

I am available to groups and individuals for pharmacology help and discussions by appointment.

lgolds01@nyit.edu

Do.
Make.
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Reinvent the Future.



pharmacodynamics

What
the
drug
does to
the
body
and how

pharmacokinetics

What
the
body
does to
the
drug
and how



phillipmartin.com

Learning objectives for Pharmacodynamics Part 1.

After completing ScholarRx Bricks, Guided Reading Questions (GRQs), Practice Questions, and optional preparation materials, students will be able to:

1. Apply the physiology of receptors, effectors, and signaling mechanisms to pharmacology.
2. Define all terms listed in the handout and give examples of the specific actions for therapeutic drug effects.
3. Relate how the physicochemical characteristics (structural features) of a drug determine its ability to initiate a cellular effect (the effector mechanism).
4. Correlate drug concentration, receptor occupancy, and the magnitude of pharmacologic response by comparing dose-dependent therapeutic and toxicologic responses.
5. Describe the concepts of intrinsic activity, affinity, potency and efficacy by comparing the graded dose-response curves of two or more drugs.

Links

Scholar Rx Bricks:

Cellular and Molecular Biology: Cell signaling <https://exchange.scholarrx.com/brick/cell-signaling>

General Pharmacology:

Pharmacology: Foundations and Frameworks <https://exchange.scholarrx.com/brick/pharmacology-foundations-and-frameworks>

Receptor Agonists and Antagonists <https://exchange.scholarrx.com/brick/receptor-agonists-and-antagonists>

Enzymes as Drug Targets <https://exchange.scholarrx.com/brick/enzymes-as-drug-targets>

Access Medicine Katzung's Basic & Clinical Pharmacology, 16e, 2024; Chapter 1: Introduction > The Nature of Drugs

<https://accessmedicine-mhmedical-com.nyit.idm.oclc.org/content.aspx?bookid=3382§ionid=281746718>

and

Chapter 2: Drug Receptors & Pharmacodynamics

<https://accessmedicine-mhmedical-com.nyit.idm.oclc.org/content.aspx?bookid=3382§ionid=281746891>

THE REVIEW BOOKS BELOW INCLUDE PRACTICE QUESTIONS.

Access Medicine Katzung's Pharmacology: Examination and Board Review, 14e, 2024; Chapter 1: Introduction > The Nature of Drugs

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<https://accessmedicine-mhmedical-com.nyit.idm.oclc.org/content.aspx?bookid=3461§ionid=285588872>

LWW Health Library Premier Basic Sciences, Lippincott Illustrated Reviews: Pharmacology, 8e, 2023: Chapter 2: Drug-Receptor Interactions and Pharmacodynamics

<https://premiumbasicsciences-lwwhealthlibrary-com.nyit.idm.oclc.org/content.aspx?sectionid=253325200&bookid=3222>

Questions help learning. Questions help to master a topic.

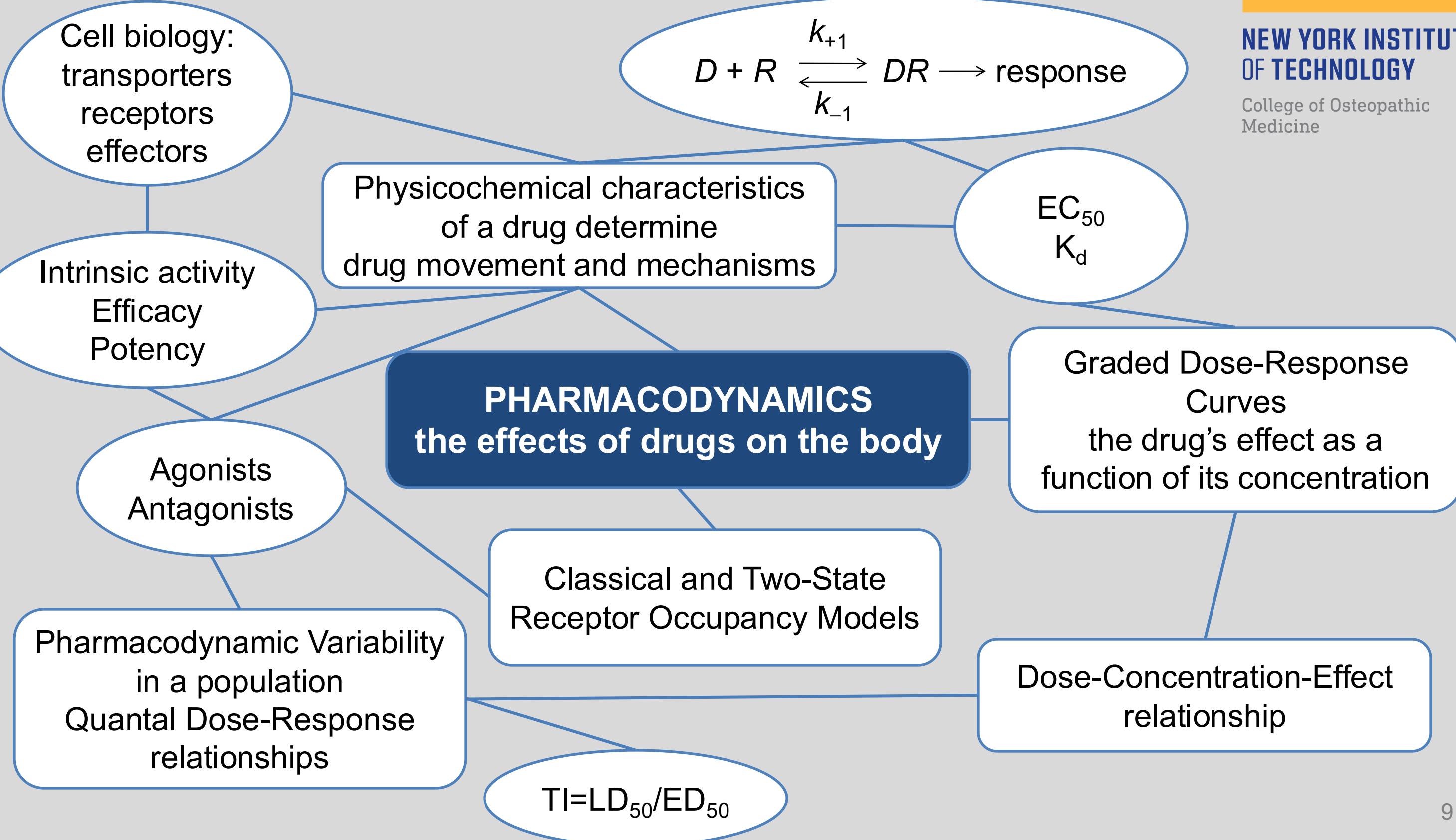
- 1. Guided Reading Questions** are intended to help you identify what you NEED to know.
 - Other information in the reading provides context to help you understand concepts within the big picture.
- 2. Practice Questions** are for your own assessment of your learning – for applying what you have learned in the context of clinical case vignettes.
 - Practicing through case vignettes will help you bridge pharmacology science and pharmacotherapeutics (clinical concepts).
- 3. Write down your own questions** as you study. This practice helps you identify where you are strong and where you weak so you can focus your efforts.

Tips for effective LEARNING (not rote memorizing).

- Identify and define key ideas/concepts.
- Rephrase MAIN ideas in your OWN WORDS.
- Convert MAIN points to questions.
- Relate the ideas to what you already know.

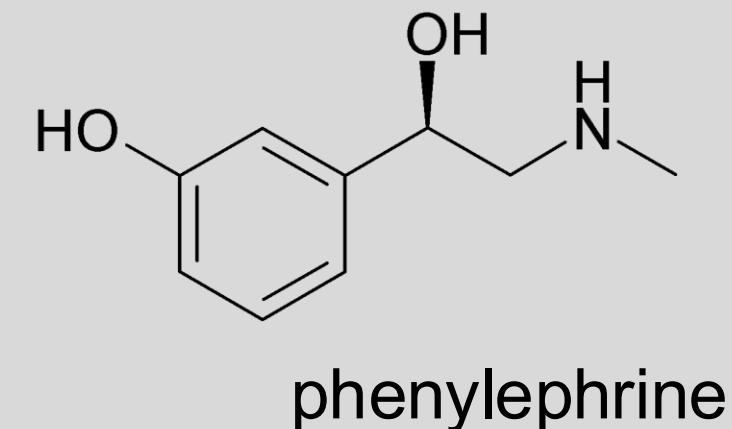
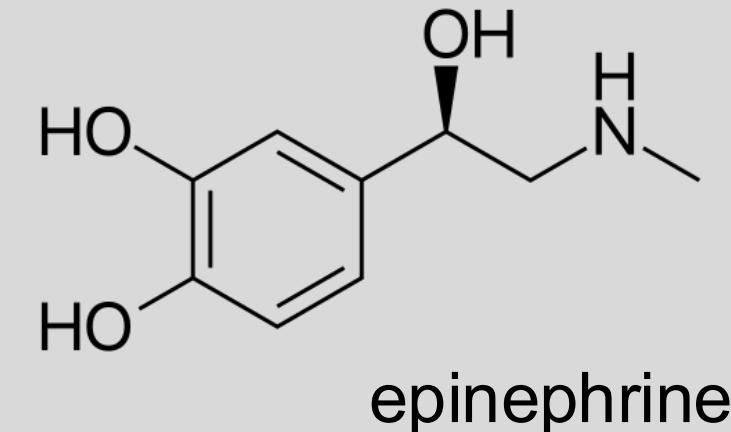
Techniques:

- Spacing your practice – reviewing material and questions after a period of time improves learning by giving your mind time to make connections.
- Mixing multiple subjects (interleaving) while you are studying improves learning and problem solving skills by forcing the brain to continually retrieve knowledge.
- Use / invent memory devices (mnemonics) to help you remember.



