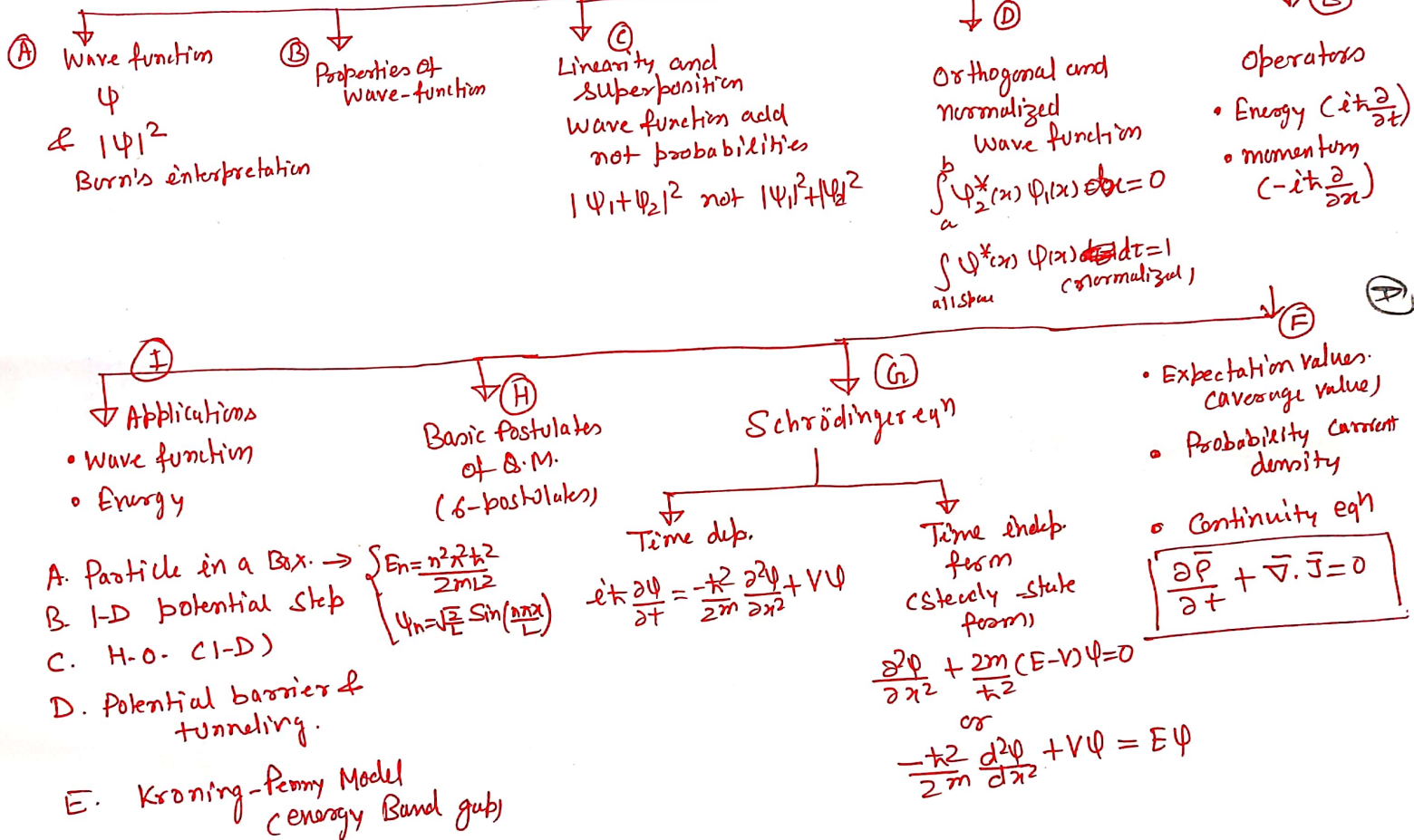


Unit-2



① Wave function: - (Statistical interpretation); - is the wavefunction. The quantity with which quantum mechanics is concerned. is the wavefunction.

② What exactly is the 'wave function' and what does it do for you once you've got it?

↓ Answer is given by Born's statistical interpretation.

