

Alternative und elektrifizierte Fahrzeugantriebe (AEF)

Alternative Vehicle Propulsion Systems (AVPS)

ANSPRECHPARTNER / ORGANIZATIONAL TEAM

Fragen, Lob und Kritik richten Sie bitte an
Questions or feedback? Contact us at
FachgruppeAEF@vka.rwth-aachen.de



Note: DE → Deutsch (German), EN → English

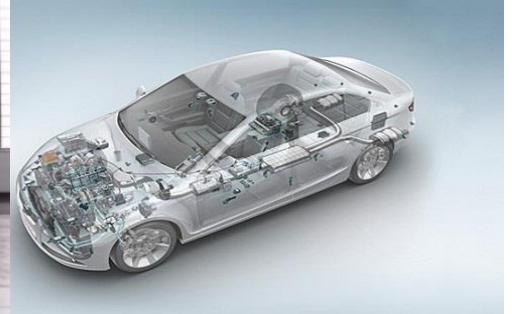
VKA – Excursion 2020



MAHLE



BOSCH



Contact:

Theodoros Kossioris, M.Sc. (Kossioris@vka.rwth-aachen.de)

Mehdi Hosseininasab, M.Sc. (Hosseininasab@vka.rwth-aachen.de)

Participants:

18 Students

Costs:

To be announced incl. Bus transfer, Accommodations and Fees

Registration:

Via email to: **exkursion_vka@vka.rwth-aachen.de**

Required Information: **Name / Matriculation Nr. / Nationality / Field of Study / Phone Nr.**

VKA – Excursion 2020

Schedule

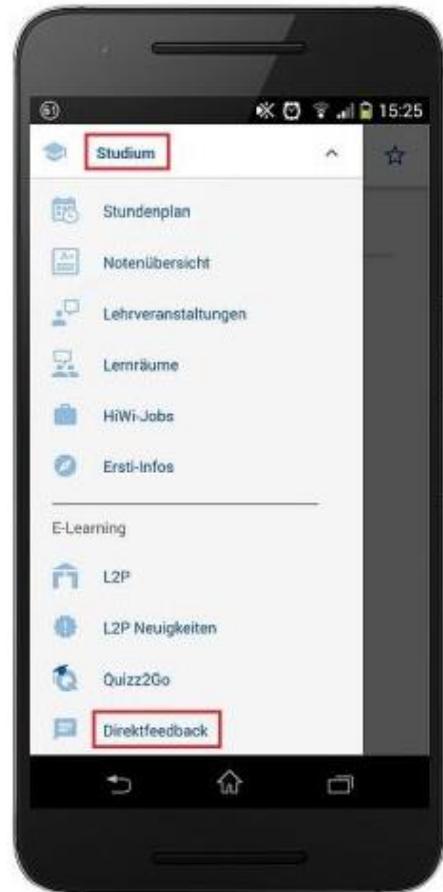
Arrival#	Departure from	Departure time	Destination	Arrival time
1	VKA Forckenbeckstraße 4 52074 Aachen	01.06.2020 approx. 15:00	Jugendherberge Stuttgart Haußmannstr. 27 70188 Stuttgart	01.06.2020 approx. 19:30
2	Jugendherberge Stuttgart Haußmannstr. 27 70188 Stuttgart	02.06.2020 approx. 9:00	Mercedes-AMG GmbH Daimlerstraße 1 71563 Affalterbach	02.06.2020 approx. 10:30
3	Mercedes-AMG GmbH Daimlerstraße 1 71563 Affalterbach	02.06.2020 approx. 15:00	Jugendherberge Stuttgart Haußmannstr. 27 70188 Stuttgart	02.06.2020 approx. 17:00
4	Jugendherberge Stuttgart Haußmannstr. 27 70188 Stuttgart	03.06.2020 approx. 9:00	Robert Bosch GmbH 70442 Stuttgart	03.06.2020 approx. 10:30
5	Robert Bosch GmbH 70442 Stuttgart	03.06.2020 approx. 15:00	Jugendherberge Stuttgart Haußmannstr. 27 70188 Stuttgart	03.06.2020 approx. 17:00
6	Jugendherberge Stuttgart Haußmannstr. 27 70188 Stuttgart	04.06.2020 approx. 7:00	Forschungs- und Innovationszentrum (FIZ) Knorrstraße 147 80937 München	04.06.2020 approx. 10:00
7	Forschungs- und Innovationszentrum (FIZ) Knorrstraße 147 80937 München	04.06.2020 approx. 12:30	BMW Welt, Am Olympiapark 1, 80809 München	04.06.2020 approx. 13:00
8	BMW Welt, Am Olympiapark 1, 80809 München	04.06.2020 approx. 16:30	Jugendherberge Stuttgart Haußmannstr. 27 70188 Stuttgart	04.06.2020 approx. 19:00

VKA – Excursion 2020

Schedule

Arrival#	Departure from	Departure time	Destination	Arrival time
9	Jugendherberge Stuttgart Haußmannstr. 27 70188 Stuttgart	05.06.2020 approx. 9:00	MAHLE International GmbH Pragstraße 26-46, 70376 Stuttgart	05.06.2020 approx. 10:30
10	MAHLE International GmbH Pragstraße 26-46, 70376 Stuttgart	05.06.2020 approx. 15:00	VKA Forckenbeckstraße 4 52074 Aachen	05.06.2020 approx. 19:30

Do you have any questions? Use the "Direct Feedback" in the RWTH app to ask questions anonymously in real time!



Login

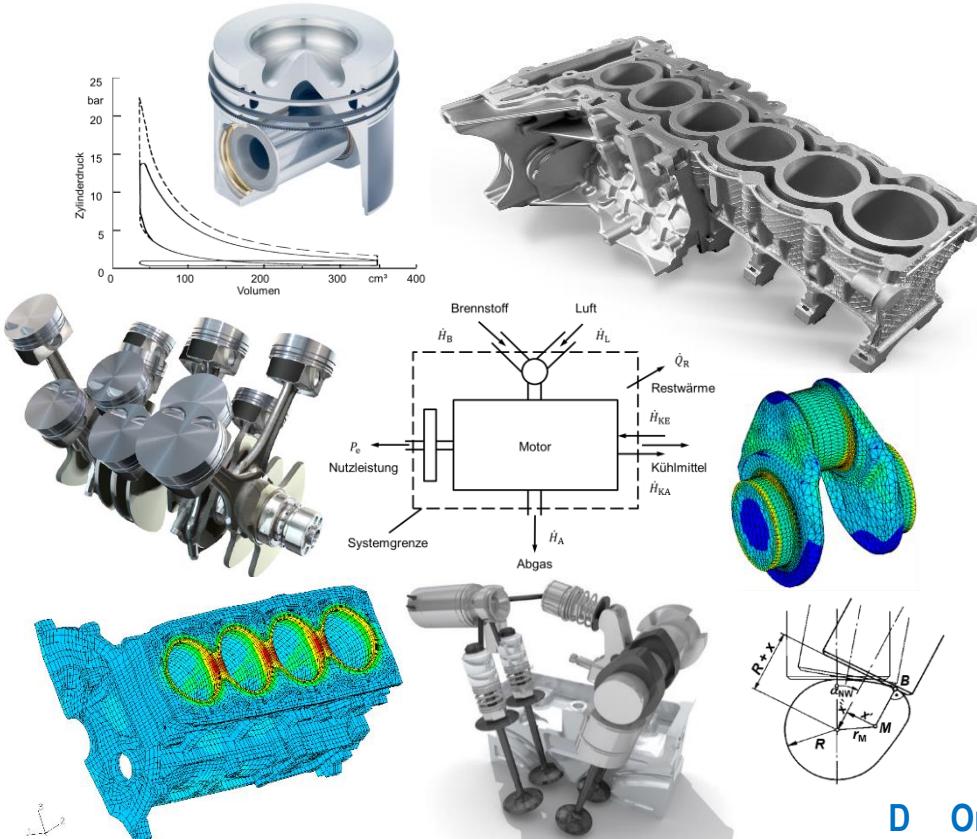
To be able to use all functions, this [link to an organizational unit in CAMPUS](#) is necessary. You can then log in to your browser at <https://app.rwth-aachen.de/das> with your TIM ID and the corresponding password.

Access for students is via the RWTHApp. Direct feedback channels can be found under „**Studium**“ > „**Direktfeedback**“ .

Passwort: aef1app

Lecture

Internal Combustion Engines: Design and Mechanics



Univ.-Prof. Dr.-Ing. S. Pischinger

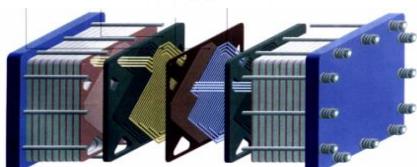
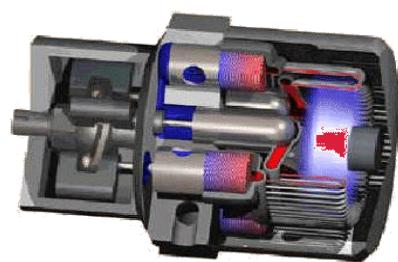
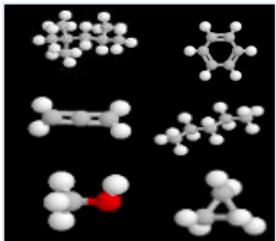
Heat Release Analysis
Valve Train
Energy Balance
Heat Transfer and Similarity
Mass Forces and Engine Design

D Ort: H04 (1385|103)

E Loc.: Room TD (1300|408)

Zeit: Wed 12:30 – 14:00 Uhr

Time: Tue 08:30 – 10:00 a.m.



Lecture **Alternative Vehicle Propulsion Systems**

Univ.-Prof. Dr.-Ing. Stefan Pischinger
Univ.-Prof. Dr.-Ing. Lutz Eckstein

Energy Carriers and Properties

Fuel Characteristics
Production of Alternative Fuels
Alternative Fuels and Application

Energy Conversion Processes and Implementation

Thermodynamic Energy Conversion
Electrochemical Energy Conversion: Fuel Cells
Electrochemical Energy Conversion: Battery
Thermoelectric Energy Conversion

D Ort: C.A.R.L. H04 (1385|103) **Zeit:** Mi 14:30 – 16:00 Uhr

E Loc.: EA (1270|109) **Time:** Tue 10:30 – 12:00 p.m.

Lecture

Electronics for Combustion Engines

Prof. Dr.-Ing. Jakob Andert



Electronic Influence of Combustion Engines

Key Sensors

Key Actuators

ECU Hardware

Bus Systems

Electric Machine Fundamentals

12 V / 48 V Powernet

Hybrid Technology

Loc.: H 212 (1400|212)

Time: Mo 11:30–13:00 Uhr

Lecture

Series Development of Transmissions for Passenger Cars and Light-Duty Vehicles

Dr.-Ing. Ingo Steinberg

Dr.-Ing. Gereon Hellenbroich

Transmission Types

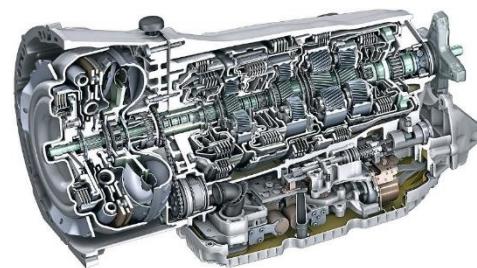
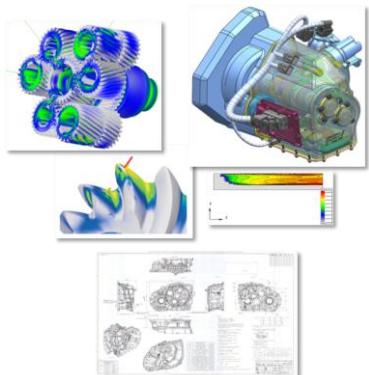
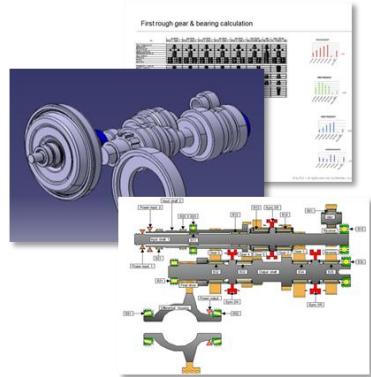
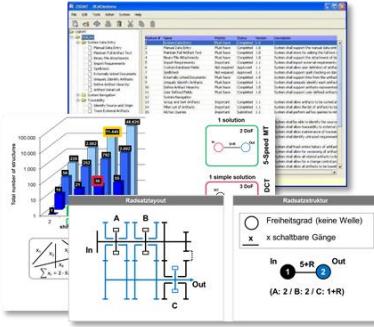
Design

Components

Development Process

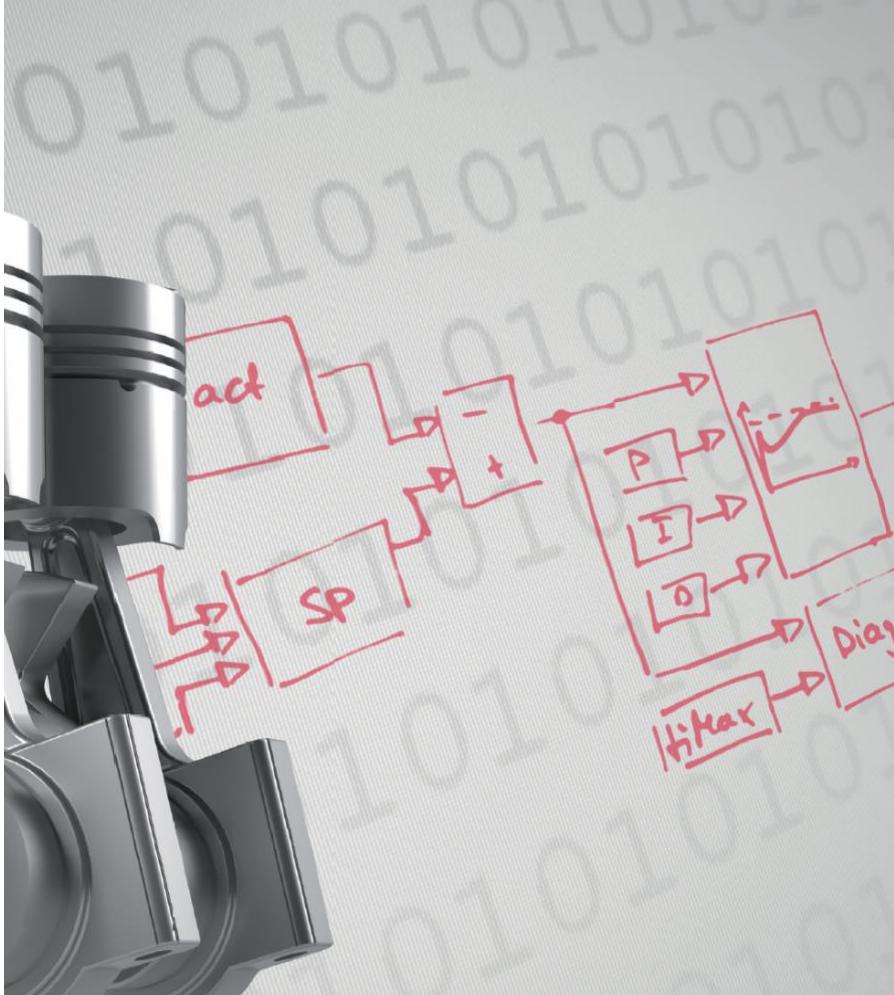
Testing

Series Production



Loc.: C.A.R.L. S11 (1385|211)

Time: Thu 12:30 – 14:00

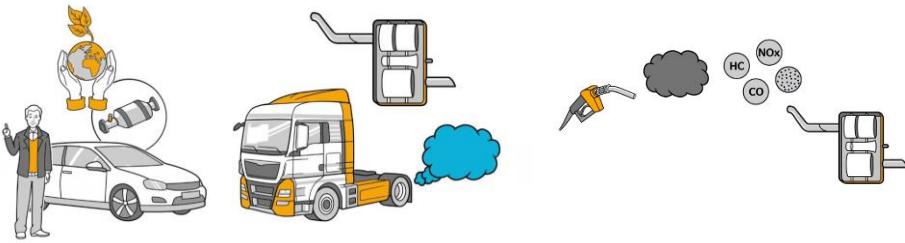


Lecture
Software for Combustion Engines
Dr.-Ing. Johannes Richenhagen

Software-Development-Process
Architecture Development
Requirements Development
Modeling of Software, Code Generation, Integration
Toolchain Development
Software-Testing
Vehicle / Powertrain Application
Agile Development Methods

Loc.: TD (1300|408)

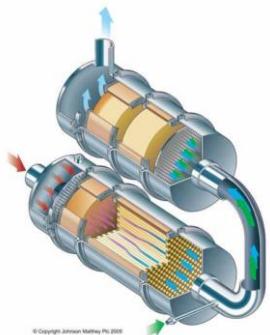
Time: Di 10:30–13:00 Uhr



Lecture

Catalytic Exhaust Gas Aftertreatment of Internal Combustion Engines

Dr. Lynzi Marie Robb



Fundamentals of Exhaust Gas Aftertreatment

Structure of Motor Vehicle Exhaust Gas Aftertreatment Components

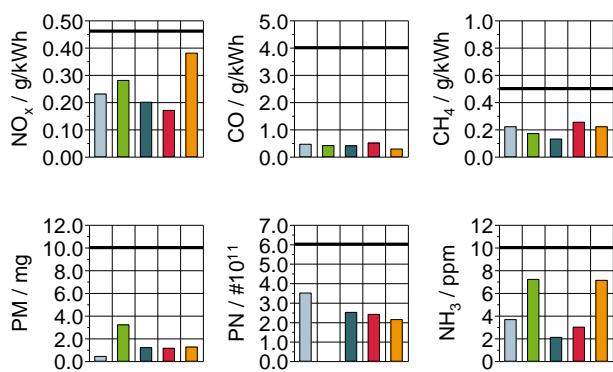
Oxidation and 3-Way Catalyst

NOx-Storage Catalyst

SCR-Catalyst

Particulate Filter

Complex Exhaust Gas Aftertreatment Systems



Loc.: SE 002 (1580|002)

Time: Mo 10:30–12:00 Uhr

Lecture

International Patent, Trademark and Registered Design Rights

Dipl.-Ing. Matthias Rößler, LLM

Introduction to International Innovation Protection

International Contracts concerning Intellectual Property

System of the Patent Cooperation Treaty (PCT)

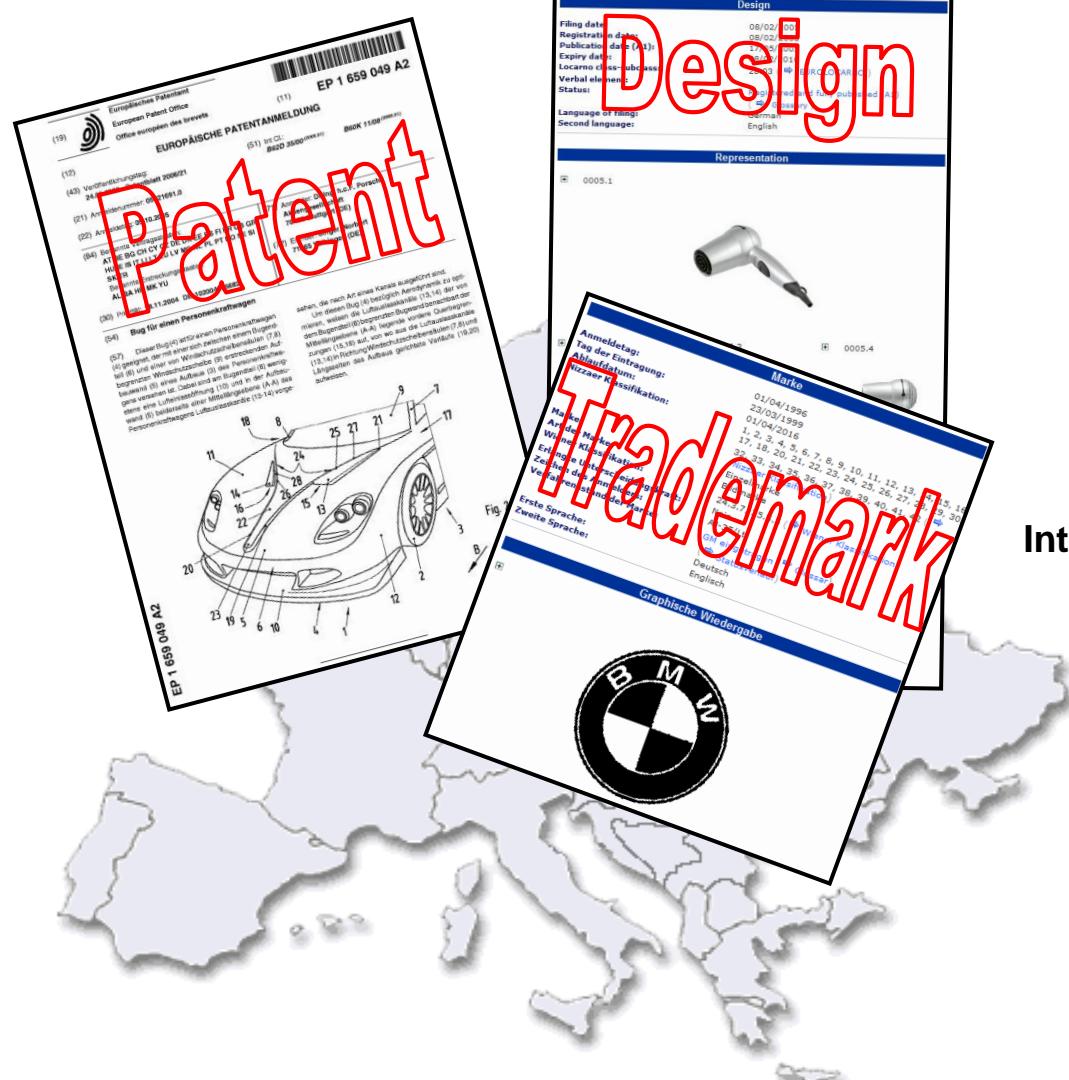
European Patent Convention

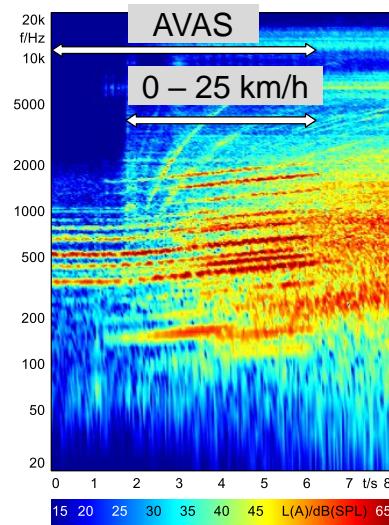
European Trademark Regulations

Community Design Regulations

Loc.: PPS H2 (2315|001)

Time: Wed. 14:30–16:00





Lecture

Acoustics of Mobile Propulsion Systems

Dr.-Ing. Christoph Steffens

Acoustic Fundamentals

Acoustic Measurement Equipment

Acoustic Behavior of Combustion and Electric Engines

Acoustic Assessment of Propulsion Concepts

Sound Quality

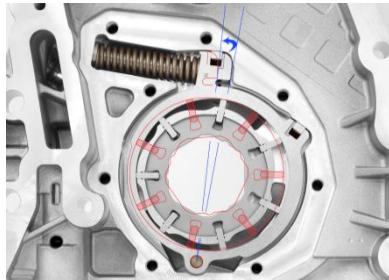
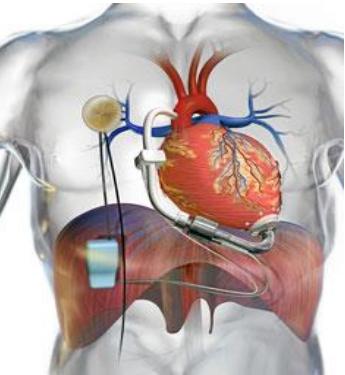
Simulation

Legal Requirements

Sound Design

Loc.: C.A.R.L. S11 (1385|211)

Time: Tue 08:30 – 10:00



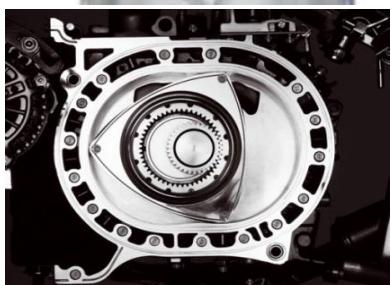
Lecture **Pumps and Compressors**

Dr. Klaus Hoff

Design and Construction Characteristics

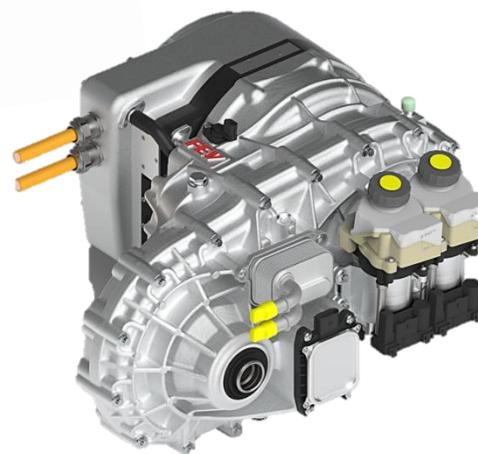
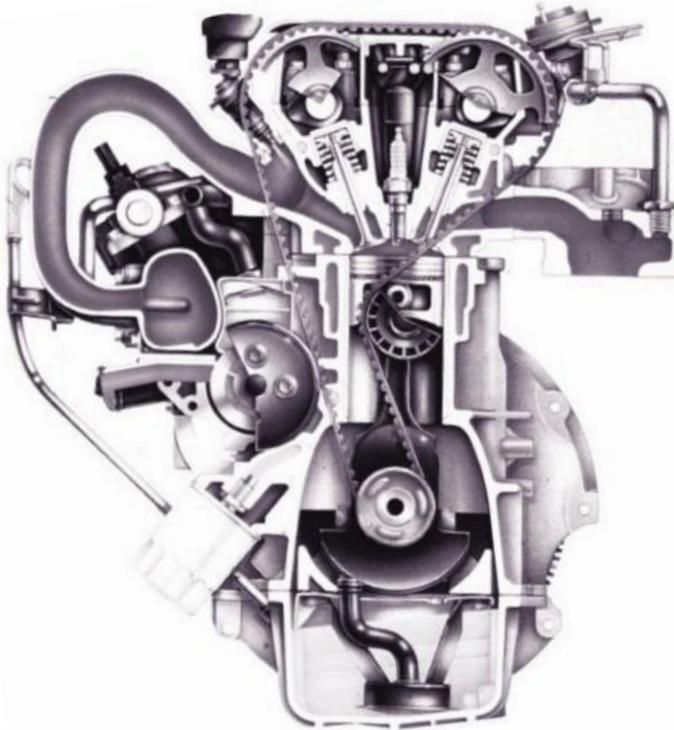
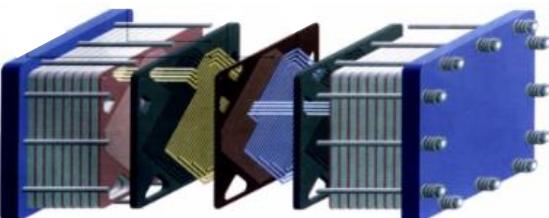
Classification according to Design Features and Calculation