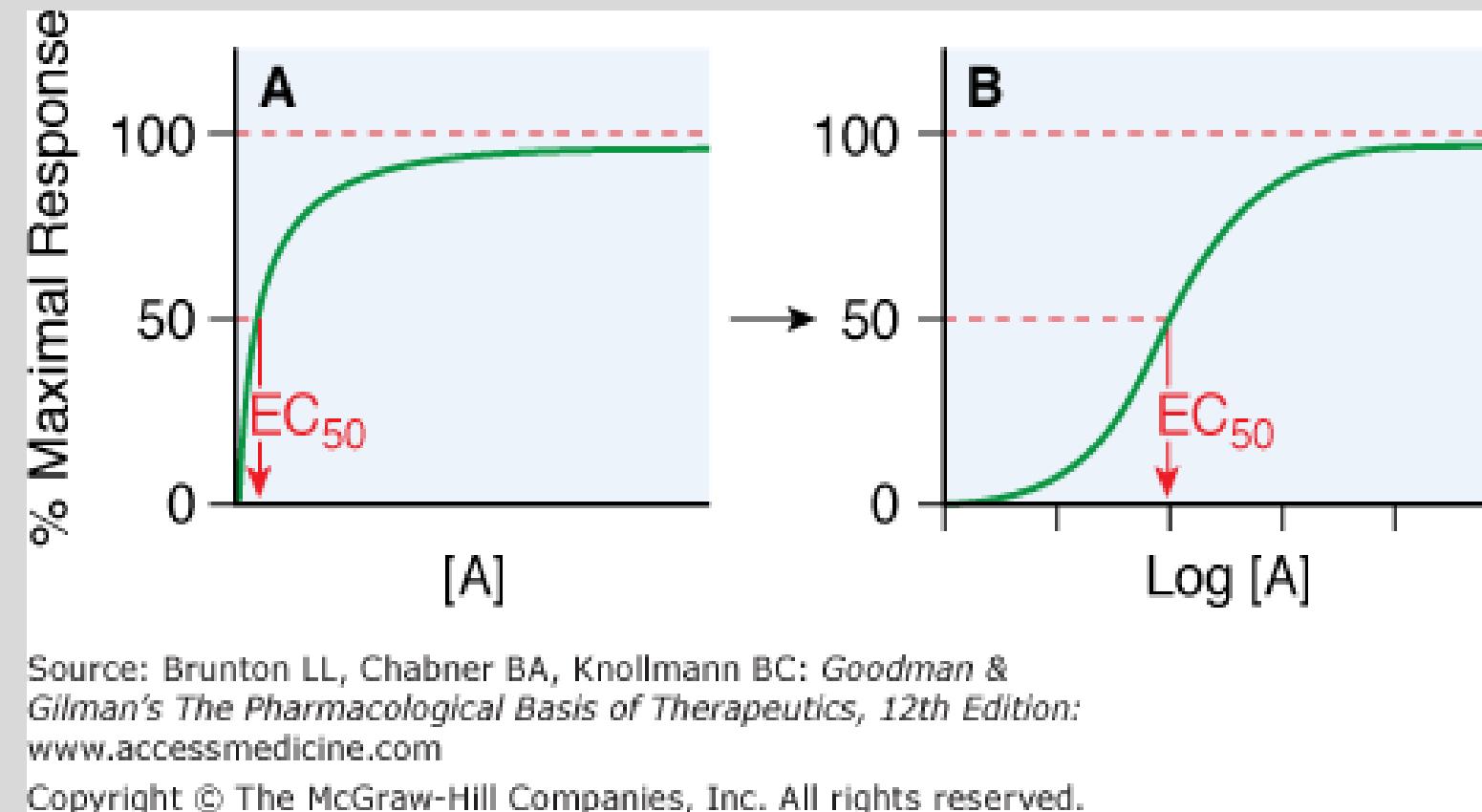
Structure-Activity Relationship

- Structure-activity relationship is the **relationship** between the chemical or 3D **structure** of a molecule and its biological **activity**.
- In the example at the right, epinephrine is a natural hormone. Phenylephrine is a synthetic chemical with a similar structure and similar action.



Graded Dose-Responses Curves

The effect of a drug as a function of its concentration

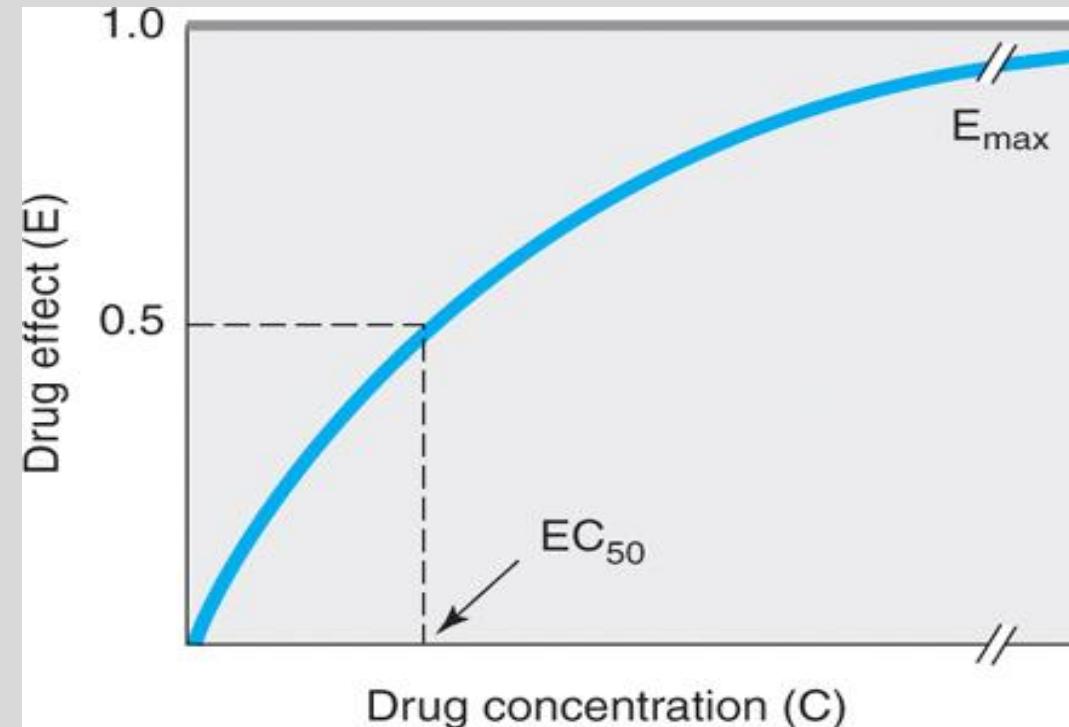


EC_{50} : the concentration of drug that produces 50% of maximal response

Plotting the data on a logarithmic concentration axis
↓
sigmoid curve
↓
simplifies mathematical manipulation of the dose-response data

Concentration-Effect & Receptor Binding Curves

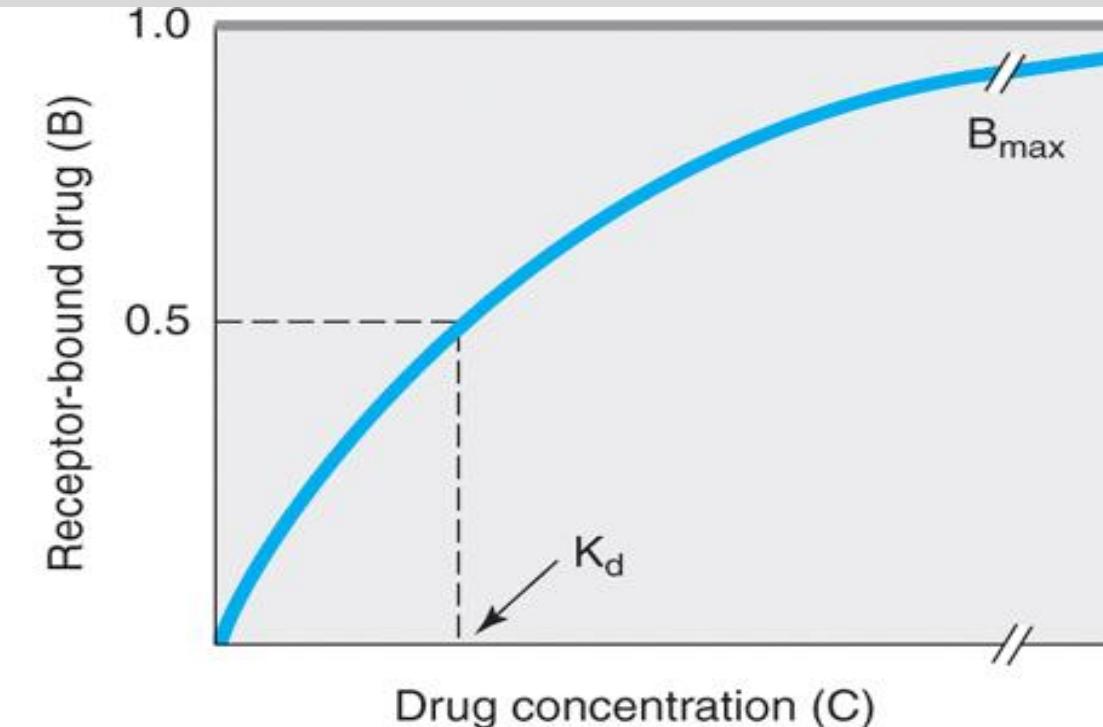
Drug Concentration-Effect Relationship



A

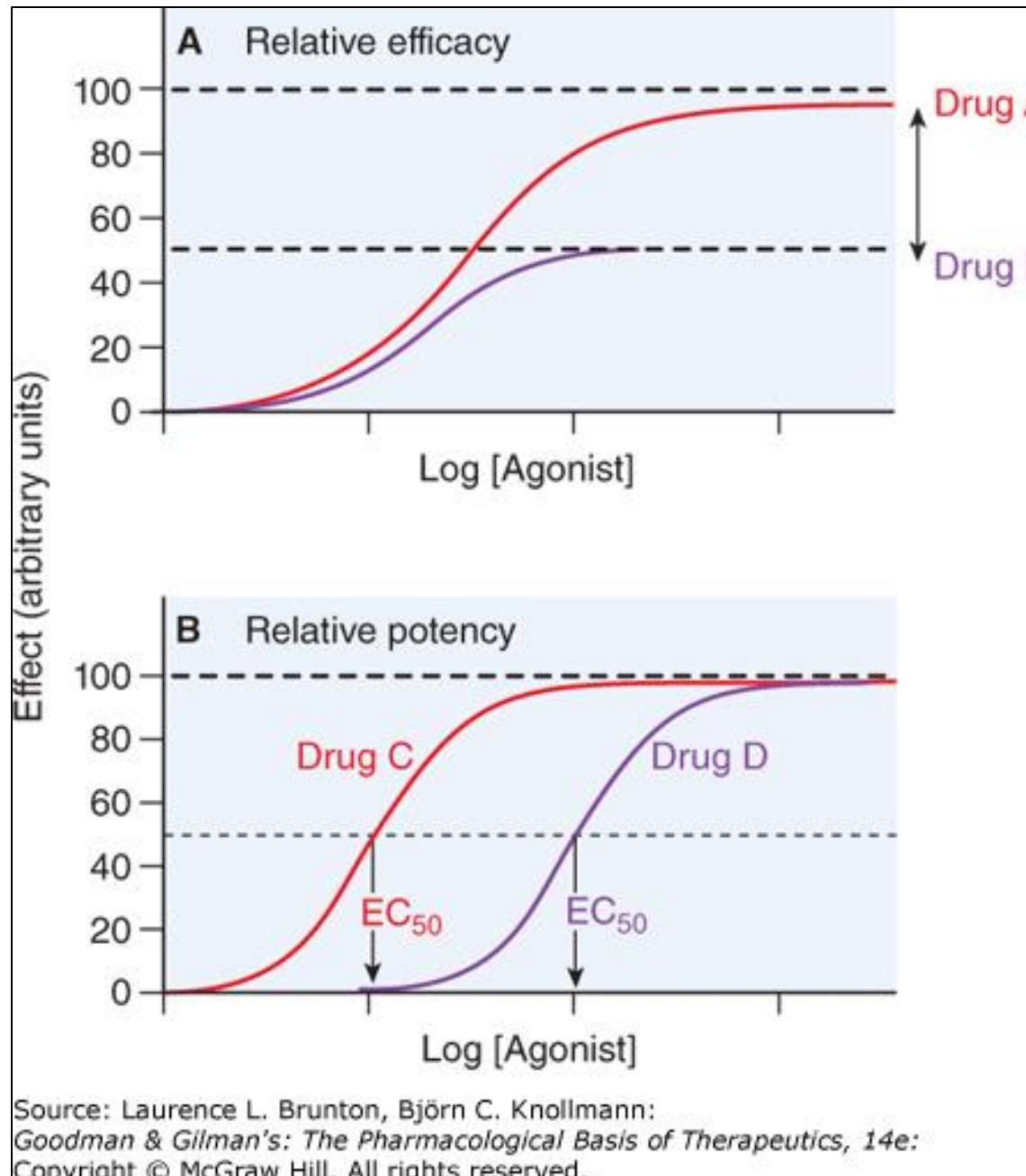
EC₅₀: Concentration of drug that produces 50% of maximal response
E_{max}: Maximal effect of agonist

