

# workshop 6

## problem 5.2

See book.

💡 Tip

$$1 \text{ dyne} \cdot \text{cm} = 10^{-7} \text{ J}$$

Soln:

$$a) W = \int p dv \cong 314 \text{ J}$$

b) Since there's only 1 mole,  $pV = RT$ .

$$\Delta E = C_V \Delta T = \frac{3}{2} R \left( \frac{P_C V_C}{R} - \frac{P_A V_A}{R} \right)$$

$$= \frac{3}{2} (6 - 2) \times 10^9 \text{ dynes} \cdot \text{cm}$$

$$= 600 \text{ J},$$

c)  $Q = \Delta E + W$  where  $\Delta E$  is as calculated in (b), which is 600 J.

$W = \text{area under curve}$

$$= 400 + \frac{600}{2} \pi$$

$$\cong 557 \text{ J}$$

$$\text{so } Q = 600 + 557 = 1157 \text{ J}$$

## problem 5.5

See book.

Soln:

a) Thermally insulated  $\Rightarrow Q=0$

$$\text{so } \Delta E = -W$$


since  $V \uparrow$ , the gas is doing work to the piston,  $\Delta E \downarrow$

since  $E \propto T$ , when  $E \downarrow$ ,  $T$  also  $\downarrow$ .  
 $\uparrow$  (proved in Eqn 3.12.11 in book).

Thus  $T \downarrow$ .

b). Entropy of gas  $\uparrow$  since process is irreversible.

$$\begin{aligned} \text{c). } \Delta E = -W &= -mgh = -mg \frac{V_f - V_0}{A} \\ \Delta E &= \nu C_v (T_f - T_0) \end{aligned} \quad \left. \vphantom{\begin{aligned} \Delta E = -W \\ \Delta E = \nu C_v (T_f - T_0) \end{aligned}} \right\} \nu C_v (T_f - T_0) = -mg \frac{V_f - V_0}{A}$$

In equilibrium,  $P = \frac{mg}{A}$ ,  $P V_f = \nu R T_f \Rightarrow V_f = \frac{\nu R T_f}{mg} A$  

$$\nu C_v (T_f - T_0) = -mg \frac{\frac{\nu R T_f}{mg} A - V_0}{A}$$

$$= -\nu R T_f + \frac{mg V_0}{A}$$

$$\nu (C_v + R) T_f = \nu C_v T_0 + \frac{mg V_0}{A}$$

$$\Rightarrow T_f = \frac{\nu C_v T_0 + \frac{mg V_0}{A}}{\nu (C_v + R)} = \frac{T_0 + \frac{mg V_0}{\nu C_v A}}{1 + \frac{R}{C_v}}$$

## problem 5.6

### ① Note

This is a variant of the Ruchardt experiment. I don't know if it's mentioned in class but see following video for more information.

[https://www.youtube.com/watch?v=e1VDAa4ttOc&ab\\_channel=Caltech%27sFeynmanLectureHall](https://www.youtube.com/watch?v=e1VDAa4ttOc&ab_channel=Caltech%27sFeynmanLectureHall)

Soln:

EOM:  $m\ddot{x} = PA - mg - P_0A$

since the process is approximately adiabatic,

$$pV^\gamma = \text{constant} = (P_0 + \frac{mg}{A}) V_0^\gamma \text{ as it was originally.}$$

$$\text{also } m\ddot{x} = (P_0 + \frac{mg}{A}) \left(\frac{V_0}{V}\right)^\gamma A - mg - P_0A$$

$$= (P_0 + \frac{mg}{A}) \left(\frac{V_0}{Ax}\right)^\gamma A - mg - P_0A \quad (1)$$

$$x = \frac{V_0}{A} + \eta \quad \text{where } \eta \text{ is the small perturbation displacement}$$

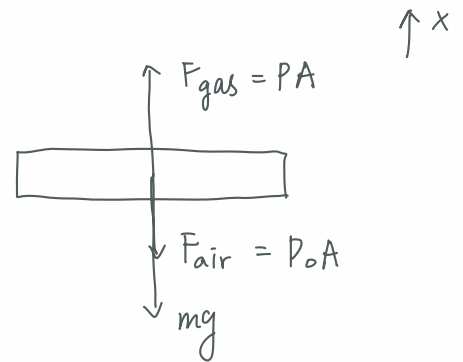
Expand about  $\frac{V_0}{A}$ :

$$\frac{1}{x^\gamma} = \frac{1}{\left(\frac{V_0}{A} + \eta\right)^\gamma}$$

Expansion of  $(1-x)^\gamma = 1 + (-\gamma)(-x)^1 + \frac{-\gamma(-\gamma-1)}{2!} (-x)^2 + \dots$

$$= \frac{1}{\left(1 + \eta \frac{A}{V_0}\right)^\gamma} \left(\frac{A}{V_0}\right)^\gamma$$

$$= \left(\frac{A}{V_0}\right)^\gamma \left(1 - \eta \frac{A}{V_0} + \dots\right) \approx \left(\frac{A}{V_0}\right)^\gamma \left(1 - \eta \frac{A}{V_0}\right)$$



put it back to ① we have

$$\begin{aligned}
 m\ddot{x} &= m\dot{\eta} = \left(P_0 + \frac{mg}{A}\right) \left(\frac{V_0}{Ax}\right)^\gamma A - mg - P_0 A \\
 &= \left(P_0 + \frac{mg}{A}\right) \frac{AV_0^\gamma}{A^\gamma} \frac{A^\gamma}{V_0^\gamma} \left(1 - \gamma \eta \frac{A}{V_0}\right) - mg - P_0 A \\
 &= \cancel{P_0 A} + \cancel{mg} - \cancel{mg} - \cancel{P_0 A} - \left(P_0 + \frac{mg}{A}\right) \gamma \eta \frac{A}{V_0}
 \end{aligned}$$

$$\Rightarrow m\ddot{\eta} = - \left(P_0 + \frac{mg}{A}\right) \gamma \eta \frac{A}{V_0}$$

Recall  $\ddot{x} + \omega^2 x = 0$

$$\text{so } \omega = \frac{\omega}{2\pi} = \frac{\sqrt{\left(P_0 + \frac{mg}{A}\right) \frac{\gamma A^2}{V_0 m}}}{2\pi}$$

$$\gamma = \frac{(2\pi\omega)^2 V_0 m}{\left(P_0 + \frac{mg}{A}\right) A^2} = \frac{4\pi^2 \omega^2 V_0 m}{P_0 A^2 + mgA}$$