

Explaining the Stern-Gerlach Experiment with Quantum Mechanics

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Abstract

The Stern-Gerlach experiment has played an extremely essential role in the physical micro-world. In this article, we'll attach high importance to explain Stern-Gerlach experiment by quantum mechanics.

1 Introduction

The Stern-Gerlach experiment, which started from Otto Stern in 1921 and then finished by Otto Stern and Walter Gerlach in 1922. It has an inhomogeneous magnetic field made by a magnet with a pointed pole tip and then send a beam through the apparatus, the beam of particles will be split into a number of beams. Surprisingly, experimental observations is completely beyond the expectations of the experimental designers. How to understand and explain the S.G experiment phenomenon is beneficial for us to learn quantum mechanics. We'll explain the result through a theoretical framework of QM.

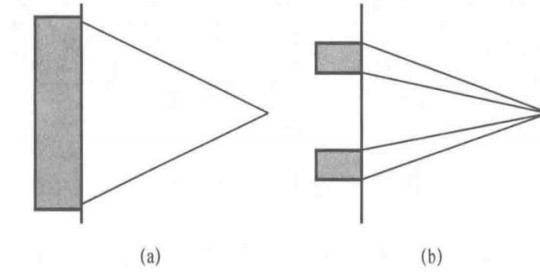
2 Stern-Gerlach experiment

2.1 Single S.G experiment

The earliest motivation of this experimentation was proving a conception from Bohr. At that time, the beam should be dispersed over a vertical distance in classical physics prediction, such as picture a. However, It has been observed that the beam magically splits into two parts, one corresponding to the spin up and one corresponding to the spin down as picture b shows. We must work out the effect of the magnetic field on the beam. Because the interaction energy of the magnetic field is just $-\mu \cdot B$, the z-component of the force experienced by the atom is given by a equation(1). The observed z component of spin has only two possible values, spin up and down, respectively called S_z+ and S_z- . The two

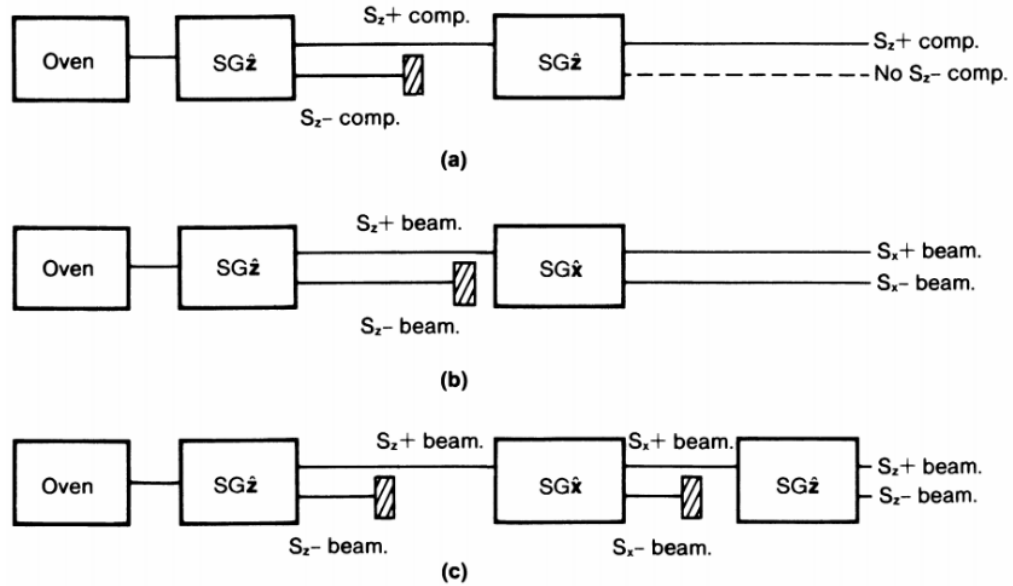
possible values are multiples of the basic angular momentum unit. Therefore, the reason, why classical physic gets a wrong prediction, is Spatial quantization. It means that S.G experiment revealed quantization of electron spin angular momentum. It has been known that the orbital angular momentum quantum number is an integer, while the Stern-Galach experiment shows that the spin angular momentum quantum number is a half integer.

$$F_z = \frac{\partial(\mu \cdot B)}{\partial z} = \mu_z \cdot \frac{\partial B_z}{\partial z} \quad (1)$$



2.2 Sequential Stern-Gerlach Experiments

Now, let's consider a sequential Stern-Gerlach experiment. As following figure shows, the beam goes through two or more SG apparatuses in sequence. [1]The first arrangement we consider is relatively straightforward. Like figure (a), we subject the beam coming out of the oven to the arrangement, where SG^z stands for an apparatus with the inhomogeneous magnetic field in the z -direction. Then we block the S_z^- component coming out of the first SG^z apparatus and let the remaining S_z^+ component be subjected to another SG^z apparatus. This time there is only one beam component coming out of the second apparatus—just the S_z^+ component. This is perhaps not so surprising; after all if the atom spins are up, they are remain so. The second arrangement is interesting shown in Figure 1.b. The S_z^+ beam that enters the second apparatus (SG^x) is split into two components, an S_x^+ component and an S_x^- component, with equal intensities. The third step is shown in Figure (c) which most dramatically illustrates the peculiarities of quantum-mechanical systems. This time we add to the arrangement of Figure (b) yet a third apparatus, of the SG^z type. It is observed experimentally that two components emerge from the third apparatus, not one; the emerging beams are seen to have both an S_z^+ component and an S_z^- component. Obviously, we can't determine both S_z and S_x simultaneously. More precisely, we can say that the selection of the S_x^+ beam by the second apparatus (SG^x) completely destroys any precious information about S_z .



3 Conclusion

Stern-Gerlach experiment phenomenon could be explained by a basic principle in quantum mechanics. The principle is blowing: Any atomic system can be separated by a filtering process into a certain set of what we will call base states, and the future behavior of the atoms in any single given base state depends only on the nature of the base state—it is independent of any previous history.

References

- [1] University of Texas at Austin. Dept. of Germanic Languages and University of Texas at Austin. Graduate School. *Dimension: Contemporary German Arts and Letters*, volume 20. Department of Germanic Languages, University of Texas at Austin., 1994.