

Tutorial 5, 19114018, Ayushman Tripathy

① Steam doesn't condense :

$$\text{Assuming } P = 500 \text{ kPa} = \text{const} \\ = 5 \text{ bar}$$

Initially steam, at 5 bar, 300°C , superheated-vap
Finally saturated vap. at 5 bar

$$h_{\text{final}} = h_{g, 5 \text{ bar}} = 2747.7 \text{ kJ/kg}$$

$$h_{\text{initial}} = h_{5 \text{ bar}, 300^\circ\text{C}} = 3064.20 \text{ kJ/kg}$$

$$\Rightarrow Q = \Delta H = m (h_{\text{initial}} - h_{\text{final}}) \\ \text{to water} \\ = m (3064.20 - 2747.7) = 316.5 m$$

$$\Delta h_{\text{water}} = h_{75} - h_{15} \quad \left\{ \text{from subcooled liq. table} \right\}$$

$$= h_{75} - h_{15} \quad \left\{ \text{from sat. table} \right\}$$

$$= 313.94 - 62.94 = 251.00 \text{ kJ/kg}$$

$$\Delta H_{\text{per kg}} = 251.00 \text{ kJ}$$

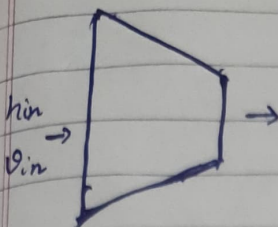
Equating both the heat values; we have.

$$m (316.5) = 251.00 \times 1$$

$$\Rightarrow m = \frac{251}{316.5} = 0.79304 \text{ kg}$$

$$\therefore \boxed{0.7930}$$

②

Inlet: $P = 10 \text{ bar}$ $\{1 \text{ MPa}\}$ $T = 400^\circ\text{C}$

Steam

$$h_{in} = 3263.88 \text{ kJ/kg}$$

$$\dot{m}_{in} = 0.1 \text{ kg/s}$$

Outlet: $P = 5 \text{ bar}$ $T = 350^\circ\text{C}$

Steam

$$h_{out} = 3167.65 \text{ kJ/kg}$$

We have

$$\dot{m}_{in} \left(h_{in} + \frac{v_{in}^2}{2} + g z_{in} \right)$$

$$= \dot{m}_{out} \left(h_{out} + \frac{v_{out}^2}{2} + g z_{out} \right)$$

For a nozzle $\dot{m}_{in} = \dot{m}_{out}$

as total mass inside = const

4 also z is same for inlet and outlet

$$\Rightarrow \dot{m} \left(h_{in} + \frac{v_{in}^2}{2} + g z \right) = \dot{m} \left(h_{out} + \frac{v_{out}^2}{2} + g z \right)$$

$$\Rightarrow h_{in} + \frac{v_{in}^2}{2} = h_{out} + \frac{v_{out}^2}{2}$$

 $v_{in} \ll v_{out} \Rightarrow$ can be neglected

$$\Rightarrow h_{in} = h_{out} + \frac{v_{out}^2}{2}$$

$$\Rightarrow v_{out} = \sqrt{2(h_{in} - h_{out})} = \sqrt{2(3263.88 - 3167.65) \times 10^3}$$

$$= 13.8729 \times \sqrt{1000} = 438.7026$$

$$\therefore \boxed{v_{out} = 438.7026 \text{ m/s}}$$

$$\dot{m} = 0.1 \text{ kg/s} = A_{\text{out}} v_{\text{out}} \rho$$

$$\Rightarrow A_{\text{out}} = \frac{\dot{m}}{\rho v_{\text{out}}} = \frac{0.1}{438.7026} v_{\text{steam}} = 2.279 \times 10^{-4} \text{ m}^2$$

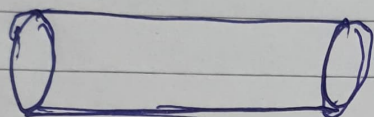
$$= 2.279 \times 10^{-4} \times 0.57012 \text{ m}^2$$

$$= 1.2995 \times 10^{-4} \text{ m}^2 = 1.2995 \text{ cm}^2$$

$$\therefore \begin{cases} v_{\text{out}} = 438.7026 \text{ m/s} \\ A_{\text{out}} = 1.2995 \text{ cm}^2 \end{cases}$$

