```
Fe examples:
                                                                 fe -> fe2+ +2e-, Fe2+2420 +2le-> fe(04)2+24cl
                                                                     Fe00H + 420 -> Fe3+ + 30H"
                                                                     FROOM + U- -> FROCE + OH-
                                                                           FeOCE + 420 -> Fe3++ CL-+ 20H-
  Post-Minor 2
   LECTURE 21 (28/03/2023) : { Given in Leibure Stides }
             NOTE: In electrochemical setups, capacitonce can form commonly due to presence on form commonly due to presence of surface charges
                                                                                                                    t = electrolyte
* NOTE! In Major: Complete eyllabus
      MOTE: Pitting very commonly occurs in le environment
        LECTURE 22 (01/04/223)
      42] Ny-quist Plot (Impedona, Z = Z'+iZ")
                                                                this is the plat

| Impendance and Imaginary part

| Impendance | Impe
                                                                                                                                                                                                                                        * { The Z'value represents ]
                                                                                                                                                                                                                                                      Resistance/
                                  Electrochemical Double Layer
                                                                               (EDL) Inreality the
                                                                                                                            cell behaviour la not
to simple
                                                                                                                             it also has some capacitance
                                                                    doub formation
                                                                                                                                       Electrochemical poulde Layer (EDLC) *
```

44] , Re: resistance of electrolyte Capacitance: from EDLC " Polarization Resistance" Rp; resistance exchange of e's due to the exchange resistance") this is always parelled to capacitance: the plat for this will be : the plat Seni-circular shape (at min! f max! w, z"iso) Now, for the case we have with Re, Rp and Capacillance: the plot obtained is - Re+ Rp-There are types of lolarization Achivation Concentration
[(i.e. Diffusion) if this occurs: it leads to "Warburg Resistance"

eg: We have some output plat given: R3 Ry R5 There are 3 By Analysis: Semi circles thus 3 capacitors Thus, the plat imply the following would! m High Him H R2 R3 Here we are observing perfect Semicircles But in reality we may also observe Depressed semistre this kind of cure is observed in Solar cells Bode Plat Plot involving phose (\$), injedence (Z) and frequency 7 leg /2/ it is a Double-y plat there will be two curres drawn. log/z/ (eg: for y, -) in blue (igo) for yz in red whome) frequency (either for 27 f)

LECTURE 23 { Given in Lecture Stidles: Topic - Crevice Cornovion } LECTURE-24 E Given in Lecture Slides } * All lectures here onwards are given in stides.

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