

Find an expression for $(\frac{\partial s}{\partial v})_T$ for a gas whose equation of state is $(P - \frac{a}{v^2})(v - b) = RT$

According to the Maxwell relations is $(\frac{\partial s}{\partial v})_T = (\frac{\partial P}{\partial T})_v$ and with the equation of state $(P - a/v^2)(v - b) = RT \rightarrow P = \frac{RT}{v-b} + \frac{a}{v^2}$ it follows that $(\frac{\partial s}{\partial v})_T = (\frac{\partial P}{\partial T})_v = \frac{R}{(v-b)}$