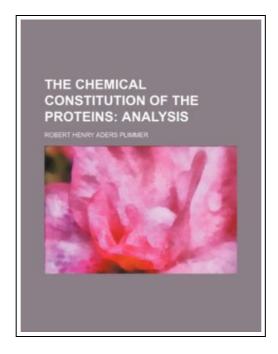
The Chemical Constitution of the Proteins; Analysis



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Reviews

Absolutely one of the best book I have ever study. It is actually writter in simple terms rather than confusing. I realized this pdf from my dad and i suggested this pdf to understand.

(Garry Quigley)

THE CHEMICAL CONSTITUTION OF THE PROTEINS; ANALYSIS



Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand ******. This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1912 edition. Excerpt: . prepared from sec. butyliodide and sodiumacetoacetic ester: --CH, CH, CO CH. CO CH, I L II HCNa +1--C--CH.CH, = NaI + HC--CH--CH2--CH, I I I COOC2He H COOC2He This compound, when treated with nitrosyl sulphate was decomposed into acetic acid and a-oximino-sec-butylacetic acid, which, on reduction with zinc dust and hydrochloric acid in alcohol, gave a 60 per cent, yield of dl-isoleucine: --CH3 I NOH NH2 CO 11 I I / CH3 I / CH- H--C--CH-C--CH(-H--C--CH(I C2H6 I C2H6 J C H5 COOC2H8 COOH COOH Locquin has since obtained d-isoleucine from this racernic compound which was identical with Ehrlich s natural product, and this therefore has the above constitution. By the same series of reactions which Fischer and Schmitz employed in the preparation of leucine, F. Ehrlich synthesised isoleucine in 1908 from malonic ester and secondary butyliodide, i.e., according to the following scheme: --C2H6. CHI. CH, + CHNa(COOC2H6)2-C2He. CH(CH3). CH(COOC2H3)2 C2H6. CH(CH3). CH(COOH), -C2H, . CH(CH3). CBr(COOH)2 C2HB. CH(CH3). CHBr. COOH-C2H6. CH(CH3). CH(NH2). COOH The same synthesis has also been carried out by Brasch and Friedmann. Phenylalanine. In a note published in 1879 Schulze mentioned a substance which he had obtained from the seedlings oiLupinus luteus; two years later he and Barbieri showed that this substance had the composition C9HnN02, and they described it as phenylamidopropionic acid, because, on oxidation, it gave benzoic acid, and, when heated, it lost carbon dioxide and gave a base, C8HnN. In its properties it closely resembled Tiemann...



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