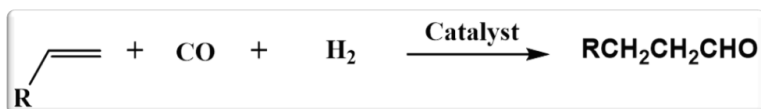
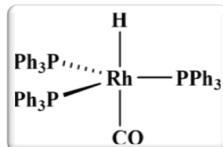
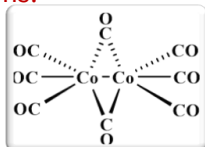


Hydroformylation reaction



- A less common, but more appropriate name is hydrocarbonylation
- Both cobalt and rhodium complexes are used as catalysts.
- Alkene isomerization, alkene hydrogenation and formation of branched aldehydes are the possible side reactions.



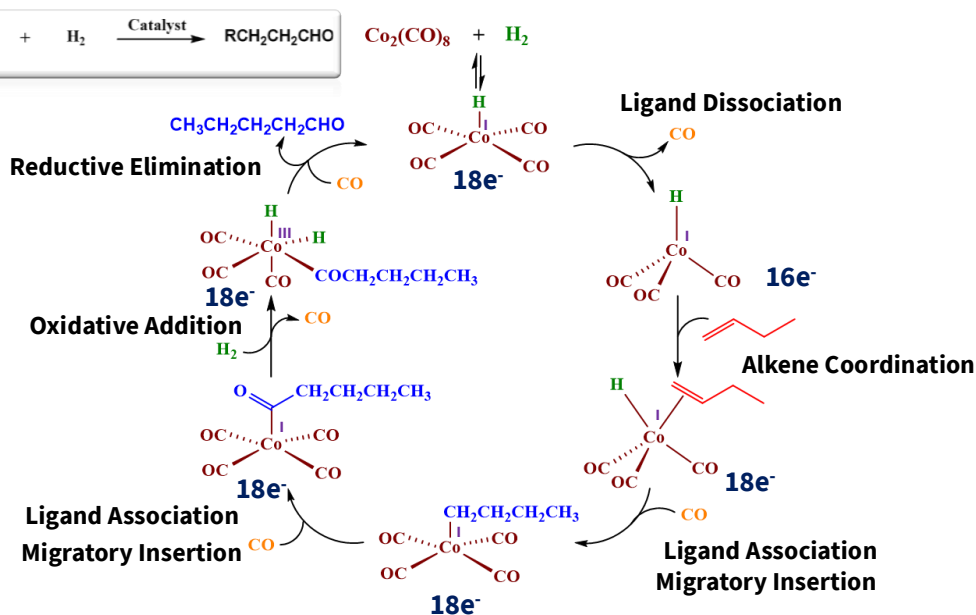
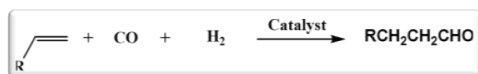
d⁹

27	Co
	Cobalt
45	Rh
	Rhodium
77	Ir
	Iridium

- Cobalt catalysts operate at 150 °C and 250 atm, whereas Rhodium catalysts operate at moderate temperatures and 1 atm.
- Rhodium catalysts promotes the formation of linear aldehydes. Cobalt catalysts do so if modified with alkylphosphine ligands.

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Hydroformylation reaction

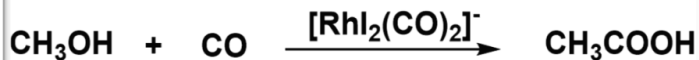


d⁹

27	Co
	Cobalt
45	Rh
	Rhodium
77	Ir
	Iridium

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Monsanto Acetic Acid Synthesis



d⁹

27	Co
	Cobalt
45	Rh
	Rhodium
77	Ir
	Iridium

- All three members of the group 9 (Co, Rh and Ir) can catalyze this reaction.
- A cobalt complex was initially used, which was replaced with the rhodium complex later on.
- Methanol initially reacts with hydroiodic acid to give methyl iodide and H₂O. Methyl iodide reacts with the 16e⁻ catalyst, which is the rate-determining step.



- The final product formed after the catalytic cycle is acetyl iodide, which is hydrolyzed by water to acetic acid and HI.



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Monsanto Acetic Acid Synthesis

