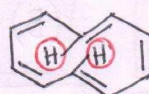
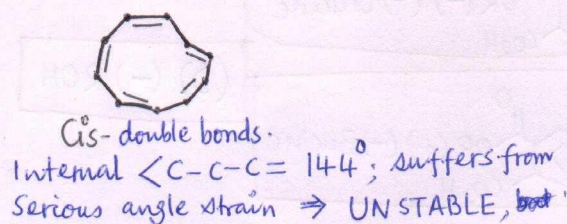


Model Answer
MID-SEMESTER EXAMINATION, WINTER: 2021-2022
Subject: Chemistry (Common) (CYI 101) for B. Tech 1st year

Part B (22)

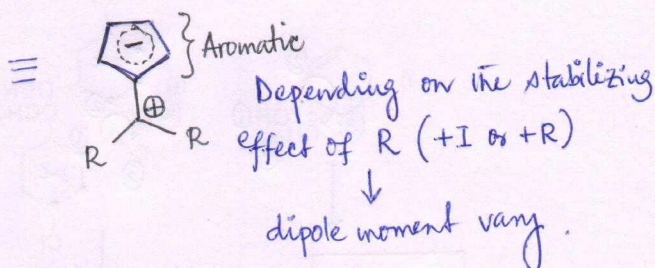
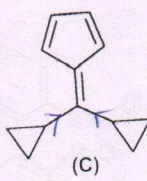
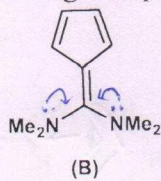
- 1 a) [10]-Annulene is non-aromatic, although bearing $[4n+2]$ π -electrons - Explain.

2×5=10



Suffers from strong
 steric crowding between
 the two peripheral H-atoms
 \downarrow
 NON-PLANNER str.

- b) Arrange the following compounds in increasing order of dipole moment.



Order: $B > C > A$

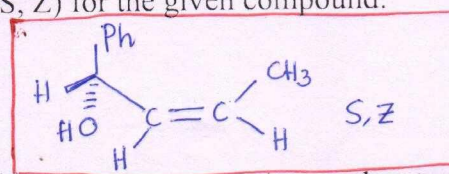
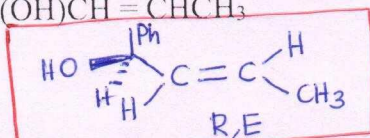
- c) What is the percentage composition of a mixture of two enantiomers of 2-butanol whose rotation is $+2.7^\circ$? The specific rotation of enantiomerically pure (+)-2-butanol is $+13.5^\circ$.

$$\% \text{ opt. ee} = \frac{\text{Sp. rotation of enantiomeric mixture}}{\text{Sp. rotation of pure enantiomer}} \times 100 = \frac{+2.7^\circ}{+13.5^\circ} \times 100 = 20\% \text{ w.r.t. (+) isomer.}$$

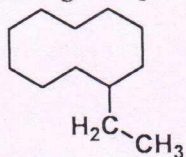
So, remaining 80% exists as racemic modification.

So, % composition of (+) isomer = 60%
 " " " (-) isomer = 40%

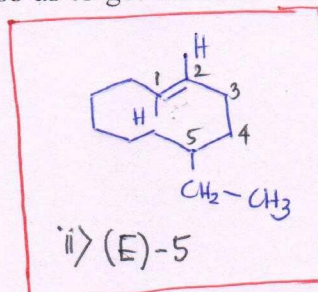
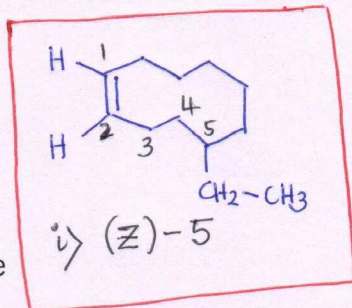
- d) Draw the possible configurations (R, E and S, Z) for the given compound:
 Ph-CH(OH)CH=CHCH₃



- e) Introduce a double bond in the following carbon skeleton so as to get the structure of the following compounds:

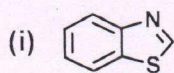


- i) (Z)-5-Ethylcycloclododecene
 ii) (E)-5-Ethylcycloclododecene

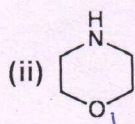


- 2 a) Mention the following compounds as aromatic, non-aromatic, and antiaromatic.

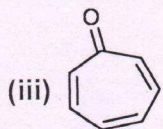
3×4=12



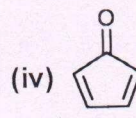
(i) Aromatic



(ii) Non-aromatic

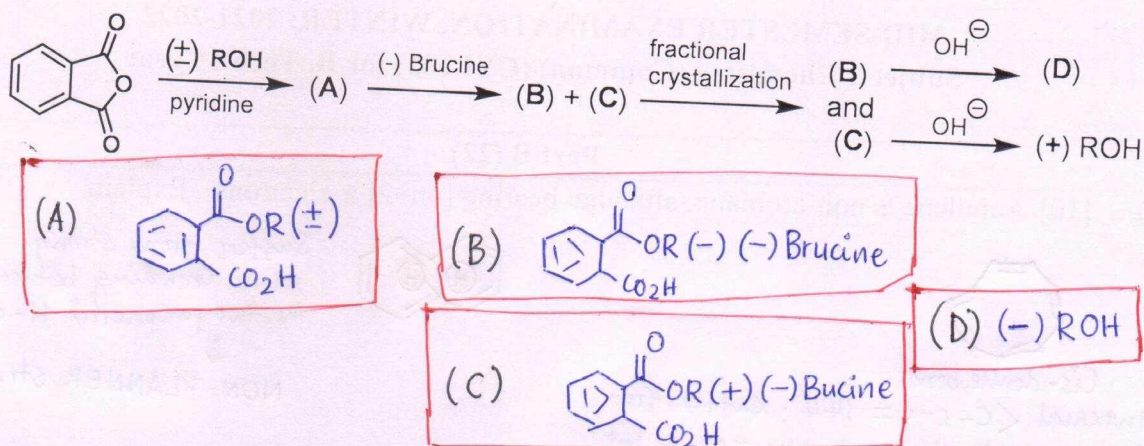


(iii) Aromatic



(iv) Anti-aromatic

b) Identify (A) to (D)



c) Identify the following compounds with R/S, R_a/S_a notations.