Which says that $|\psi(x,t)|^2$ gives the probability of finding the particle at point x , at time t - or, more precisely,

$$\int_a^b |\psi(x,t)|^2 dx = \left\{ \begin{array}{l} \text{probability of finding the particle} \\ \text{between } a \text{ and } b, \text{ at time } t \end{array} \right\}$$

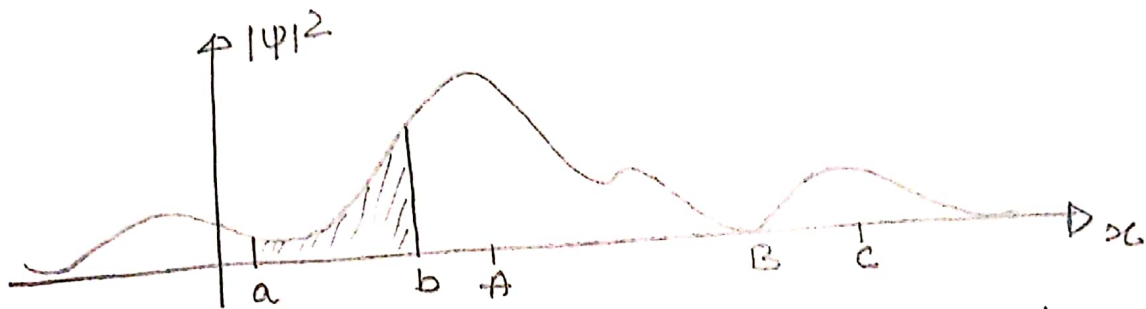


Fig: - A typical wave function. The shaded area represents the probability of finding the particle between a and b . The particle would be relatively likely to be found near A, and unlikely to be found near B.

↓
All quantum mechanics has to offer is statistical information about the possible results.

↓
The indeterminacy has been profoundly disturbing to physicists and philosophers alike; and it is natural to wonder whether it is fact of nature, or a defect in the theory.

Suppose I do measure the position of the particle, and I find it to be at point C. Where was the particle just before I made the measurement?

↓ Three possible answers

1. The realist position: the particle was at C.

2. The Orthodox position: the particle was not really anywhere.

Wave function:

①

The quantity with which quantum mechanics is concerned is the wave function ψ of a body.

$\psi \rightarrow$ itself has no physical interpretation,

$|\psi|^2 \rightarrow$ The square of its absolute magnitude evaluated at a particular place at a particular time is proportional to the probability of finding the body there at that time.



The linear momentum, angular momentum, and energy of the body are other quantities that can be established from ψ .

\Rightarrow Wave function are usually complex with both real and imaginary parts. A probability, however, must be positive real quantity.

$$\psi = A + iB$$

$$\psi^* = A - iB$$

$$|\psi|^2 = \psi^* \psi = A^2 - i^2 B^2 = A^2 + B^2$$

$|\psi|^2 \rightarrow \psi^* \psi \rightarrow$ Always a positive real quantity.

Well-Behaved Wave functions:-

1. ψ must be continuous and single-valued everywhere.

2. Derivative

$\frac{\partial \psi}{\partial x}, \frac{\partial \psi}{\partial y}, \frac{\partial \psi}{\partial z}$ must be continuous and single-valued everywhere.

3. ψ must be normalized, which means that ψ must go to 0 as $x \rightarrow \pm\infty, y \rightarrow \pm\infty, z \rightarrow \pm\infty$ in order that $\int |\psi|^2 dV$ over all space be a finite constant.

$\psi(x)$ must be finite.
 $\psi(x)$ must be single-valued.
 $\psi(x)$ must be continuous

$\frac{d\psi(x)}{dx}$ must be finite.

$\frac{d\psi(x)}{dx}$ must be single valued

$\frac{d\psi(x)}{dx}$ must be continuous

Linearity and Superposition

(2)

Wavefunctions add, not probabilities

If ψ_1 and ψ_2 are two solutions (i.e. wave function) that satisfy the Schrödinger eqn, then

$$\psi = a_1 \psi_1 + a_2 \psi_2$$

~~This~~ ~~Then~~ ~~is~~ is also a solution.

Thus the wave functions ψ_1 and ψ_2 obeys the superposition principle that other waves do.

Thus we conclude that interference effects can occur for wave functions just as they can for light, sound, water and electromagnetic waves

