# Charge transport properties in organic electronics

#### Ram Dhari Pandey

Faculty of physics, Astronomy and Informatics Nicolaus Copernicus University Torun Poland

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### Introduction

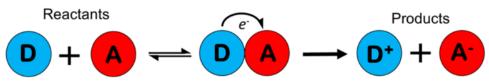
- movement of charge carriers
- Carbon based material
- Eco-friendly and low cost
- flexibility

## Main objectives

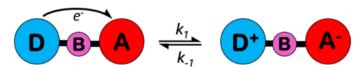
- The design, implementation, and testing of charge transfer models
- The design and implementation of a graphical user interface
- Quantum entanglement and correlation analysis

#### Electron transfer

#### Intermolecular Electron Transfer

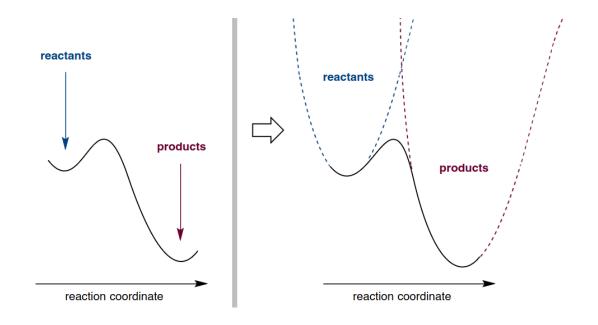


#### Intramolecular Electron Transfer

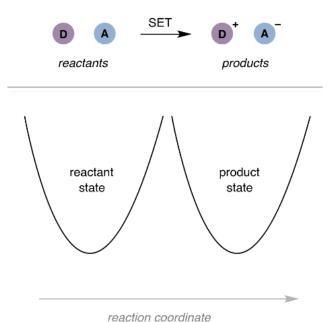


(Top):Formation of encounter complex (Bottom):Rigid molecular bridge

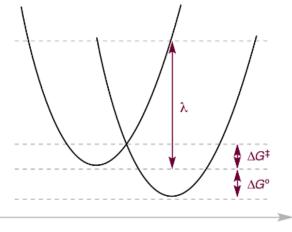
# Marcus theory



# Marcus theory

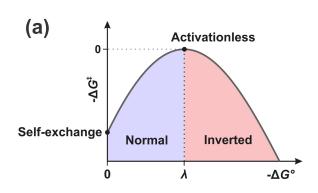


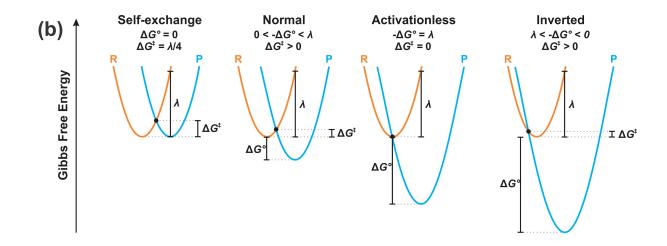
 $\Delta G^{\ddagger} = \frac{\left(\lambda + \Delta G^{\circ}\right)^{2}}{4\lambda}$ 



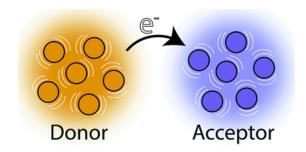
reaction coordinate

# Marcus theory





# Marcus-theory



The electron transfer rate...

$$k_{ET} = \frac{2\pi}{\hbar^2} t^2 \sqrt{\frac{1}{4\lambda \pi k_B T}} \exp\left[\frac{-(\Delta G^{\circ} + \lambda)^2}{4\lambda k_B T}\right]$$

controlled by  $\Delta G^{\circ}$ ,  $\lambda$ , and t

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### Reorganization energy

The reorganization energy  $\lambda$  is defined as

$$\lambda = (E_{neutralgeometry}^{charged} - E_{neutralgeometry}^{neutral}) + (E_{chargedgeometry}^{neutral} - E_{chargedgeometry}^{charged})$$

