

Tutorial Sheet-3

[Laser Technology and Applications, 16B1NPH533, Odd Semester 2020]

1. [CO 1] If energy levels 1 and 2 are separated by an energy $E_2 - E_1$, such that the corresponding transition frequency falls in the middle of the visible range, calculate the ratio of the populations of the two levels in thermal equilibrium at room temperature.
[Ans: 1.1577×10^{-38}]
2. [CO 1] The wavelength of emission is 600 nm and the lifetime is 10^{-6} s. Determine the coefficient for the stimulated emission.
[Ans: 1.3×10^{19} m/kg]
3. [CO 1] For an optical source at thermal equilibrium (1000 K), having wavelength 500 nm, calculate the ratio of the number of spontaneous to stimulated emissions.
[Ans: 5×10^{12}]
4. [CO 1] For the $2P \rightarrow 1S$ transition in the hydrogen atom, the wavelength of the transition is 121.5 nm. The lifetime of the $2P$ state for spontaneous emission is 1.6×10^{-9} s. Calculate the Einstein's A and B coefficients.
[Ans: $6.25 \times 10^8 \text{ s}^{-1}$, 6.73×10^{19} m/kg]
5. [CO 1] The orange krypton line ($\lambda = 6058 \text{ \AA}$) has a coherence length of 20 cm. Calculate the line width in terms of wavelength.
[Ans: 0.018 \AA]
6. [CO 1] The coherence time for the red cadmium line ($\lambda = 6438 \text{ \AA}$) is about 10^{-9} s. Estimate the Monochromaticity of the line.
[Ans: 2×10^{-6}]
7. [CO 1] The sun rays subtends an angle of about $32'$ (32 minutes) on earth and fall at double slit arrangement with wavelength 5000 \AA by using appropriate filter. What should be the separation between the two slits in order to obtain good contrast fringes on the screen?
[Ans: $55.55 \mu\text{m}$]
8. [CO 1] Find the coherence length for white light source (400 nm to 700 nm).
[Ans: $1.008 \mu\text{m}$]
9. [CO 1] A laser beam has a wavelength of 720 nm and aperture 5 mm. The laser beam is focused towards moon the distance of which from earth is 4×10^8 m. Calculate (i) the angular spread and (ii) Axial (or Aerial) spread when the beam reaches to moon.
[Ans: (i) 1.75×10^{-4} radian, (ii) $1.55 \times 10^{10} \text{ m}^2$]