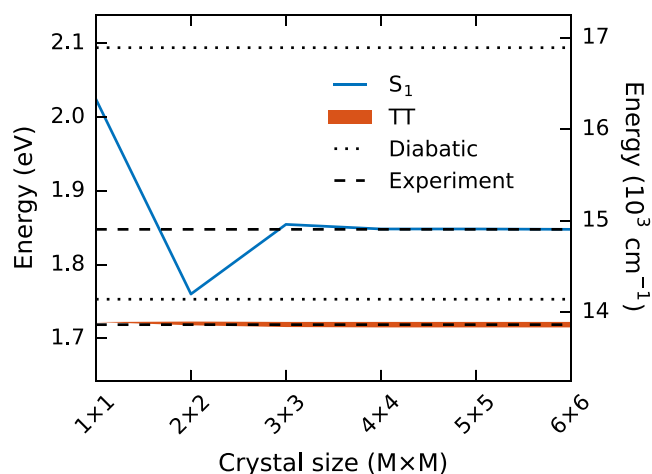


FIG. 3. Adiabatic S_1 (blue curve) and TT_1 (red shaded area) energies as a function of the crystal size $M \times M$. Also shown are the experimentally determined adiabatic energies (dashed),^{15,30} and the diabatic energies E_{s_1} and $E_{t_1t_1}$ applied in our calculations (dotted lines at 2.09 eV and 1.75 eV, respectively).

The adiabatic energy separation between S_1 and TT_1 is considered as a primary factor impacting singlet fission.³ Nevertheless, values assumed for this separation in pentacene vary from 200 meV (Ref. 3) to full resonance.¹⁴ This lack of consensus arises to a great extent from the very weak optical transition between TT_1 and the (singlet) ground state, which makes the correlated triplet pair energy extremely challenging to detect. Only very recently has this optical transition directly been determined by means of 2DES.¹⁵ In our modeling, the diabatic energies are adjusted so that the converged energy of TT_1 matches this experimentally determined value, similar to the approach used to align the energy of S_1 to the experimental value from linear absorption.³⁰ Importantly, our results show that although the splitting between the measurable adiabatic states is 130 meV, the actual splitting between the diabatic energies E_{S_1} and $E_{t_1t_1}$ is found to be 340 meV, as evinced by Fig. 3. Such a determination of diabatic energies is the key to obtaining reliable results from dynamical calculations using a microscopic diabatic basis, especially with regards to the recent indications that resonances between vibrational modes and electronic splittings facilitate rapid singlet fission.^{15,16} In this respect, it is interesting to note that a different diabatic splitting of 170 meV was applied in Ref. 26 to yield a near-resonance between the adiabatic states.²⁶ A slightly larger diabatic splitting of 200 meV was employed in the dynamic study by Berkelbach *et al.*,²⁹ however, the weaker couplings employed in that work mitigate the redshift of S_1 , resulting in a substantial energy offset from TT_1 . Given these quantitative differences in energy splittings and couplings, it is of great interest to dynamically evaluate singlet fission using our model, which will be the topic of a later study.

C. Characterizing the correlated triplet pair state

In addition to its energetic offset from S_1 , other aspects of the correlated triplet pair state are of importance to singlet fission dynamics. In particular, the degree of spatial separation between the two triplets constituting TT_1 expectedly has a profound influence on the dissociation of this intermediate into free triplet excitons.⁵⁵ Conventionally, these constituents are envisioned to be located at adjacent molecules, owing to the short range of the charge overlap integrals that couple between S_1 and TT_1 .^{3,4} However, recent magnetic field measurements on tetracene give indications of an inter-triplet separation of two to four times the intermolecular separation.³² Relatively little experimental data are available to elaborate on these contrasting viewpoint, in part due to the dark nature of TT_1 . Nonetheless, through the microscopic basis of our calculations, information about the spatial anatomy of the correlated triplet pair state in pentacene is readily available.

As a quantitative measure of the spatial composition of TT_1 , we consider the triplet-triplet correlation operator defined as

$$\hat{n}_{t_1t_1}(\mathbf{l}) = \frac{1}{2} \sum_{m \neq m'} |(t_1)_m, (t_1)_{m'}\rangle \langle (t_1)_m, (t_1)_{m'}| \delta_{\mathbf{l}, \mathbf{l}' - \mathbf{l}''}. \quad (12)$$

Here, the summation extends over all triplet pairs located at molecules m and m' , and the prefactor corrects for double

counting. The vectors \mathbf{l} , \mathbf{l}'' , and \mathbf{l}' refer to molecular positions in terms of the crystallographic lattice vectors.