$$\lambda = \lambda_1 + \lambda_2$$

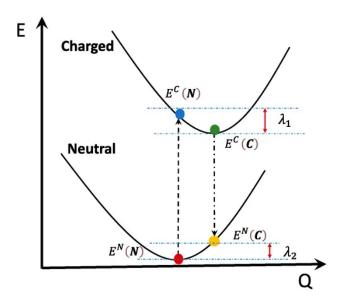


Figure 1: Adiabatic potential energy curves

## Charge transfer integral

$$V = \frac{J - S(e1 + e2)/2}{1 - S^2} - --- > (a)$$

$$J = \left\langle \phi_1 \middle| \hat{H} \middle| \phi_2 \right\rangle - --- > (b)$$

$$S = \left\langle \phi_1 \middle| \phi_2 \right\rangle - --- > (c)$$

$$e_1 = \left\langle \phi_1 \middle| \hat{H} \middle| \phi_1 \right\rangle - --- > (d)$$

$$e_2 = \left\langle \phi_2 \middle| \hat{H} \middle| \phi_2 \right\rangle - --- > (e)$$

### ESID Method

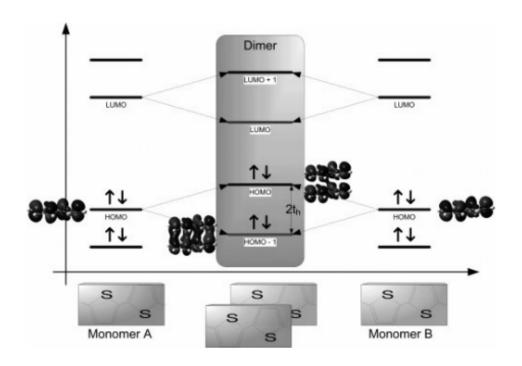


Figure 2: Splitting of the HOMO and HOMO-1 in the dimer

#### ESID method

Transfer integral for hole

$$t_{\text{hole}} = \frac{1}{2} \sqrt{(E_H - E_{H-1})^2 - (\varepsilon_1 - \varepsilon_2)^2}$$

$$t_{\text{hole}} = \frac{E_H - E_{H-1}}{2} \tag{1}$$

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## Tight-binding Hamiltonian

Relationship between charge-transport properties and geometric structure

$$H = \sum_{m} \epsilon_m a_m^+ a_m + \sum_{m \neq n} t_{mn} a_m^+ a_n$$

# Research Methodology

#### pCCD based methods

- pCCD-LCCD/LCCSD
- fpCCSD/fpCCD
- IP-EOM-pCCD
- DIP-EOM-pCCD

PyBEST's support of molecular fragments...

Methods...

- Exact approach
- ESID method
- Approximate model

Testing...

# Graphical user interface (GUI)

- A user-friendly graphical user interface...
- charge transport properties
- electron cross platform
- JavaScript Object Notation format
- Seprate Git-Lab repository.

### Critical path and risk assessment

• Convergence-related difficulties in pCCD based calculations.

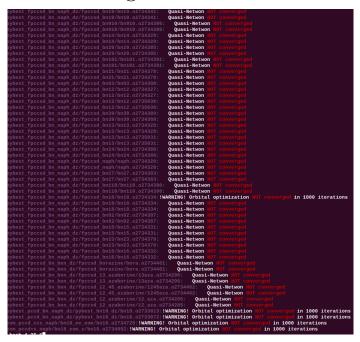


Fig-1 Quasi-Newton not converged

- Might have numerical difficulties (for instance, solving the amplitude equations for pCCD-LCCSD)
- Unreliable electronic structures.

DMRG code...

# Application

$$+ \left\langle \begin{array}{c} \stackrel{R}{\longrightarrow} - \stackrel{H}{\longrightarrow} - \stackrel{N}{\longrightarrow} - - \stackrel{N}{\longrightarrow} - - \stackrel{N}{\longrightarrow} - - \stackrel{N}{\longrightarrow} - - \stackrel{N}{\longrightarrow$$

Figure 3: Polyaniline(Top) and its derivatives(Bottom)

## Application

#### Leucoemeraldine base (fully reduced; colorless)

#### Emeraldine base (half oxidized; violet-blue)

#### Pernigraniline base (fully oxidized; purple)

Figure 4: Polyaniline in its idealized oxidation states

### Conclusion

#### Work package-A

• Charge transport module in PyBEST

#### Work package-B

• GUI for modeling charge transport properties

#### Work package-C

• Calculation of electronic structures and charge transport properties

# Acknowledgement

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