Regulations



Loc.: S12 (1385|212)

Time: Do 10:30–12:00 Uhr



Lecture

Mobile Propulsion Fundamentals

Univ.-Prof. Dr.-Ing. Stefan Pischinger

Combustion Engines

Fuel Cells

Electrical Machines

Classification

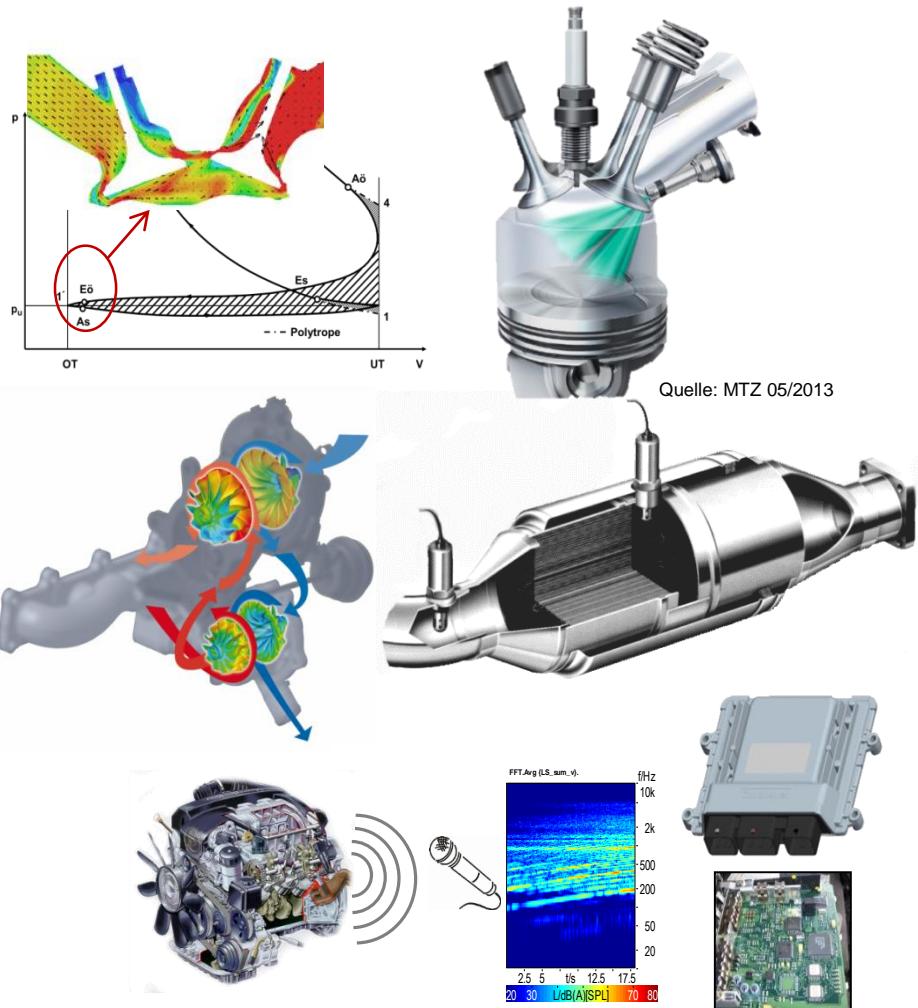
Characteristic Values

Operational Behavior

Lecture

Internal Combustion Engines II

Univ.-Prof. Dr.-Ing. Stefan Pischinger



Charge Cycle

Turbocharging

Exhaust Gases and Exhaust Gas Aftertreatment

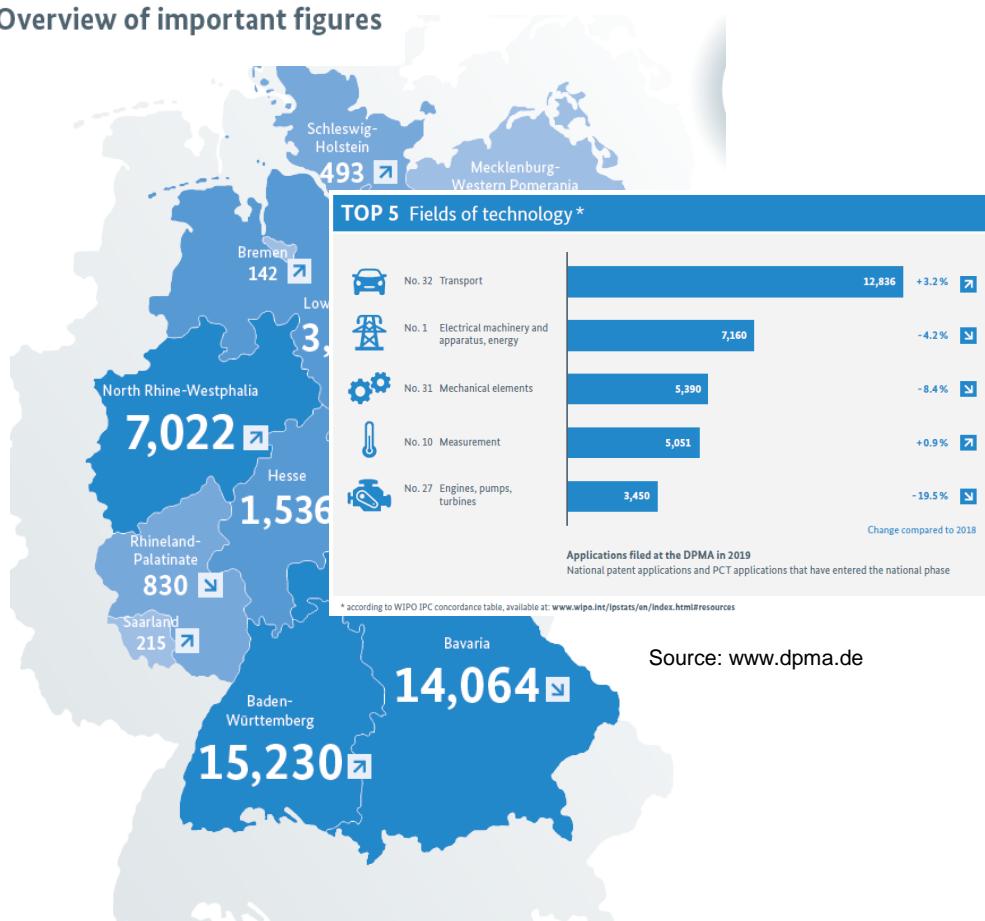
Engine Management

Mixture Formation

Engine Acoustics

Patents

Overview of important figures



Lecture

Fundamentals of Patent and Utility Model Law

Dipl.-Ing. Matthias Rößler

Basics of German Patent and Utility Model Law

Granting, effect and enforcement of technical Intellectual Property rights in Germany

Licensing contract law

German Employee Invention Act

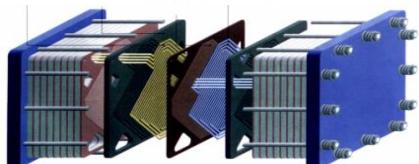
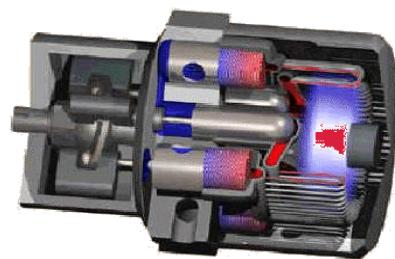
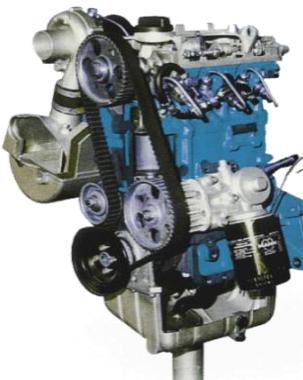
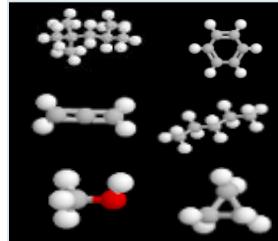
Winter Term

vka | cmp | RWTH AACHEN UNIVERSITY

Lecture

Alternative Vehicle Propulsion Systems

Univ.-Prof. Dr.-Ing. S. Pischinger



Energy carriers and properties

Properties of fuels

Production of alternative fuels

Alternative fuels and applications

Energy conversion cycles and realization

Thermodynamical energy conversion

Electrochemical energy conversion: Fuel cell

Electrochemical energy conversion : Battery

Thermoelectrical energy conversion

Content of lecture

1. Introduction
 - 1.1 Definition
 - 1.2 Motivation
 - 1.3 Classification of alternative propulsion systems
2. Energy sources and properties
 - 2.1 Properties of fuels
 - 2.2 Production of alternative fuels
 - 2.3 Alternative fuels and applications
3. Energy conversion cycles and their realization
 - 3.1 Thermodynamical energy conversion
 - 3.2 Electrochemical energy conversion: Fuel cell
 - 3.3 Electrochemical energy conversion: Battery
 - 3.4 Thermoelectrical energy conversion

Content of lecture II (ika)

4. Vehicle energy demand
5. Structure of alternative drive systems
6. Energy carriers
7. Storage systems
8. Energy converter
9. Torque converter (transmissions)
10. Examples and energy balances

Introduction – energy conversion chain

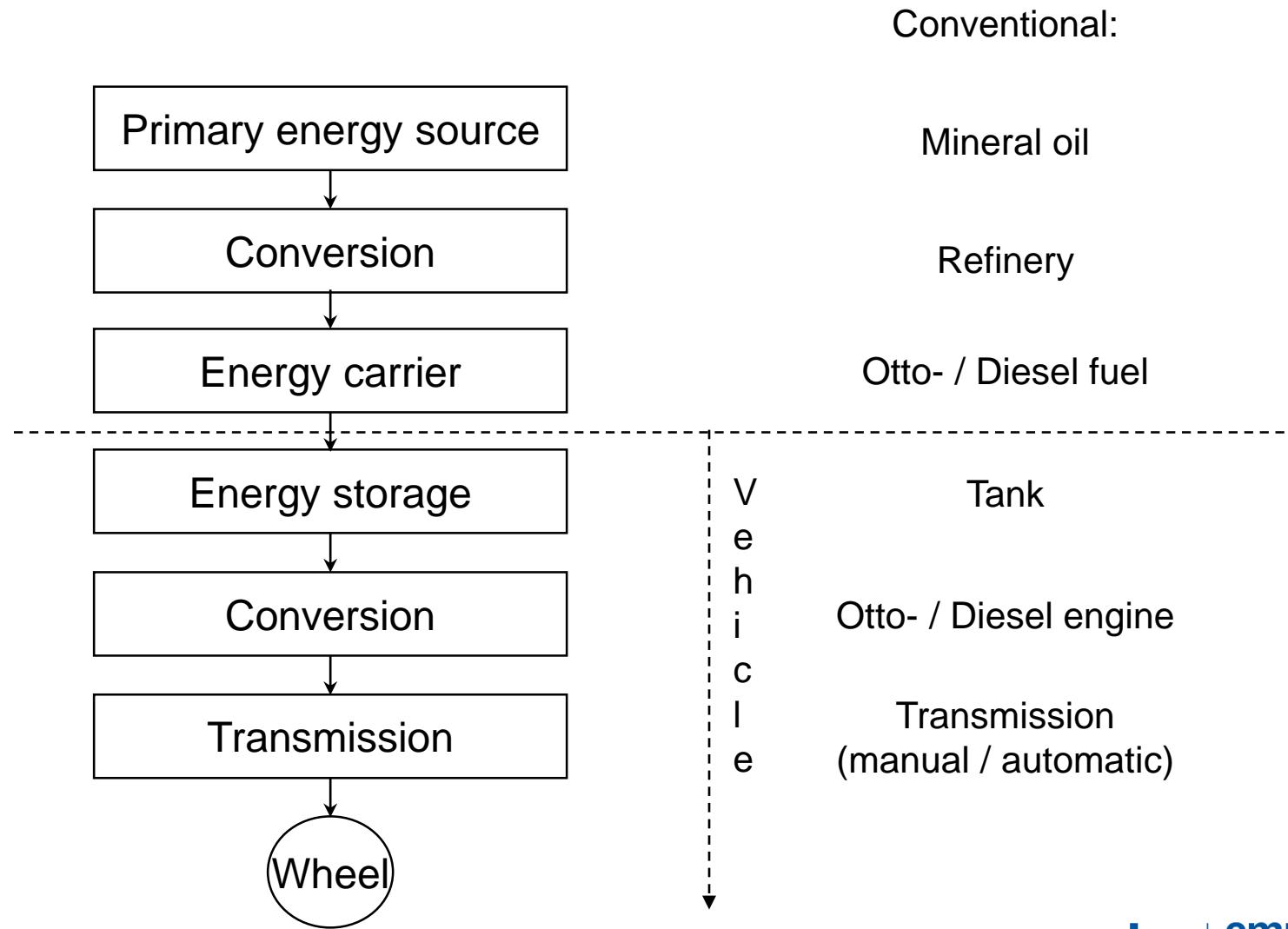
Definition

- 98 % of today's vehicle propulsion systems consist of:
 - Liquid fuel tank
 - Internal combustion engine
 - Otto engine
 - Diesel engine
 - Speed converter / torque converter
 - Manual transmission + clutch
 - Automatic transmission + torque converter
 - Propulsion systems, that consist of other components are termed as:
Alternative vehicle propulsion systems

Well to Wheel

- Consideration of complete energy conversion chain

Energy conversion chain (well to wheel)



Rankine-Cycle in steam engine Application

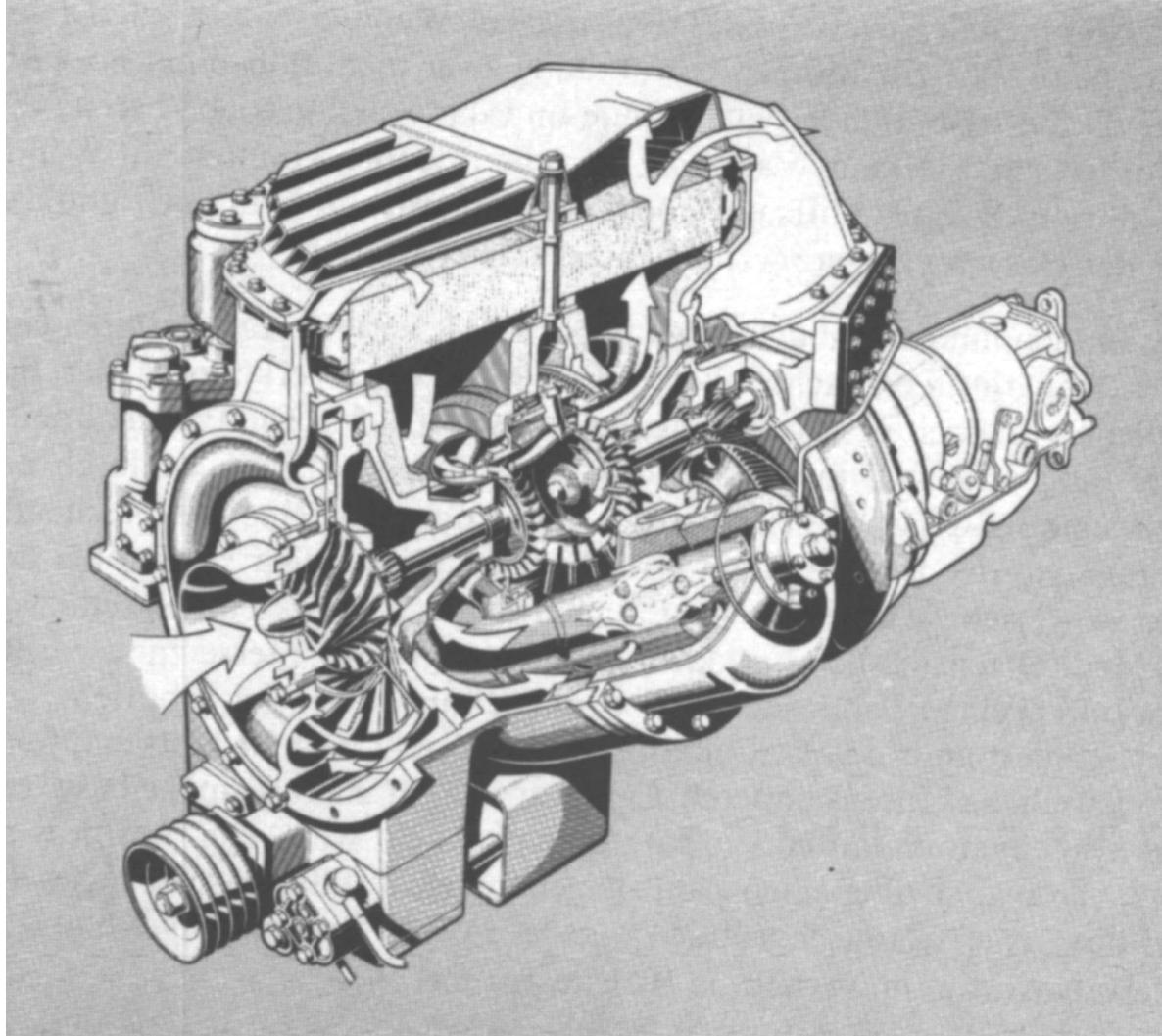
California Clean Car (1975)



Steam engine in Citroen
(self made, 1979)



Daimler-Benz high temperature gas turbine



Variable compression

Saab variable compression SVC, 5-Zyl. (2000)

$\varepsilon = 8 \dots 14$

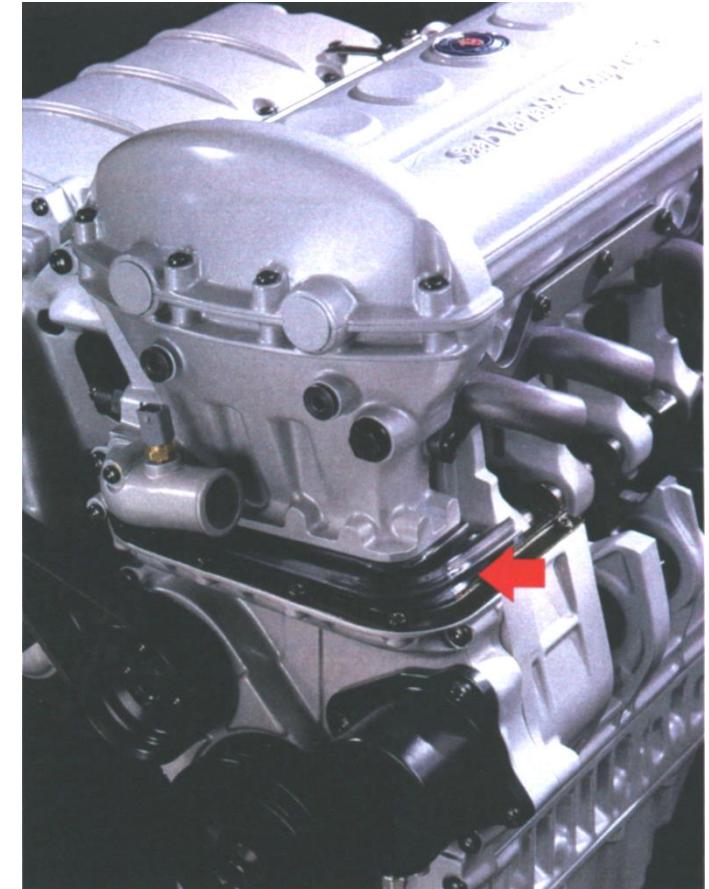
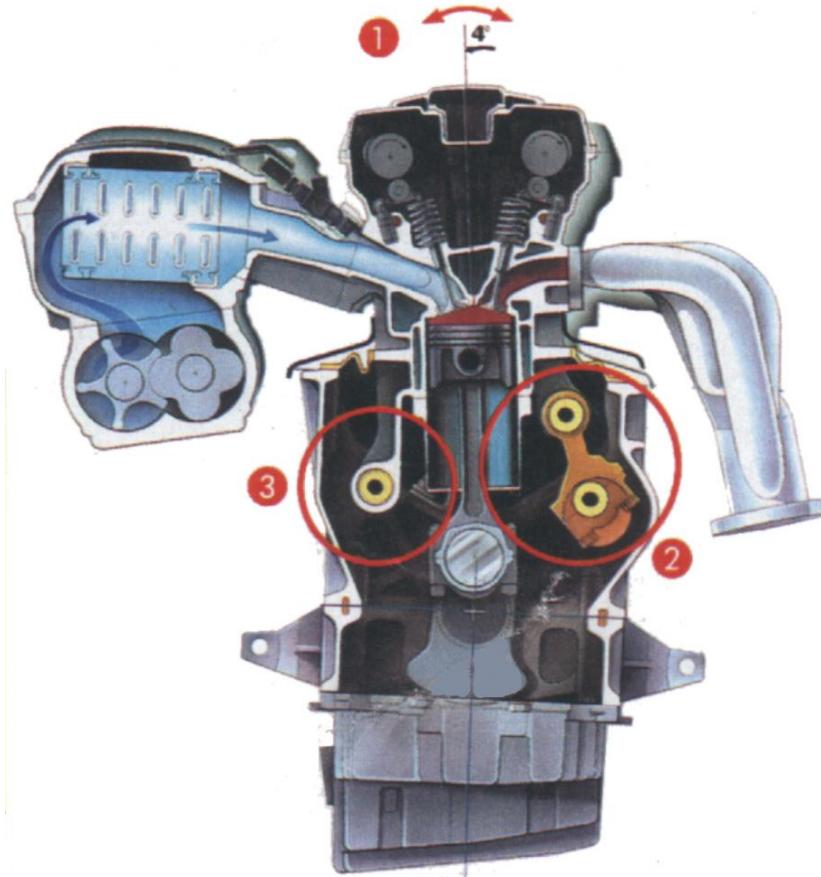
$p_{\text{sau},\text{max}} = 2.8 \text{ bar}$

(mech. boost)

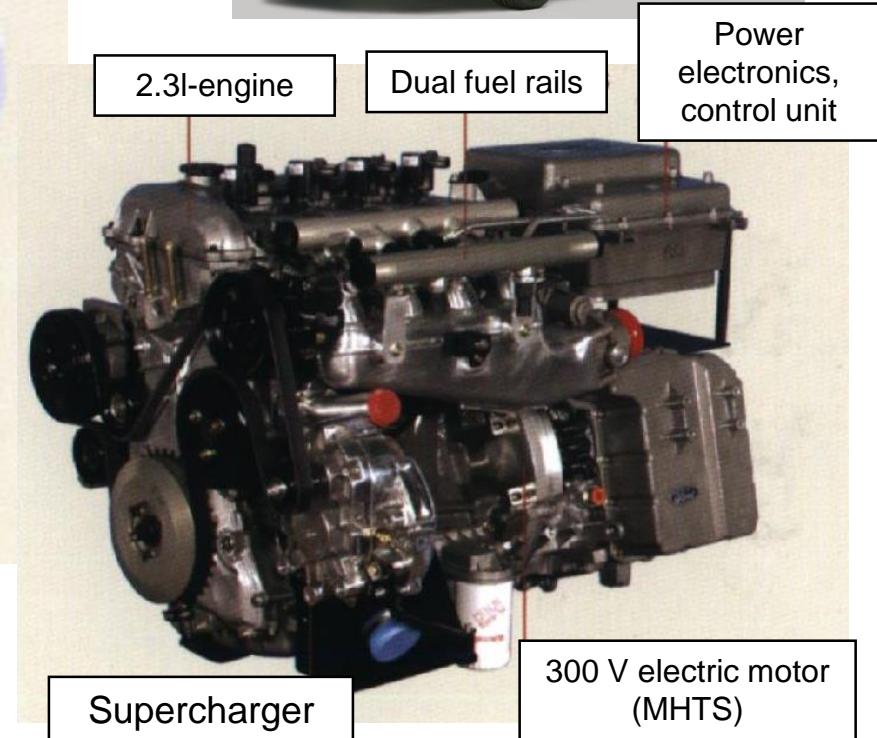
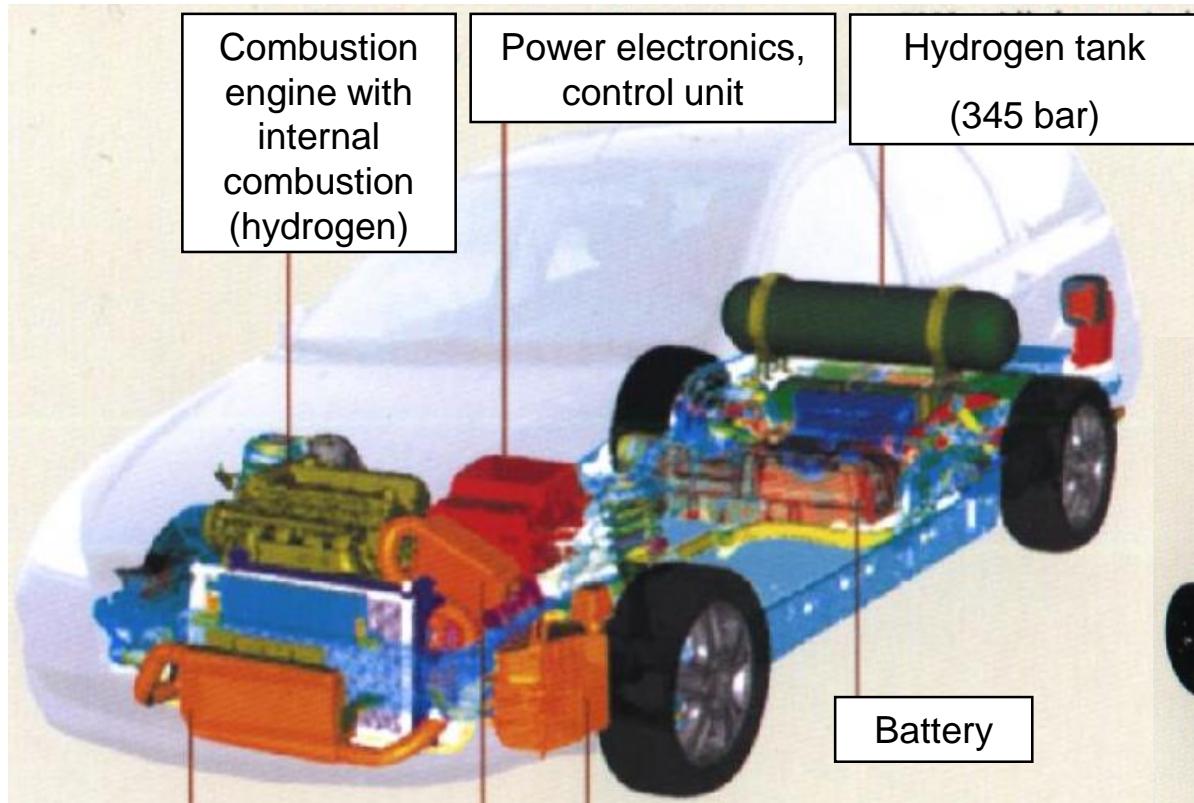
$V_h = 1.6 \text{ l}$

