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Photosynthesis and Electronic Properties of Fenna-Matthews-Olson Light Harvesting Complexes

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Photosynthesis is a key process used by plants and other organism to convert light energy from sun to the chemical energy which is then used to fuel the organism. The light harvesting pigment, chlorophyll, is the most abundant in the organism during the process of photosynthesis. It absorbs light from sun as well as converts it into the chemical energy, and the bacteriochlorophyll molecules play a vital role in this. Although they are known to be usually water insoluble, there exists a particular type of bacteriochlorophyll protein which is soluble in water. It is known as the Fenna-Matthews-Olson complex and it is found in the green sulfur bacteria. The particular structure of the Fenna-Matthew-Olson complex is shown in figure 1(a). The FMO protein complex contains seven bacteriochlorophyll (Bchl)a molecules wrapped in a string bag of protein. Many experimental and computational studies have been done on the structure of chlorophyll applying various techniques including solid state and liquid state NMR to understand its role in the photosynthesis [4, 2]

In order to understand the process of photosynthesis in detail, the study on the electronic structure and physical properties of the chlorophyll is essential. In view of this, we have studied energetic and the spectroscopic properties of the seven different types of the bacteriochlorophylls using density functional theory. The presence of the seven different bacteriochlorophylls is demonstrated in the model structure of the FMO complex and Chlorophyll in Figure 1.

Here we note that the chemical shift of a nucleus is defined by the relation [1]

$$\delta = \sigma_{iso} - \sigma, \quad (1)$$

where σ is the shielding tensor which describes the magnetic polarizability of a molecule of the nucleus under the study and the σ_{iso} is the isotropic value of the shielding tensor of the standard reference taken in the NMR experiment. Mathematically, the shielding tensor is defined by the relation

$$\sigma_{\alpha\beta} = \left(\frac{\partial^2 E}{\partial \mu_\alpha \partial \beta} \right). \quad (2)$$

The calculation of spectra was done by using the functional PBE1PBE of Purdew-Burke-Edword with basis set 6-311G. The structures used in our studies have been taken from the protein data bank which was determined by the X-ray crystallographic method [3]. We have used the Gaussian broadening to calculate the NMR spectra. The geometries obtained from the structures have been optimized using density functional

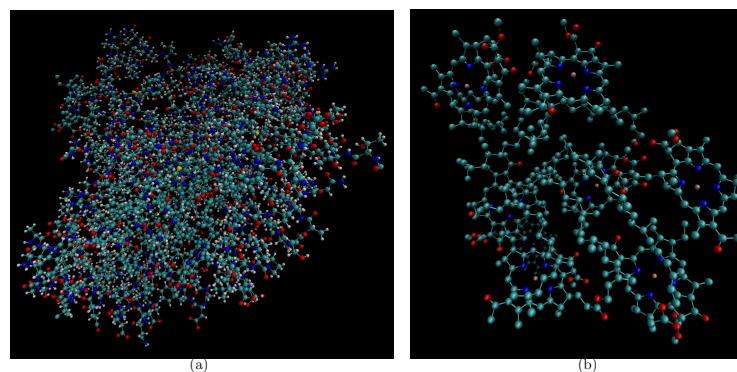


Fig. 1. VMD generated structure of the Fenna-Mathew-Olsen light harvesting complex (a) The whole structure and (b) The model structure for Chlorophyll

theory implemented by the Gaussian 09 set of programs. The chemical shift has been calculated by taking the difference in shielding tensors from the chemical shift of the carbon atom in TMS. Since the crystal structure of the FMO complex was from the X-ray spectra, the missing hydrogen atoms have been added to saturate the dangling bonds. Also, the separation of seven different chlorophylls has been done by using the computational visualization software VMD. In our calculation we have taken the tetramethylsilane (TMS) as a reference system. Furthermore, we also study the infrared spectrum for each of these components using density functional theory. We also determine the energy for all the bacteriochlorophylls present in the FMO complex. In order to address the nuclear properties of the complex and get a better insight into electronic properties, we have estimated the quadrupole coupling constant and the asymmetry parameters for the pyrrole nitrogen and the magnesium at the center of the porphyrin ring of Chlorophyll. The results obtained from our theoretical investigations have been compared to the experimental results available in the literature.

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