Drug Concentration-Receptor Occupancy Relationship



B

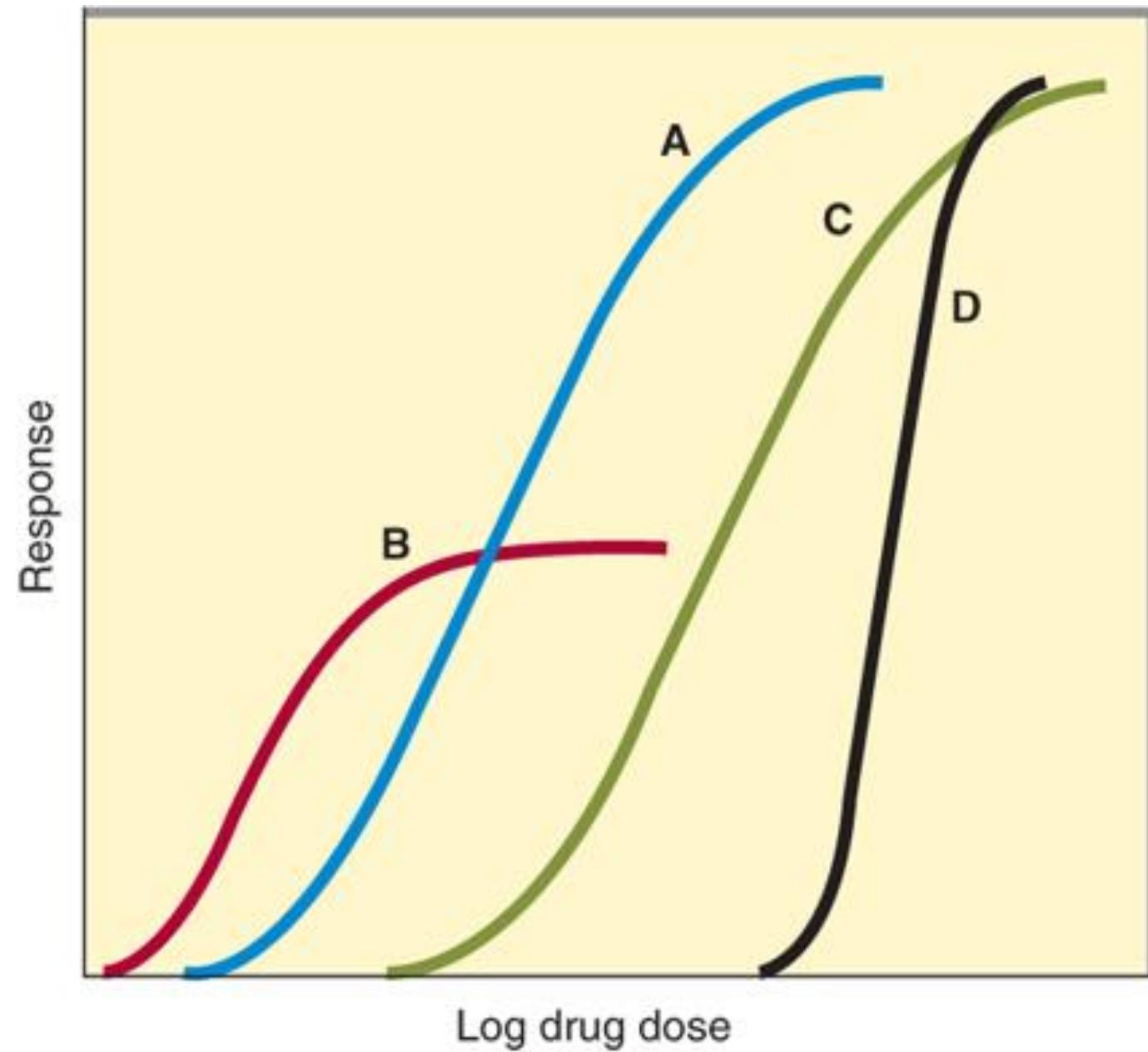
K_d: Concentration of free drug at which half-maximal binding is observed
B_{max}: total concentration of receptor sites



Relative Efficacy and Relative Potency

- **Efficacy:** The observed maximal responses of Drug A and Drug B binding to the same receptor is a function of their relative intrinsic activity (efficacy)
 - **the drug with greater efficacy produces a greater magnitude of effect than the drug with lesser efficacy**
- **Potency:** The relative potency of two agonists, Drug C and Drug D, obtained in the same tissue (same receptor) is a function of:
 - **their relative affinities**

Two ways of quantifying agonism. 1. The relative efficacy of two agonists (Drug A = red line; Drug B = purple line) for a given type of receptor in the same cell or tissue is evaluated based on a comparison of responses. The asymptotic response of Drug A is two times that of Drug B; hence, Drug A is twice as efficacious as Drug B. 2. The EC₅₀ of Drug C is one-tenth that of Drug D; hence, Drug C is 10-fold more potent than drug D.

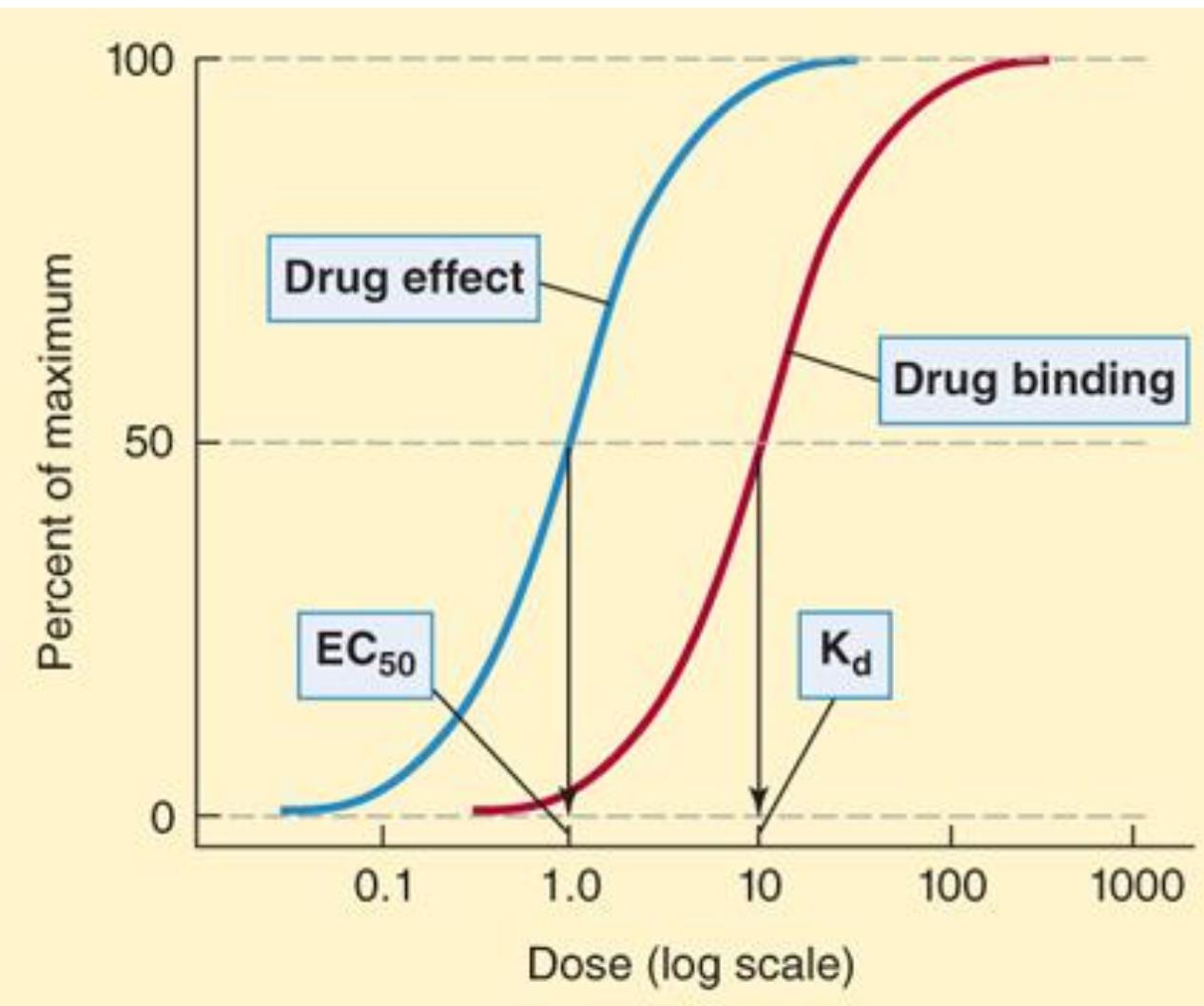


Source: Todd W. Vanderah:
Basic & Clinical Pharmacology, Sixteenth Edition
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Shape of Dose-response Curves

- Drugs A and B are more potent than drugs C and D.
- Drug B is more potent than Drug A but Drug A has a larger maximal efficacy.
- Drugs A, C, and D have equal maximal efficacy.
- Drug D has a very steep dose-response curve, which may have clinical consequences if the upper portion represents an undesirable magnitude of response, such as coma caused by a sedative hypnotic – there are increasingly higher risks of toxic responses with relatively small increases in dose.

Graded dose-response curves for four drugs, illustrating different pharmacologic potencies and different maximal efficacies.



Source: M. Kruidering-Hall, B. G. Katzung, R. L. Tuan, T. W. Vanderah:
Katzung's Pharmacology Examination & Board Review, 14th Edition
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In a system with spare receptors, the EC_{50} is lower than the K_d , indicating that to achieve 50% of maximal effect, less than 50% of the receptors must be activated. This might result from one of two mechanisms. First, the duration of the *effector activation* may be much greater than the duration of the *drug-receptor interaction*. Second, the actual number of receptors may exceed the number of effector molecules available. The presence of spare receptors increases sensitivity to the agonist because the likelihood of a drug-receptor interaction increases in proportion to the number of receptors available.