$p_{\text{me},\text{max}} = 23.8 \text{ bar}$

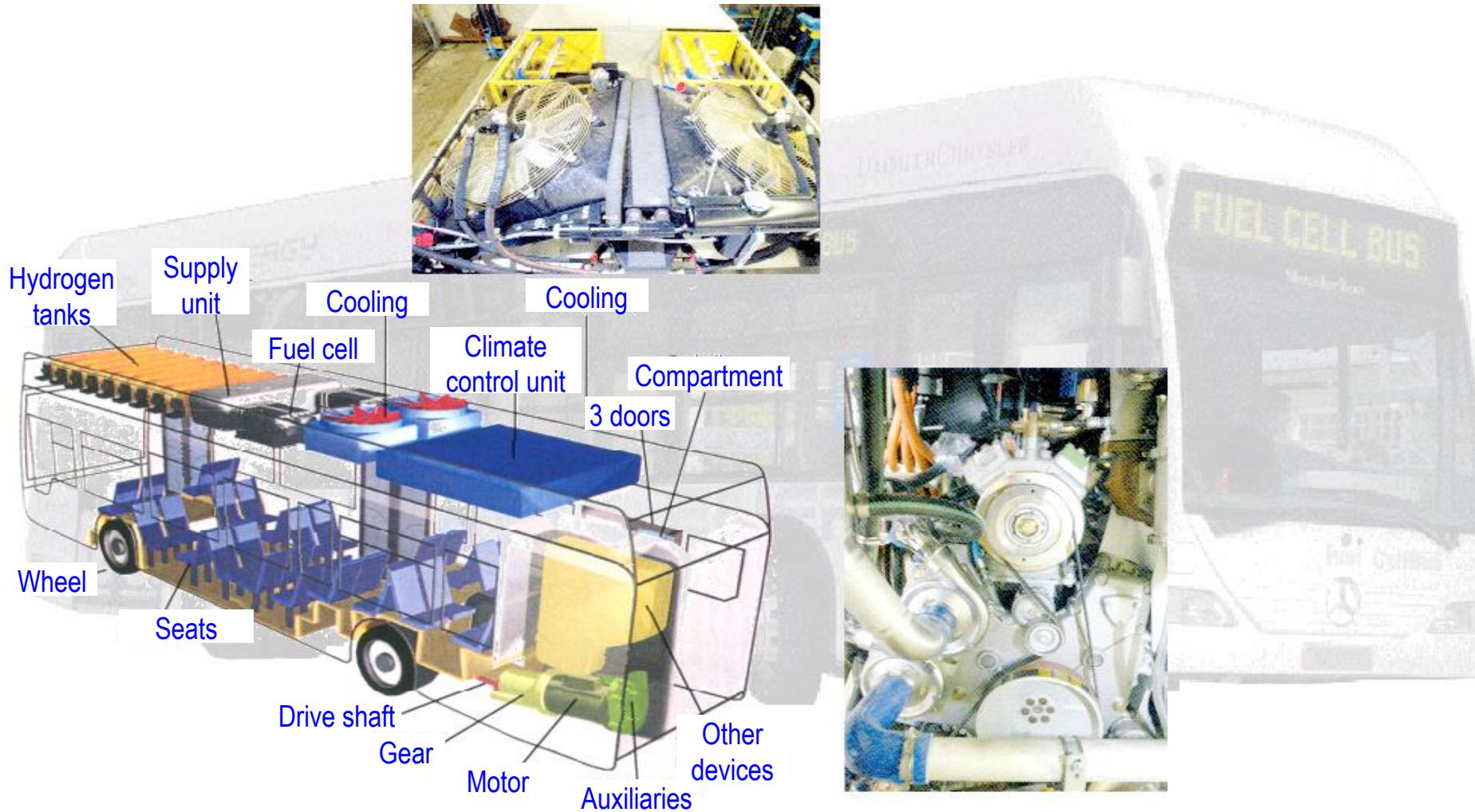
$P_e = 168 \text{ kW}$



Hydrogen hybrid vehicle (Ford H2RV)



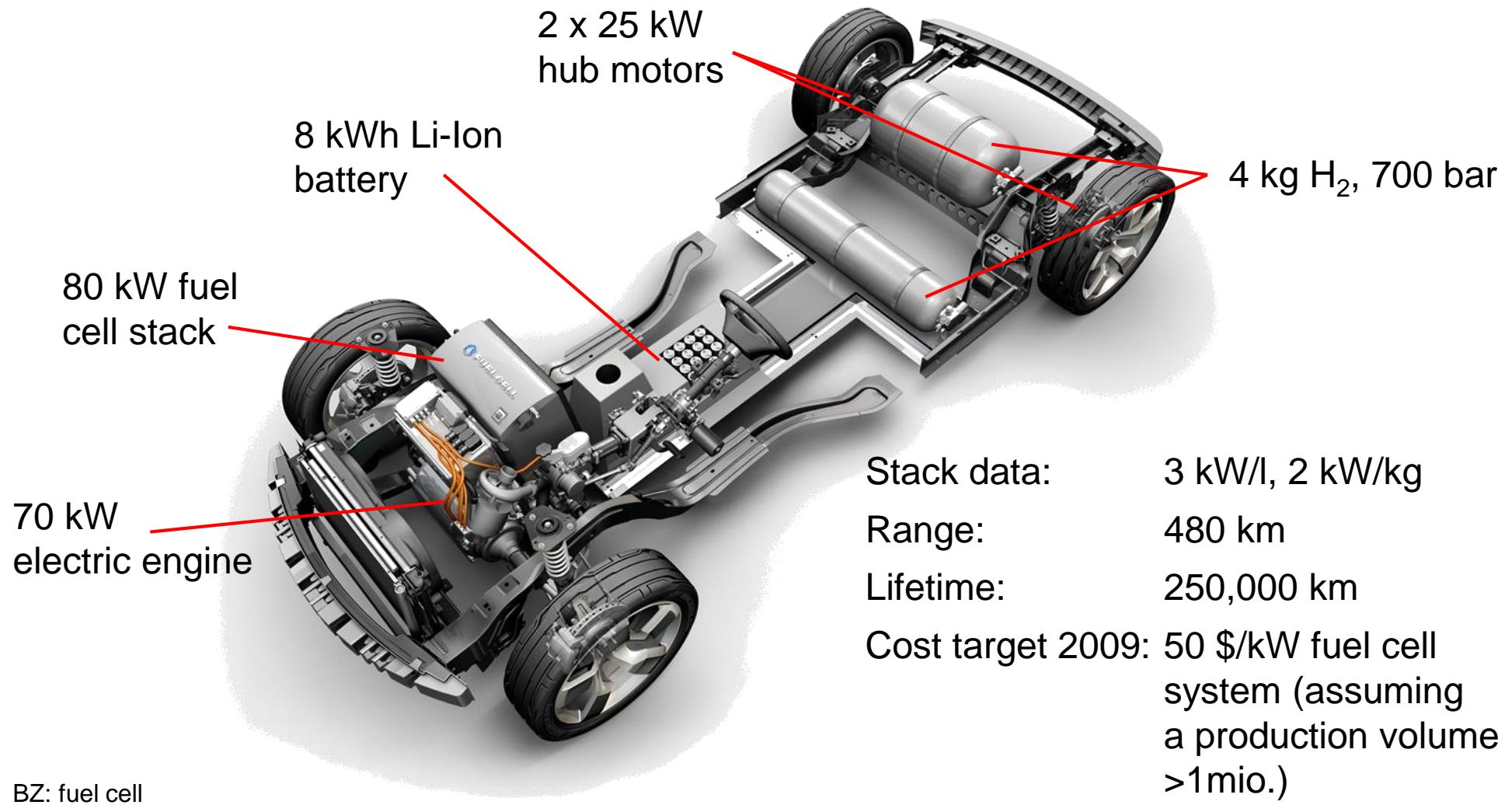
Daimler fuel cell bus (Development 2003 / series 2009)



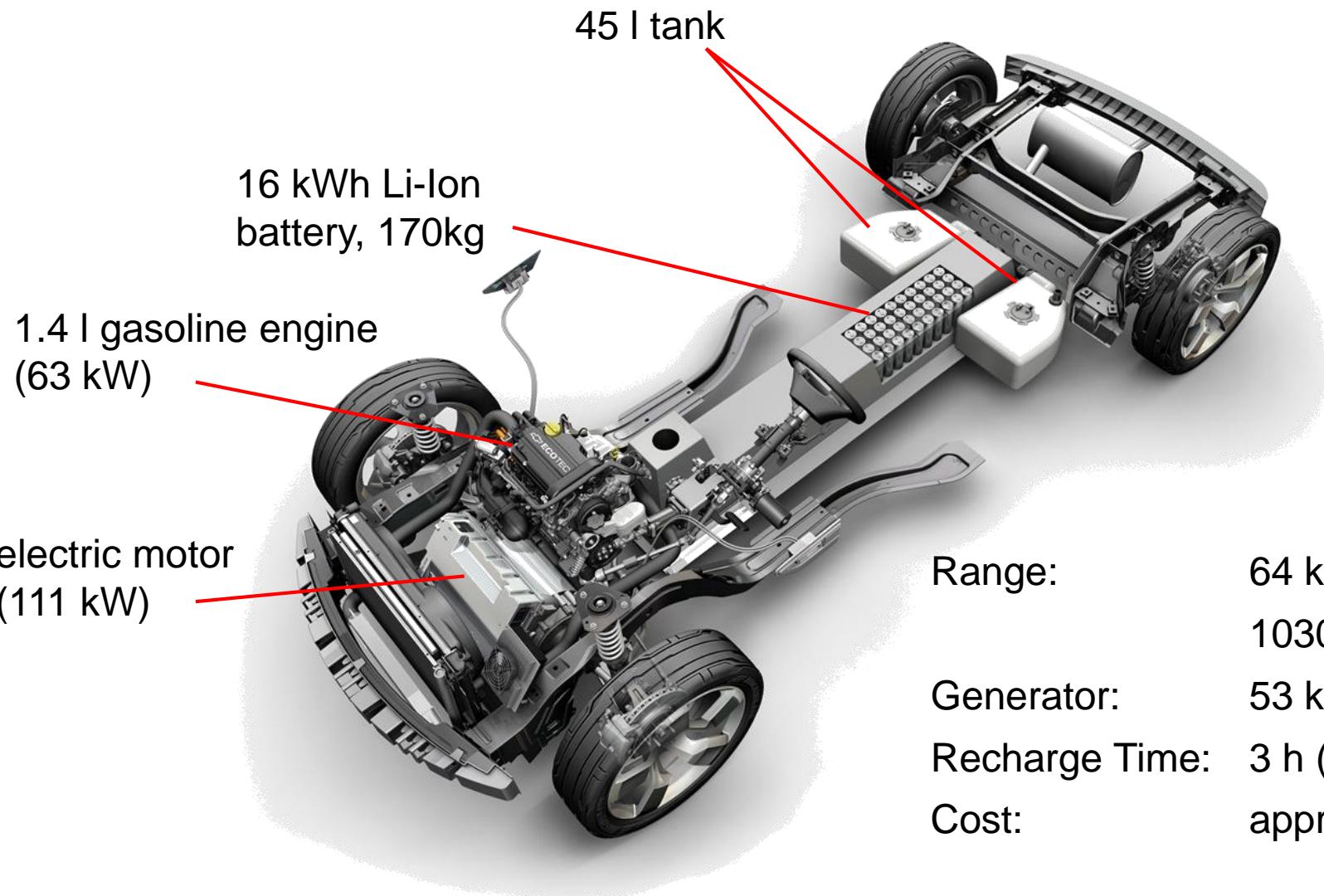
BMW vehicle with hydrogen engine (2006)



GM E-Flex fuel cell vehicle (2007)



GM Volt (2009)



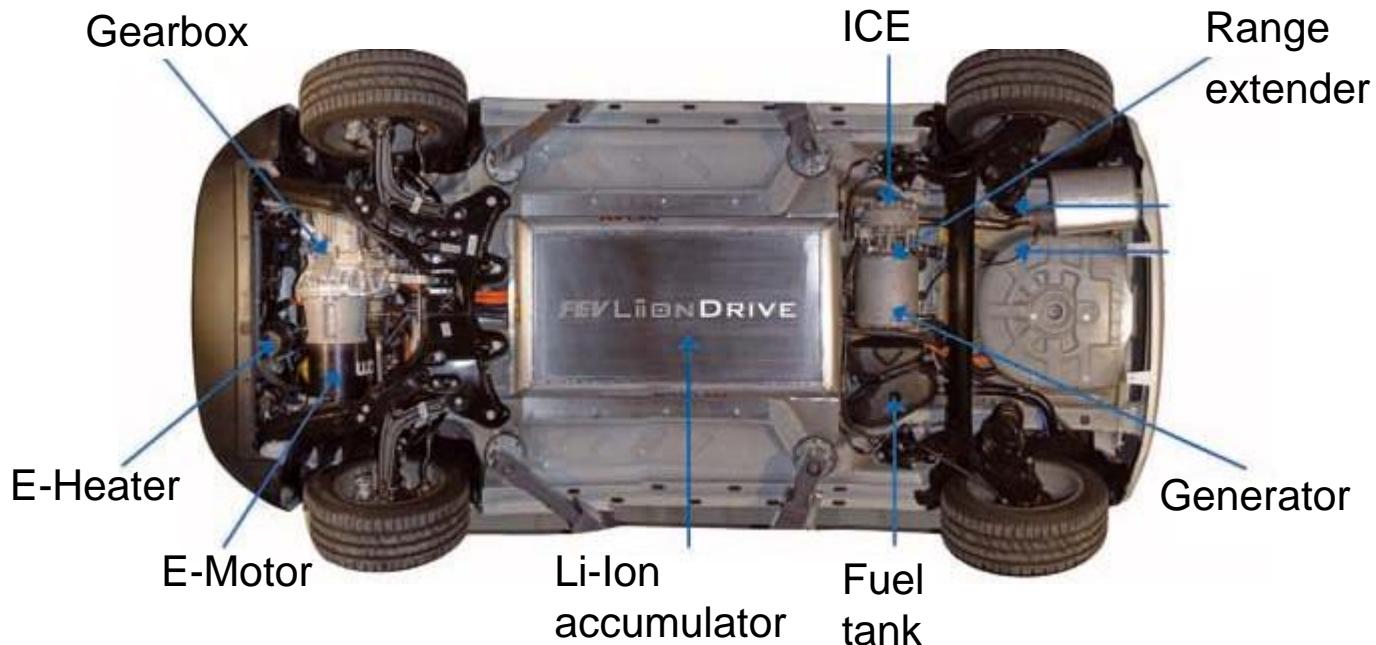
Tesla Model S



- Four pole three-phase induction motor with copper rotor
- Curb weight: 2100 kg
- Energy consumption 18.1 kWh/100km

Battery capacity in kWh	75 (2017)	85	100D	P100D
Driving range NEFZ in km	401	520	632	613
0 to 100 km/h in sec	5.8	4.4	4.3	2.7
Maximum velocity in km/h	225	225	250	250
Maximum Power in kW	235	270	310	449

FEV Plug-In Hybrid with Range Extender



Accumulator:

- 12 KWh Li-Ion accumulator for 60 km urban traffic

Range extender:

- 20 KW Wankel engine
- Range with range extender 300 km

E-Motor:

- 75 KW
- Maximum speed 120 km/h
- Acceleration from 0 to 100 km/h in less than 10 s



BMW i3 (Range Extender) (since 2013)

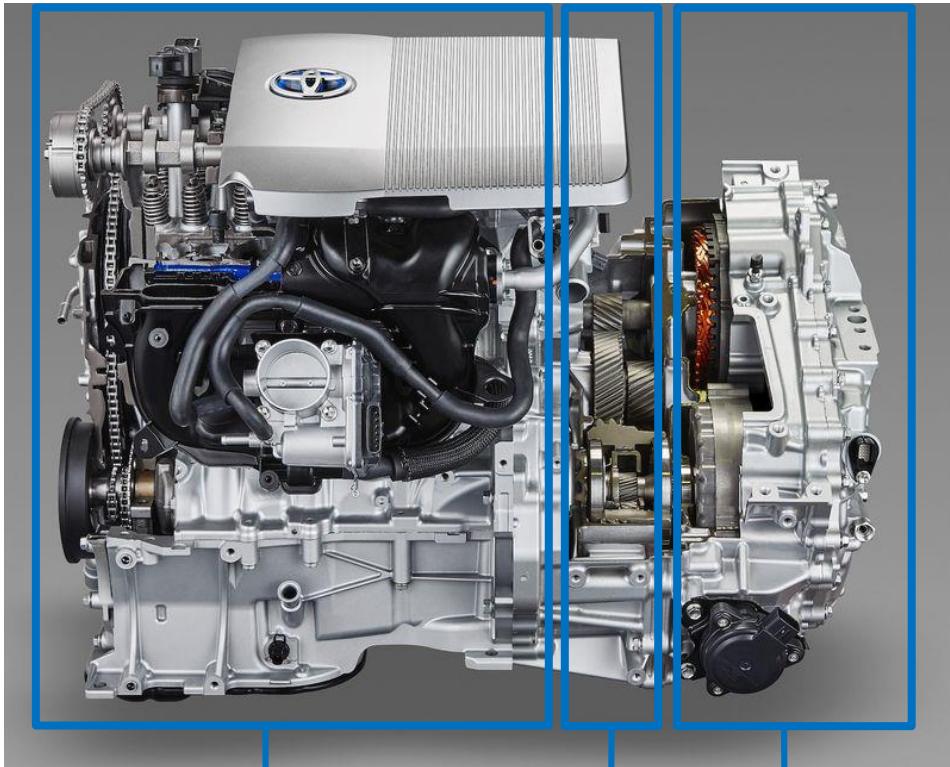


- E-Motor: Synchronous electric machine
- Range Extender: 2-cylinder-Otto

Technical Data

Battery capacity (Li-Ion) in kWh	18.8
Vehicle mass in kg	1315
Electrical travel distance (NEDC) in km	170
Maximum range (most efficient driving mode) in km	ca. 340
0 to 100 km/h in sec	7.9
Maximum velocity in km/h	150
Maximum power (E-Motor) in kW (hp)	125 (170)
Maximum power (Range Extender) in kW (hp)	25 (34)

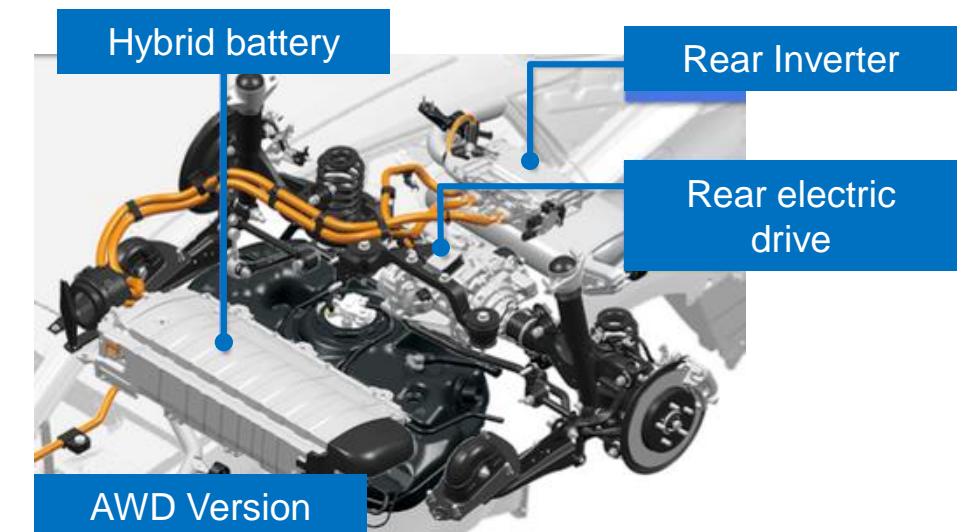
Cutaway model of a Toyota hybrid propulsion system (Toyota Prius IV, since 2016, Prius I 1997)



Combustion Engine

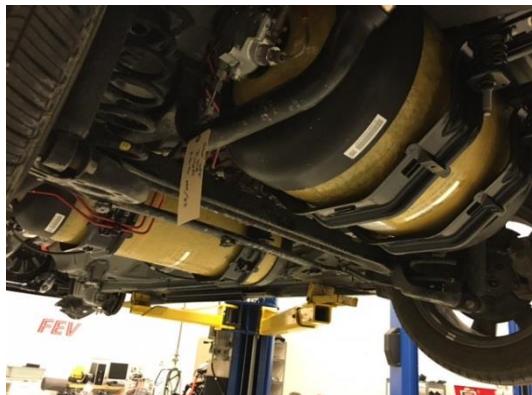
Electric drive

Planetary gear unit



Quelle: Greencarsreports.com 2016 Toyota Prius: A Few Details On Engine and Hybrid System

Toyota Mirai – Fuel cell vehicle (2014)



- Electric machine:
Synchronous electric machine
- 370-cell stack

Technical Data

Battery capacity (Li-Ion) in kWh	1.6
Vehicle weight in kg	1850
Price in € (ca.)	79,000
Maximum travel distance (most efficient driving mode) in km	ca. 500
0 to 100 km/h in sec	9.6
Maximum velocity in km/h	178
Maximum Power (E-Motor) in kW (hp)	114 (155)
Refueling in min	5

Source: <http://www.toyota.com/fuelcell/fcv.html>

Mercedes-Benz GLC F-Cell (2018)



Vehicle Data

Maximum velocity	160	km/h
Maximum Power	147	kW
H ₂ tank capacity	4.4	kg
Battery capacity (Li-Ion)	13.7	kWh
Max. travel distance (48 battery only)	437	km
Vehicle weight	2130	kg
Price	799 p.m (leasing only)	€

Sources: The New Mercedes-Benz Fuel Cell Plug-in-Hybrid Drive – 27th Aachen Colloquium Automobile and Engine Technology 2018
<https://www.adac.de/der-adac/motorwelt/reportagen-berichte/auto-innovation/mercedes-glc-fuel-cell-test>

Hyundai Nexo (2018)



Propulsion system and Fuel Cell Stack Data

Max. Power (E-Motor)	120 (160)	kW (hp)
Battery Power Output	40	kW
Fuel Cell Power Output	95	kW
Number of cells in FC stack	440	-
Stack Density	3.1	kW/L
Cold-Start Temperature	-30	° C

