

1. **The Ising model.** The Ising model is named after Ernst Ising, Ph.D. in Physics (1924) from the University of Hamburg under the supervision of Wilhelm Lenz. Ising solved the one-dimensional (1D) Ising model exactly to find no phase transition. He also provided arguments on why there would not be a phase transition in higher dimensions either. In 1936, Peierls argued that both 2D and 3D Ising models admit phase transitions. Read the paper of Peierls [1] and explain the arguments.

The Ising Hamiltonian can be written as,

$$\mathcal{H} = -J \sum_{\langle ij \rangle} \sigma_i \sigma_j.$$

- (a) The spins σ_i can take values ± 1 on the i th lattice site.
- (b) $\langle ij \rangle$ implies nearest-neighbor interaction only,
- (c) $J > 0$ is the strength of exchange interaction.

The system undergoes a second order phase transition at the critical temperature T_c in two and higher dimension. For temperatures less than T_c , the system magnetizes in an ordered state. For temperatures greater than T_c , the system is in the disordered state. In this case, there are no long-range correlations between the spins. The order parameter

$$m = \frac{1}{N} \sum_i \sigma_i$$

for this system is the average magnetization. The order parameter distinguishes the two phases realized by the systems. It is zero in the disordered state, while non-zero in the ordered state.

- (a) The Ising Model has no explicit dynamics. Simulate (or download a code from the internet [2]) Ising model using the Metropolis algorithm, whose steps are:
 - i. Prepare an initial configuration of N spins
 - ii. Flip the spin of a randomly chosen lattice site.

- iii. Calculate the change in energy dU .
 - iv. If $dU < 0$, accept the move. Otherwise accept the move with probability $e^{-\beta dU}$. Here $\beta = 1/(k_B T)$
 - v. Repeat (ii)-(iv).
- (b) Check that the Metropolis algorithm satisfies the condition of detailed balance.
- (c) Plot the following quantities as a function of T from your simulation in one and two-dimensions.
- Order parameter (magnetisation): m
 - Energy $U = \langle E \rangle$
 - Specific heat $C = \frac{\partial U}{\partial T} = \frac{\langle E^2 \rangle - \langle E \rangle^2}{k_B T^2}$
 - Susceptibility: $\chi = \frac{\partial m}{\partial h} = \frac{\langle M^2 \rangle - \langle M \rangle^2}{k_B T}$
- (d) Solve the Ising model exactly in one-dimension using the transfer matrix method.
2. Solve problem 1 of chapter 1 of Kardar's book (Statistical Physics of Fields).
3. Solve problem 2 of chapter 1 of Kardar's book (Statistical Physics of Fields).
4. **Legendre transformation and thermodynamic potentials.**
- (a) Explain the concept of Legendre transformations.
 - (b) Explain the concept of thermodynamic potentials.
 - (c) What are extensive and intensive variables in thermodynamics?
 - (d) The change in internal energy U of a system is:

$$dU = TdS - PdV + \mu dN \quad (1)$$

Here μ is the chemical potential. The internal energy U can be considered as a function of S , V and μ . In some situations of experimental interest, it may be useful to have other independent variables. Find the independent variables for the following thermodynamics potentials:

- Helmholtz Free energy: $F = U - TS$
- Gibbs Free energy: $G = F + PV = U - TS + PV$

- Enthalpy: $H = U + PV$

(e) Show that: $G = \mu N$.

(f) The variations of intensive coordinates variables is constrained:

$$SdT - VdP + Nd\mu = 0 \quad (2)$$

The above is called the Gibbs-Duhem relation. Prove the relation and explain its significance.

5. **van der Waals equation of state.** The van der Waals equation is given as

$$\left(P + \frac{a}{v^2}\right)(v - b) = k_B T.$$

Here v is the volume per unit particle. The van der Waals equation of state reduces to the idea gas equation $Pv = k_B T$ when the constants a and b are zero.

- Explain the origin of the constants a and b ?
- Determine change in pressure with volume at constant temperature $(\partial P / \partial v)_T$
- Find the critical temperature T_c below which the mechanical stability is lost.
- Sketch P as a function of V for temperatures above and below T_c .

6. **The exact equivalence between the Lattice Gas Model and Ising Model.**

In the lattice gas model, each site maybe occupied by a fluid molecule. The fluid molecules interact via the potential $V(r)$, which is given as:

$$V(r) = \begin{cases} \infty & (r = 0) \\ -U & (r = a) \\ 0 & \text{otherwise} \end{cases}$$

Thus, two fluid particle on nearby site reduce energy by U . Otherwise there is no interaction if nearby site is empty. The Hamiltonian for the Lattice Gas Model is written as:

$$H = - \sum_{\langle ij \rangle}^N U c_i c_j - \mu \sum_{i=1}^N c_i \quad (3)$$

- (a) Show that the above model in Eq.(3) is exactly equivalent to the Ising model:

$$H = - \sum_{\langle ij \rangle}^N J \sigma_i \sigma_j - h \sum_i \sigma_i \quad (4)$$

Here $J > 0$ and $\sigma = \pm 1$.

- (b) The transformation can be achieved by the transformation: You may use the transformation $\sigma_i = f(c_i)$. Find the exact form of $f(c_i)$.
- (c) Show that the transformation results in the following:

$$J = \frac{U}{4}, \quad 2h = zU + \mu. \quad (5)$$

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- [1] Peierls, Rudolf. [On Ising's model of ferromagnetism](#). 1936.
- [2] See for example: <https://rajeshrinet.github.io/blog/2014/ising-model/>.