In our evaluation of the triplet-triplet correlation function, we did not truncate the basis with respect to the triplet-triplet and charge separation in order to avoid artificially confined triplet pair states. As this results in a rapidly growing basis set with increasing crystal size, we restricted ourselves to a 3×3 pentacene crystal. For these parameters, TT_1 is found to be represented by four near-degenerate triplet-dominated adiabatic states. Fig. 4 shows the triplet-triplet correlation function for each state. From this figure, it is evident that triplets reside predominantly on neighboring molecules in either the $(l_a, l_b) = (1/2, 1/2)$ or the $(-1/2, 1/2)$ direction, while some mixing between these two configurations is observable. Importantly, the contribution from triplet pairs separated beyond these nearest-neighbor distances is negligible. As such, these findings justify the triplet-triplet truncation radius of 6.2 Å applied in the remainder of our calculations. More importantly, however, they confirm the conventional idea of TT_1 being largely composed of adjacent triplets. Still, the possibility remains that an enhanced separation is induced by the two-electron couplings that transfer triplet excitons between neighboring molecules in accordance with the well-known Dexter energy transfer mechanism. We have explored this possibility by repeating our calculations while applying typical values for such couplings. The results (shown in the [supplementary material](#)) show no quantitative differences with respect to Fig. 4, so

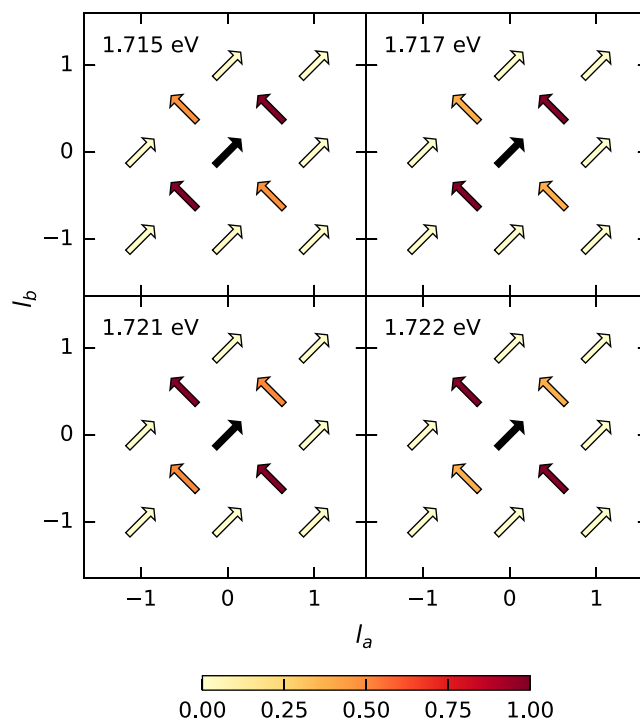


FIG. 4. Triplet-triplet correlations of the adiabatic states represented by TT_1 , calculated for a 3×3 pentacene crystal (associated adiabatic energies shown in top-left corner). For each state, the expectation value of Eq. (12) is shown as a function of $\mathbf{l} = (l_a, l_b)$ by means of colored arrows (loosely referring to the herringbone structure formed by the short molecular axes in the ab -plane). The results are normalized individually in each panel for the ease of demonstration. Note that the correlation function is undefined for $l_a = l_b = 0$ (corresponding arrow shown as black).

that this idea can safely be discarded. For tetracene, whose difference to pentacene lies mostly in the energetic location of the S_1 state, we anticipate a qualitatively similar composition for TT_1 . We therefore hypothesize that the magnetic field measurements reported in Ref. 32 relate to the subsequent step in the singlet fission process, in which TT_1 is in the process of dissociating into spatially separated triplets. The initial stage of this dissociation, and the potential role of Dexter transfer integrals as well as the effective binding energy between triplets,^{18,56} will be a topic of interest in our future work.