$$P_1 = |\psi_1|^2 = \psi_1^* \psi_1$$

$$P_2 = |\psi_2|^2 = \psi_2^* \psi_2$$

The probability density at screen is therefore,

$$P = |\psi|^2 = |\psi_1 + \psi_2|^2 = (\psi_1^* + \psi_2^*)(\psi_1 + \psi_2)$$

$$= \psi_1^* \psi_1 + \psi_2^* \psi_2 + \psi_1^* \psi_2 + \psi_2^* \psi_1$$

$$= P_1 + P_2 + \underbrace{\psi_1^* \psi_2 + \psi_2^* \psi_1}_{\downarrow}$$

S_2
 S_1

S_1 open
 \downarrow



$|\psi_1|^2$

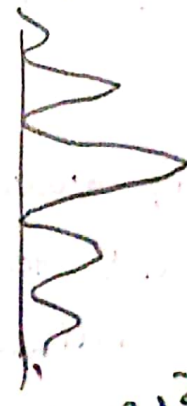
S_2 open
 \downarrow



$|\psi_2|^2$



$|\psi_1|^2 + |\psi_2|^2$



$|\psi_1 + \psi_2|^2$
✓

Orthogonal and Normalized wave function :-

If the product of a wavefunction $\psi_1(x)$ and complex conjugate $\psi_2^*(x)$ of a wave function $\psi_2(x)$ vanishes when integrated with respect to x over the interval $a \leq x \leq b$ i.e. if

$$\int_a^b \psi_2^*(x) \psi_1(x) dx = 0$$

Then $\psi_1(x)$ and $\psi_2(x)$ are said to be orthogonal in the interval (a, b) .

Probability

$$\psi \psi^* dv$$

The total probability of finding the particle in the entire space is, of course, unity i.e.

$$\int |\psi|^2 dv = 1$$

Where the integration is extends over all space.

We can
also
write

$$\int \psi \psi^* dv = 1$$

Any wave function satisfy the above eqⁿ is said to be normalized to unity or simply normalized.

Very often ψ is not normalized.

$$\psi \rightarrow A \psi \rightarrow \text{This is also soln of wave eqⁿ}$$

③

Now, the problem is to choose the proper value of A such that new wavefunction is normalized function.

In order that it is a normalized function, it must meet the requirement

$$\int (A\psi)^* A\psi \, dx \, dy \, dz = 1$$

$$\text{or } |A|^2 \int \psi \psi^* \, dx \, dy \, dz = 1$$

$$\text{or } |A|^2 = \frac{1}{\int \psi \psi^* \, dx \, dy \, dz}$$

$|A| \rightarrow$ normalizing const.

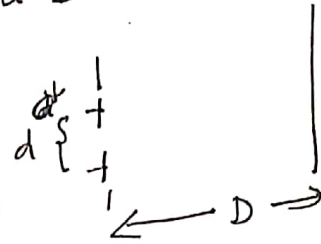
Operator \rightarrow from Notes

(F)

Operator \rightarrow Quantum Mechanics is a operator Mechanics.

Let us consider the two-slit expt. with the beam of mono-energetic electrons. We can determine the wavelength λ of the electrons by measuring the fringe width β of the interference pattern on the screen the distance d between two slit and distance D

A plane ~~wave~~ propagating classical wave of wave vector k and angular frequency propagating in $+x$ direction is represented by



But for the moment, let us consider a plane propagating (matter/electron) wave represented by a complex function

$$\psi(x,t) = A \cos(kx - \omega t)$$

$$\psi(x,t) = A e^{i(kx - \omega t)} \rightarrow (A)$$

Now let us see that the operator $-i\hbar \frac{\partial}{\partial x}$ operating on the $\psi(x,t)$ gives

$$\begin{aligned} -i\hbar \frac{\partial}{\partial x} \psi(x,t) &= -i\hbar \frac{\partial}{\partial x} [A e^{i(kx - \omega t)}] \\ &= (-i\hbar)(ik) A e^{i(kx - \omega t)} \\ &= \hbar k \psi(x,t) \end{aligned}$$

\downarrow
linear momentum (p)

\downarrow linear momentum of electron

Thus $-i\hbar \frac{\partial}{\partial x} \rightarrow$ linear momentum operator

$k \rightarrow$ wave vector

To be more specific, if ~~the~~ an electron wave represented by the wave function $\psi(x,t) = A e^{i(5x - \omega t)}$ The operator $-i\hbar \frac{\partial}{\partial x}$ operating on $\psi(x,t)$ shall give

$$-i\hbar \frac{\partial}{\partial x} \psi(x,t) = 5\hbar \psi(x,t)$$

\downarrow
momentum where $k=5$

On the other hand, when the mono-energetic electron beam represented by the wave function $\psi(x,t) = A e^{i(5x - \omega t)}$ falls on two-slits, it shall produce interference pattern with a fringe width such that λ comes out to be $\frac{2\pi}{5}$ from which we get $k=5$ and linear momentum $= 5\hbar$,

one hand doing ... other hand, doing operator algebra \Rightarrow Both gives same result.

