Hydrogen atom with Klein-Gordon.

$$E^2 - p^2 = m^2$$
 $c = 1 \quad k = 1$

$$E \rightarrow i\partial_{\xi} \qquad \vec{p} = -i\partial_{\vec{x}} = -i\vec{\nabla}$$

$$(i\partial_t)^2 + (i\nabla)^2 \psi = m^2 \psi$$

$$(E^2-m^2)\psi = -\Delta\psi - \frac{2Ze^2\psi}{rE\psi} - \frac{Z^2e^4\psi}{r^2\psi}$$

Consider months: Check:
$$E_b \ll m = \frac{E^2 - m^2}{2m} \simeq E_b \uparrow$$
, $\frac{Z_b^2 \chi^2}{2mr^2} \sim \frac{e^2 \perp}{2r} \frac{136eV}{511 \text{ NeV}} = \frac{136eV}{2r} = \frac{1$

$$-\frac{1}{2m}\left(\frac{\partial^2}{\partial r^2} + \frac{2}{r}\partial_r - \frac{\ell(\ell+1)}{r^2}\right)\psi - \frac{Ze^2E}{mr}\psi - \frac{Ze^h}{2mr^2}\psi = \frac{E^2-m^2}{2m}\psi$$

$$-\frac{1}{2m}\left(\frac{3^24}{3r^2}+\frac{2}{r}\partial_r\psi\right)+\frac{\ell(\ell+1)-2\tilde{e}^4}{2mr^2}\psi-\frac{Z\tilde{e}^2E}{mr}\psi=\frac{E^2-m^2}{2m}\psi$$

$$\hat{Z}(\hat{l}+1) = l(l+1) - Ze^4$$

$$\hat{Z}\hat{e}^2 = \frac{Ze^2E}{m}$$

$$\hat{E} = \frac{E^2 - m^2}{2m}$$

$$\Rightarrow -\frac{1}{2m} \left(\frac{3^2 \psi}{3r^2} + \frac{2}{r} \partial_r \psi \right) + \frac{\hat{\varrho}(\hat{\varrho}+1)}{2mr^2} \psi - \frac{\hat{\varrho}(\hat{\varrho}+1)}{r} \psi - \frac{\hat{\varrho}(\hat{\varrho}$$

$$\frac{E^2 - m^2}{Z^{4}m} = -\frac{Z^{2}e^4 E^2}{M^2} \frac{m}{Z(n_r + \hat{l} + 1)^2}$$

$$E^{2} = \frac{m^{2}}{1 + \frac{z^{2}e^{4}}{(n_{r}+\ell+1)^{2}}}$$

$$\hat{\ell} = \frac{1}{2} + \hat{\ell} - \ell(\ell+1) + \frac{2}{2}e^{4} = 0$$

$$\hat{\ell} = \frac{-1 \pm \sqrt{1 + 4\ell^{2} + 4 - 42\ell^{2}}e^{4}}{2}$$

$$l = -\frac{1}{2} \pm \sqrt{(l + \frac{1}{2})^2 - 2^2 e^4}$$

$$\hat{l} = -\frac{1}{2} + \sqrt{(l+\frac{1}{2})^2 - 2^2 e^4}$$

Restore units:

$$l = -\frac{1}{2} + \sqrt{(l+\frac{1}{2})^2 - 2^2 \alpha^2}$$

$$E = \frac{mc^2}{\sqrt{1 + \frac{z^2 \alpha^2}{(N_r + \hat{l} + 1)^2}}}$$

if
$$z^2 z^2 > \frac{137}{4}$$
 $z > \frac{137}{2}$ we

(low every consequence "(low energy consequence)"

Snall Z:

$$\hat{l} \simeq -\frac{1}{2} + (l + \frac{1}{2}) \left(1 - \frac{z^2 \alpha^2}{2(l + \frac{1}{2})^2} \right) = l - \frac{z^2 \alpha^2}{(2l + 1)} = l - \delta_{\ell}$$

$$\delta_{\ell} = \frac{z^2 \alpha^2}{(2l + 1)}$$

$$E = mc^{2} \left(1 - \frac{1}{2} \frac{z^{2} z^{2}}{(n_{e} - \delta_{e})^{2}} + \frac{3}{8} \frac{z^{4} z^{4}}{(n - \delta_{e})^{4}} \right)$$

$$= mc^{2}\left(1-\frac{1}{2}\frac{z^{2}a^{2}}{n^{2}}\left(1+\frac{z\delta_{1}}{n}\right)+\frac{3}{8}\frac{z^{4}a^{4}}{n^{4}}\right)$$

$$E_{b} = -\frac{mc^{2}}{2} \frac{z^{2} x^{2}}{n^{2}} - \frac{z^{2} x^{2}}{8} \frac{z^{2} x^{2}}{n^{3}} + \frac{3}{8} \frac{mc^{2} z^{4} y^{4}}{n^{4}}$$

$$E_{b} = -\frac{m^{2}e^{4}}{2n^{2}t_{1}^{2}} + mc^{2} 2^{4} \sqrt{\frac{3}{8n^{4}} - \frac{1}{n^{3}(2l+1)}}$$

I does not agre with experiment should be 2j+1