IV. CONCLUSIONS

In summary, this article presents a vibronic exciton theory of fission materials which combines a microscopic set of electronic degrees of freedom with an explicit quantum-mechanical treatment of an intramolecular vibrational mode. As such, this article forms a starting point of our ongoing pursuit to unravel the mechanistic principles of singlet fission, in particular, addressing the recent experimental indications of the functional relevance of vibronic coupling.^{15,16,20} The vibronic model is partly derived from a recent parametrization of pentacene single crystals through polarized linear absorption,³⁰ which we extended to include the triplet excitons constituting the correlated triplet pair state TT_1 through which singlet fission proceeds. In doing so, the associated energy is adjusted in accordance with the recent detection of TT_1 by means of 2DES.¹⁵ We examined the excited states constituting the absorption band of pentacene, paying particular attention to the dependence of adiabatic energies on the applied crystal sizes. Besides providing the diabatic transition energies required to reliably simulate excited state dynamics, our findings revealed that a crystal consisting of 3×3 unit cells in the *ab*-plane suffices to account for the photophysics of pentacene in the spectral region of S_1 and TT_1 . Lastly, we provided an insight into the spatial configuration of TT_1 demonstrating that the constituent triplets are predominantly located at adjacent molecules, which contradicts a recent experimental report predicting a triplet-triplet separation of several intermolecular distances.³²

In Paper II⁵⁷ of this series, we focus on the observation of singlet fission through time-resolved spectroscopic experiments, and in particular, the recent direct observation of the correlated triplet pair state through 2DES measurements on pentacene.¹⁵ To this end, the vibronic exciton model introduced in the present article is employed to simulate and analyze 2DES. Our investigation of the crystal size convergence is instrumental to realize such simulations at manageable computational costs. The importance of this investigation is perhaps even more pronounced for the dynamical calculations through which we plan to identify the key factors making successful fission materials, to be presented in future work.

SUPPLEMENTARY MATERIAL

See [supplementary material](#) for the electronic couplings and Coulomb energies of crystalline pentacene applied in our modeling, and for results.

ACKNOWLEDGMENTS

The authors thank Nick Hestand, Frank Spano, and Tim Berkelbach for providing us with the electronic couplings and Coulomb energies of crystalline pentacene, and for fruitful discussions. R.T. acknowledges The Netherlands Organisation for Scientific Research NWO for support through a Rubicon grant. D.R.R. acknowledges funding from NSF Grant No. CHE-1464802.

