

X757/75/11

Physics Relationships Sheet

THURSDAY, 22 MAY 9:00 AM - 11:00 AM





$$E_p = mgh$$

$$E_k = \frac{1}{2}mv^2$$

$$Q = It$$

$$V = IR$$

$$R_T = R_1 + R_2 + \dots$$

$$\frac{1}{R_T} = \frac{1}{R_1} + \frac{1}{R_2} + \dots$$

$$V_2 = \left(\frac{R_2}{R_1 + R_2}\right) V_s$$

$$\frac{V_1}{V_2} = \frac{R_1}{R_2}$$

$$P = \frac{E}{t}$$

$$P = IV$$

$$P = I^2 R$$

$$P = \frac{V^2}{R}$$

$$E_h = cm\Delta T$$

$$p = \frac{F}{A}$$

$$\frac{pV}{T}$$
 = constant

$$p_1V_1 = p_2V_2$$

$$\frac{p_1}{T_1} = \frac{p_2}{T_2}$$

$$\frac{V_1}{T_1} = \frac{V_2}{T_2}$$

$$d = vt$$

$$v = f\lambda$$

$$T = \frac{1}{f}$$

$$A = \frac{N}{t}$$

$$D = \frac{E}{m}$$

$$H = Dw_R$$

$$\dot{H} = \frac{H}{t}$$

$$s = vt$$

$$d = \overline{v}t$$

$$s = \overline{v} t$$

$$a = \frac{v - u}{t}$$

$$W = mg$$

$$F = ma$$

$$E_w = Fd$$

$$E_h = ml$$

Additional Relationships

Circle

circumference = $2\pi r$

area =
$$\pi r^2$$

Sphere

$$area = 4\pi r^2$$

volume =
$$\frac{4}{3}\pi r^3$$

Trigonometry

$$\sin \Theta = \frac{\text{opposite}}{\text{hypotenuse}}$$

$$\cos \Theta = \frac{\text{adjacent}}{\text{hypotenuse}}$$

$$\tan \theta = \frac{\text{opposite}}{\text{adjacent}}$$

$$\sin^2\theta + \cos^2\theta = 1$$

Electron Arrangements of Elements

| | | 87 Fr 2,8,18,32, 18,8,1 Francium | 55 Cs 2,8,18,18, 8,1 Caesium | Rubidium | 7 8 18 8 1 | 37 | Potassium | 2.8.8.1 | 5 19 | Sodium | 2,8,1 | Na | 11 | Lithium | , <u> </u> | <u>.</u> . | Hydrogen | | エ → | (1) | Group 1 |
|---|--|--|--|----------------------|-------------------------|----------|-----------|-------------------------------|------------------|------------|-------|---------------------|----|------------|-------------------------|------------|----------------------|--------|---------------|------|-------------------------------|
| Actinides | Lanthanides | 88 Ra 2,8,18,32, 18,8,2 Radium | 56 Ba 2,8,18,18, 8,2 Barium | Strontium | Sr | 38 | Calcium | 2.8.8.2 | 20 | Magnesium | 2,8,2 | Μg | 12 | Bervllium | י ר | 4 0 | <u> </u> | (2) | | | Group 2 |
| | | 89 Ac 2,8,18,32, 18,9,2 Actinium | 57 La 2,8,18,18, 9,2 Lanthanum | Yttrium | 7 8 18 9 7 | 39 | Scandium | 2.8.9.2 | S 21 | (3) | | | | | | | | | | | |
| 89 Ac 2,8,18,32, 18,9,2 Actinium | 57 La 2,8,18, 18,9,2 Lanthanum | 104 Rf 2,8,18,32, 32,10,2 Rutherfordium | 72 Hf 2,8,18,32, 10,2 Hafnium | 10,2 Zirconium | Zr 2,8,18, | 40 | Titanium | 2.8.10.2 | : | (4) | | | | | | | | | Key | | |
| 90 Th 2,8,18,32, 18,10,2 Thorium | 58 Ce 2,8,18, 20,8,2 Cerium | 105 Db 2,8,18,32, 32,11,2 Dubnium | 73 Ta 2,8,18, 32,11,2 Tantalum | 12,1 Niobium | Nb 2,8,18, | 41 | Vanadium | v 2.8.11.2 | 23 Y | (5) | | | | | | | Electro | | Ato | | r |
| 91 Pa 2,8,18,32, 20,9,2 Protactinium | 59 Pr 2,8,18,21, 8,2 Praseodymium | 106 Sg 2,8,18,32, 32,12,2 Seaborgium | 74 W 2,8,18,32, 12,2 Tungsten | | Mo 2,8,18,13, | 42 | Chromium | 2.8.13.1 | 24 | (6) | | _ | | | 2 | Name | Electron arrangement | Symbol | Atomic number | | |
