Tutorial 9 (Physics 2-15B11PH211)

Assignment 6: Show that FCC crystals are more closely packed than BCC crystals.

- 1. Write the miller indices of different planes of the family of {1 0 0 }. [CO1]
- 2. Find the Miller Indices of the plane that intercepts \vec{a} , \vec{b} , and \vec{c} axes at 3 A⁰, 4 A⁰, and 3 A⁰, respectively, in a tetragonal crystal with c/a = 1.5. [CO3]
- 3. A plane makes intercepts of 1 A⁰ and 2 A⁰ on \vec{a} and \vec{b} axes respectively, but parallel to \vec{c} axis. Find Miller Indices if a:b:c is 3:2:1. [CO3]
- 4. The interatomic separation of Pb crystal having fcc structure is 3.5 A⁰, calculate the number of atoms in 1 mm² area of (100) plane. **[CO4]**
- 5. From the X-rays diffraction (XRD) data, Fe is found to have a cubic cell parameter of 2.87 A⁰. Its density and mass are 7870 km/m³ and 55.85 amu, respectively. Find out the Bravias lattice structure of Fe and interatomic separation of Fe atoms. [CO4]
- 6. Calculate the packing fraction in **hexagonal close-packed (**HCP) crystal lattice. **[CO2]**
- 7. A BCC crystal is used to measure the wavelength of some X-rays. The Bragg's diffraction angle for the reflection from (110) plane is 20.2°. Calculate the wavelength of X-rays if the lattice constant of the crystal is 3.15 A°. [CO4]