Vehicle Data

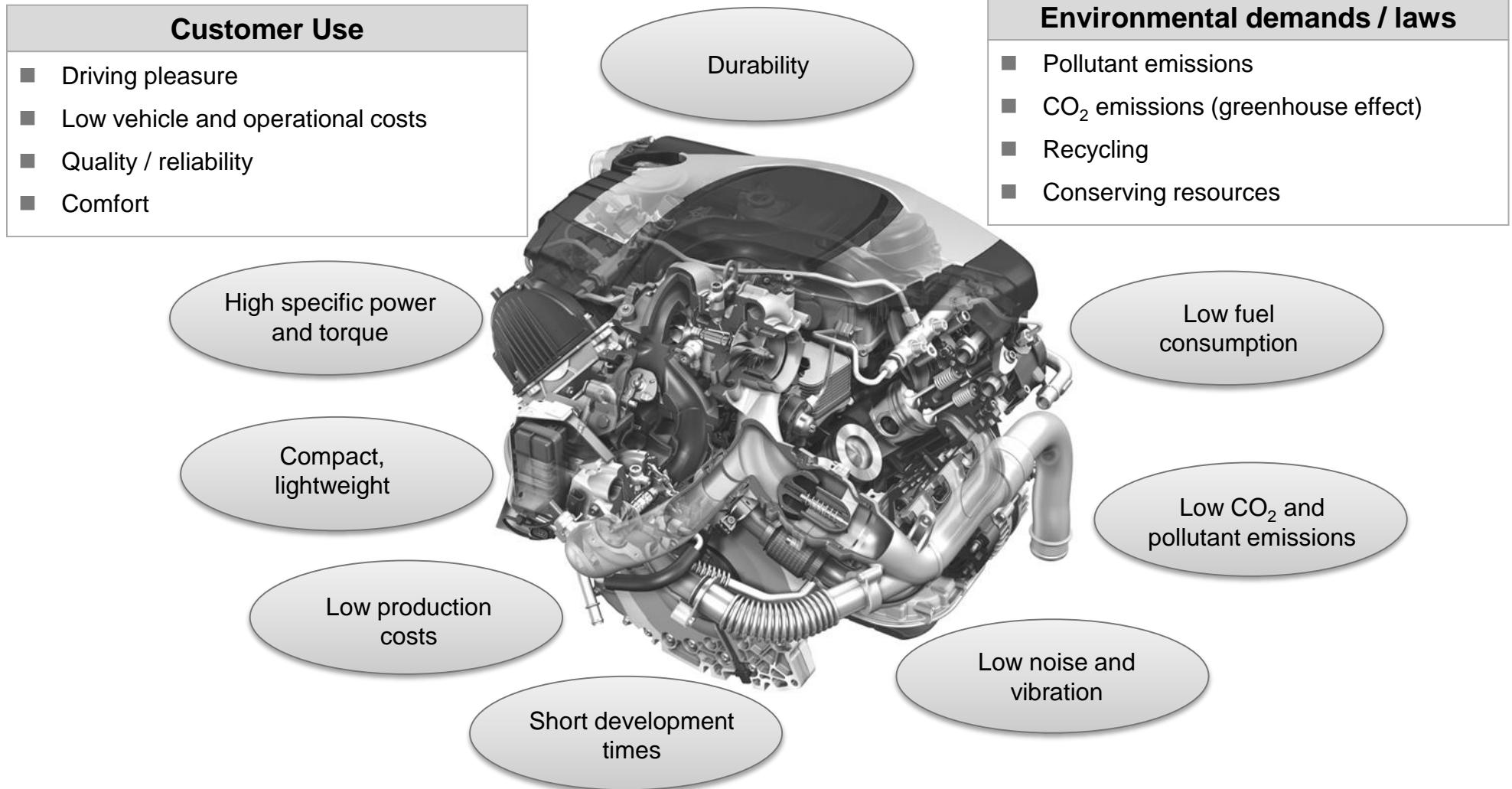
Maximum velocity	179	km/h
Acceleration from to 100 km/h	9.2	sec
H ₂ tank capacity	156.6 (3 x 52.2)	L
Battery capacity (Li-Ion)	1.56	kWh
Max. travel distance	756 (NEDC) 666 (WLTP)	km
Refueling time	5	min
Vehicle weight	1,814-1,873	kg
Price	69,000	€

Source: <https://www.hyundai.news/eu/press-kits/all-new-hyundai-nexo-technical-specifications/>

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Requirements for mobile propulsion systems

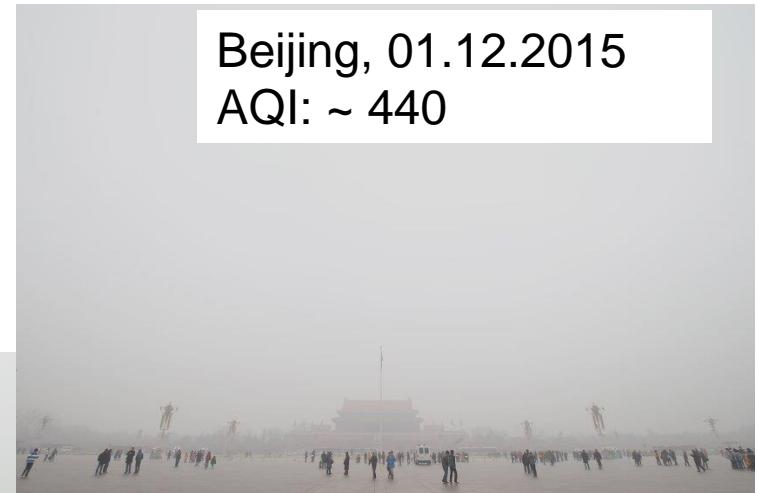


Atmospheric load due to exhaust gases



Beijing, 09.10.2014
AQI: ~ 380

AQI	Air Pollution Level
0 - 50	Good
51 - 100	Moderate
101-150	Unhealthy for Sensitive Groups
151-200	Unhealthy
201-300	Very Unhealthy
300+	Hazardous

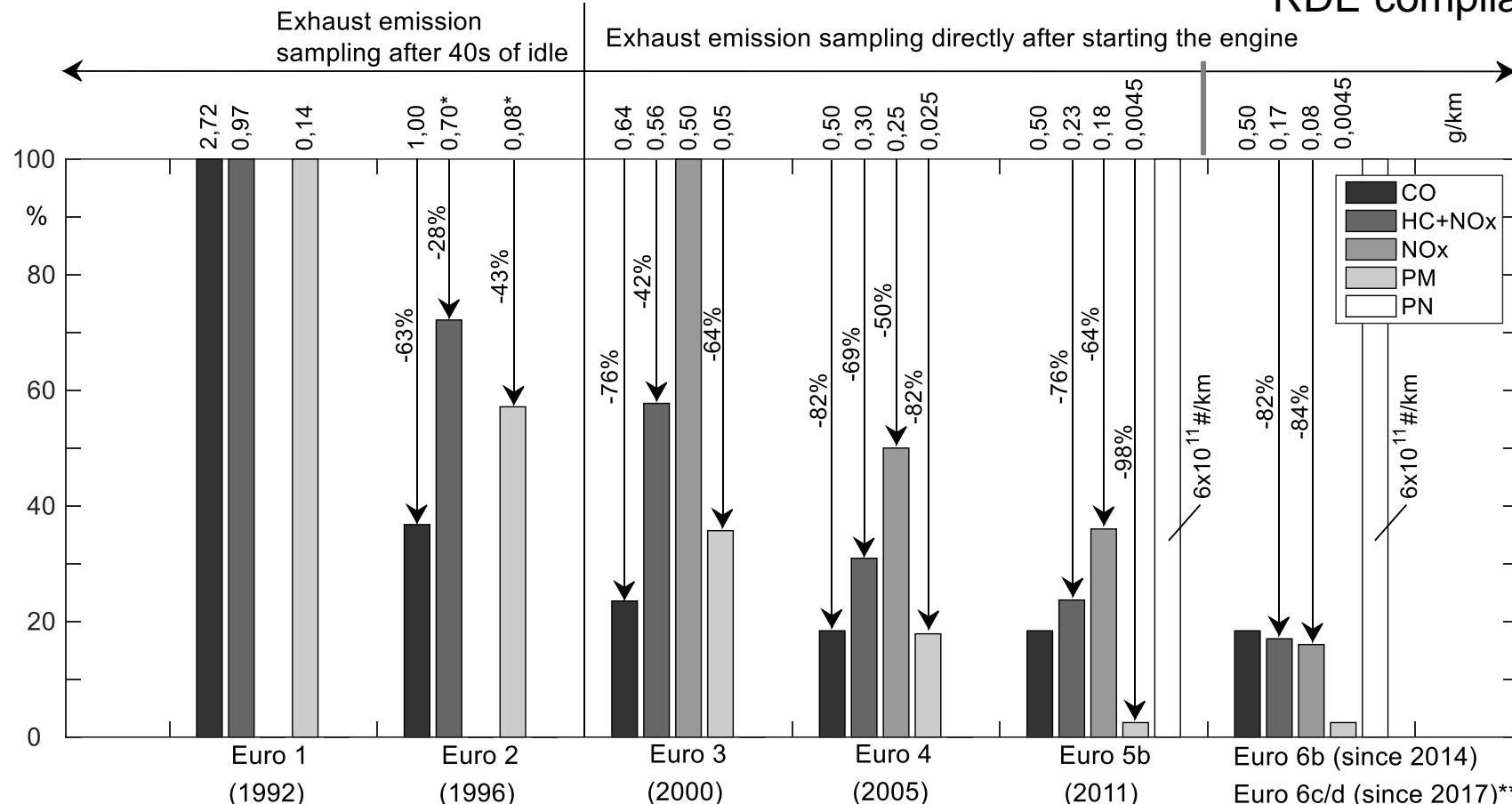


Legal motivation for alternative propulsion

- Limitation of toxic emissions through state regulation
 - EU6 temp, EU7, TIER 2 Bin 2 (USA)
- Penalties for car producers when their fleets CO2 emissions exceed a limit
 - In the EU the fleet's CO2 emission limit is set at 95 g/km from 2020/20 with a penalty of 95 € per g/km for every vehicle produced
- Tax incentives for vehicle with lower CO2-Emissions
 - In Germany CO2-based car taxes start in 2009, CO2-based fine in Europe
- EU-Guideline to renewable energy sources
 - Rate of renewable energies used in transportation sector (10% until 2020)
- Lower taxes for bio fuels
 - Lower taxes for pure Bio diesel and E-85 in Germany until 2015, CNG/LPG till 2018
- Emissions dependent road charge
 - Road charge on highways for trucks in Germany
 - Congestion charge for cars in London
- Local emission dependent driving ban
 - Pollution-free zone in German cities
- Purchase incentives / promotion
 - Subsidy for electric vehicles
 - Tax benefits for e-company cars

Development of the European Pollutant Emission Limits for Passenger Car Diesel Engines

New European Driving Cycle (NEDC) since 1992

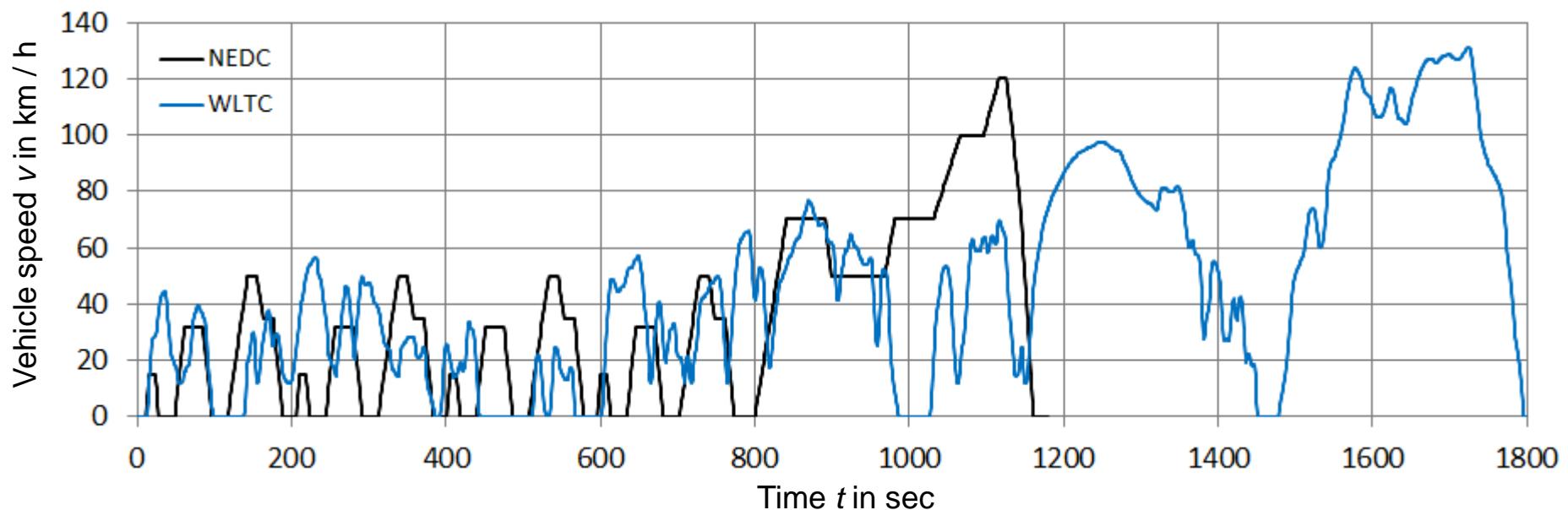


* without Diesel-DI

** Change from NEDC to WLTC

Worldwide harmonized Light vehicles Test Procedures (WLTP)

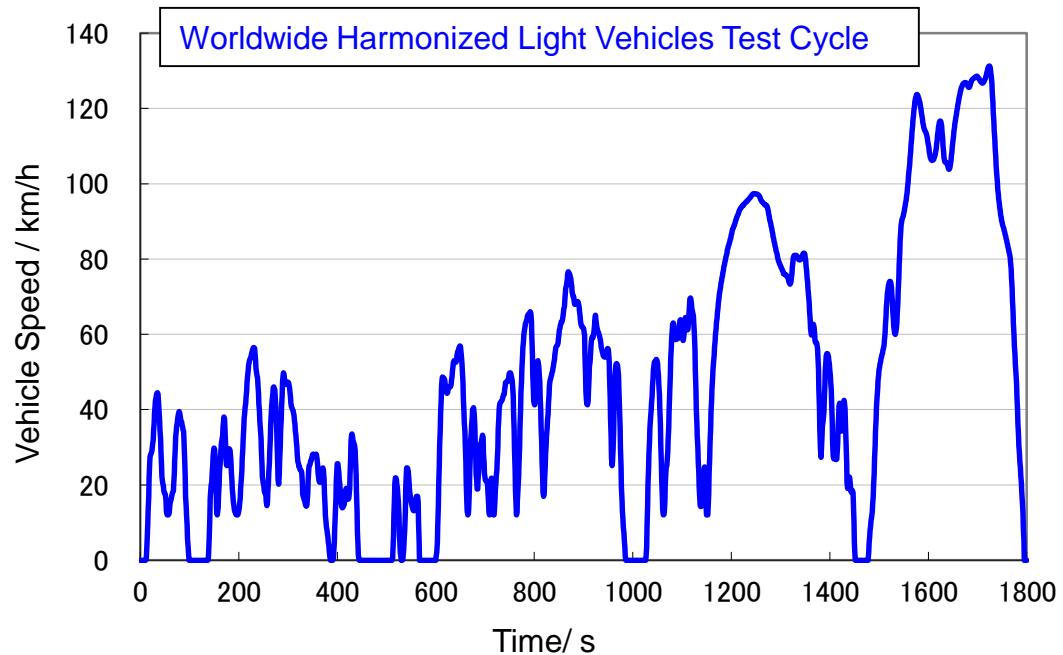
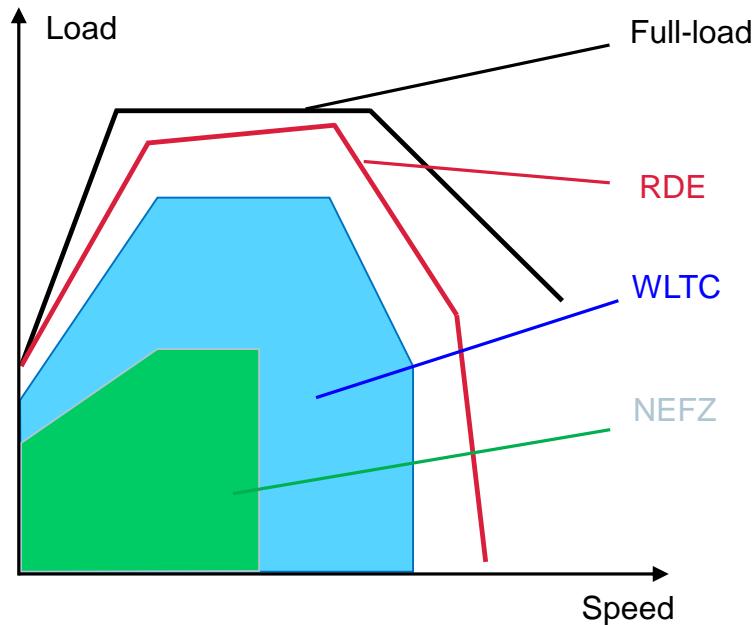
Parameter	NEDC	WLTP
Test duration	1180 s	1800 s
Test phases	Urban / Extra Urban	Low-, Middle-, High- und Extra-High
Max. speed	120 km/h	131,3 km/h
Max. acceleration	1,04 m/s ²	1,6 m/s ²
Test route length	11 km	23,262 km



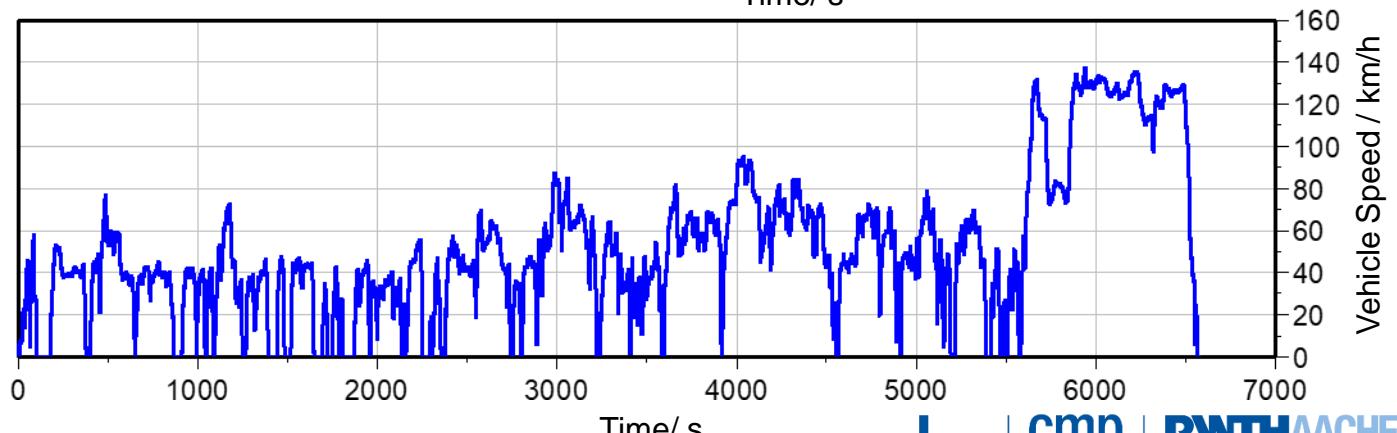
- The same cycle for cars and light commercial vehicles, but with different switching points

Introduction of new test procedures

Worldwide harmonized Light vehicles Test Procedures (WLTP) and RDE



RDE (Real Driving Emissions):

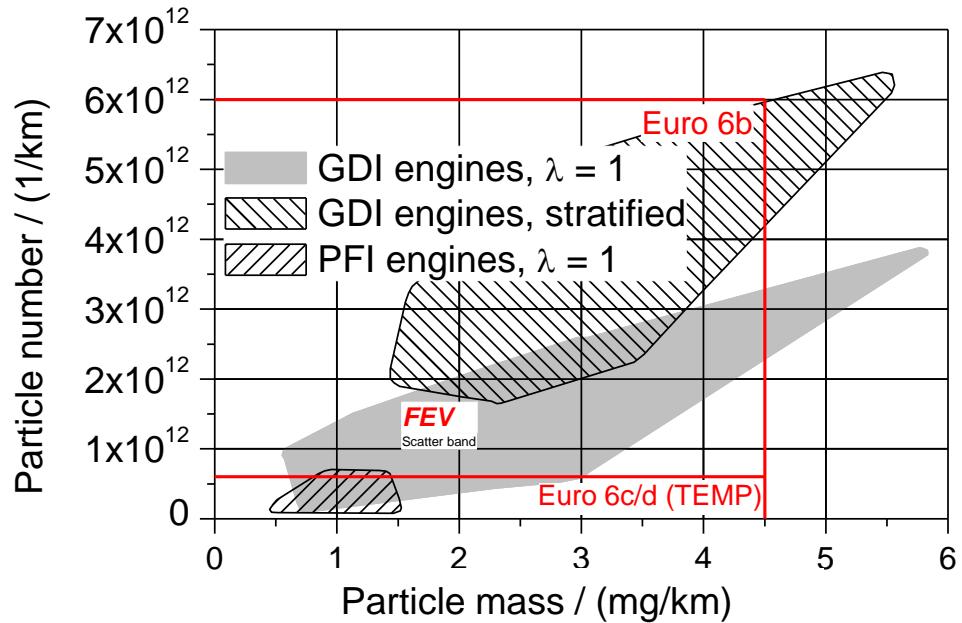


Partikelemissionen von Ottomotoren - Motivation

Legislative Requirements

Europe: Euro

- 6b - 09/2014: 6×10^{12} particles/km (DI) (NEFZ)
- 6c - 09/2017: 6×10^{11} particles/km (DI) (WLTP)
- 6d TEMP – 09/2017: 6×10^{11} particles/km (DI) (WLTP + RDE compliance)



Customer Acceptance

Diesel with DPF



Gasoline PFI

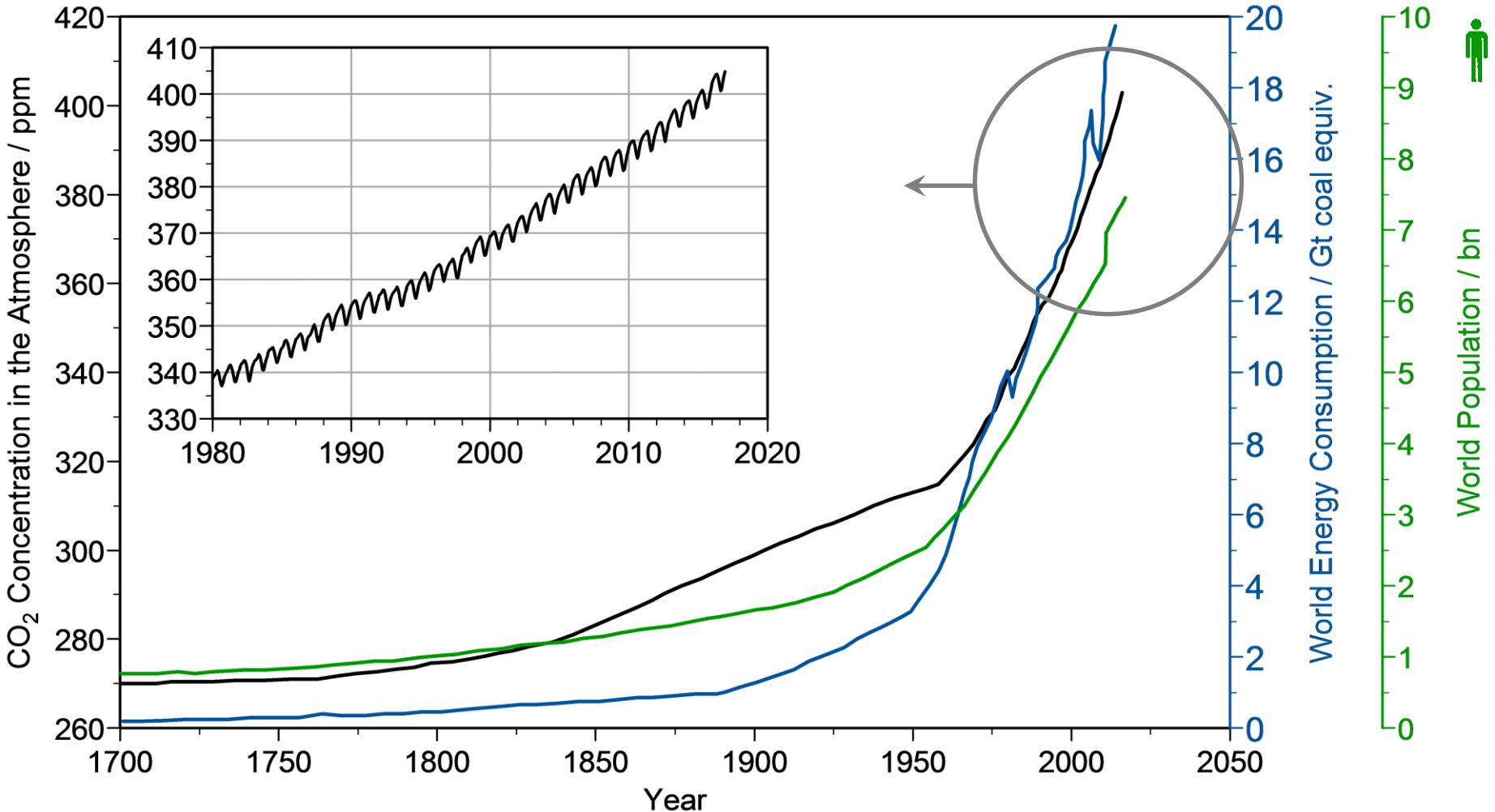


Gasoline DI



Legislative and customer demands require particle emission reduction for gasoline engines

