Name	Date
Physics 216: Homework for Lab 10: Ideal Gases and the First Law of Thermodynamics (Due at the beginning of Lab 11)	
Read through the Experimental Uncertainty han 1. In Activity 1-3, I measured the inner diameter length as 37.5 ± 0.5 cm. a) From these data, calculate the volume of the your work. Report the volume and the uncertainty	er of the tube as 3.5 ± 0.1 mm and the tube and its absolute uncertainty. Show
b) Which of the two measurements contributes volume? How can you tell?	more to the overall uncertainty in the

c) From the first part of Activity 1-3, I determined from the graph that the volume of the tube was $3.1 \pm 0.2~\text{cm}^3$. Does this measurement agree with the volume obtained from the direct measurement (what you calculated in a)? Explain your reasoning.