Spare Receptors are said to exist if the maximal drug response (E_{max}) is obtained at less than 100% occupation of the receptors (B_{max}).

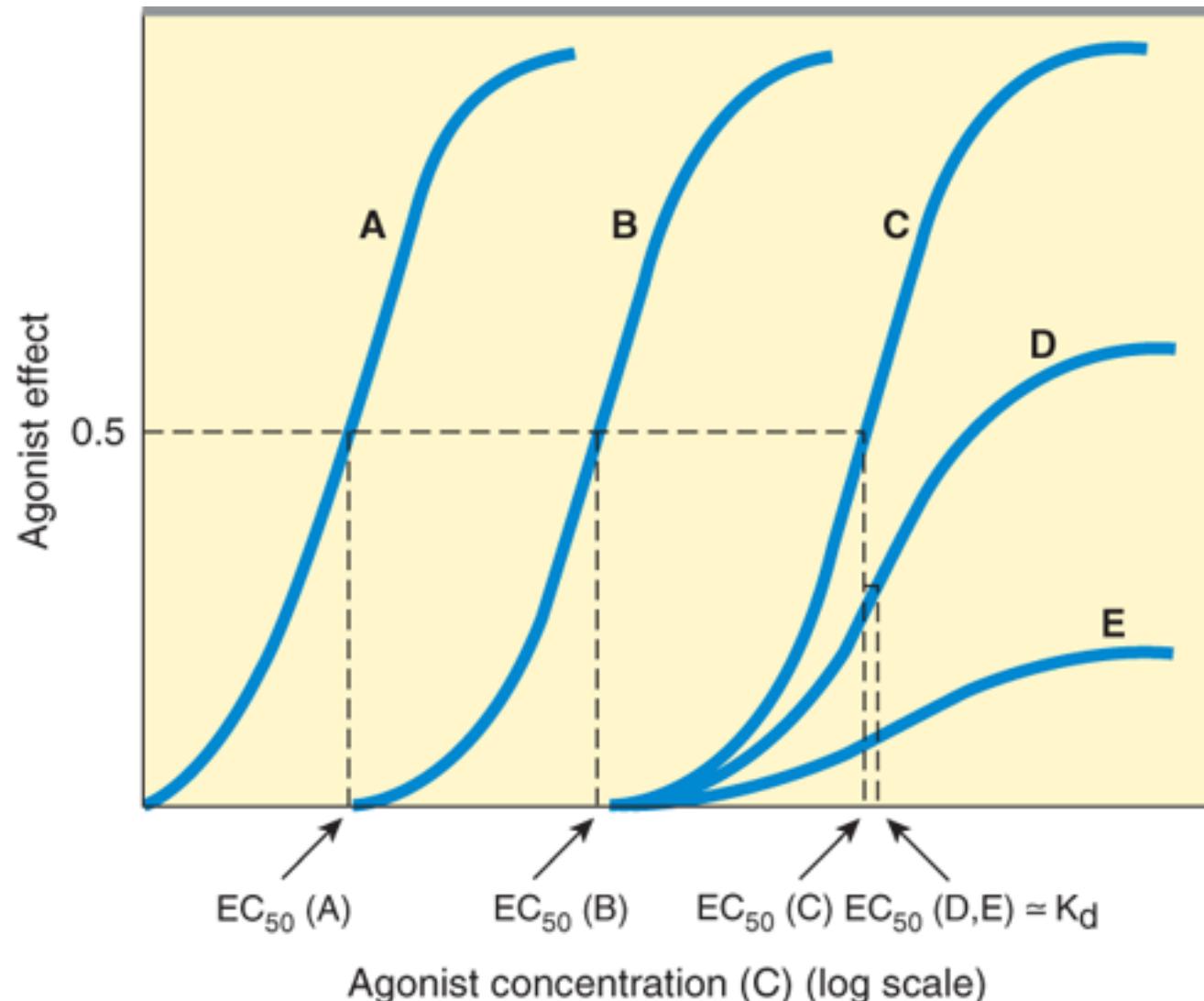
Said another way,

- Only a fraction of the total receptors are needed to elicit a maximal biologic response.

Thus,

- **The remaining receptors that are not bound by agonist are referred to as spare receptors.**

Spare receptors can be demonstrated experimentally by using an irreversible antagonist to prevent binding of the agonist to a proportion of available receptors.



A: Agonist response in absence of antagonist

B: Low concentration of antagonist: shift to right, maximal response preserved

Even with the higher concentration of agonist, activation of only a fraction of receptors in the receptor pool is adequate to achieve maximal response: the agonist's EC_{50} is still $< K_d$.

C: Larger concentration of antagonist: shift to right, max response preserved

The number of remaining receptors is just enough to achieve the maximal response: the agonist's $EC_{50} \approx K_d$.

D, E: Higher antagonist concentrations reduce the number of available receptors; response is diminished.

the agonist's $EC_{50} \approx K_d$.

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Basic & Clinical Pharmacology, Sixteenth Edition
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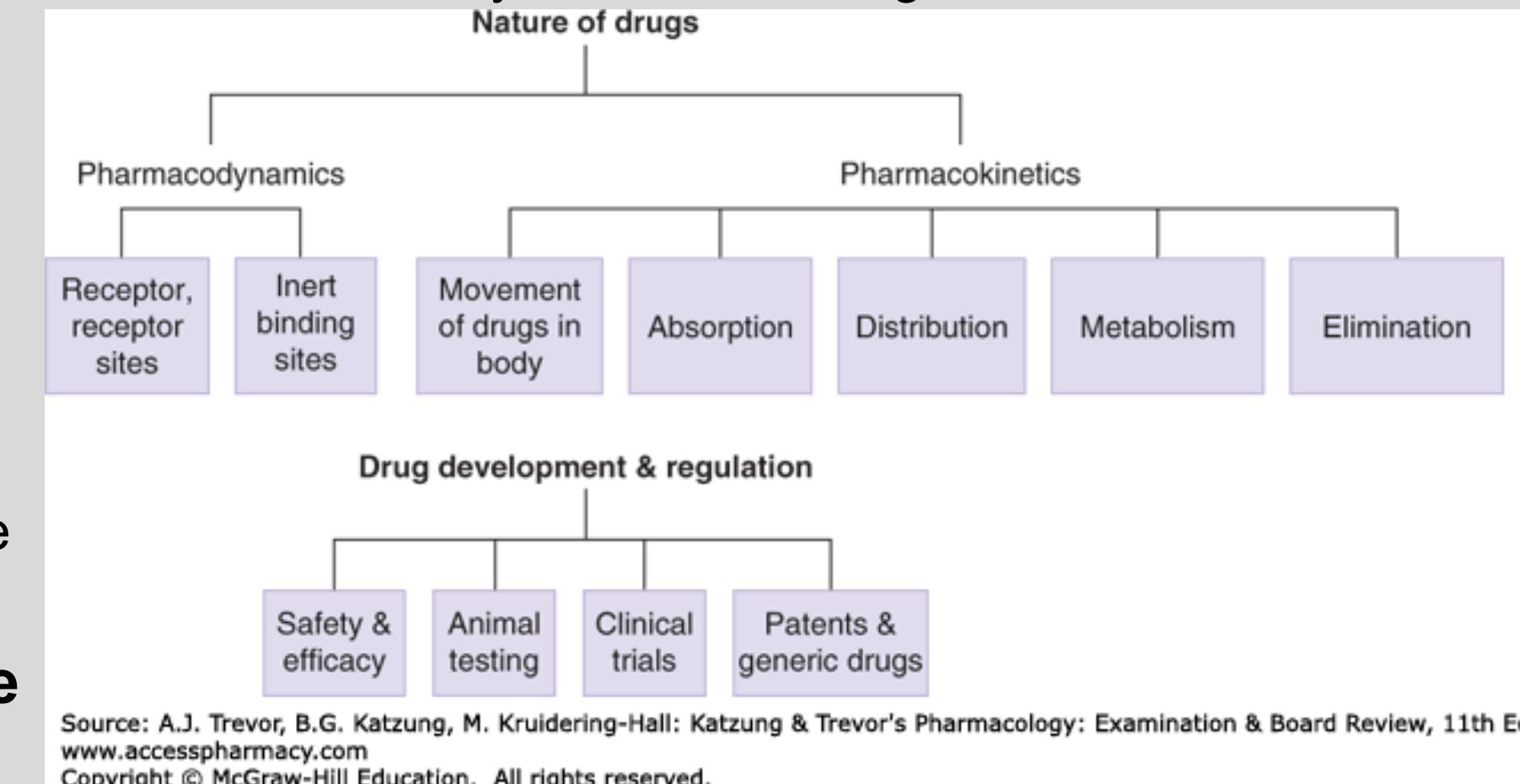
SUMMARY OF PHARMACODYNAMICS CONCEPTS

Part 1

Pharmacology is the body of knowledge concerned with the action of chemicals on biologic systems.