- ¹S. Singh, W. J. Jones, W. Siebrand, B. P. Stoicheff, and W. G. Schneider, *J. Chem. Phys.* **42**, 330 (1965).
- ²P. Avakian and R. E. Merrifield, *Mol. Cryst.* **5**, 37 (1968).
- ³M. B. Smith and J. Michl, *Chem. Rev.* **110**, 6891 (2010).
- ⁴M. B. Smith and J. Michl, *Annu. Rev. Phys. Chem.* **64**, 361 (2013).
- ⁵M. C. Hanna and A. J. Nozik, *J. Appl. Phys.* **100**, 074510 (2006).
- ⁶W. Shockley and H. J. Queisser, *J. Appl. Phys.* **32**, 510 (1961).
- ⁷H. Najafov, B. Lee, Q. Zhou, L. C. Feldman, and V. Podzorov, *Nat. Mater.* **9**, 938 (2010).
- ⁸P. Irkhin and I. Biaggio, *Phys. Rev. Lett.* **107**, 017402 (2011).
- ⁹P. J. Jadhav, A. Mohanty, J. Sussman, J. Lee, and M. A. Baldo, *Nano Lett.* **11**, 1495 (2011).
- ¹⁰B. Ehrler, M. W. B. Wilson, A. Rao, R. H. Friend, and N. C. Greenham, *Nano Lett.* **12**, 1053 (2012).
- ¹¹D. N. Congreve, J. Lee, N. J. Thompson, E. Hontz, S. R. Yost, P. D. Reusswig, M. E. Bahlke, S. Reineke, T. Van Voorhis, and M. A. Baldo, *Science* **340**, 334 (2013).
- ¹²L. Yang, M. Tabachnyk, S. L. Bayliss, M. L. Böhm, K. Broch, N. C. Greenham, R. H. Friend, and B. Ehrler, *Nano Lett.* **15**, 354 (2015).
- ¹³M. W. B. Wilson, A. Rao, J. Clark, R. S. S. Kumar, D. Brida, G. Cerullo, and R. H. Friend, *J. Am. Chem. Soc.* **133**, 11830 (2011).
- ¹⁴W.-L. Chan, M. Ligges, A. Jailaubekov, L. Kaake, L. Miaja-Avila, and X.-Y. Zhu, *Science* **334**, 1541 (2011).
- ¹⁵A. A. Bakulin, S. E. Morgan, T. B. Kehoe, M. W. B. Wilson, A. W. Chin, D. Zigmantas, D. Egorova, and A. Rao, *Nat. Chem.* **8**, 16 (2016).
- ¹⁶N. R. Monahan, D. Sun, H. Tamura, K. W. Williams, B. Xu, Y. Zhong, B. Kumar, C. Nuckolls, A. R. Harutyunyan, G. Chen, H.-L. Dai, D. Beljonne, Y. Rao, and X.-Y. Zhu, "Dynamics of the triplet-pair state reveals the likely coexistence of coherent and incoherent singlet fission in crystalline hexacene," *Nat. Chem.* **9**, 341 (2016).
- ¹⁷P. M. Zimmerman, F. Bell, D. Casanova, and M. Head-Gordon, *J. Am. Chem. Soc.* **133**, 19944 (2011).
- ¹⁸X. Feng, A. V. Luzanov, and A. I. Krylov, *J. Phys. Chem. Lett.* **4**, 3845 (2013).
- ¹⁹S. M. Parker, T. Seideman, M. A. Ratner, and T. Shiozaki, *J. Phys. Chem. C* **118**, 12700 (2014).
- ²⁰A. J. Musser, M. Liebel, C. Schnedermann, T. Wende, T. B. Kehoe, A. Rao, and P. Kukura, *Nat. Phys.* **11**, 352 (2015).
- ²¹P. M. Zimmerman, Z. Zhang, and C. B. Musgrave, *Nat. Chem.* **2**, 648 (2010).
- ²²R. W. Havenith, H. D. de Gier, and R. Broer, *Mol. Phys.* **110**, 2445 (2012).
- ²³D. Casanova, *J. Chem. Theory Comput.* **10**, 324 (2014).
- ²⁴H. Yamagata, J. Norton, E. Hontz, Y. Olivier, D. Beljonne, J. L. Brédas, R. J. Silbey, and F. C. Spano, *J. Chem. Phys.* **134**, 204703 (2011).
- ²⁵P. E. Teichen and J. D. Eaves, *J. Phys. Chem. B* **116**, 11473 (2012).
- ²⁶D. Beljonne, H. Yamagata, J. L. Brédas, F. C. Spano, and Y. Olivier, *Phys. Rev. Lett.* **110**, 226402 (2013).
- ²⁷T. C. Berkelbach, M. S. Hybertsen, and D. R. Reichman, *J. Chem. Phys.* **138**, 114102 (2013).
- ²⁸T. C. Berkelbach, M. S. Hybertsen, and D. R. Reichman, *J. Chem. Phys.* **138**, 114103 (2013).
- ²⁹T. C. Berkelbach, M. S. Hybertsen, and D. R. Reichman, *J. Chem. Phys.* **141**, 074705 (2014).
- ³⁰N. J. Hestand, H. Yamagata, B. Xu, D. Sun, Y. Zhong, A. R. Harutyunyan, G. Chen, H.-L. Dai, Y. Rao, and F. C. Spano, *J. Phys. Chem. C* **119**, 22137 (2015).
- ³¹After completion of our manuscript and during the formulation of our dynamical modeling, we became aware of Y. Fujihashi, L. Chen, A. Ishizaki, J. Wang, and Y. Zhao, *J. Chem. Phys.* **146**, 044101 (2017), which non-perturbatively explores the effect of high-frequency modes on singlet fission using dynamical calculations based on a phenomenological model. A comparison of this work with the dynamics that emerge