$k = \frac{2\pi}{\lambda}$

→ Another operator

$$\cancel{i\hbar \frac{\partial}{\partial t}} \quad i\hbar \frac{\partial}{\partial t}$$

$$i\hbar \frac{\partial \psi(x,t)}{\partial t} = i\hbar \frac{\partial}{\partial t} [A e^{i(5x - \omega t)}]$$

$$= (\hbar \omega) \psi(x,t)$$

energy of the particle,

$$\text{for } \omega = \frac{h}{2\pi} \cdot 2\pi \nu = h\nu$$

~~Obs~~ operator for energy $\rightarrow i\hbar \frac{\partial}{\partial t}$

Physical quantity \rightarrow corresponding operator.

66 So in quantum mechanics, making a statement about the value of the physical quantity of the system is equivalent to an experimental measurement process on the system for that physical quantity.

linear momentum $\xrightarrow{\text{operator}} -i\hbar \nabla \quad (-i\hbar \frac{\partial}{\partial x}; -i\hbar \frac{\partial}{\partial y}; -i\hbar \frac{\partial}{\partial z})$

energy $\xrightarrow{i\hbar \frac{\partial}{\partial t}}$

kinetic energy $T = \frac{p^2}{2m} \rightarrow -\frac{\hbar^2}{2m} \nabla^2$

energy of the particle $\hat{E} \rightarrow i\hbar \frac{\partial}{\partial t}$
~~potential energy~~

Pot. energy, $V(r,t) \rightarrow V(r,t)$

Total energy, $\frac{p^2}{2m} + V(r,t) \rightarrow -\frac{\hbar^2}{2m} \nabla^2 + V(r,t)$

Angular momentum $\vec{L} = \vec{r} \times \vec{p} \rightarrow -i\hbar \vec{r} \times \vec{\nabla}$

$$T = \frac{p^2}{2m} = -\frac{\hbar^2}{2m} \nabla^2$$

Expectation Value:-

$$\Psi(x,t) = A e^{i(c k x - \omega t)}$$

$$P(x,t) dx = |\Psi(x,t)|^2 \rightarrow \text{Probability of finding the particle in the length } dx \text{ about point } x \text{ at time } t.$$

↓
Therefore the quantity

$$\left[-i\hbar \left(\frac{\partial}{\partial x} \right) \right] \left[|\Psi(x,t)|^2 \right] dx$$

↓
Information about the probable value of the linear momentum of the particle in the region x of $x dx$ at time t .

↓ Entire space

$$\int_{-\infty}^{\infty} \left(-i\hbar \frac{\partial}{\partial x} \right) |\Psi(x,t)|^2 dx$$

or

$$\int_{-\infty}^{\infty} \Psi^*(x,t) \left(-i\hbar \frac{\partial}{\partial x} \right) \Psi(x,t) dx$$

↓
gives us information about the probable value or average value or expectation value of the linear momentum of the particle in state $\Psi(x,t)$.

→ The x component of linear momentum

$$\langle p_x \rangle = \int_{-\infty}^{\infty} \Psi^*(x,t) \left(-i\hbar \frac{\partial}{\partial x} \right) \Psi(x,t) dx$$

$$A = \langle \hat{A} \rangle = \int_{-\infty}^{\infty} \psi^*(x,t) \hat{A} \psi(x,t) dx$$

Q: A particle is described by the normalized wave function

$$\psi(x) = A x e^{-bx} \quad x > 0$$

$$= 0 \quad x < 0$$

(a) Find the normalization constant A.

(b) Find $\langle x \rangle$ and $\langle x^2 \rangle$

Soln. (a) $\int_{-\infty}^{\infty} \psi^*(x) \psi(x) dx = A^2 \int_0^{\infty} x^2 e^{-2bx} dx$

$$= A^2 \frac{2}{(2b)^3} = 1$$

$$\left| \begin{aligned} \int_0^{\infty} x^n e^{-2bx} dx \\ = \frac{n!}{(2b)^{n+1}} \end{aligned} \right.$$

$$A = 2(b)^{3/2}$$

(b) $\langle x \rangle = \int_0^{\infty} 2b^{3/2} x e^{-bx} (x) 2b^{3/2} x e^{-bx} dx$

$$= 4b^3 \int_0^{\infty} x^3 e^{-2bx} dx$$

$$= 4b^3 \frac{3!}{(2b)^4} = \frac{3}{2b}$$

$$\langle x^2 \rangle = 4b^3 \int_0^{\infty} x^4 e^{-2bx} dx$$

$$= 4b^3 \frac{4!}{(2b)^5} = \frac{3}{b^2}$$

Ans