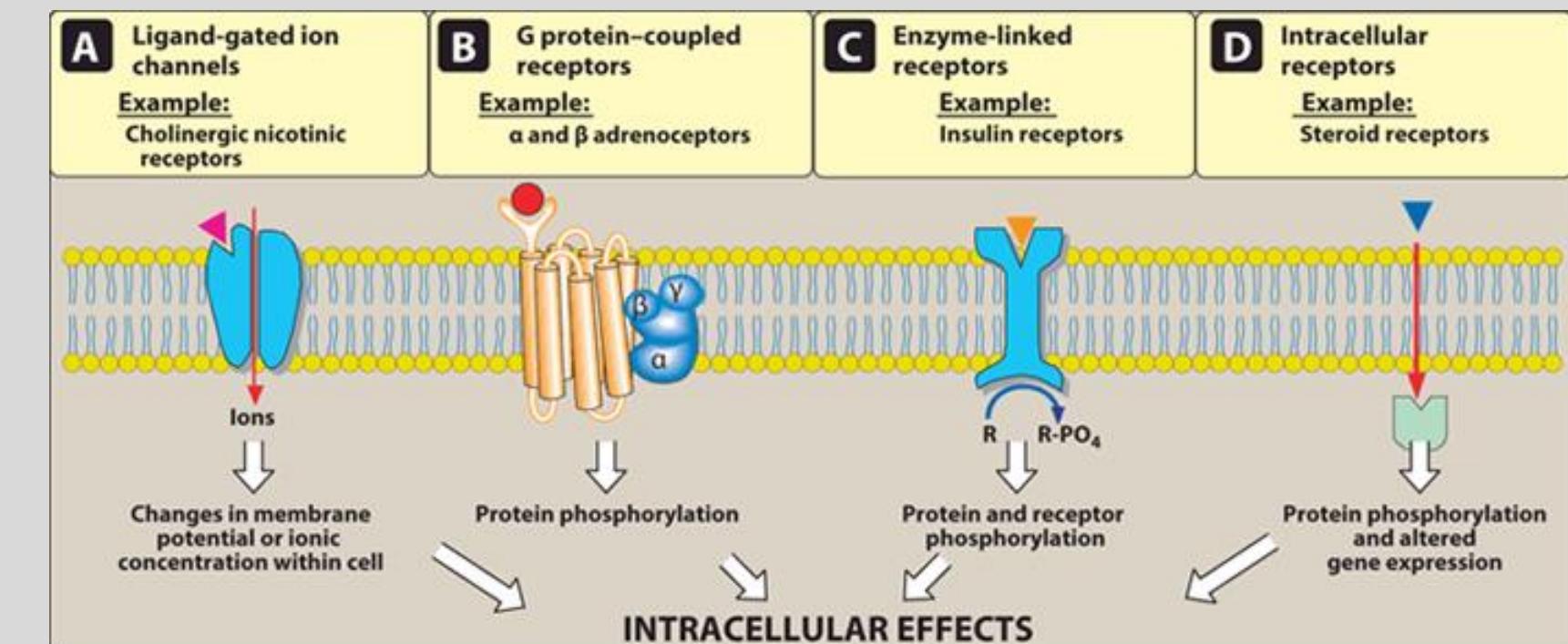
1. Pharmacodynamics is the study of the biochemical, cellular, and physiologic effects of drugs and their mechanisms of action. “What the drug does to the body, and how it does it.”
 2. Pharmacokinetics is the study of the movement and fate of drugs in the body: absorption, distribution, metabolism, and excretion. “What the body does to the drug, and how it does it.”
- Knowledge of biochemical, cellular, physiologic, and pathologic processes is essential for understanding pharmacodynamics and pharmacokinetics.
 - Knowledge of pharmacology is essential for the safe and effective use of drugs:

Optimizing drug therapy in the individual patient.



The Nature of Drugs: Structure and Function

- The physicochemical characteristics of a drug determines its ability to move through the body and initiate a cellular effect.
- The physicochemical characteristics of a drug that predict its movement and availability at sites of action are its molecular size, shape, and chirality, degree of ionization, and relative lipid solubility of its ionized and non-ionized forms.
- The structure-activity relationship is the **relationship** between the chemical or 3D **structure** of a molecule and its **biological activity**.
- The interaction of a drug with its receptor is the fundamental event that initiates the action of the drug.



Pharmacodynamics Part 2:

- Quantifying Agonism and Antagonism
- Receptor Dynamics
- Pharmacodynamic Variability in Populations

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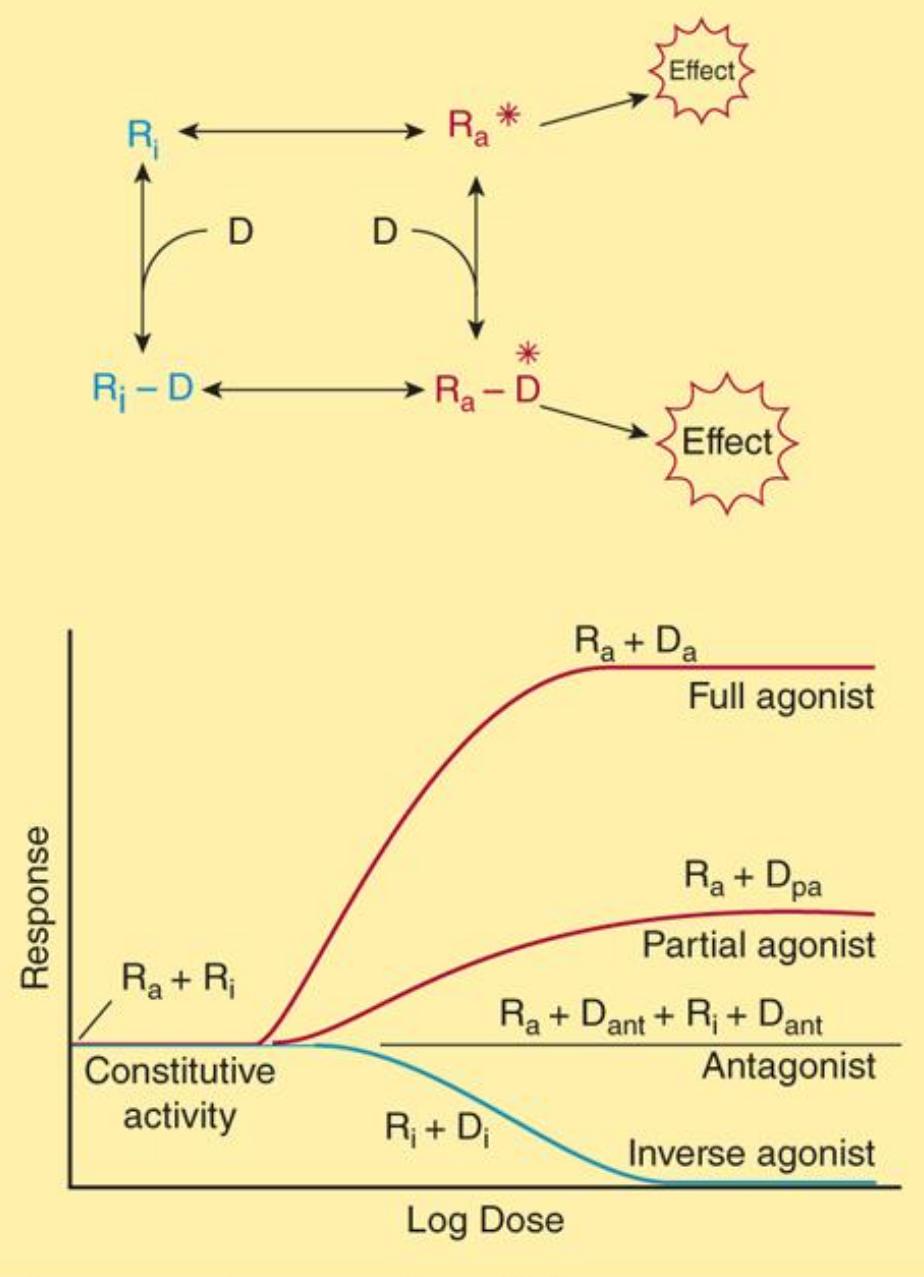
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LWW Health Library Premier Basic Sciences, Lippincott Illustrated Reviews: Pharmacology, 8e, 2023: Chapter 2: Drug-Receptor Interactions and Pharmacodynamics

<https://premiumbasicsciences-lwwhealthlibrary-com.nyit.idm.oclc.org/content.aspx?sectionid=253325200&bookid=3222>



Source: Bertram G. Katzung, Todd W. Vanderah:
Basic & Clinical Pharmacology, Fifteenth Edition
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Two-state model of drug-receptor interaction. The hypothetical receptor is able to assume two conformations.

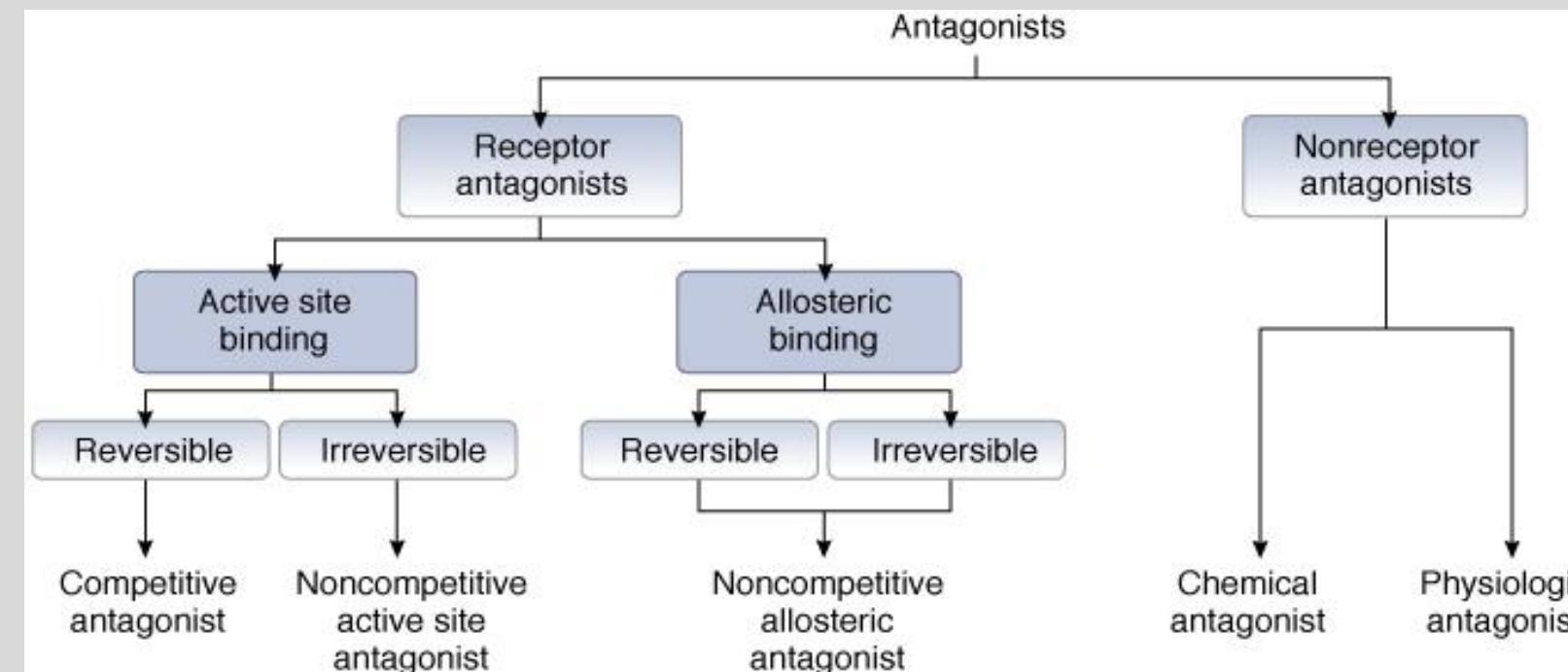
The receptor in the R_i conformation is inactive and produces no effect, even when combined with a drug molecule. In the R_a conformation, the receptor can activate downstream mechanisms that produce a small observable effect, even in the absence of drug (constitutive activity). In the absence of drugs, the two isoforms are in equilibrium, and the R_i form is favored.

- **Full agonists** have a much higher affinity for the R_a conformation and shift almost all of the receptors to the R_a -D pool, which can activate the receptor-effector systems to the maximal effect.
- **Partial agonists** have slightly greater affinity for R_a than for R_i . They do not stabilize the R_a configuration as fully as full agonists; a significant fraction of receptors exist in the R_i -D pool. They prevent the full agonist from accessing the site of action.
- **Neutral antagonists** have equal affinity for both receptor forms and maintain the same level of constitutive activity. The drug will appear to be without effect, but will block the agonist from accessing the site of action.
- **Inverse agonists** have a much higher affinity for the R_i form, stabilize a large fraction in the R_i -D pool, and reduce constitutive activity, resulting in an effects opposite those of the agonist.

