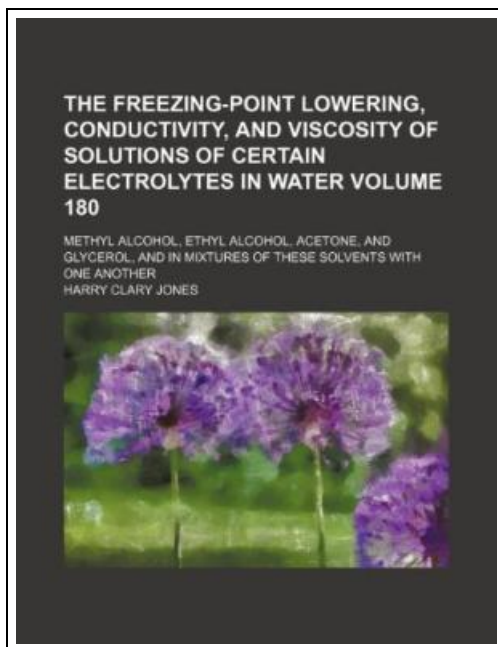


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1913 Excerpt: . gives the conductivities of sodium bromide in methyl alcohol at both 0 and 25. At 0 we have complete dissociation probably at 6,400 liters. At 25 complete dissociation is not attained until a dilution of 12,800 liters is reached. Table 74 gives the conductivities of sodium bromide in ethyl alcohol at 0 and 25. Complete dissociation is reached at both temperatures at a dilution of 12,800 liters. At the volume 102,400 there is a marked decrease in conductivity, probably due to the great solvation at this extremely high dilution. The same fact may be noticed in several other tables where ethyl alcohol is used as a solvent. Table 74 gives the conductivity of lithium bromide in methyl alcohol. At 0 complete dissociation probably is reached at 6,400 liters. At 25 there is no maximum in conductivity, but the rate of increase is much smaller above 3,200 liters than it is at greater concentrations, indicating that the maximum is nearly reached. Table 74 gives the conductivity of lithium bromide in ethyl alcohol. At 0 there is complete dissociation at 3,200 liters. At 25 it is complete at 6,400 liters. For both temperatures at higher dilutions the conductivity remains almost constant up to a dilution of 102,400 liters, where, at 25, there is a marked decrease in the conductivity. Table 74 gives the conductivity of potassium sulphocyanate in methyl alcohol. This is a repetition of work previously done. We have now obtained more concordant results. Here, complete...



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