③

$$P = 15 \text{ bar}$$



$$\rightarrow P = 2 \text{ bar}$$

$$T = 150^\circ\text{C}$$

$$v_{\text{in}} = 5 \text{ m/s}$$

Here, as KE is to be neglected

$$\Rightarrow \dot{m} \left(h_{\text{in}} + \frac{KE_{\text{in}}}{\dot{m}} + PE \right) = \dot{m} \left(h_{\text{out}} + \frac{KE_{\text{out}}}{\dot{m}} + PE \right)$$

$$= h_{\text{in}} + \frac{KE_{\text{in}}}{\dot{m}} = h_{\text{out}} + \frac{KE_{\text{out}}}{\dot{m}}$$

$$\Rightarrow \boxed{h_{\text{in}} = h_{\text{out}}} \quad \left\{ \text{As we neglect KE} \right\}$$

Here, $T < T_{\text{sat}}$ at 15 bar $\left\{ \therefore \text{subcooled liq} \right\}$
 $\{150^\circ\} \quad \{198.3^\circ\}$

$$h_{\text{in}} = h = h_f = 632.15 \text{ kJ/kg}$$

$\left\{ \text{As no vol in subcooled tube} \right\}$

$$\therefore h_{\text{in}} = h_{\text{out}} = 632.15 \text{ kJ/kg}$$

At 2 bar

$$h_f = 504.7 \text{ kJ/kg}$$

$$h_g = 2706.4 \text{ kJ/kg}$$

$$h_{fg} = 2201.7 \text{ kJ/kg}$$

$$h = 632.15 \text{ kJ/kg}$$

$$\therefore h_f < h < h_g$$

 \Rightarrow Sat. liq. vap mixture

$$x = \frac{h - h_f}{h_{fg}} = \frac{632.15 - 504.7}{2201.7} = 0.057887$$

 $\dot{m} = \text{constant}$

$$\Rightarrow \dot{m}_{in} A v_{in} = \dot{m}_{out} A v_{out}$$

$$\Rightarrow \frac{1}{v_{in}} \cdot A v_{in} = \frac{1}{v_{out}} A v_{out} \quad \left\{ \begin{array}{l} \dot{m}_{in} = \frac{1}{v_{in}} \\ v_{in} = \text{sp. volume} \\ \text{SH flow} = \frac{1}{v_{out}} \end{array} \right.$$

$$\Rightarrow v_{out} = \left(\frac{v_{out}}{v_{in}} \right) v_{in}$$

$$= v_{in} \frac{(v_f + x v_{fg})}{(v_{f, 2 \text{ bar}})} = 5 \times \left(\frac{0.001061 + 0.057887 \times 0.884969}{0.0010906} \right)$$

$$v_{f, out} = v_{f, 2 \text{ bar}} = 0.001061 \text{ m}^3/\text{kg}, \quad v_{g, 2 \text{ bar}} = 0.884969 \text{ m}^3/\text{kg}$$

$$v_{f, in} = v_f \text{ at } 150^\circ\text{C} = 0.0010906 \text{ m}^3/\text{kg}$$

$$0.0010906$$

$$= 5 \times 47.94535$$

$$= 239.7267 \text{ m/s}$$

$$\therefore v_{out} = 239.7267 \text{ m/s}$$

$$x = 0.057887$$

$$\Rightarrow Q + m \left(h_{in} + \frac{v_{in}^2}{2} + g z_i \right) = W + m \left(h_{out} + \frac{v_{out}^2}{2} + g z_o \right)$$

$Q=0$, 4 differentiating

$$\Rightarrow \dot{W} = \dot{m} \left(h_{in} - h_{out} + \frac{v_{in}^2 - v_{out}^2}{2} + g(z_{in} - z_{out}) \right)$$

$$\Rightarrow \bar{\omega} = h_{in} - h_{out} + \frac{v_{in}^2}{2} = \frac{v_{in}^2}{2} - \Delta h$$

$\{ \text{as given } v_{out} \approx 0, z_{in} = z_{out} \}$

$p_{in} = 2000 \text{ kPa} = 20 \text{ bar}$

\Rightarrow Subcooled liq
 $\} \text{ as } P_{\text{sat}} = 0.92397 \text{ bar}$
 at 20°C

h_{out}: 100 kPa = 1 bar, 20°C
⇒ Subcooled liq.

As for liquids
change in v is
very negligible
?