Classical Receptor Occupancy Model

Classification of Antagonists

- Competitive antagonists compete for the same binding site as the agonist and affect the amount of agonist needed to achieve a maximal response.
- Noncompetitive antagonists reduce the magnitude of the maximal response that can be achieved by any amount of agonist without having an effect on the binding of the agonist to the receptor.



LWW Health Library Principles of Pharmacology, 2017 Figure 2-4: Antagonist Classification

- Physiologic antagonists are substances that act on different endogenous regulatory pathways mediated by different receptors and have opposite (counterbalancing) effects.
- Chemical antagonists are compounds that directly interact with the agonist, modifying or sequestering it so that the agonist cannot bind its receptors.

An antagonist binds the receptor but does not activate it.

An antagonist blocks the effects of the agonist.

Competitive antagonist: A drug that binds reversibly to the *agonist binding site* on the receptor and competes with the agonist for the binding site (Panel A, upper).

- Inhibitory effects are surmountable with increasing agonist dose.
- The agonist's concentration-effect curve shifts to the right. E_{max} is not reduced

Irreversible active site antagonist: A drug that binds covalently (irreversibly) or with very high affinity (pseudo-irreversibly) at the same site as the agonist prevents the agonist from activating the receptor, even at high agonist concentrations, and depresses the agonist's maximal response.

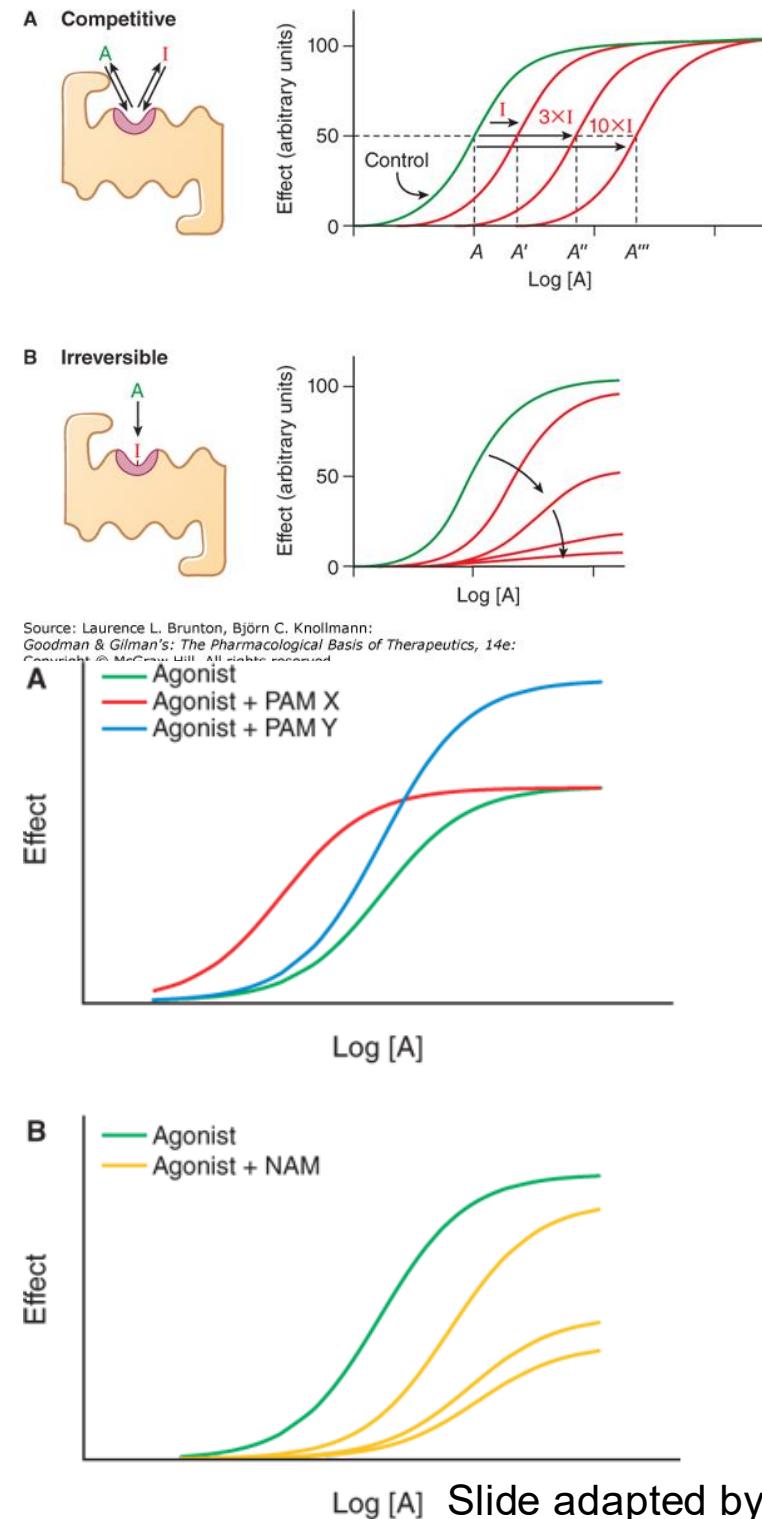
(Panel B, upper: As antagonist concentration increases, the agonist's dose-response curves shift rightward and downward, suggesting some degree of spare receptors.)

Positive allosteric modulator: A drug that binds to an allosteric site and *enhances the affinity* of the receptor for the agonist, increasing the agonist's effect.

(Panel A, lower: PAM X, shifts the agonist's EC_{50} leftward, which increases the affinity/potency of the agonist. PAM Y increases the E_{max} for the agonist.)

Negative allosteric modulator: A drug that binds to a site on the receptor other than the active site and reduces the affinity and/or the efficacy of the agonist. The EC_{50} may not be changed.

(Panel B, lower: In this set of dose-response curves, the negative allosteric modulator, NAM, has a negative impact on both EC_{50} and E_{max} , reducing affinity/potency and efficacy.)



Other Mechanisms of Antagonism

Physiologic (Functional) Antagonist

A substance that produces physiologic effects opposing the physiologic effects of another substance

- The functional actions on different regulatory pathways mediated by different receptors are opposite to each other.

Examples:

1. Insulin reduces blood sugar. Glucagon increases blood sugar.
2. Methacholine causes bronchoconstriction. Albuterol causes bronchodilation.

Chemical Antagonist

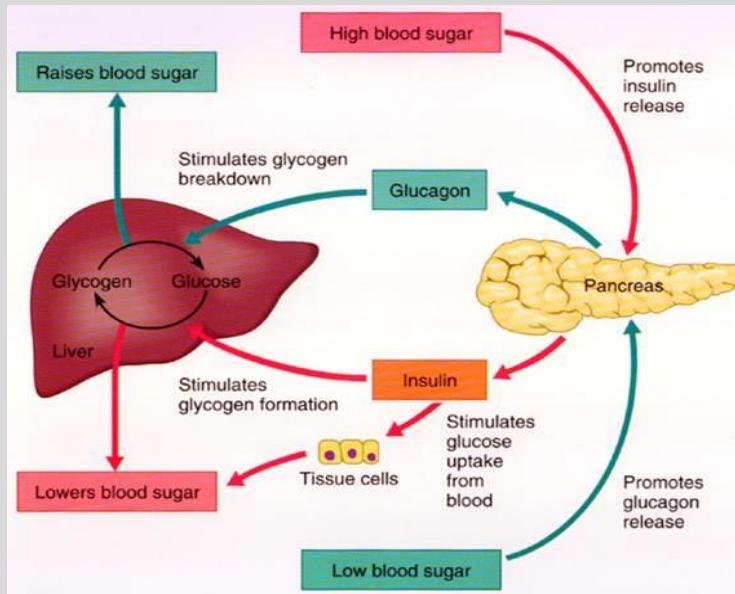
A compound that directly interacts with the agonist, modifying or sequestering it so that the drug is no longer capable of binding to its receptor.

- The product of the combination is inactive and excreted.

Examples:

1. Heparin is bound by protamine.
2. Divalent cations in antacids chelate tetracycline.
3. Cholestyramine is an cation exchange resin that sequesters many drugs in the gut, preventing absorption of that drug.

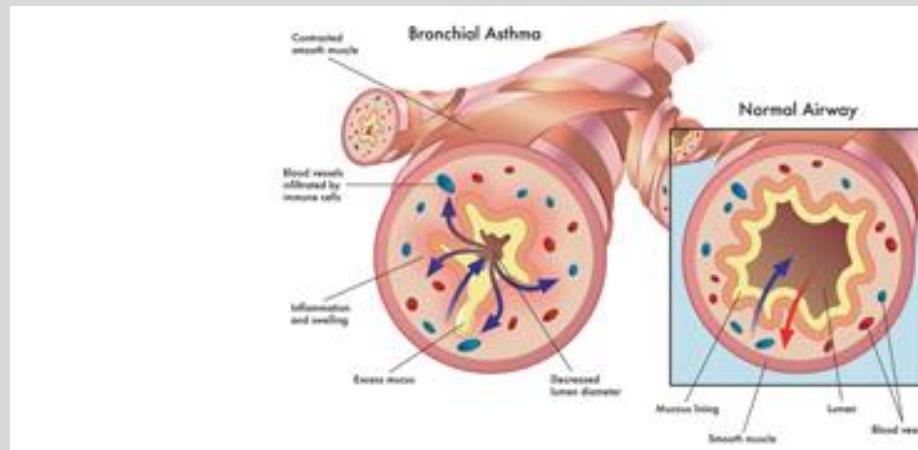
Example: The counterbalancing effects of physiologic (functional) antagonism



- Insulin lowers blood sugar by stimulating uptake of glucose by cells in the liver, skeletal muscle, and adipose tissue.
- Glucagon raises blood sugar by promoting glucose synthesis in the liver.

Clinical application: Glucagon may be administered for the treatment of severe hypoglycemia, such as from too much insulin.

- Methacholine (inhaled) is a cholinomimetic that activates muscarinic receptors in bronchial smooth muscle, which causes contraction – bronchoconstriction.
- Albuterol (inhaled) is a sympathomimetic that activates beta-2 receptors in bronchial smooth muscle, which causes relaxation – bronchodilation.