World population, consumption of primary energy, CO₂-emissions

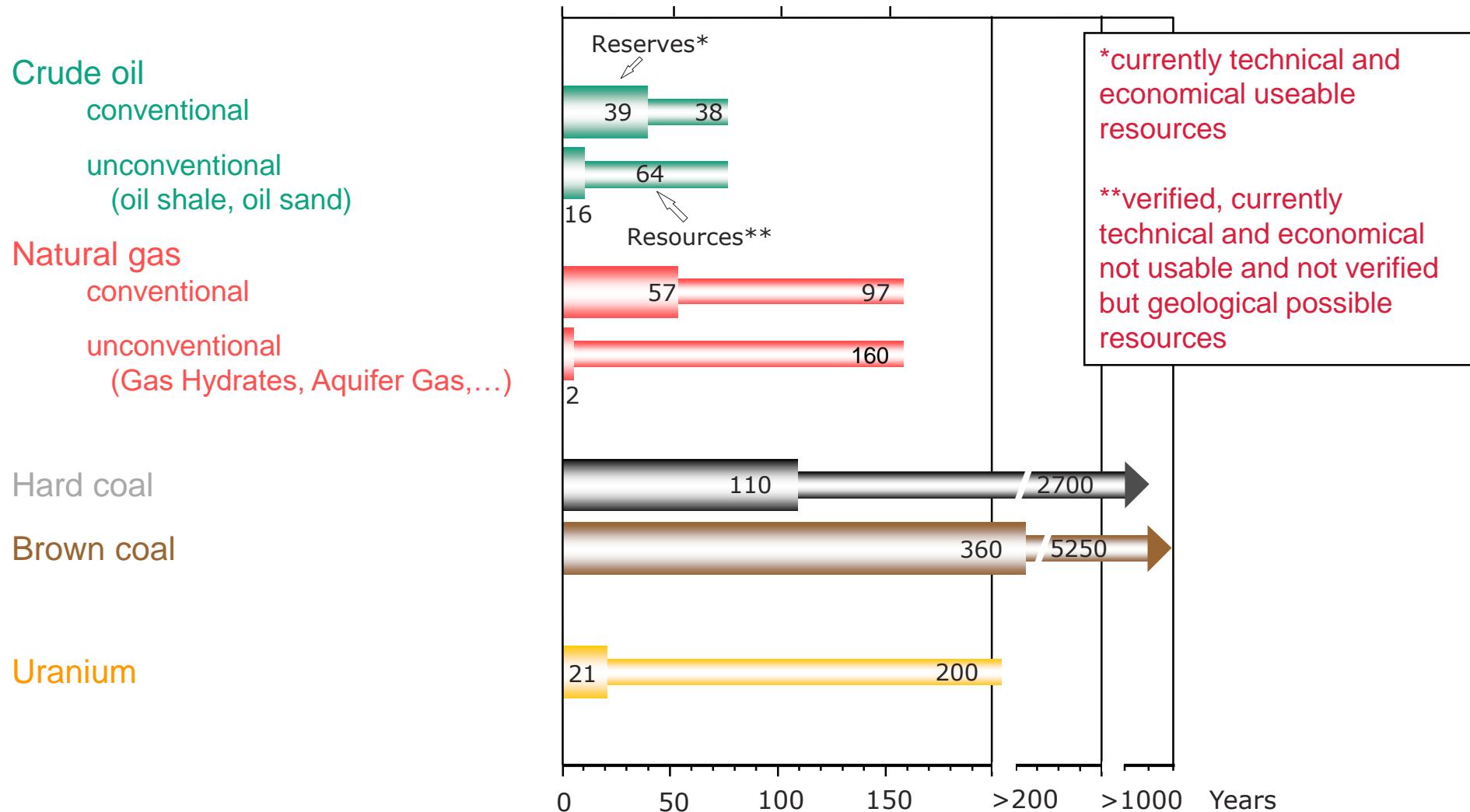


Sources: Deutsche Stiftung Weltbevölkerung 2017, International Energy Agency 2016, National Oceanic and Atmospheric Administration 2017

Script: ICE2 15.3, ICEF 8.4, AVPS 1.2

E 6818 ICEF, ICE2, AVPS

Reach/Coverage of Fossil and Nuclear Energy Sources (current consumption)



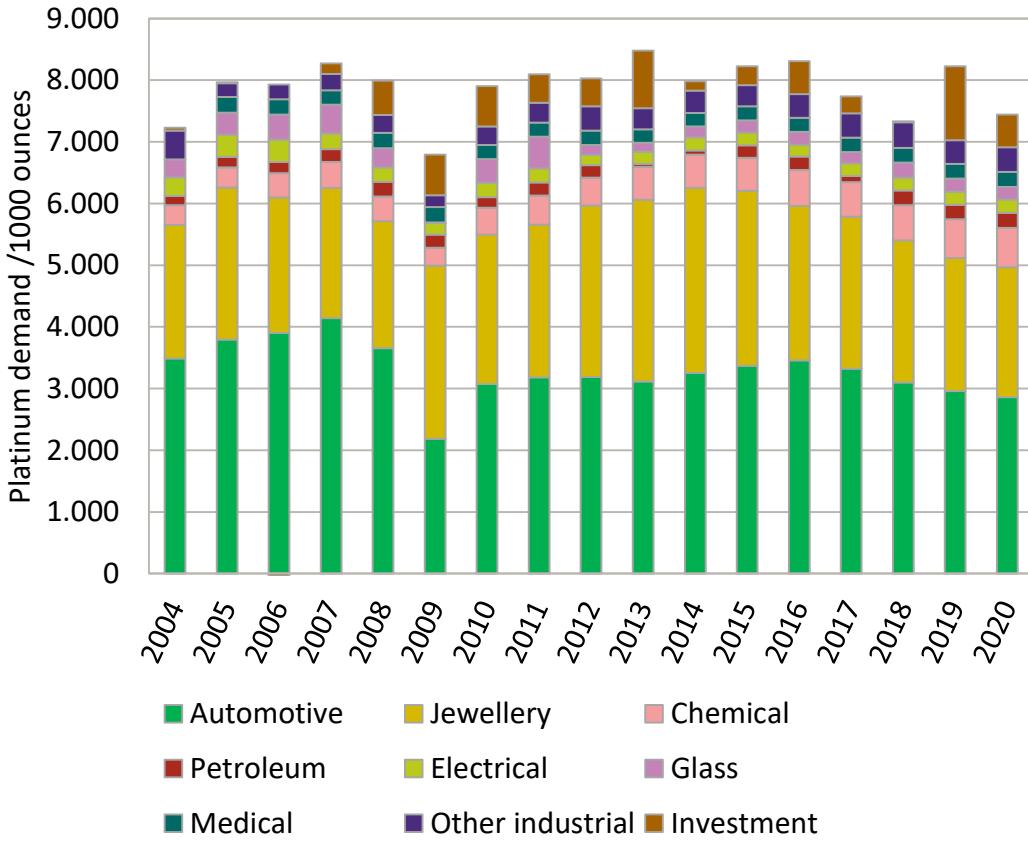
Source: Energiestudie 2017 – Daten und Entwicklungen der deutschen und globalen Energieversorgung, BGR

Lecture notes: ICE1 2.2, AVPS 1.2

E 4319 ICE1, AVPS

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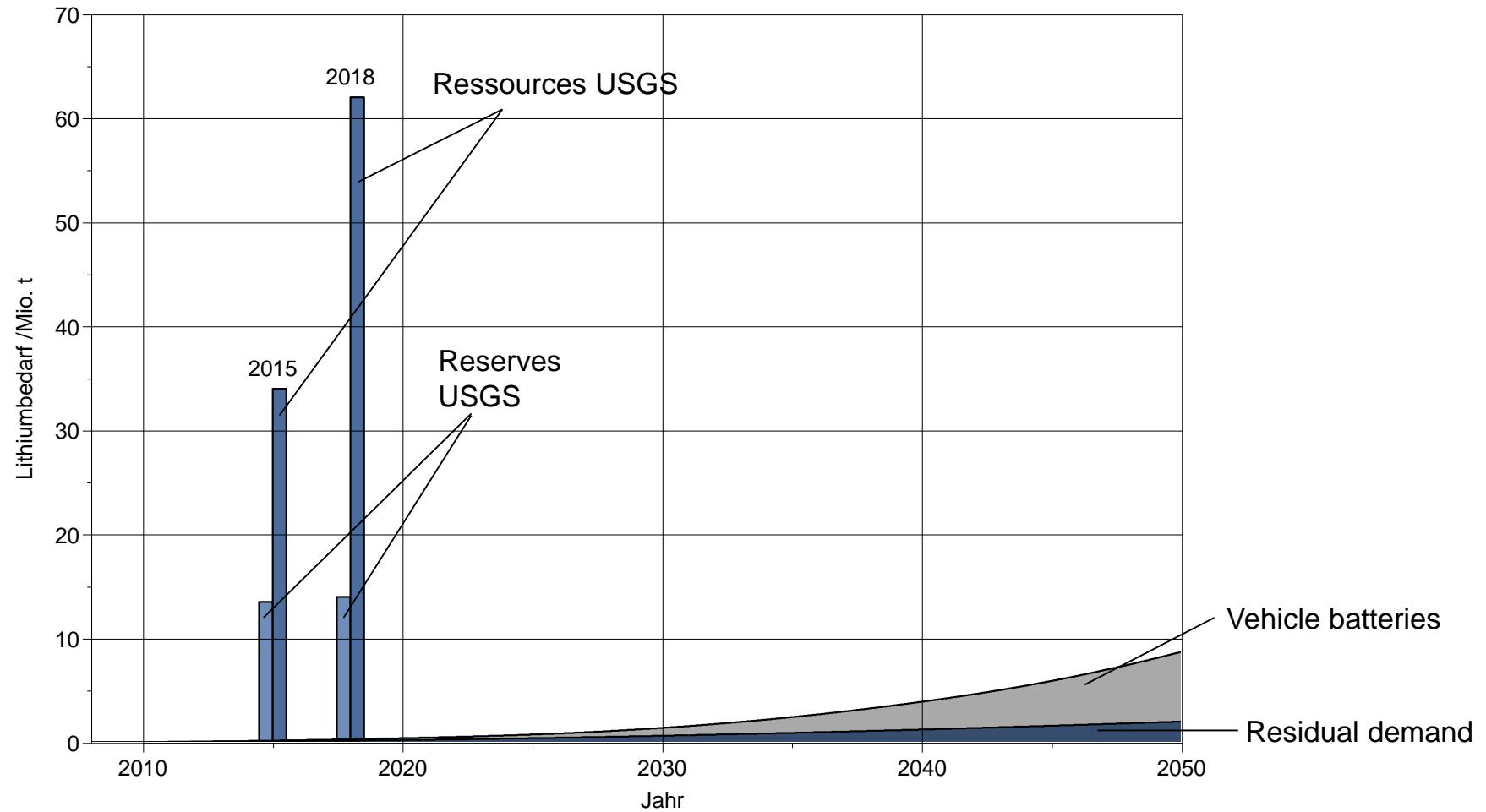
Automotive platinum consumption



Usage of platinum per vehicle

- fuel-cell technology
 - beginning: up to 1000 g
 - now: 10-20 g
(ca. 0.1 mg Pt/cm²)
- Exhaust gas after treatment for combustion engines
 - approx. 4-5 g Pt/vehicle (Europe)
 - up to 25 g Pt/vehicle
(Diesel, US Tier 2 Bin 5)

Lithium availability – demand scenario: almost complete substitution by hybrids and pure electric vehicles

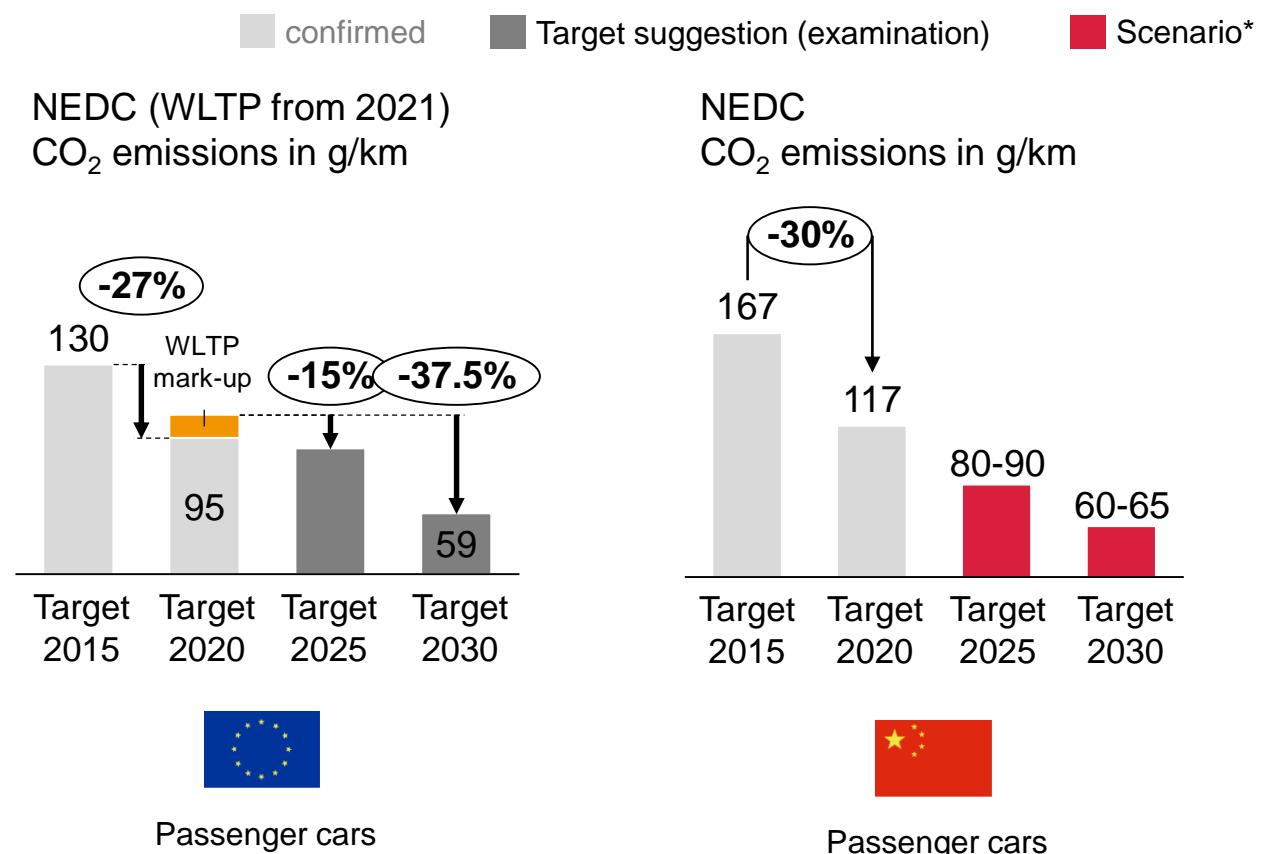
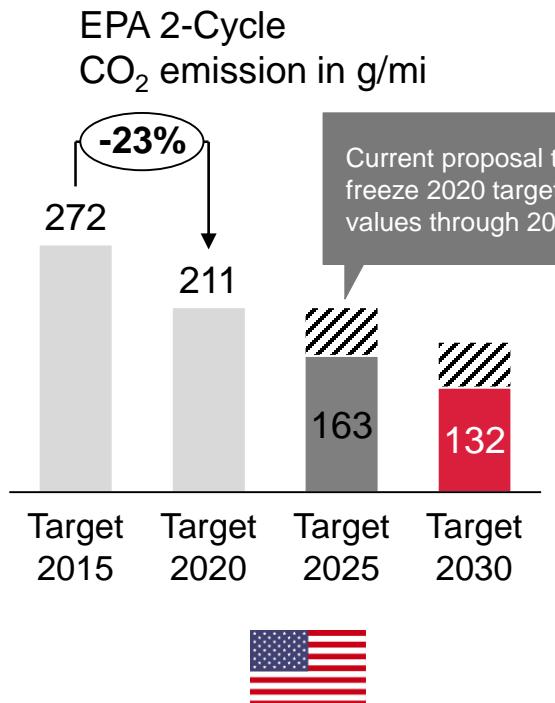


Source: Fraunhofer ISI - Lithium für Zukunftstechnologien, 2009

USGS: U.S. Department of Interior – U.S. Geological Survey – Mineral Commodity Summaries 2009 and 2016

Intensification of CO₂ emission targets (2015 – 2030)

CO₂ Emissions Regulation



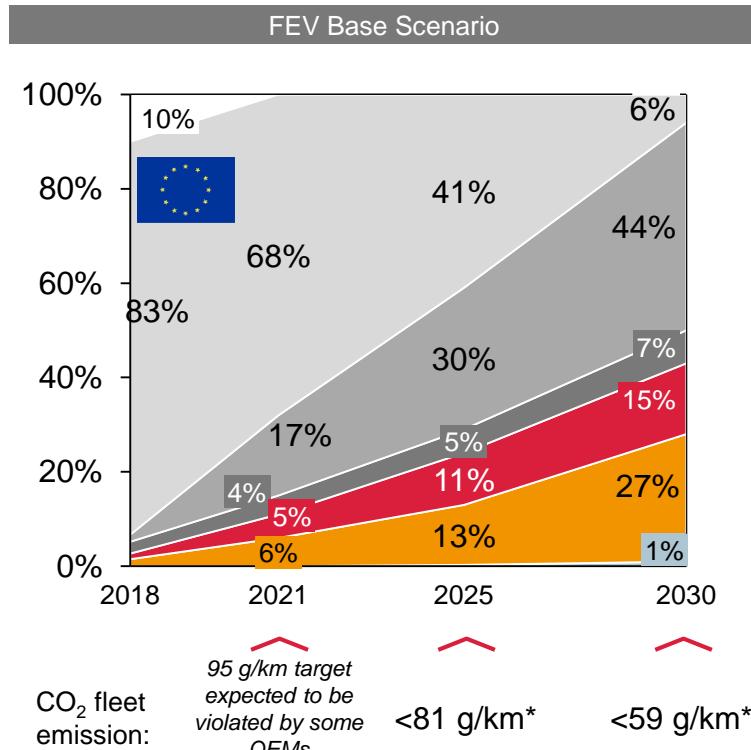
NECD = New European Driving Cycle; GHG = Greenhouse Gas

* EU: based on GHG reduction targets for transport sector by European Commission; US: 4% annual reduction assumed after 2025; China: convergence with EU targets expected

// CN figures are converted from l/km

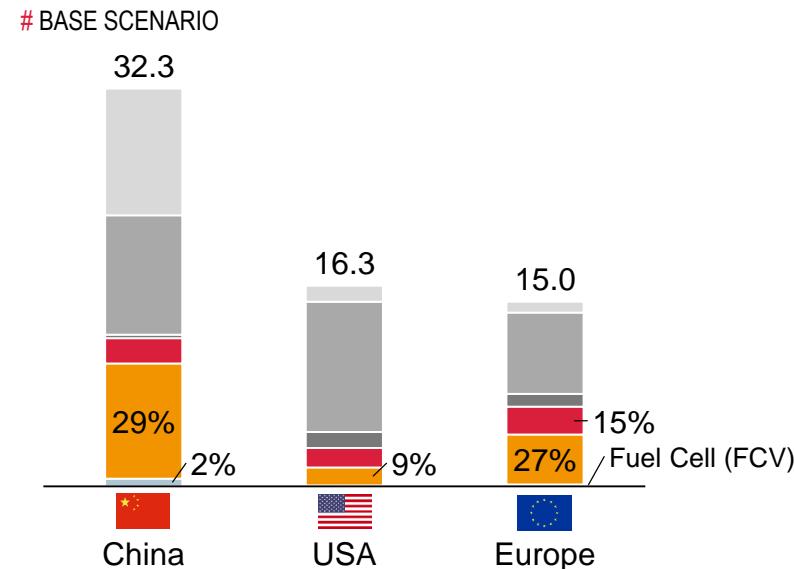
Quelle: European Commission, ACEA, FEV

Significant BEV share increase to 2030 – major market for Battery electric vehicles is China followed by EU



Passenger Car powertrain electrification scenarios – Vehicle sales

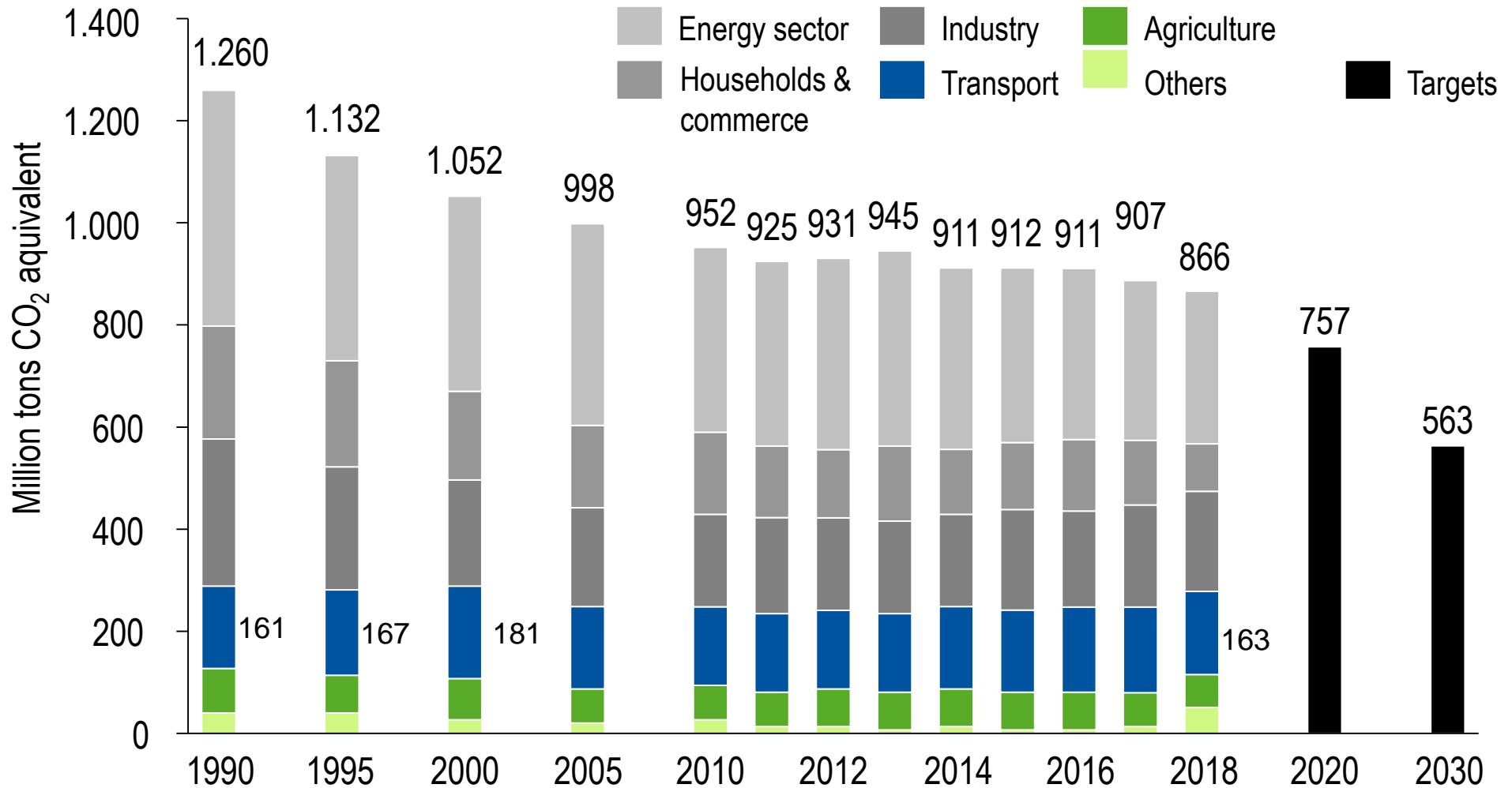
Pass. car powertrain type forecast 2030 in mio. units



Legend:
 ICE only Stop-Start & 12V Energy Mgmt Mild Hybrid Plug-In Hybrid Fuel Cell
 Full Hybrid Battery Electric

*: normalized to NEDC

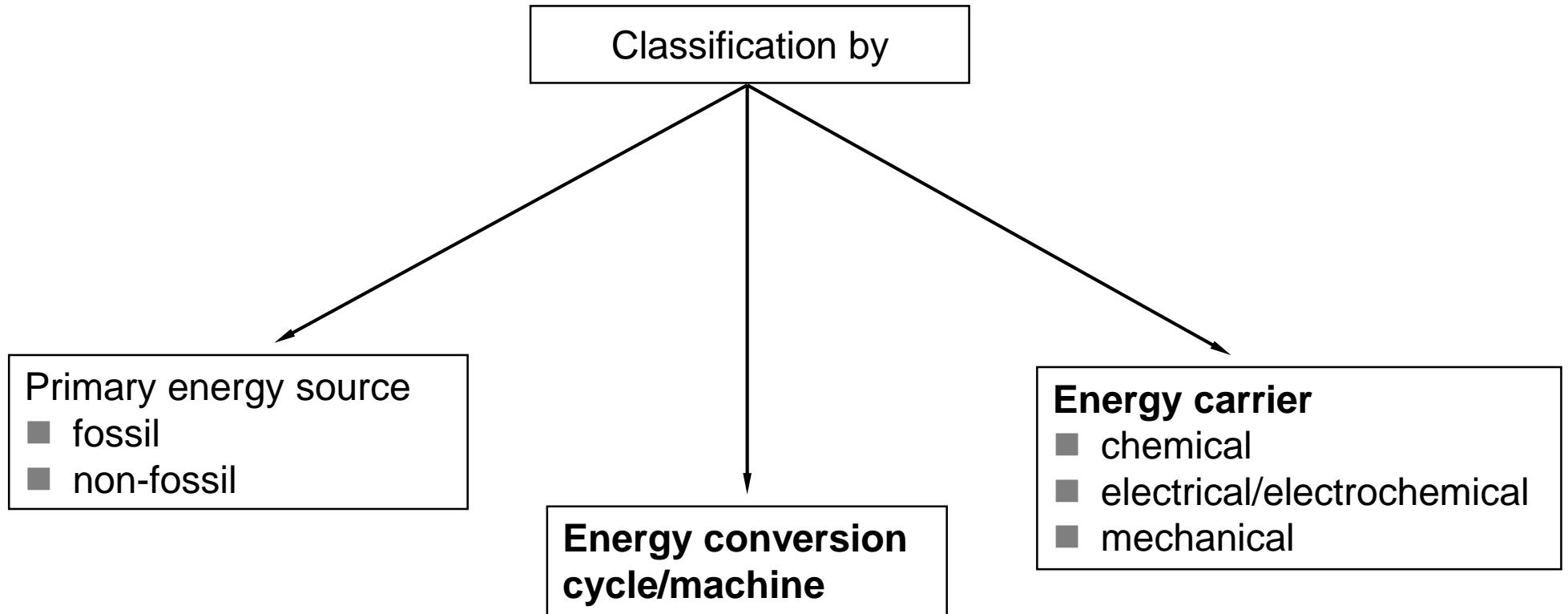
Greenhouse gas emissions in Germany



Content of lecture

1. Introduction
 - 1.1 Definition
 - 1.2 Motivation
 - 1.3 Classification of alternative propulsion systems
2. Energy sources and properties
 - 2.1 Properties of fuels
 - 2.2 Production of alternative fuels
 - 2.3 Alternative fuels and applications
3. Energy conversion cycles and their realization
 - 3.1 Thermodynamical energy conversion
 - 3.2 Electrochemical energy conversion: Fuel cell
 - 3.3 Electrochemical energy conversion: Battery
 - 3.4 Thermoelectrical energy conversion

Classification of alternative propulsion systems



- Combinations of two or more classifications are possible.

