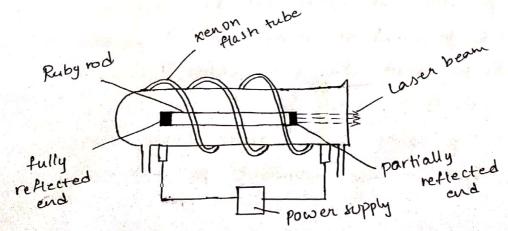
5. Ruby laser

It is a three level solid state laser and was constructed by Mainmann in 1960. Ruby (A1203 + C1203) is a crystal of Aluminium Oxide in which 0.05% of A1-3 ions are replaced by  $cr^{-3}$  ions. The color of rod is pink. The active medium in the ruby rod is  $cr^{-3}$  ions

## construction

In Ruby later 4 cm length and 5mm diameter rod is generally used Both the ends of rods are highly polished and made strictly parallel. The ends are silvered in a way, one become partially reflected and other end fully reflected The tuby rod is surrounded by Nenon Hash tube which provides the pumping light to excite the chromium ions to upper energy levels.

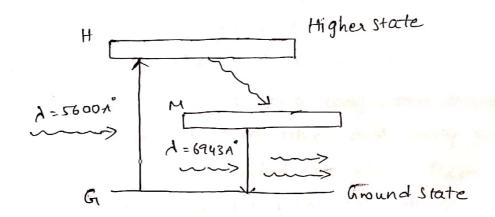


xenon flash trube emits thousands joules of energy in few milli seconds but only a part of that energy is utilized by chromium ions while test of the energy heats up the appartus a cooling temperature is provised to keep the experimental set up at normal

The energy level diagram of chromium ions shown in figure

TA TENT WATER

The chromium ions get excitation into higher energy levels by absorbing of 5600 1° of waveleng —the radiation. The excited chromium ions stay in the level H for short interval of time After their life time most of chromium ions are de-excited from H to G and a few chromium ions are ions are de-excited from H to M.



The transition. blu H&M is non-radio active transition. I-e the chromium ions gives their Energy to the lattice in form of heat. In the meta stable the life time of chromium ions is 10<sup>-3</sup> sec. The life time of chromium ions in meta stable state is so times greater than life time of chromium ions in Higher state

uses

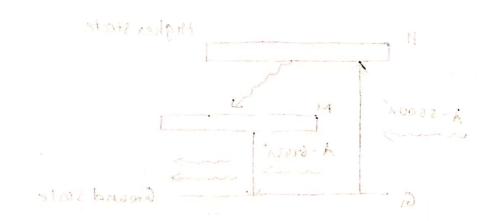
- 1. In optical photography
- 2 To remove melanin of skin
- 3. For recording of holograms

-> Laser requires high pumping power (1)

- A pulse laser

Efficiency is very small do per done personal and all missis match.

de-excited from H to G and a few chromium



The transition be a the is non-radio notive transition in the area done from the transition of heat done in the transition of the transition to the transiti

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