| 92 U 2,8,18,32, 21,9,2 Uranium | 60 Nd 2,8,18,22, 8,2 Neodymium | 107 Bh 2,8,18,32, 32,13,2 Bohrium | 75 Re 2,8,18,32, 13,2 Rhenium | 2 Technetium | Tc 2,8,18,13, | 43 | Manganese | 2.8.13.2 | 25 | (7) | | Transition Elements | | | | | ement | | ber | | רוכיניו לוו לווימופיוויוני לו |
| 93 Np 2,8,18,32, 22,9,2 Neptunium | 61 Pm 2,8,18,23, 8,2 Promethium | 108 Hs 2,8,18,32, 32,14,2 Hassium | 76 Os 2,8,18,32, 14,2 Osmium | | Ru 2,8,18,15, | 4 | Iron | 2.8.14.2 | 7 6 | (8) | | Element | | | | | | | | | |
| 94 Pu 2,8,18,32, 24,8,2 Plutonium | 62 Sm 2,8,18,24, 8,2 Samarium | 109 Mt 2,8,18,32, 32,15,2 Meitnerium | 77 Ir 2,8,18,32, 15,2 Iridium | 1 Rhodium | Rh 2,8,18,16, | 45 | Cobalt | 2.8.15.2 | 27 | (9) | | ίς | | | | | | | | | ר ני |
| 95 Am 2,8,18,32, 25,8,2 Americium | 63 Eu 2,8,18,25, 8,2 Europium | 110 Ds 2,8,18,32, 32,17,1 Darmstadtium | 78 Pt 2,8,18,32, 17,1 Platinum | 18,0 Palladium | Pd 2,8,18, | 46 | Nickel | 2.8.16.2 | 28 N : | (10) | | | | | | | | | | | 8 |
| 96 Cm 2,8,18,32, 25,9,2 Curium | 64 Gd 2,8,18,25, 9,2 Gadolinium | Rg 2,8,18,32, 32,18,1 Roentgenium | 79 Au 2,8,18, 32,18,1 Gold | 18,1 Silver | Ag 2,8,18, | 47 | Copper | 2.8.18.1 | 29 | (11) | | | | | | | | | | | |
| 97 Bk 2,8,18,32, 27,8,2 Berkelium | 65 Tb 2,8,18,27, 8,2 Terbium | 110 111 112 Ds Rg Cn 2,8,18,32, 2,8,18,32, 2,8,18,32, 32,17,1 32,18,1 32,18,2 Darmstadtium Roentgenium Copernicium | 80 Hg 2,8,18, 32,18,2 Mercury | 18,2 Cadmium | Cd 2,8,18, | 48 | Zinc | 2.8.18.2 | 30 7 5 | (12) | | | | | | | | | | | |
| 98 Cf 2,8,18,32, 28,8,2 Californium | 66 Dy 2,8,18,28, 8,2 Dysprosium | | 81 T l 2,8,18, 32,18,3 Thallium | 18,3 Indium | ln 2,8,18 | 49 | Gallium | 2.8.18.3 | 3 31 | Aluminium | 2,8,3 | ≥ | 13 | Boron | ر د | 0 U | 7 | (13) | | | Group 3 |
| 99 Es 2,8,18,32, 29,8,2 Einsteinium | 67 Ho 2,8,18,29, 8,2 Holmium | | 82 Pb 2,8,18, 3, 32,18,4 m Lead | | 2, | 50 | _ | .3 2.8.18.4 | 32 | ım Silicon | 2,8,4 | Si | _ | | ٠ ر | ٥ ر | \ \ | (14) | | | 3 Group 4 |
| 100 Fm 2,8,18,32, 30,8,2 Fermium | 68 Er 2,8,18,30, 8,2 Erbium | | 83 Bi 32,8,18, 4 32,18,5 Bismuth | 18,5 Antimony | | 51 | | .4 2.8.18.5 | > 33 | Phosphorus | 2,8,5 | P | | n Nitrogen | ² л 2 | z | 7 | (15) | | | 4 Group 5 |
| 101 Md 2,8,18,32, 31,8,2 Mendelevium | 69 Tm 2,8,18,31, 8,2 Thulium | | 84 Po 2,8,18, 32,18,6 h Polonium | 18,6 ny Tellurium | | 52 | | ع ن کا 5 2.8.18.6 | 34 | rus Sulfur | 2,8,6 | S | | n Oxvaen | , c | > ∝ | • | (16) | | | 5 Group 6 |
| 102 No 2,8,18,32, 32,8,2 Nobelium | 70 Yb 2,8,18,32, 8,2 Ytterbium | | 85 At 3, 2,8,18, 6 32,18,7 M Astatine | | 2, | \dashv | | .6 2.8.18.7 | 5 35 | . Chlorine | 2,8,7 | CI | | n Fluorine | 2 7 | П | | (17) | | | 6 Group 7 |
| 103 Lr 2,8,18,32, 32,9,2 Lawrencium | 71 Lu 2,8,18,32, 9,2 Lutetium | | 86 Rn 3, 2,8,18, 7 32,18,8 Radon | 18,8 Xenon | | | | .7 2.8.18.8 | 36 | ne Argon | 2,8,8 | Δŗ | | Neon | > T | Z = | Hellum | 2 | He 2 | (18) | 7 Group 0 |
| | | | 00-J | | <u>•</u> | | ر د | ∞ | | | , | | | | | | | | | | 0 |