Classification of alternative propulsion systems

by primary energy source:

■ fossil

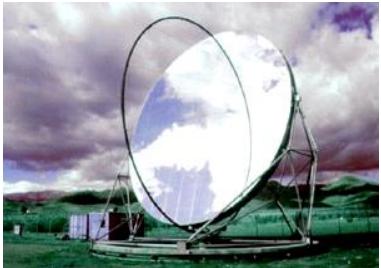


Coal

Mineral
oil

Natural gas

■ „regenerative“:



Sun

Water

Wind

Biomass

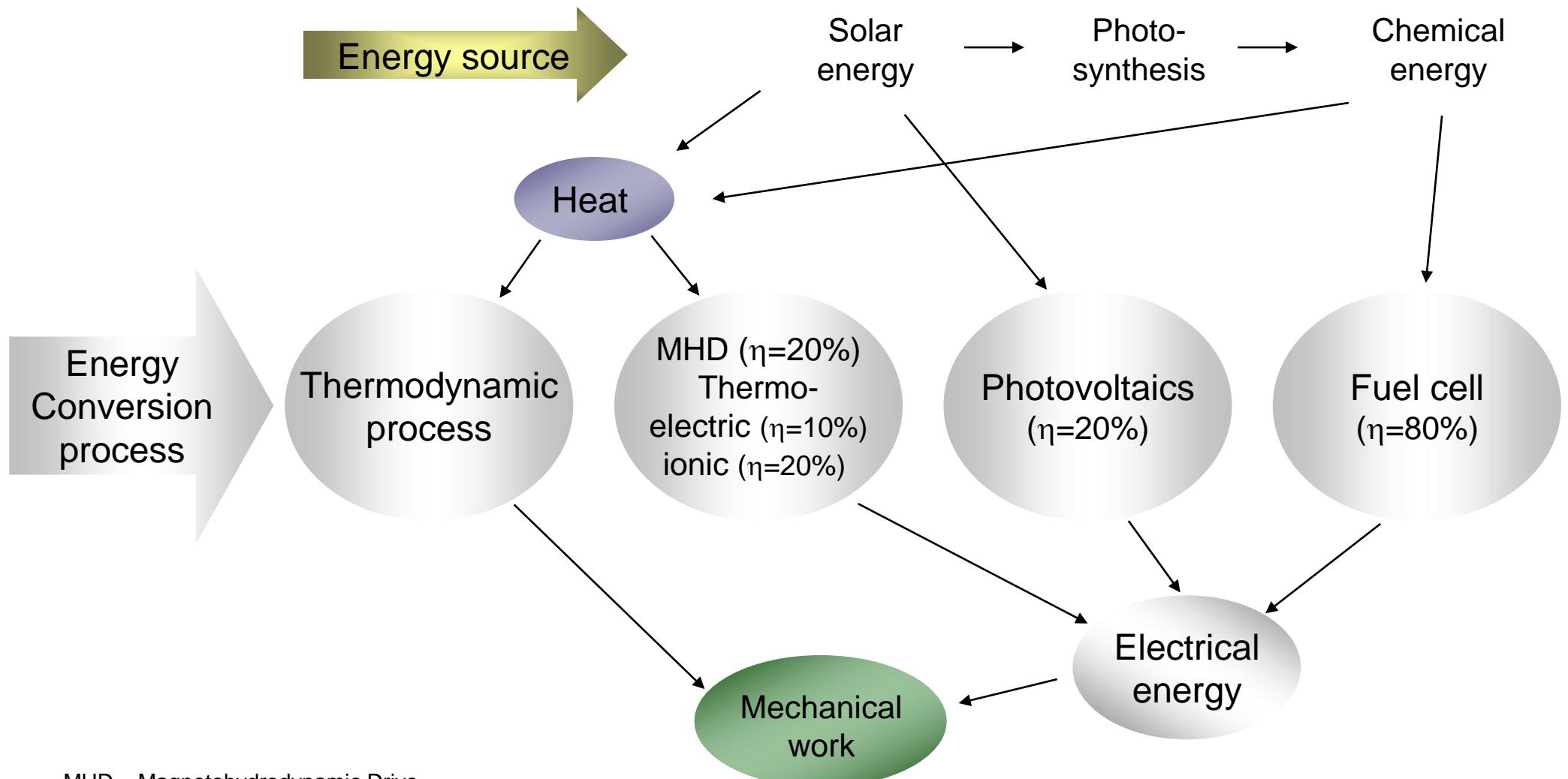
Geothermal

Classification of alternative propulsion systems

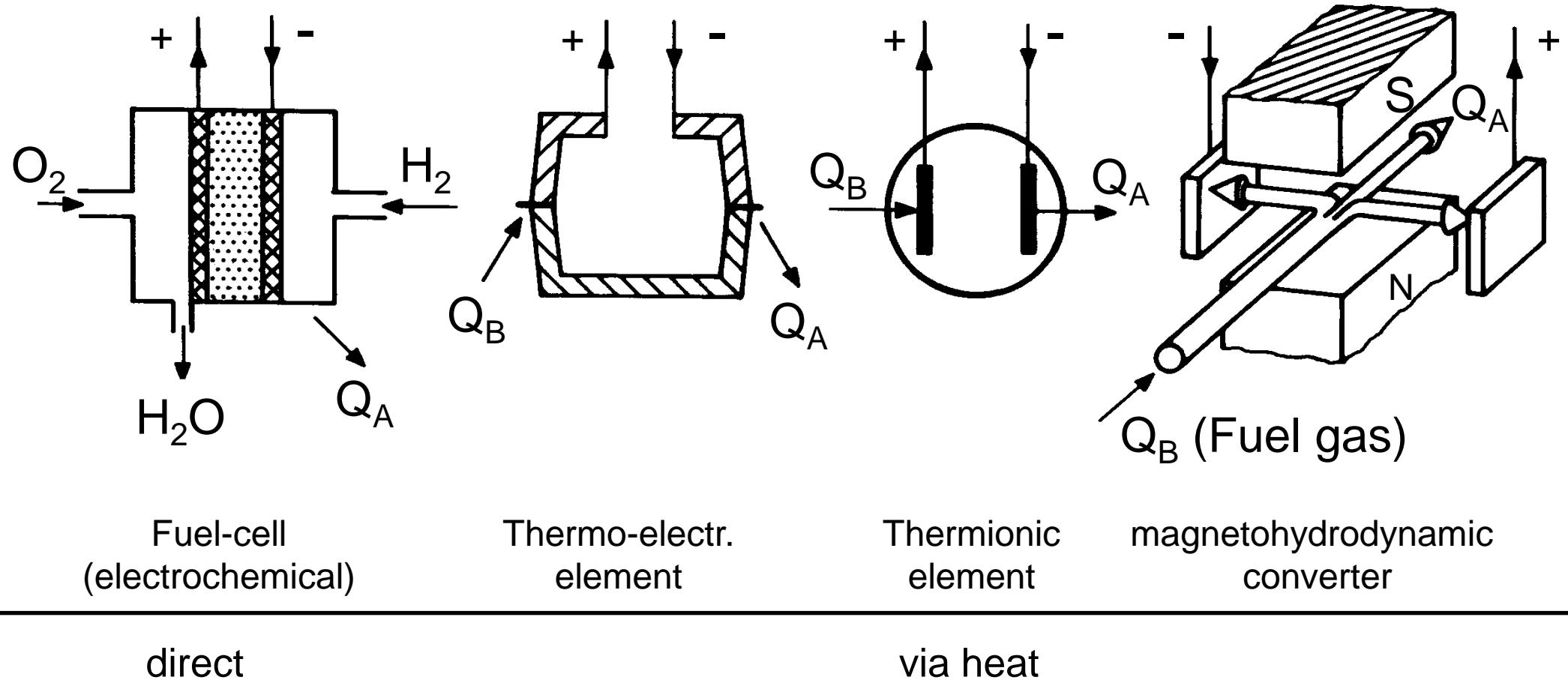
by form/ source of energy (closely related: energy storage):

- - chemical (Otto, Diesel, natural gas, methanol, ethanol, hydrogen...)
 - Storage:
 - Liquid fuel tank
 - Pressure tank
 - Hydride storage
- electrical/electrochemical
 - Storage
 - Battery
 - Super capacitor
- mechanical (pneumatic, hydraulic, kinetic)
 - Storage
 - Pressure tank
 - Flywheel

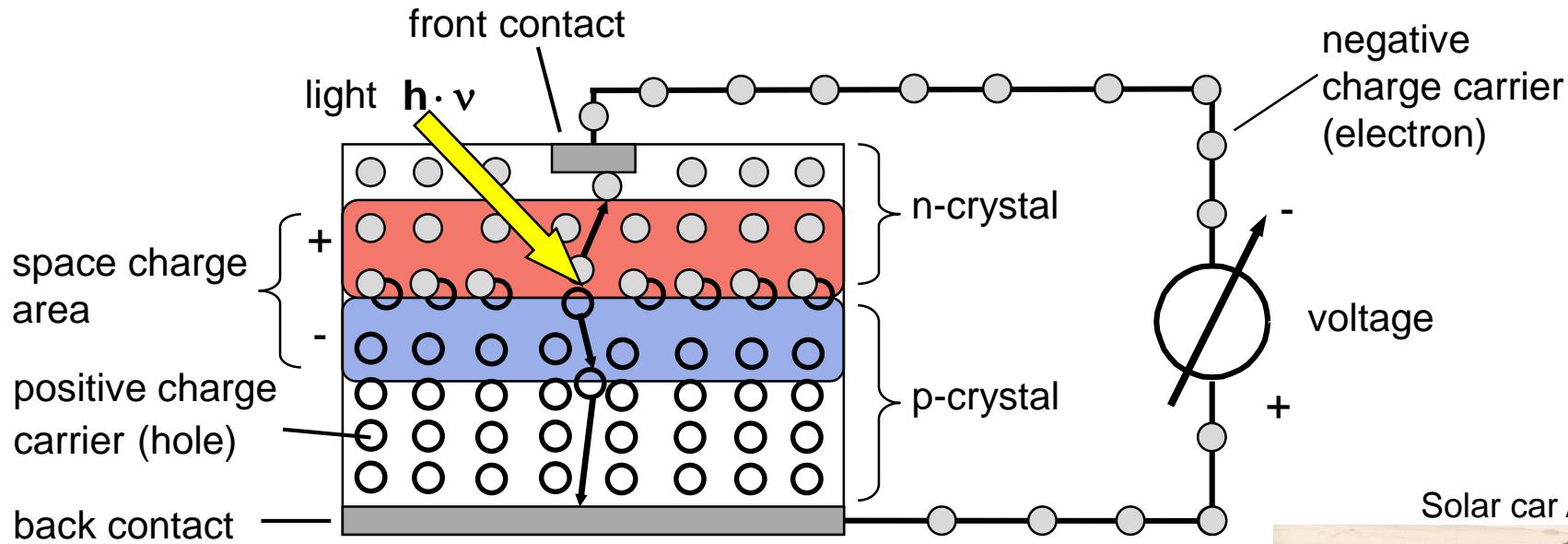
Classification of alternative vehicle propulsion systems by energy conversion process



Principles for fuel energy conversion



Photovoltaic cell



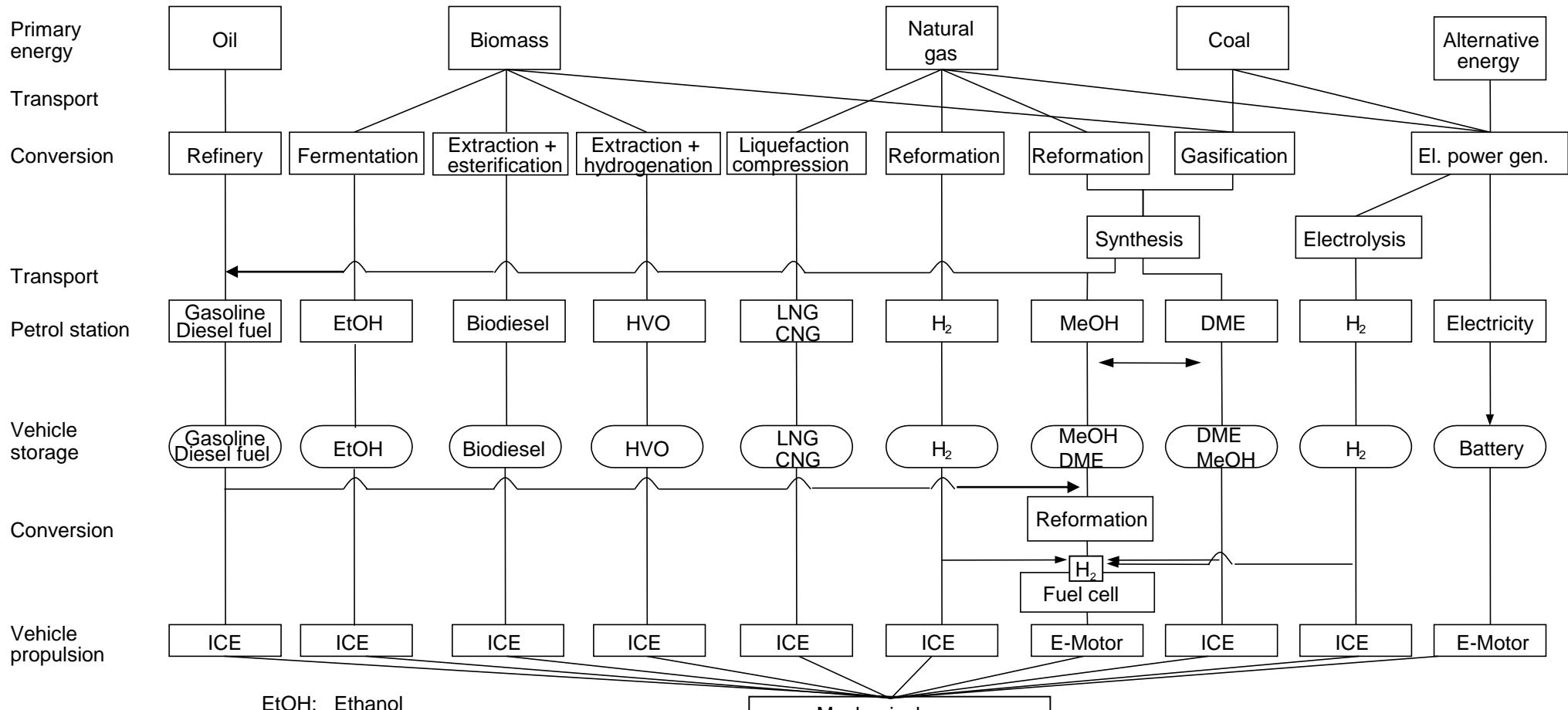
- At the p/n junction electrons diffuse into the p-crystal and holes into the n-crystal, whereby an area with less free charge carrier is generated (space charge area).
- Light enters the space charge area and produces free charge carrier (electrons/holes).
- Due to the electrical field, the charge carrier are separated and conducted to the contacts.
- A usable current is generated.



20 kg lithium ion cell
4m² PV area
135 V wheel hub motor (1.4 kW)
Weight = 200 kg
 $V_{Max} = 135$ km/h

Well-to-Wheel

New energy sources and propulsion systems for road traffic



Content of lecture

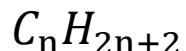
1. Introduction
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Properties of chemical energy sources

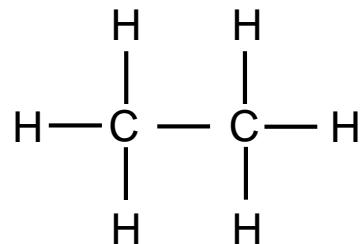
- Energy carriers
 - chemical
 - electrical / electrochemical
 - mechanical
- This chapter focuses on chemical energy sources
(storage density is much higher compared to other energy sources)
- Chemical energy sources
 - Almost only hydrocarbons (C, H, O, (N), S)
 - Requirements of energy carriers are lower for external combustion than for internal combustion

Hydrocarbon chain

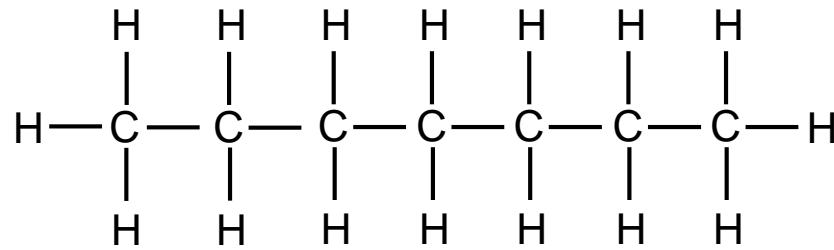
Paraffin (Alkane)



Normal Paraffins:

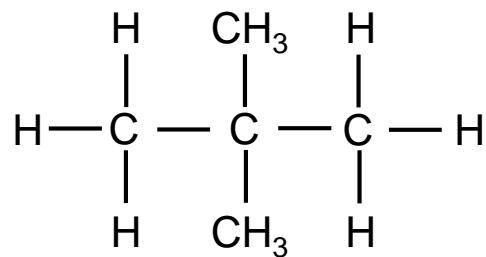


Ethane

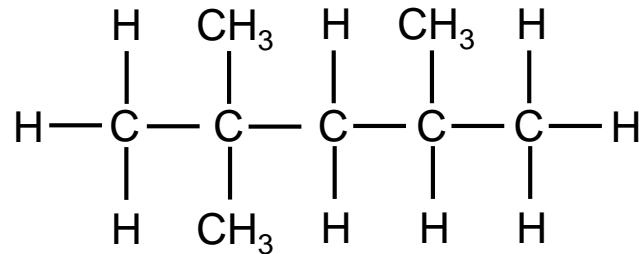


n-Heptane

Isoparaffins:



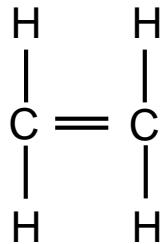
2,2 Dimethylpropane (Isopentane)



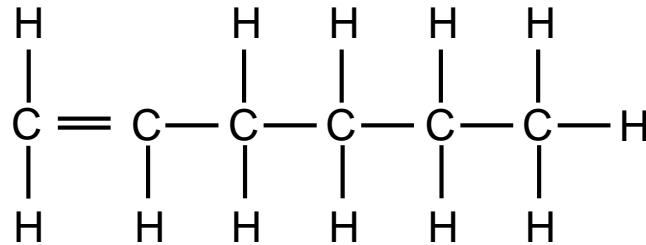
2,2,4 Trimethylpentane (Isooctane)

Hydrocarbon chain

Olefins

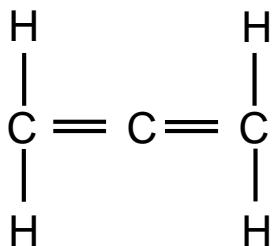


Ethylene

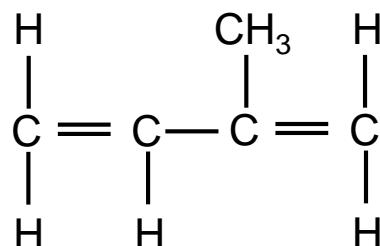


1-Hexen

Diolefins (Alkadienes)



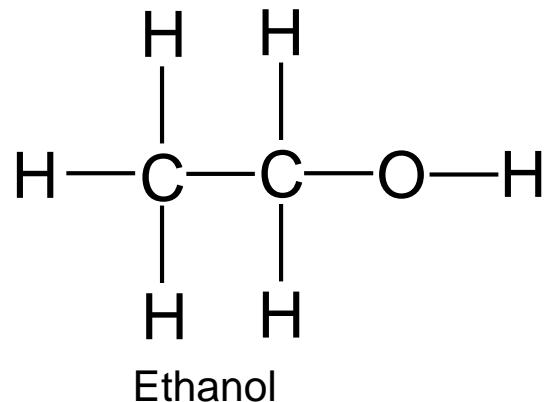
Propadiene (Allene)



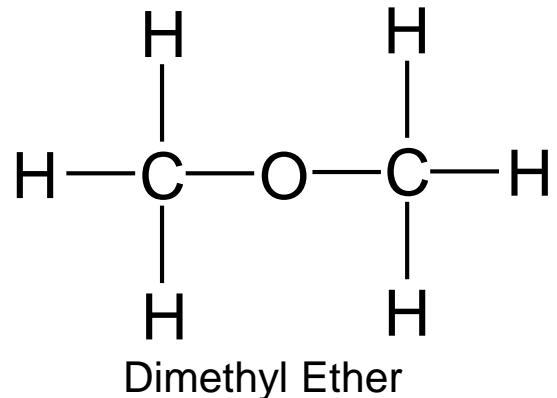
2-Methyl-1,3 Butadiene (Isoprene)

Hydrocarbon chain

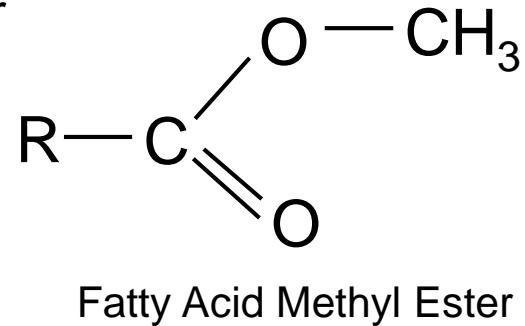
Alcohols



Ether

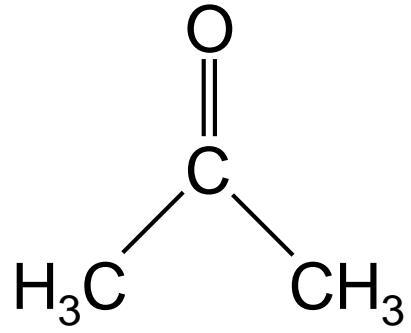


Ester

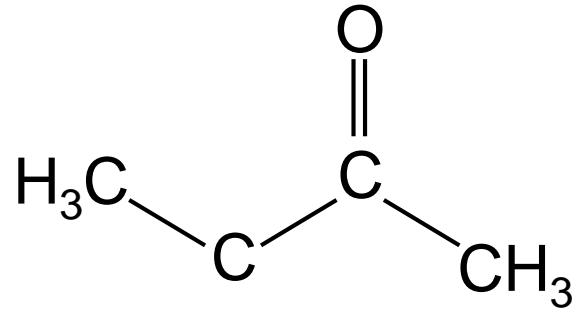


Hydrocarbon chain

Ketones



Acetone

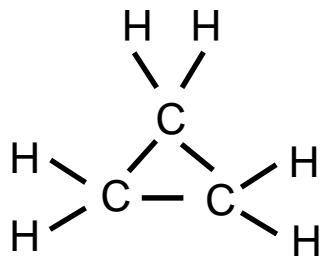


Butanone

Cyclohydrocarbons

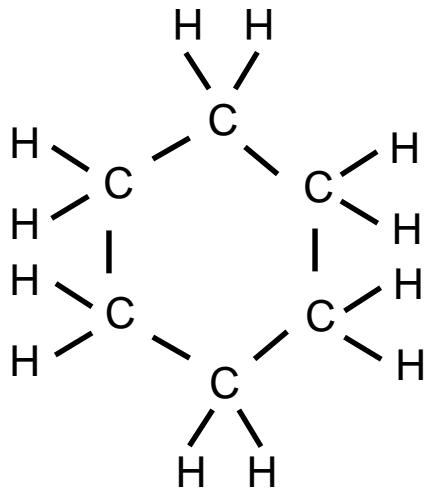
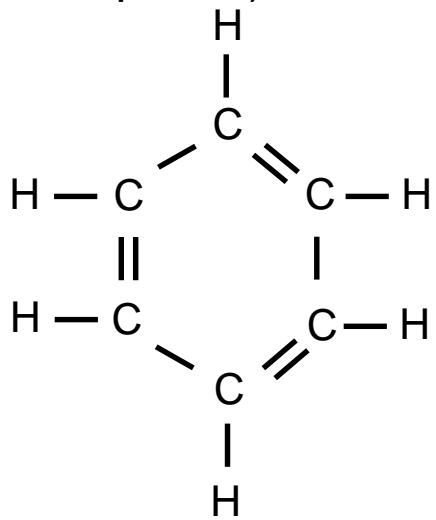
Naphthenes (Cycloalkenes)

Cyclopropane

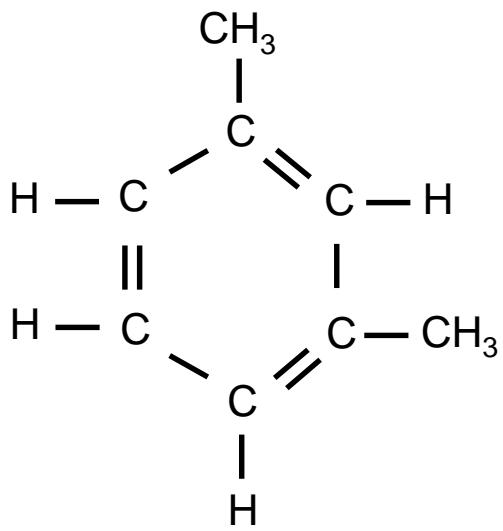


Benzene (aromatic compound)

Benzene (C_6H_6)



Cyclohexane

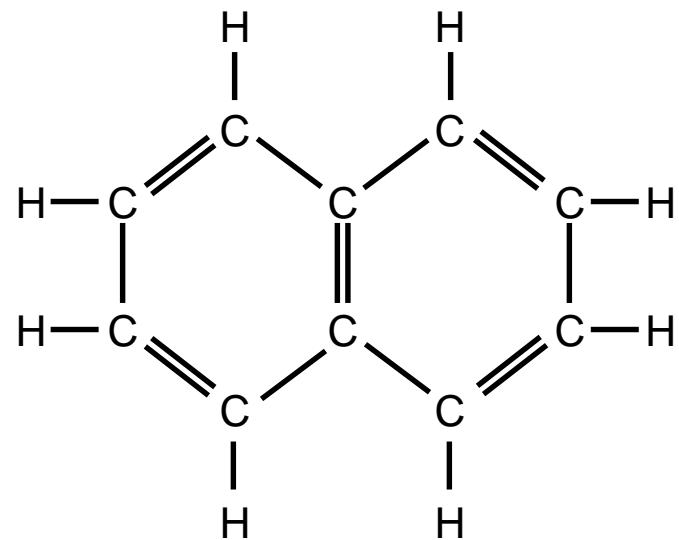


1,3 Dimethylbenzene
(*m*-Xylene)