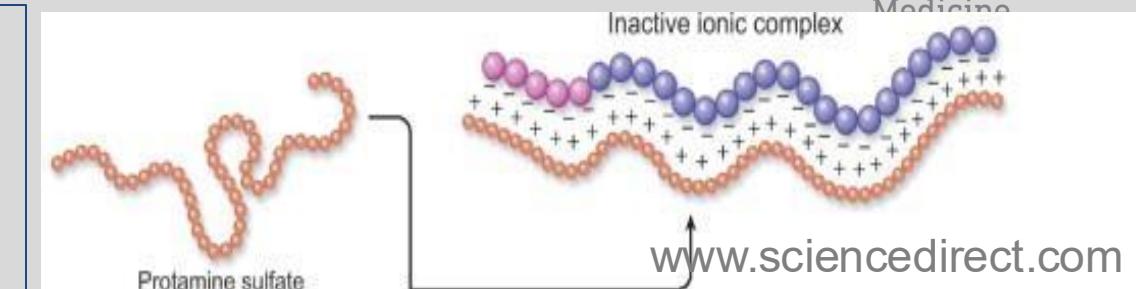
Clinical application:

Methacholine challenge is a test for bronchial hyperresponsiveness for the diagnosis of asthma.

Albuterol treats the asthma symptoms caused by the bronchoprovocation test by reversing the effects of methacholine.

Examples of Chemical Antagonists

- Positively charged protamine sulfate binds to negatively charged heparin in the bloodstream
- Heparin is not available to bind to its target
- Heparin's anticoagulant effect is neutralized



- Polyvalent metal ions chelate tetracycline forming a poorly absorbed complex.
- Antacids reduce absorption and reduce efficacy of tetracycline.

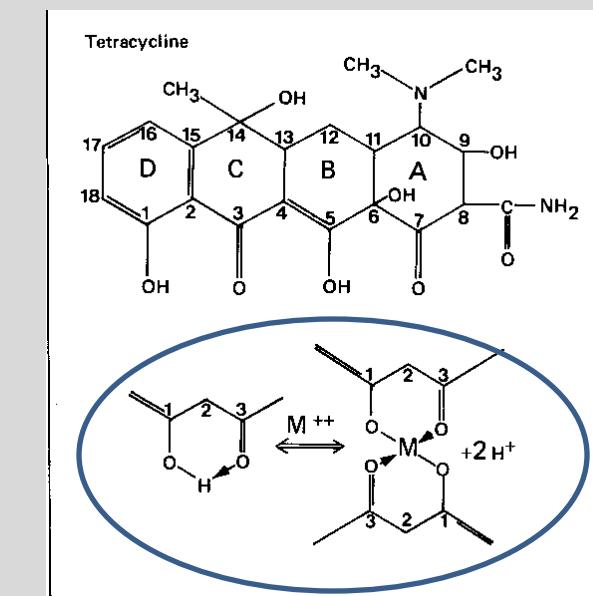
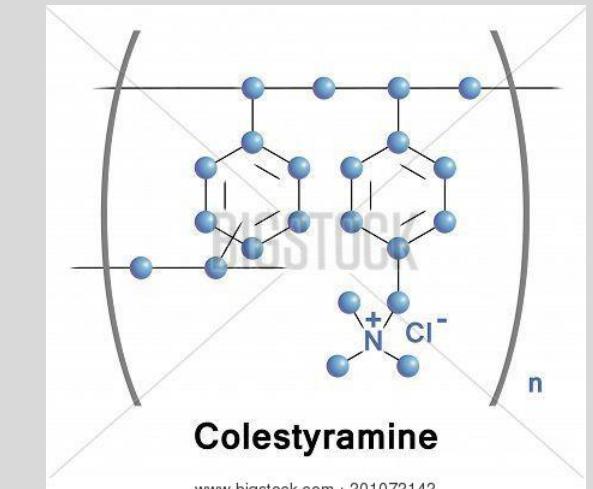


Fig. 1. Structural formula of tetracycline and the



Panel A: Response to a beta-adrenoceptor agonist vs time

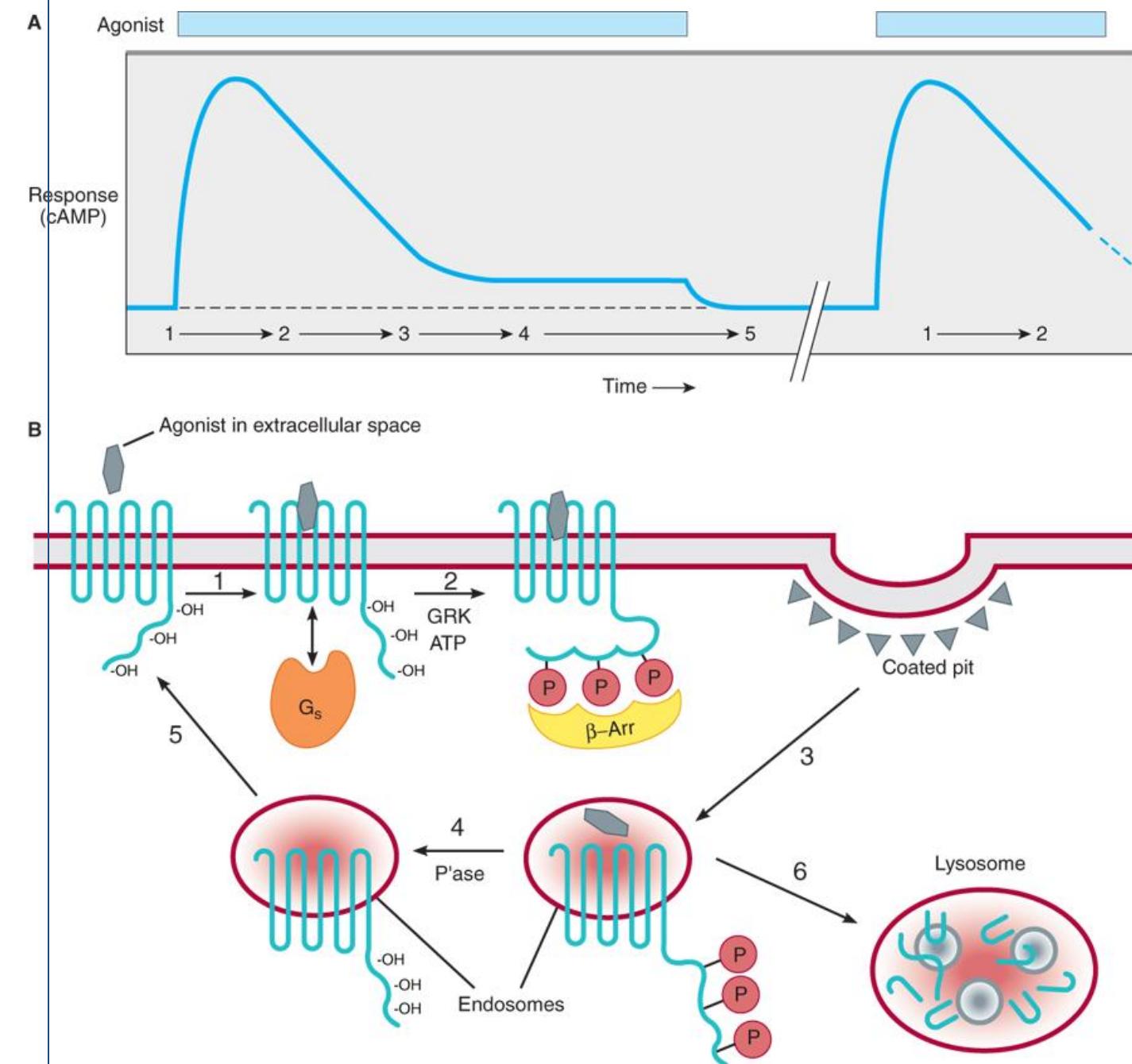
Resensitization: The cell recovers its capacity to respond when the agonist is removed for a short time.

Note: Resensitization is reduced or absent if cells are exposed to agonist repeatedly or over a more prolonged time.

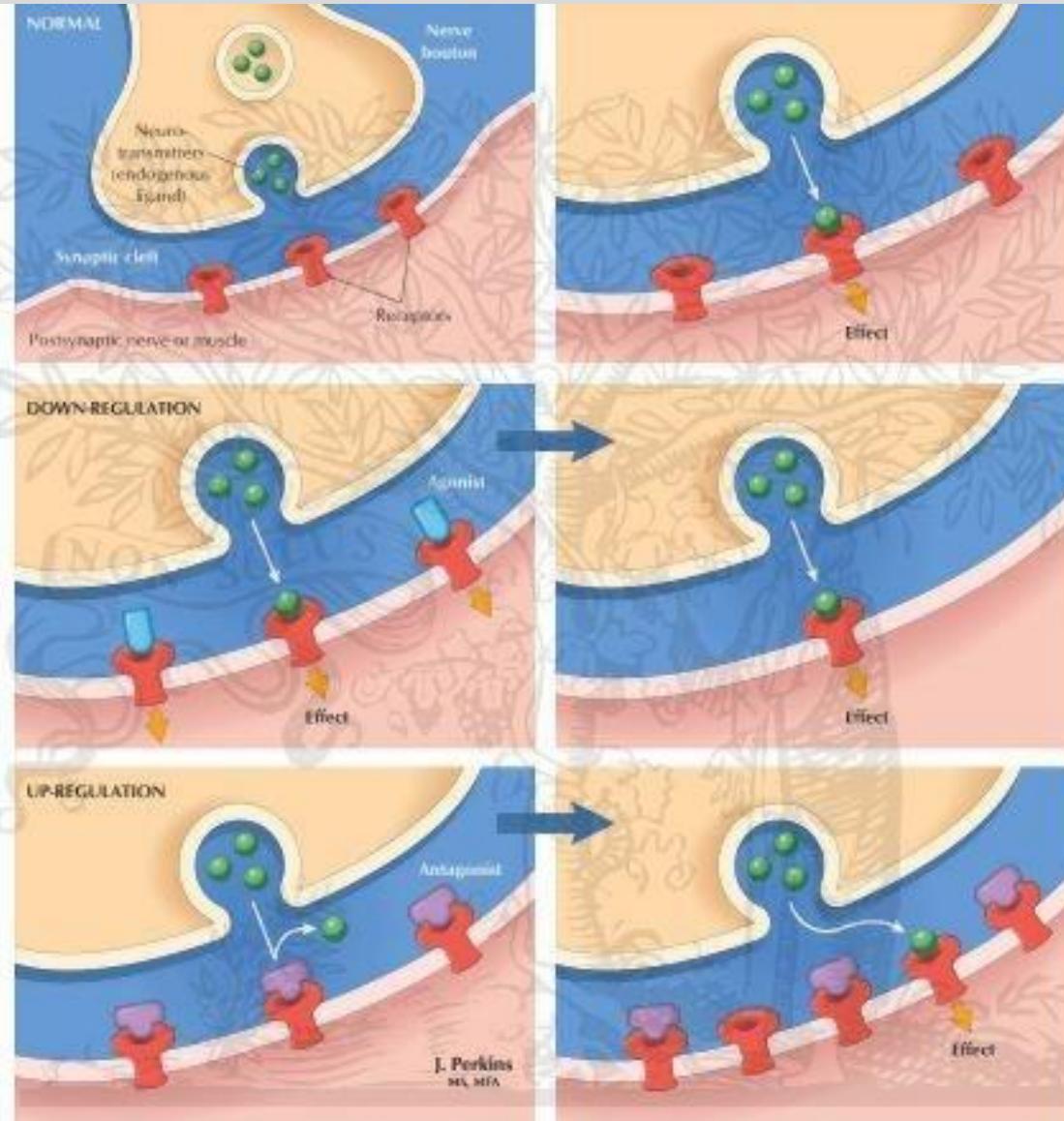
Panel B

1. Agonist binding → signaling initiated
2. Phosphorylation by a GPCR kinase (GRK) and binding of arrestin (-Arr) → prevents R interaction with G protein
3. R-arrestin complex binds to coated pits → receptor internalization
4. Agonist dissociates from internalized receptors →→ receptor dephosphorylation
5. Receptors are returned to the plasma membrane → resensitization of cellular responsiveness
6. Receptors are internalized in lysosomes (receptor downregulation) is favored by repeated or prolonged exposure of cells to agonist

Adrenoceptor desensitization, resensitization and downregulation of β adrenergic receptors



Source: Bertram G. Katzung, Todd W. Vanderah:
Basic & Clinical Pharmacology, Fifteenth Edition
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NORMAL

Receptor downregulation with exposure to an agonist:

Internalization and degradation can lead to reduced receptors on cell surface and diminished response.