∴ Δh is very negligible
∴ we have to do
from equations
{ As table would give
a difference
on approximation

$$\therefore h = u + pv \quad dh = du + p dv + v dp$$

$$= 0 + 0 + \gamma dp$$

$$\therefore \Delta h = v \Delta p$$

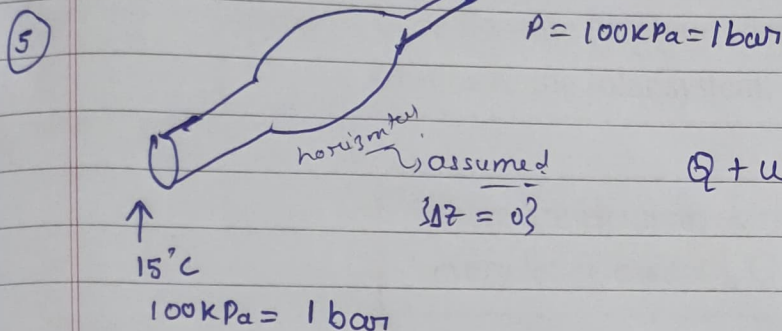
$$v = v_{\text{at } 20^\circ\text{C}} = 0.0010017 \text{ m}^3/\text{kg}$$

PTU

$$\begin{aligned}
 \Rightarrow \Delta h &= \int \Delta p \\
 &= -0.001017 \left(20 \text{ bar} - 1 \text{ bar} \right) \times 10^2 \frac{\text{kJ}}{\text{kg}} \\
 &= -1.90323 \text{ kJ/kg}
 \end{aligned}$$

$$\begin{aligned}
 w &= \frac{v_{in}^2}{2} - \Delta h = \left(\frac{15^2}{2} \right) \times 10^{-3} - \Delta h \\
 &= 0.1125 + 1.90323 \\
 &= 2.01573 \text{ kJ/kg}
 \end{aligned}$$

$$\begin{aligned}
 P &= \dot{m} \cdot w = 2 \times 2.01573 \\
 &\boxed{P = 4.03146 \text{ kW}} \\
 &\boxed{w = 2.01573 \text{ kJ/kg}}
 \end{aligned}$$



$$\begin{aligned}
 Q + w + \left(h_{in} + \frac{v_{in}^2}{2} + gz \right) m \\
 = m \left(h_{out} + \frac{v_{out}^2}{2} + gz \right)
 \end{aligned}$$

$$\begin{aligned}
 \Rightarrow Q + w &= m \left(h_{out} - h_{in} + \frac{v_{out}^2 - v_{in}^2}{2} \right) \\
 \text{As for compressed liquid} \\
 \Delta h &= \int \Delta p + \Delta u \\
 \therefore \Delta p &= 0 \quad \left\{ 1 \text{ bar} - 1 \text{ bar} \right\} \text{ \& } \Delta u_{\text{also given}} \\
 \therefore \text{here, } \Delta h &= 0 \\
 \therefore Q + w &= w = 0 + \left(\frac{v_{out}^2 - v_{in}^2}{2} \right) m \\
 \therefore w &= m \left(\Delta h + \frac{v_{out}^2 - v_{in}^2}{2} \right) \\
 \therefore w &= m \left(\Delta h + \frac{v_{out}^2 - v_{in}^2}{2} \right)
 \end{aligned}$$

$$\therefore Q + w = w = 0 + \left(\frac{v_{out}^2 - v_{in}^2}{2} \right) m$$

$$\boxed{w = \frac{m v_{out}^2}{2}}$$

$v_{in} \approx 0$ as finally short pipe

$$W = \frac{m v_{out}^2}{2}$$

$$\dot{m} = \rho A v = \frac{A v}{v} \quad \left\{ \begin{array}{l} \text{where} \\ v = \text{velocity} \\ v = \text{specific volume} \end{array} \right\}$$

$$P = \frac{dW}{dt} = \frac{v_{out}^2}{2} \cdot \frac{dm}{dt} = \frac{v_{out}^2}{2} \cdot \frac{A v_{out}}{v}$$

$$\Rightarrow \frac{v_{out}^3 \cdot A}{2v} = P$$

$$\Rightarrow v_{out} = \left(\frac{2Pv}{A} \right)^{1/3} = \left(\frac{2 \times \frac{0.0010008}{\pi \left(\frac{\pi}{4} (1 \times 10^{-2})^2 \right)} \times 10^3}{\pi \left(\frac{\pi}{4} (1 \times 10^{-2})^2 \right)} \right)^{1/3} = 29.498 \text{ m/s}$$

$$\left\{ v = 0.0010008 \text{ m}^3/\text{kg}, P = 1 \text{ kW} = 10^3 \text{ W} \right\}$$

$$\dot{m} = \frac{A v}{v} = \left(\frac{\pi d^2}{4} \right) \cdot \frac{29.498}{0.0010008} = 2.31495$$

$$\begin{array}{l} v = 29.498 \text{ m/s} \\ \dot{m} = 2.31495 \text{ kg/s} \end{array}$$