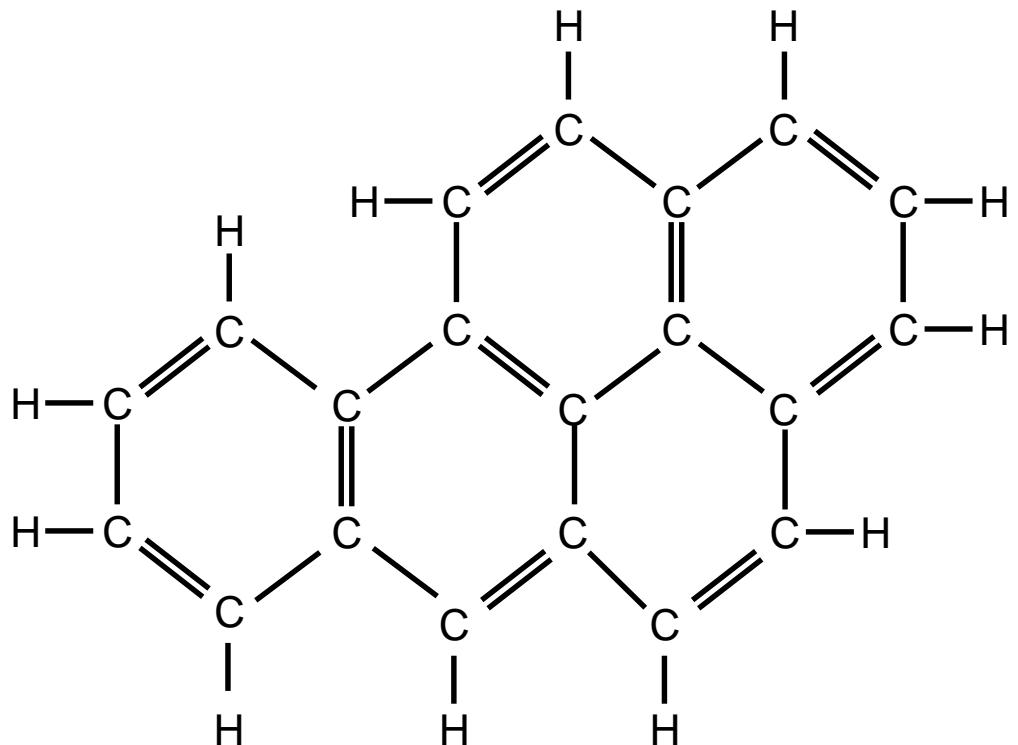
Polycyclic hydrocarbons

polycyclic compound

naphthalene



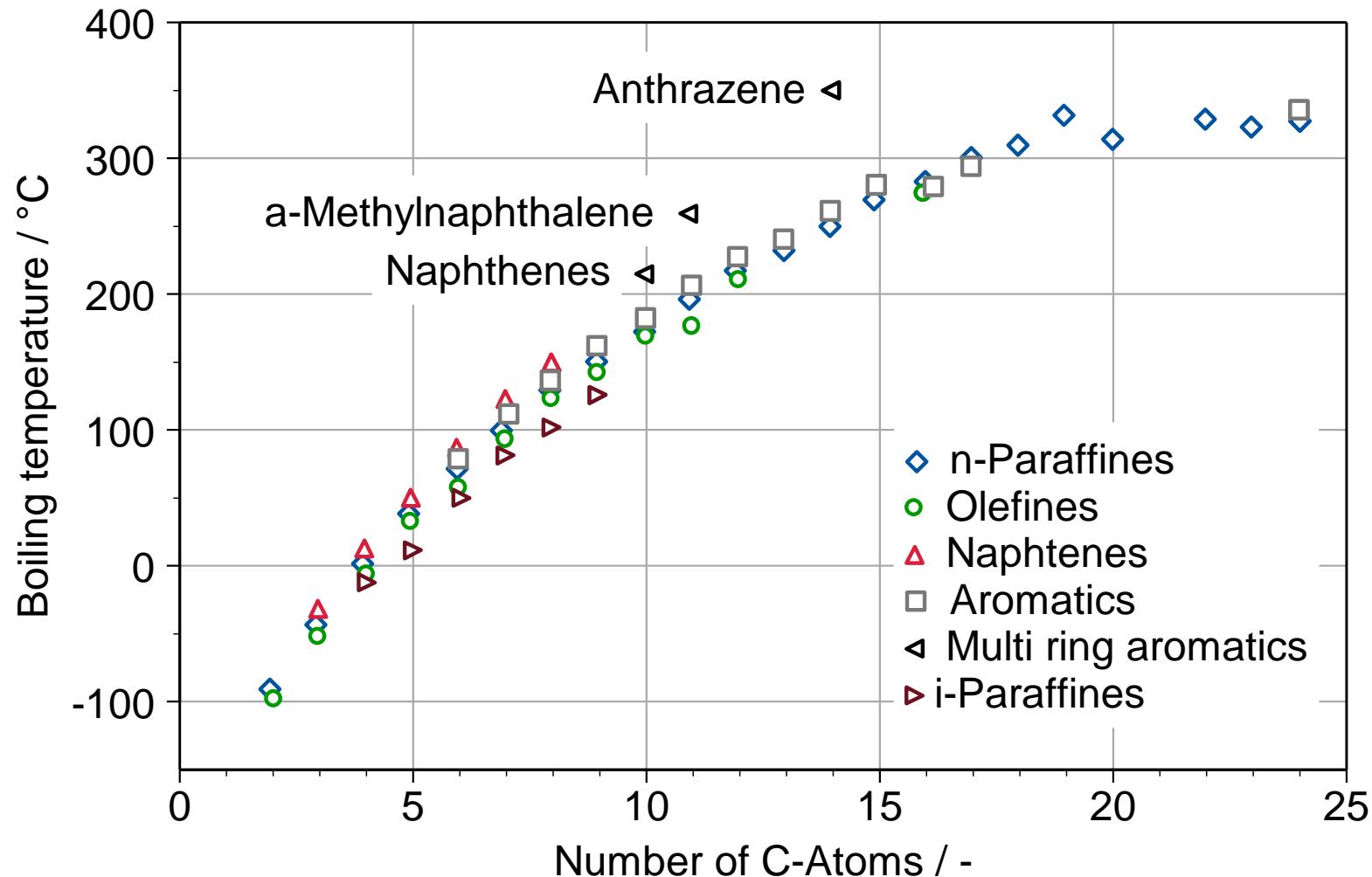
Benzo [a] pyrene



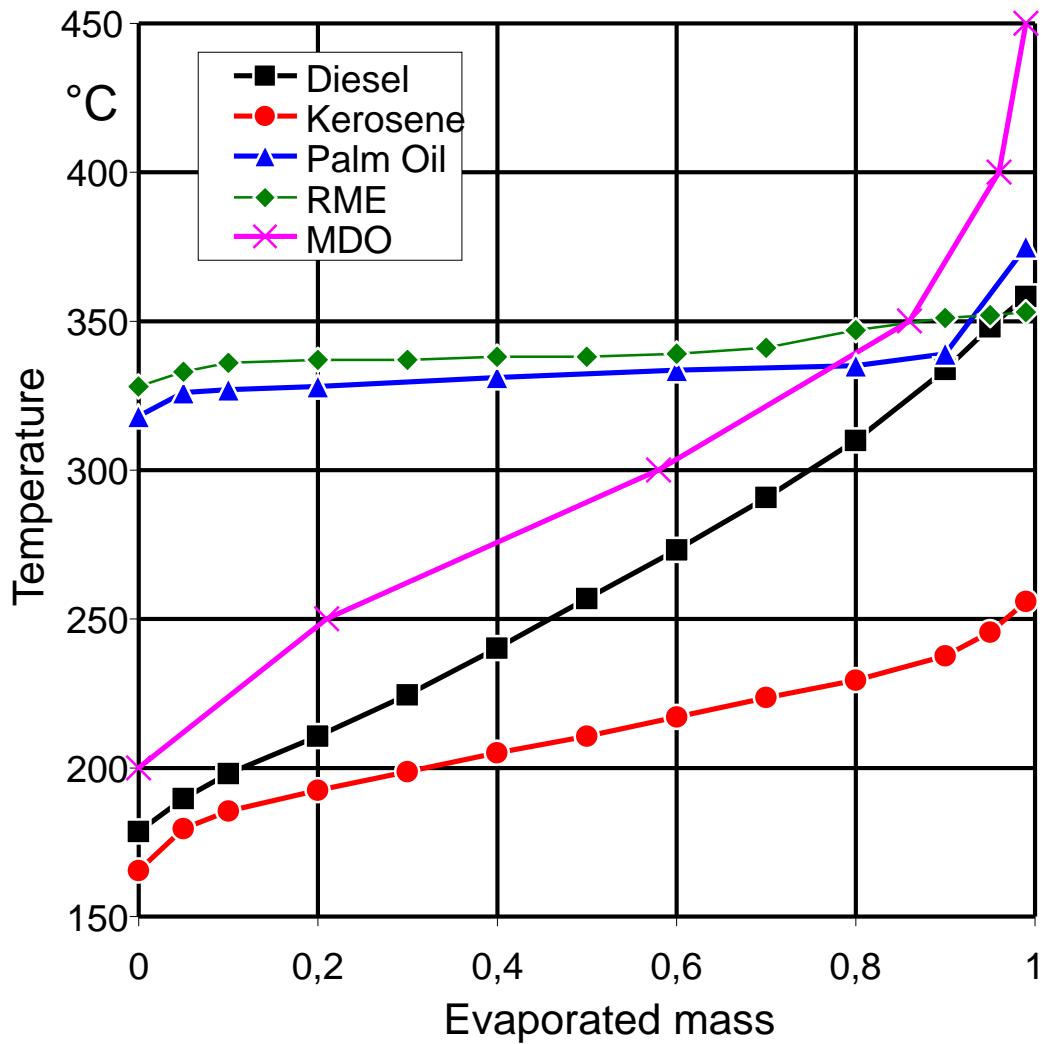
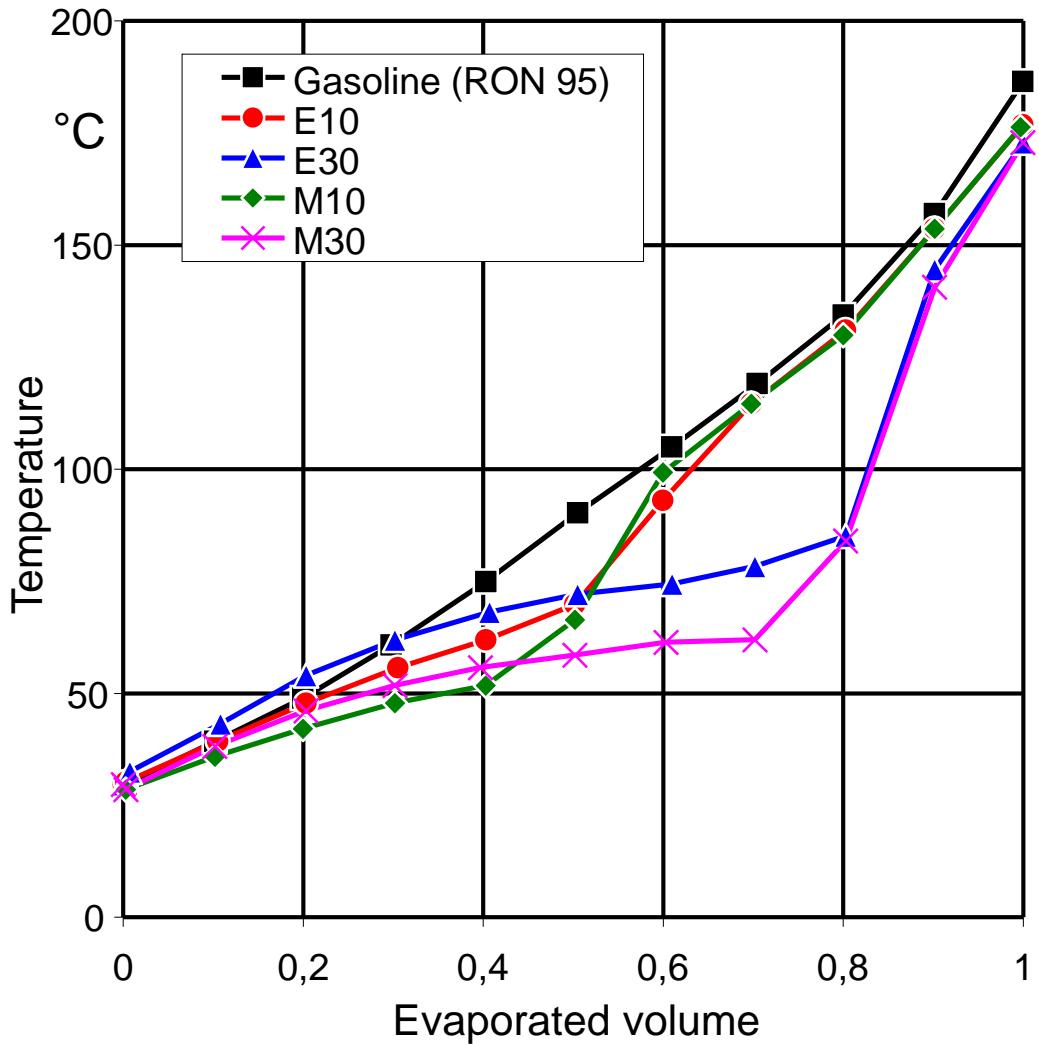
Chemical energy carriers: Main properties and relevance for vehicle propulsion

- Boiling temperature:
 - Increases with long chained molecules and is of importance for mixture formation.
 - Fuel mixtures have boiling curve
- Heat of vaporization / Vapor pressure:
 - Indicates amount of heat required for fuel vaporization; a high vaporization heat causes a high mixture cooling, thus naturally aspirated SI-engines with an exterior mixture formation will have a better cylinder charge
- Calorific value:
 - H_u , kJ/kg: Mass of stored fuel
- Stoichiometric air requirement L_{St} :
 - Indicates amount of air required for stoichiometric combustion: $L_{St} = m_{LSt}/m_B$
- Energy specific carbon content:
 - c/H_u : CO₂-Emission based on fuel energy
 - H₂ emits no CO₂
 - CH₄ has a CO₂ advantage compared to gasoline of about 20 %
- Density / State of aggregation:
 - Density increases with number of C, higher density → less tank volume
- Volume specific calorific value → tank volume