- from the microscopic Hamiltonian will be made in the third paper of this series.
- ³²R. Wang, C. Zhang, B. Zhang, Y. Liu, X. Wang, and M. Xiao, *Nat. Commun.* **6**, 8602 (2015).
 - ³³M. R. Philpott, *J. Chem. Phys.* **55**, 2039 (1971).
 - ³⁴F. C. Spano, *J. Chem. Phys.* **116**, 5877 (2002).
 - ³⁵F. C. Spano, J. Clark, C. Silva, and R. H. Friend, *J. Chem. Phys.* **130**, 074904 (2009).
 - ³⁶W.-L. Chan, T. C. Berkelbach, M. R. Provorse, N. R. Monahan, J. R. Tritsch, M. S. Hybertsen, D. R. Reichman, J. Gao, and X.-Y. Zhu, *Acc. Chem. Res.* **46**, 1321 (2013).
 - ³⁷D. Holmes, S. Kumaraswamy, A. J. Matzger, and K. P. C. Vollhardt, *Chem. - Eur. J.* **5**, 3399 (1999).
 - ³⁸C. C. Mattheus, A. B. Dros, J. Baas, A. Meetsma, J. L. de Boer, and T. T. M. Palstra, *Acta Crystallogr., Sect. C: Cryst. Struct. Commun.* **57**, 939 (2001).
 - ³⁹E. V. Tsiper and Z. G. Soos, *Phys. Rev. B* **68**, 085301 (2003).
 - ⁴⁰V. Coropceanu, J. Cornil, D. A. da Silva Filho, Y. Olivier, R. Silbey, and J.-L. Brédas, *Chem. Rev.* **107**, 926 (2007).
 - ⁴¹H. Yamagata, D. S. Maxwell, J. Fan, K. R. Kittilstved, A. L. Briseno, M. D. Barnes, and F. C. Spano, *J. Phys. Chem. C* **118**, 28842 (2014).
 - ⁴²N. J. Hestand, R. Tempelaar, J. Knoester, T. L. C. Jansen, and F. C. Spano, *Phys. Rev. B* **91**, 195315 (2015).
 - ⁴³S. Ito, T. Nagami, and M. Nakano, *J. Phys. Chem. Lett.* **6**, 4972 (2015).
 - ⁴⁴S. P. McGlynn, T. Azumi, and M. Kasha, *J. Chem. Phys.* **40**, 507 (1964).
 - ⁴⁵R. Tempelaar, A. Stradomska, J. Knoester, and F. C. Spano, *J. Phys. Chem. B* **117**, 457 (2013).
 - ⁴⁶F. Anger, J. O. Ossó, U. Heinemeyer, K. Broch, R. Scholz, A. Gerlach, and F. Schreiber, *J. Chem. Phys.* **136**, 054701 (2012).
 - ⁴⁷F. C. Spano, *Acc. Chem. Res.* **43**, 429 (2010).
 - ⁴⁸H. Marciniak, M. Fiebig, M. Huth, S. Schiefer, B. Nickel, F. Selmaier, and S. Lochbrunner, *Phys. Rev. Lett.* **99**, 176402 (2007).
 - ⁴⁹A. Rao, M. W. B. Wilson, S. Albert-Seifried, R. Di Pietro, and R. H. Friend, *Phys. Rev. B* **84**, 195411 (2011).
 - ⁵⁰M. W. B. Wilson, A. Rao, B. Ehrler, and R. H. Friend, *Acc. Chem. Res.* **46**, 1330 (2013).
 - ⁵¹R. Ruiz, D. Choudhary, B. Nickel, T. Toccoli, K.-C. Chang, A. C. Mayer, P. Clancy, J. M. Blakely, R. L. Headrick, S. Iannotta, and G. G. Malliaras, *Chem. Mater.* **16**, 4497 (2004).
 - ⁵²B. Nickel, R. Barabash, R. Ruiz, N. Koch, A. Kahn, L. C. Feldman, R. F. Haglund, and G. Scoles, *Phys. Rev. B* **70**, 125401 (2004).
 - ⁵³L. Sebastian, G. Weiser, and H. Bässler, *Chem. Phys.* **61**, 125 (1981).
 - ⁵⁴A. Rao, M. W. B. Wilson, J. M. Hodgkiss, S. Albert-Seifried, H. Bässler, and R. H. Friend, *J. Am. Chem. Soc.* **132**, 12698 (2010).
 - ⁵⁵R. D. Pensack, E. E. Ostroumov, A. J. Tilley, S. Mazza, C. Grieco, K. J. Thorley, J. B. Asbury, D. S. Seferos, J. E. Anthony, and G. D. Scholes, *J. Phys. Chem. Lett.* **7**, 2370 (2016).
 - ⁵⁶T. Zeng, R. Hoffmann, and N. Ananth, *J. Am. Chem. Soc.* **136**, 5755 (2014).
 - ⁵⁷R. Tempelaar and D. R. Reichman, *J. Chem. Phys.* **146**, 174704 (2017).