Receptor upregulation with exposure to an antagonist:

Increased expression of receptors on cell surface can lead to an exaggerated response (supersensitivity) after the antagonist is withdrawn.

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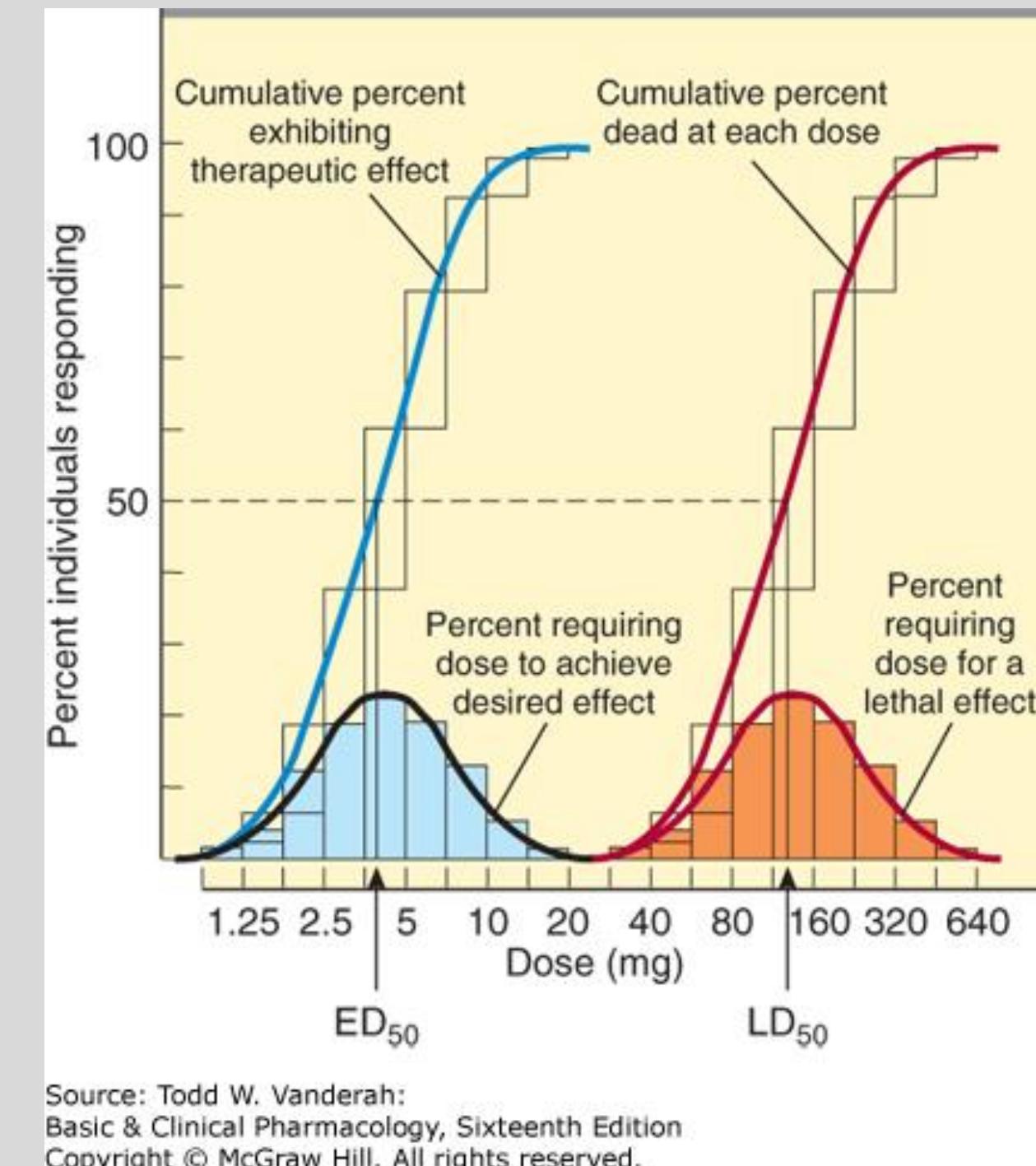
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<http://www.netterimages.com/image/5037.htm>

Pharmacology - Raffa 1E Author: Robert B. Raffa, PhD, Sc... Chapter: Basic Principles Page: 20

- In a population, there is usually some variation in the dose needed to achieve a specified drug effect.
- The response elicited with each dose is plotted against the log dose of the drug.
- The percentage of the population affected increases as the dose is raised.



Source: Todd W. Vanderah:
Basic & Clinical Pharmacology, Sixteenth Edition
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Therapeutic index (TI) is an *estimate* of the safety of a drug.

- **Median effective dose (ED_{50}):** The dose of a drug required to produce a specified effect in 50% of the population.
- **Median lethal dose (LD_{50}):** The dose of a drug that is lethal in 50% of the population.
- **Median toxic dose (TD_{50}):** The dose of a drug that produces a specified toxic effect in 50% of the population.

The therapeutic index relates the dose of a drug required to produce a desired effect to the dose that produces an undesired effect.

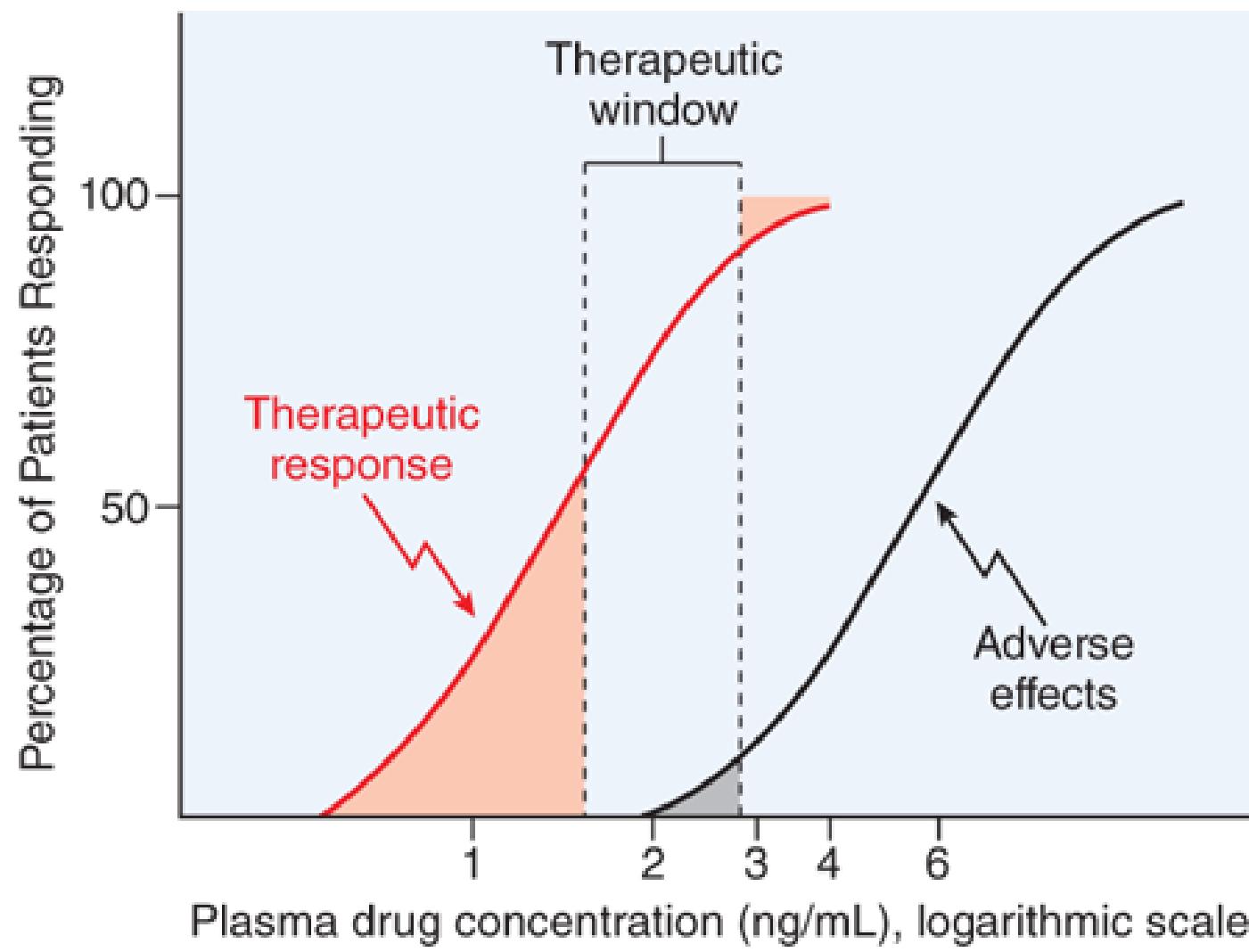
LD_{50} and TD_{50} are experimentally defined.

$$TI = LD_{50}/ED_{50}$$

$$TI = TD_{50}/ED_{50}$$

A large TI indicates a wide margin between effective doses and toxic doses.

A small TI indicates a narrow margin. Small changes in systemic concentration can lead to significant drug-related adverse effects.



Source: Laurence L. Brunton, Björn C. Knollmann:
Goodman & Gilman's: The Pharmacological Basis of Therapeutics, 14e:
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Relation of the therapeutic window of drug concentrations to therapeutic and adverse effects in the population. The ordinate is linear; the abscissa is logarithmic. This particular therapeutic window represents the difference in drug concentrations eliciting a therapeutic response in 50% of the patients and adverse effects in 10%.

The therapeutic window represents the range of steady state concentrations (dose range) at which the:

- likelihood of efficacy is high, and the
- probability of adverse effects is low.

This range does not guarantee safety or efficacy in individual patients.

Pharmacodynamics is the study of the effects of drugs on biologic systems.