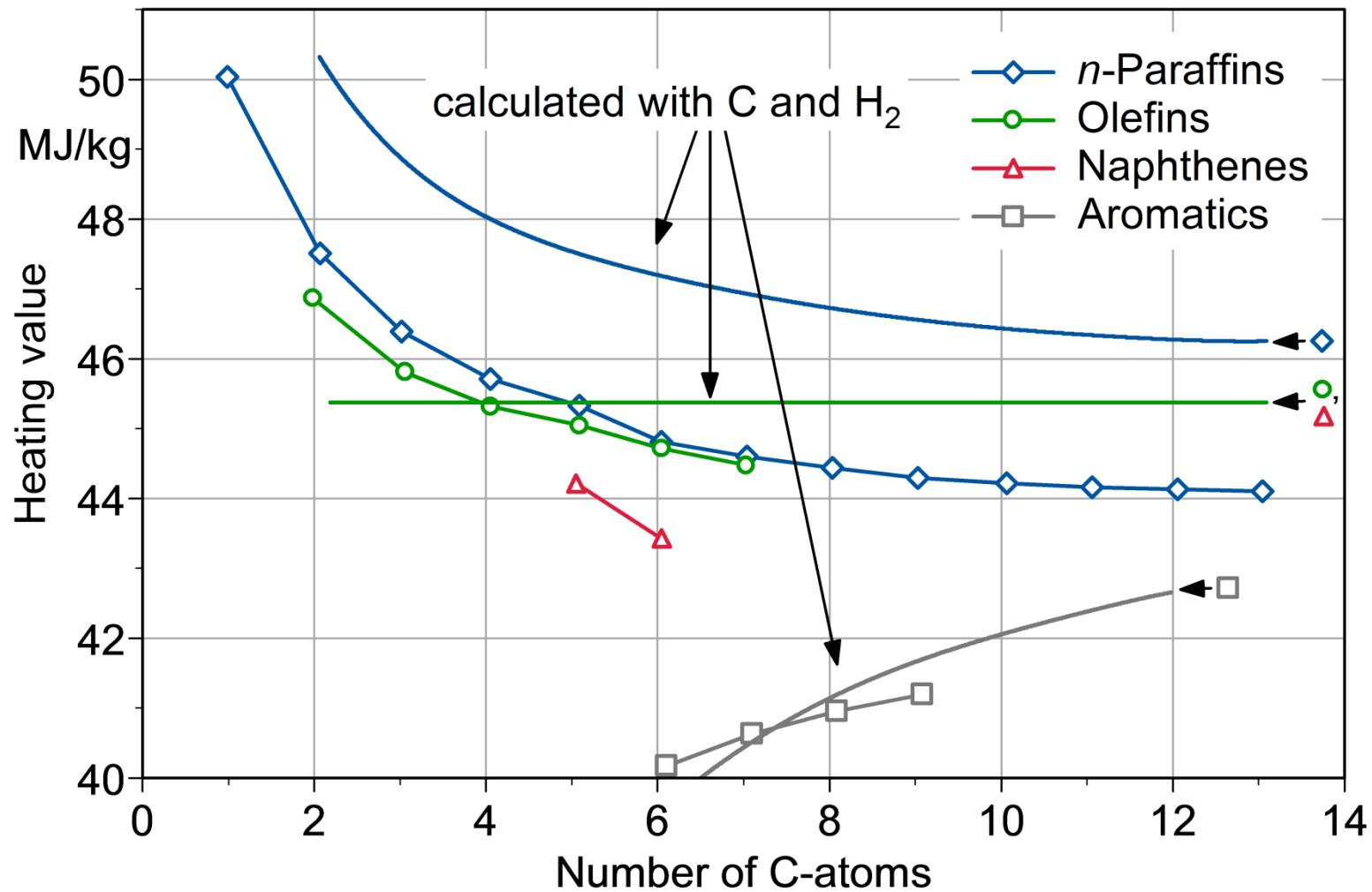
Boiling Temperature of HC Compounds



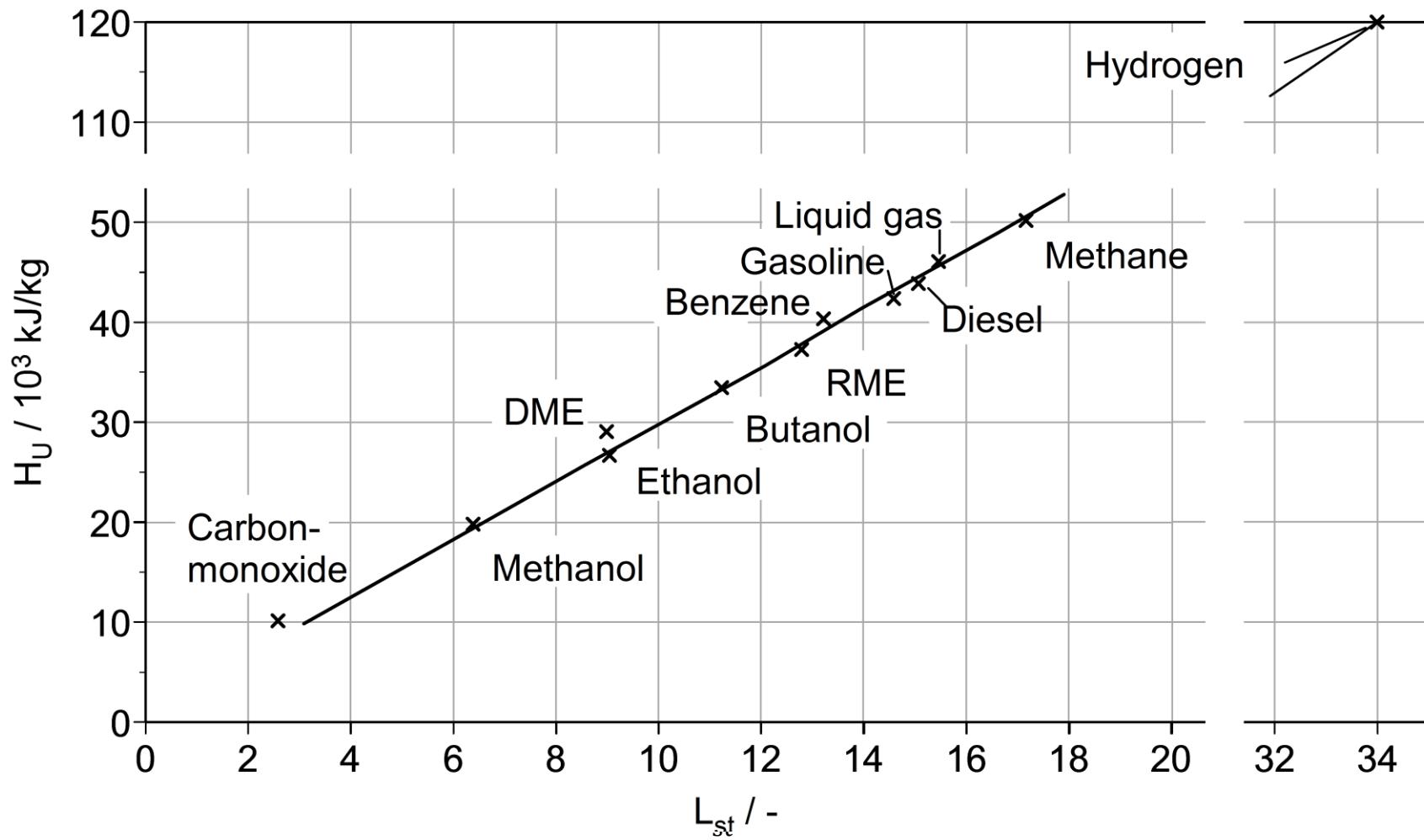
Boiling Curves of Alternative Fuels



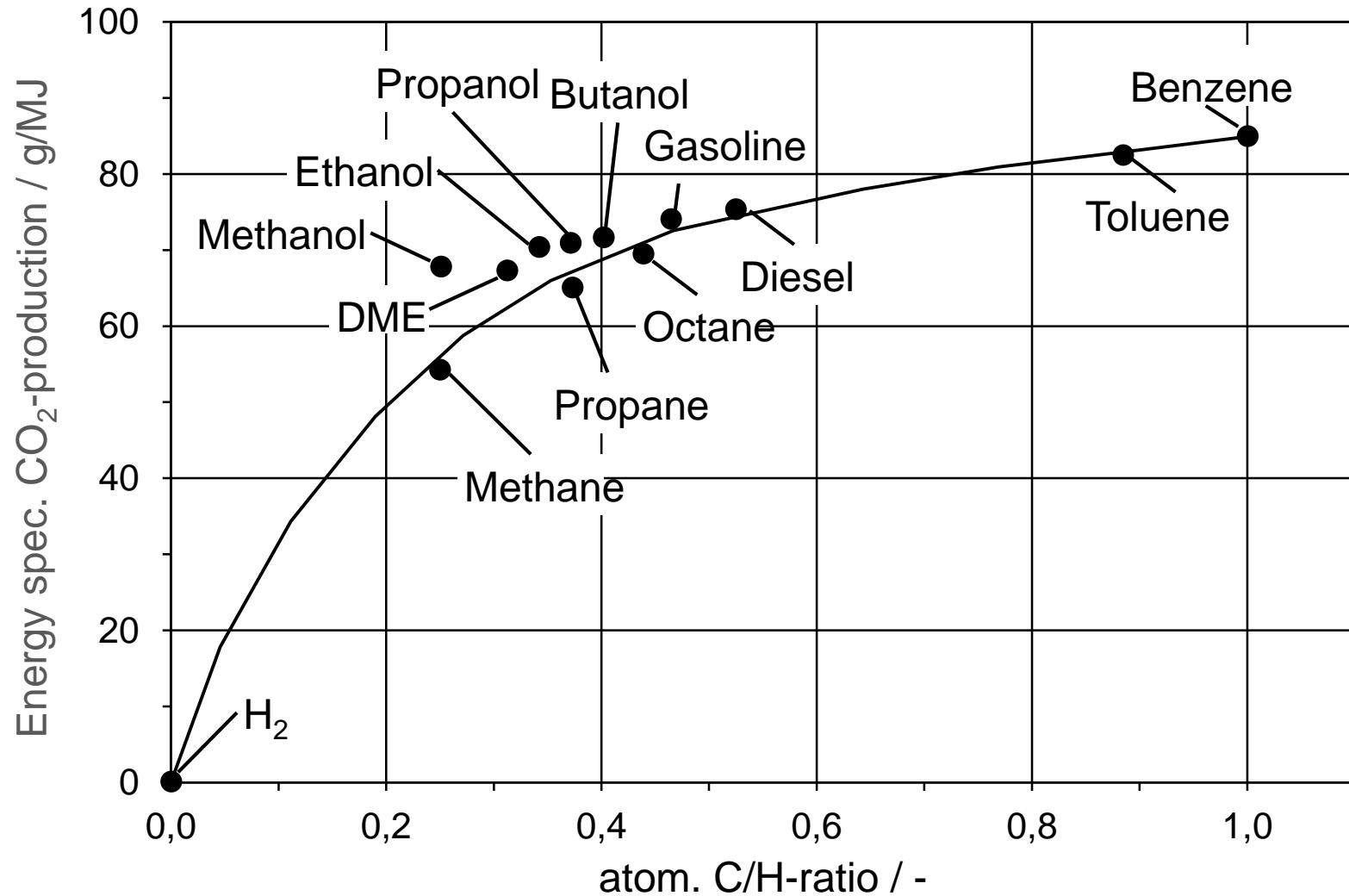
Heating value of HC compounds



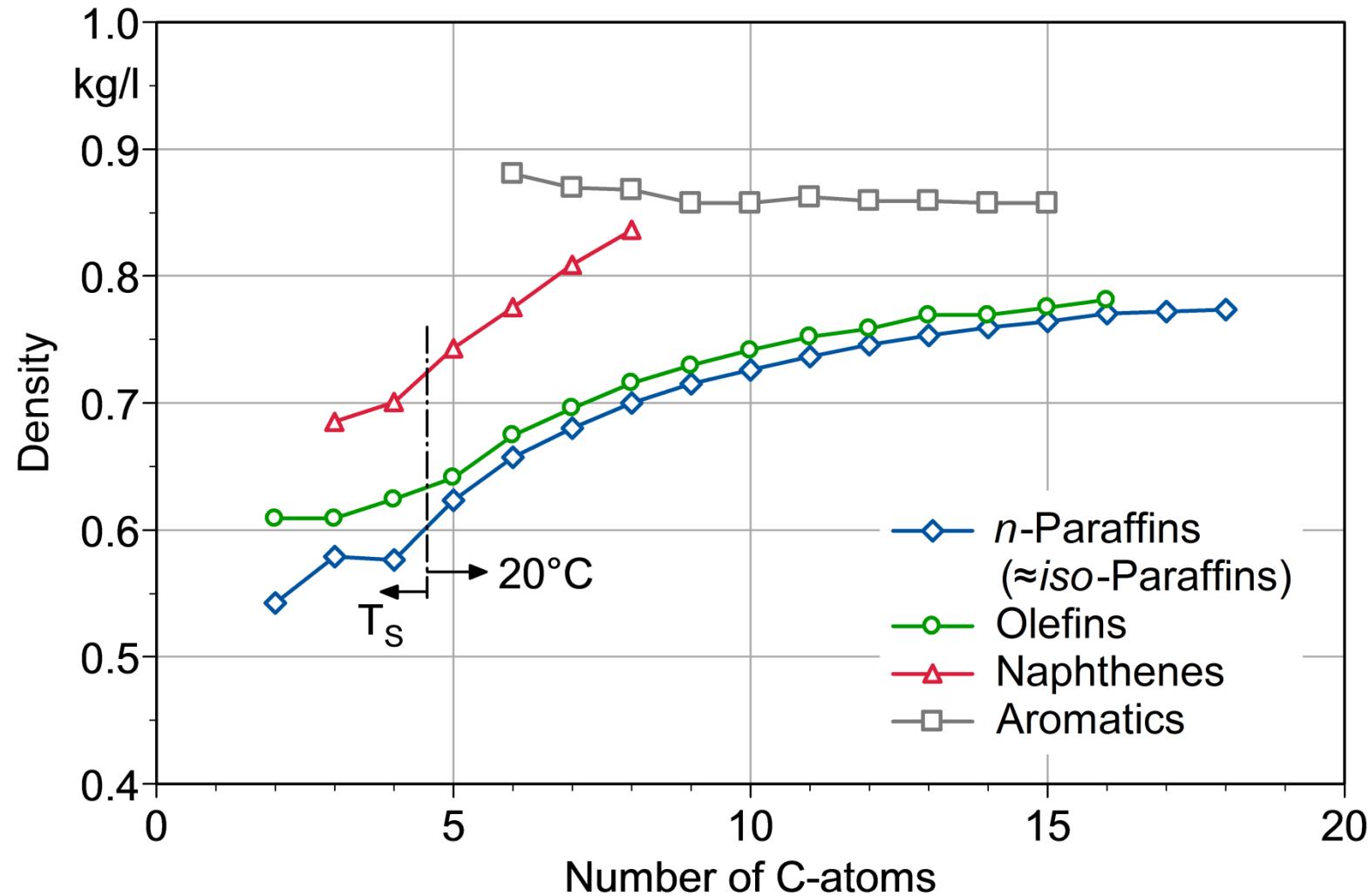
Features of chemical energy sources



Energy specific CO₂ emissions of fuels



Density of HC Compounds



Main properties and significance for vehicle propulsion

- Volume-related calorific value
 - Tank volume
- Ignitability / Antiknock property:
 - Octane number ON (n-Heptane / iso-Oktane)
 - antiknock property (Otto cycle) RON > 95 (Eurosuper, EN)
 - Cetane number CN (a-Methylnaphtalin / Cetane)
 - ignition performance (Diesel cycle) CN > 51 (Diesel, EN)
 - Methane number MN (Hydrogen / Methane)
 - antiknock property for gaseous fuels (range up to ON > 100, ON of Methane approx. 140)
- Soot formation:
 - decreases with oxygen content
- Burning velocity, Ignition limits:
 - high burning velocity
 - wide ignition limits

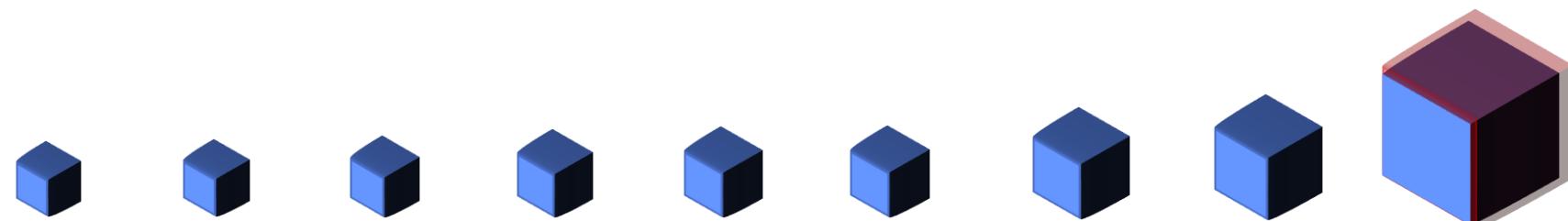
→ advantageous for homogeneous lean operation (low NO_x combustion with high efficiency)

Size of storage systems for identical vehicle range

Mass and volume for identical range (1000 km):

(* including storage system)

54 kg*	61 kg*	65 kg*	72 kg*	95 kg*	93 kg*	140 kg*	164 kg*	943 kg*
39 kg	45 kg	45 kg	41 kg	68 kg	58 kg	8 kg	37 kg	736 kg



Energy consumption NEDC in MJ / 100 km :

166	166	191	190	181	166	95	184	53
-----	-----	-----	-----	-----	-----	----	-----	----

Diesel

RME

Gasoline
(Petro)

LPG

Ethanol

DME

Hydrogen
liquid for PEM
(H₂ 700bar)

Natural Gas
(CNG 200 bar)

Battery
Li-Ion

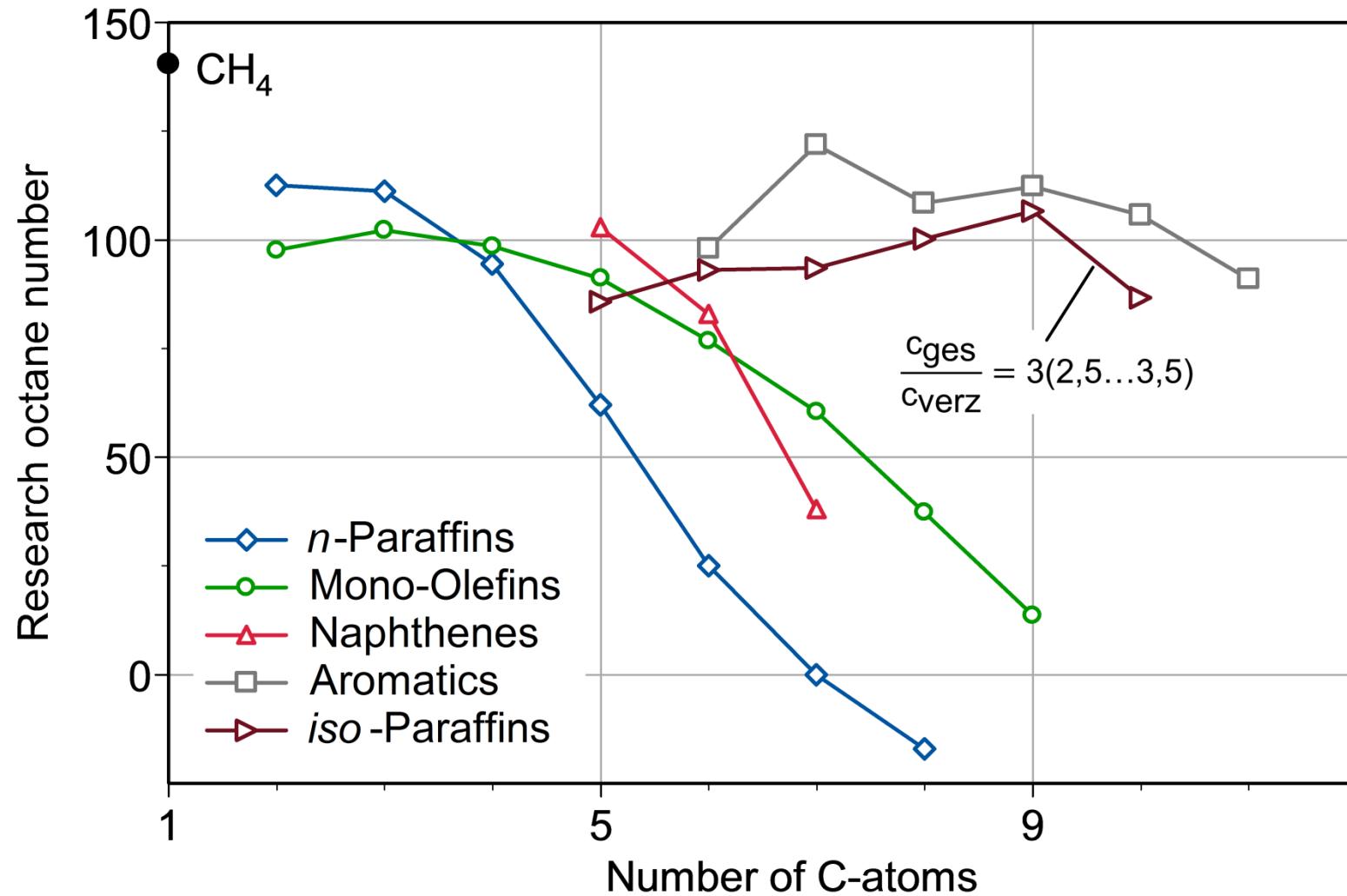
DME – dimethyl ether

RME – rapeseed methyl ester

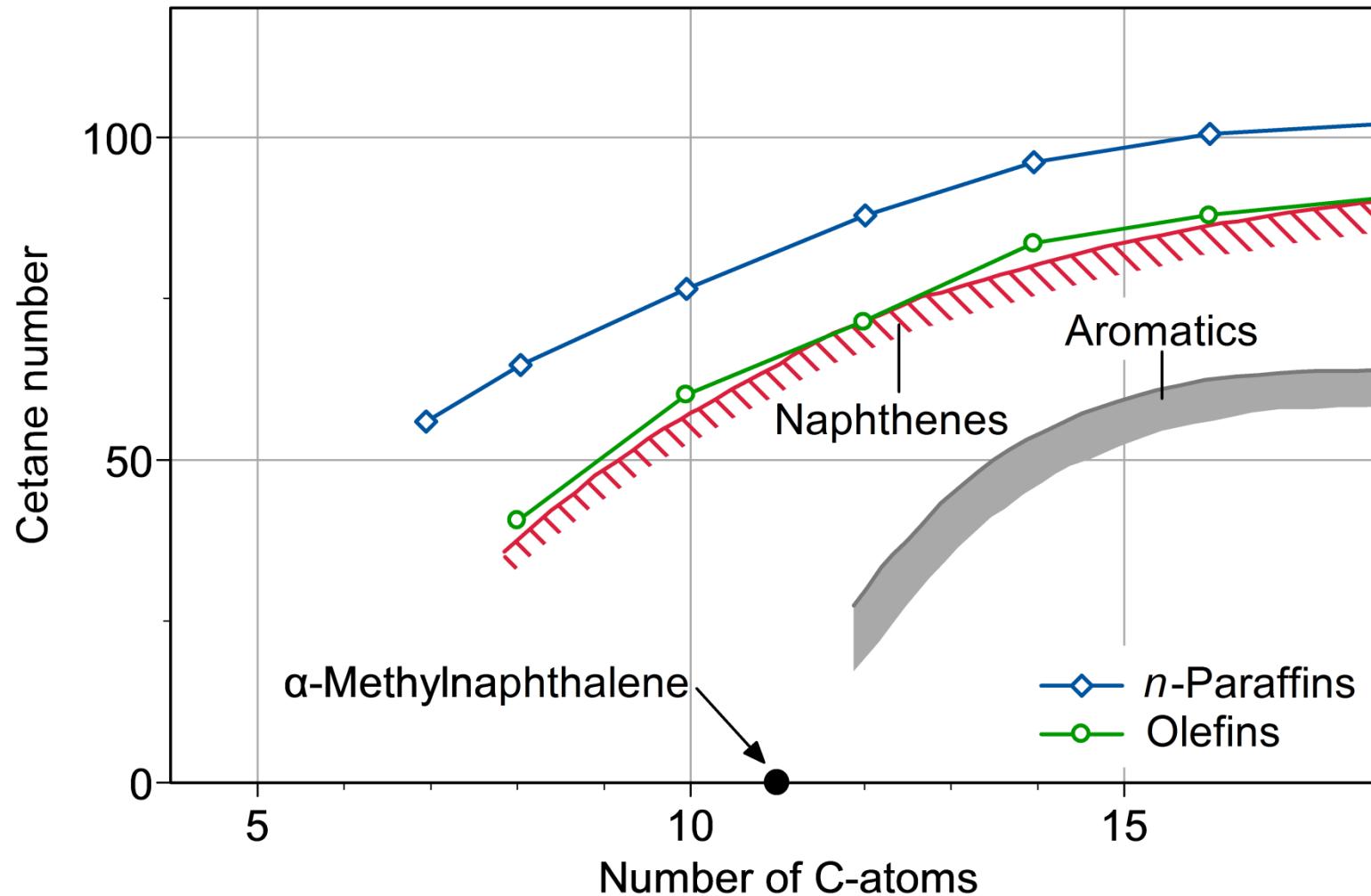
LPG – liquefied Petroleum Gas

PEM – polymer electrolyte fuel cells

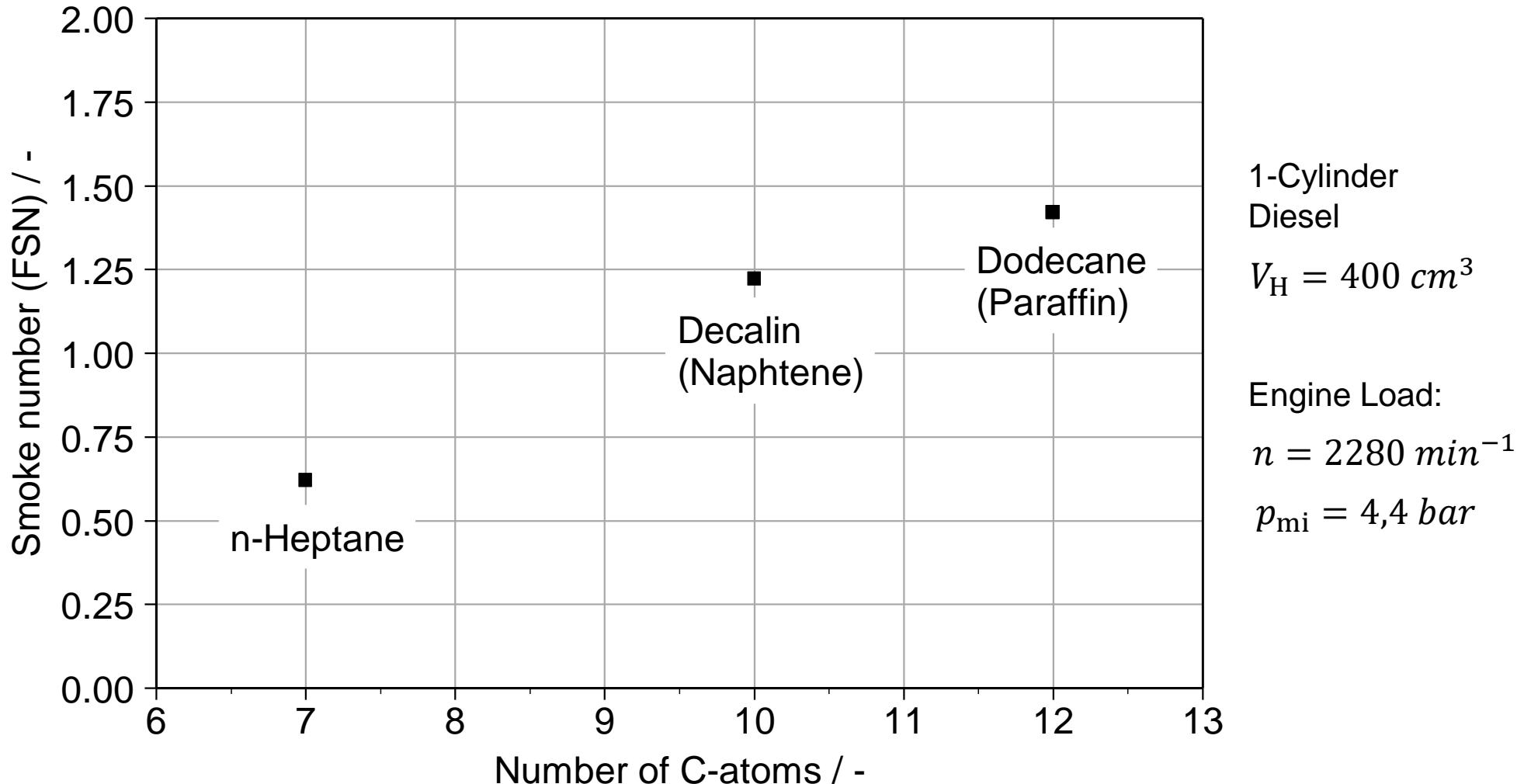
RON of HC Compounds



Cetane Number of HC compounds



Influence of different fuels on soot emissions

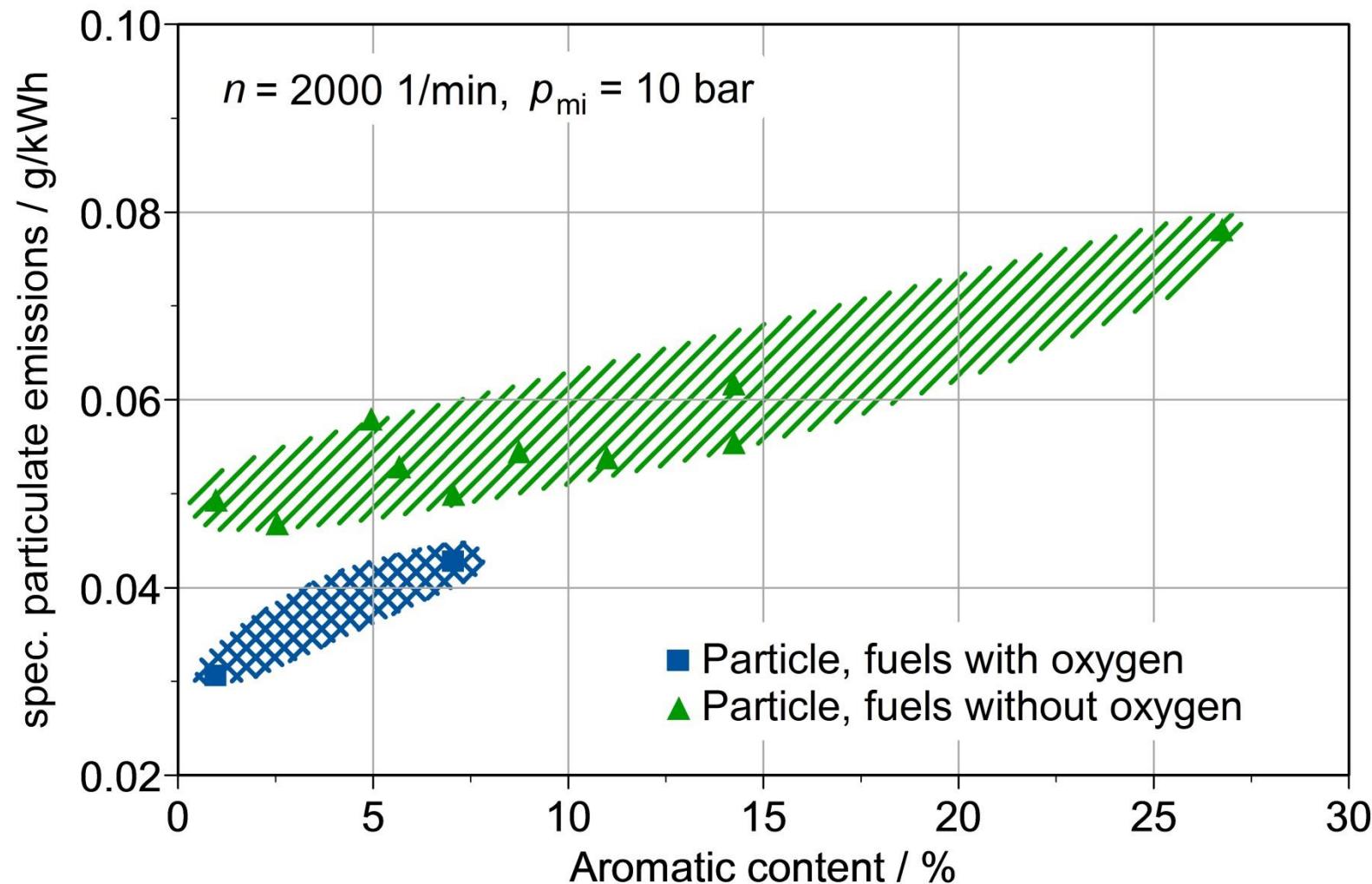


Influence of the molecular structure of fuels on the tendency to soot formation

Alkanes	Alkenes	Alcohols	Ethers	Ketones	Esters	Aromatics
						Cyclic, C-C double bond
Butane C_4H_{10}	Butene C_4H_8	Butanol $\text{C}_4\text{H}_{10}\text{O}$	Diethyl ether $\text{C}_4\text{H}_{10}\text{O}$	Butanone $\text{C}_4\text{H}_8\text{O}$	Ethyl acetate $\text{C}_4\text{H}_8\text{O}_2$	Benzene
Moderate sooting tendency	High sooting tendency due to C-C double bond	Low sooting tendency due to oxygen in functional group				Very high sooting tendency, because soot precursor

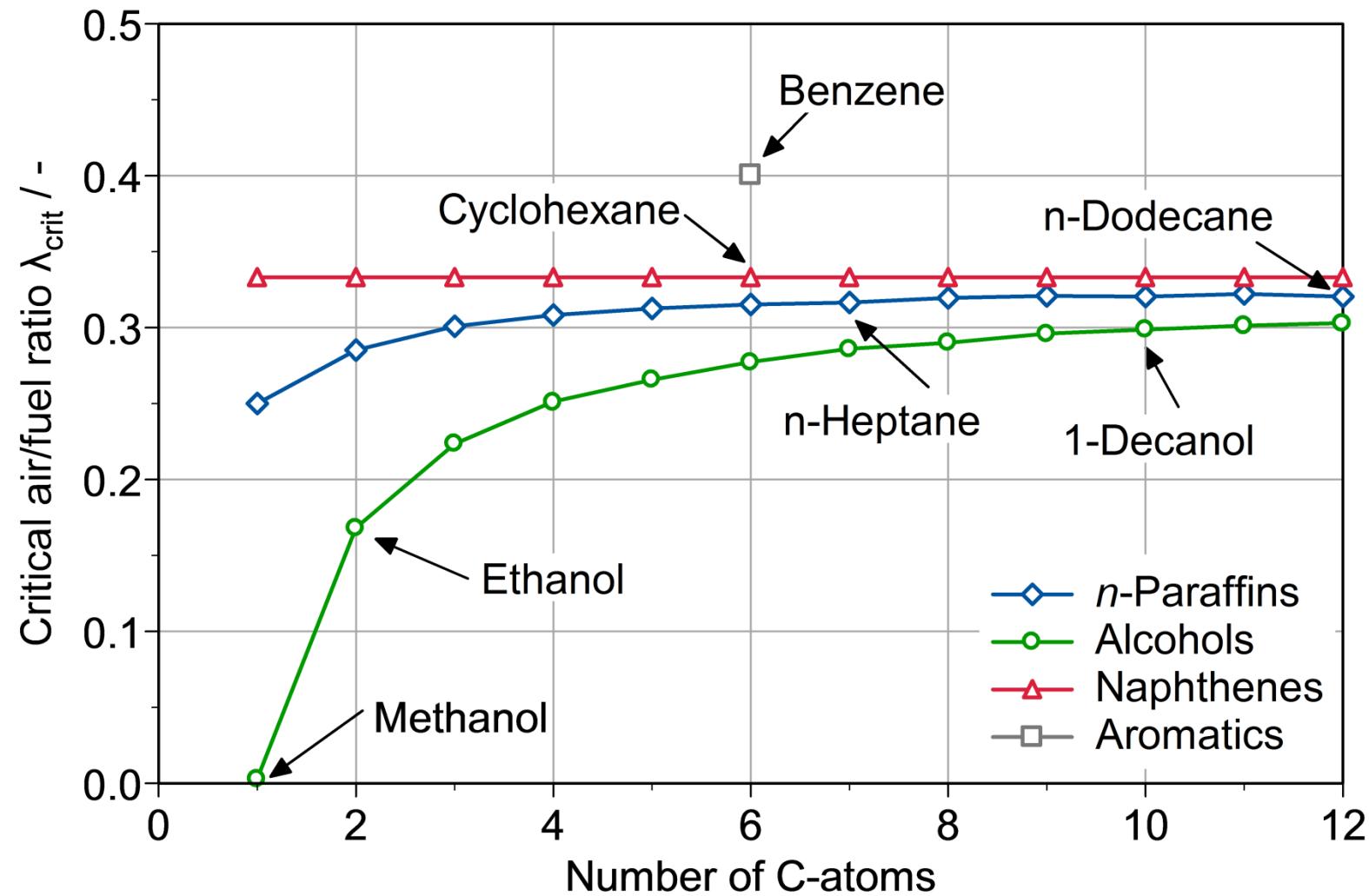
R = organic residual

Influence of different fuels on particulate emissions

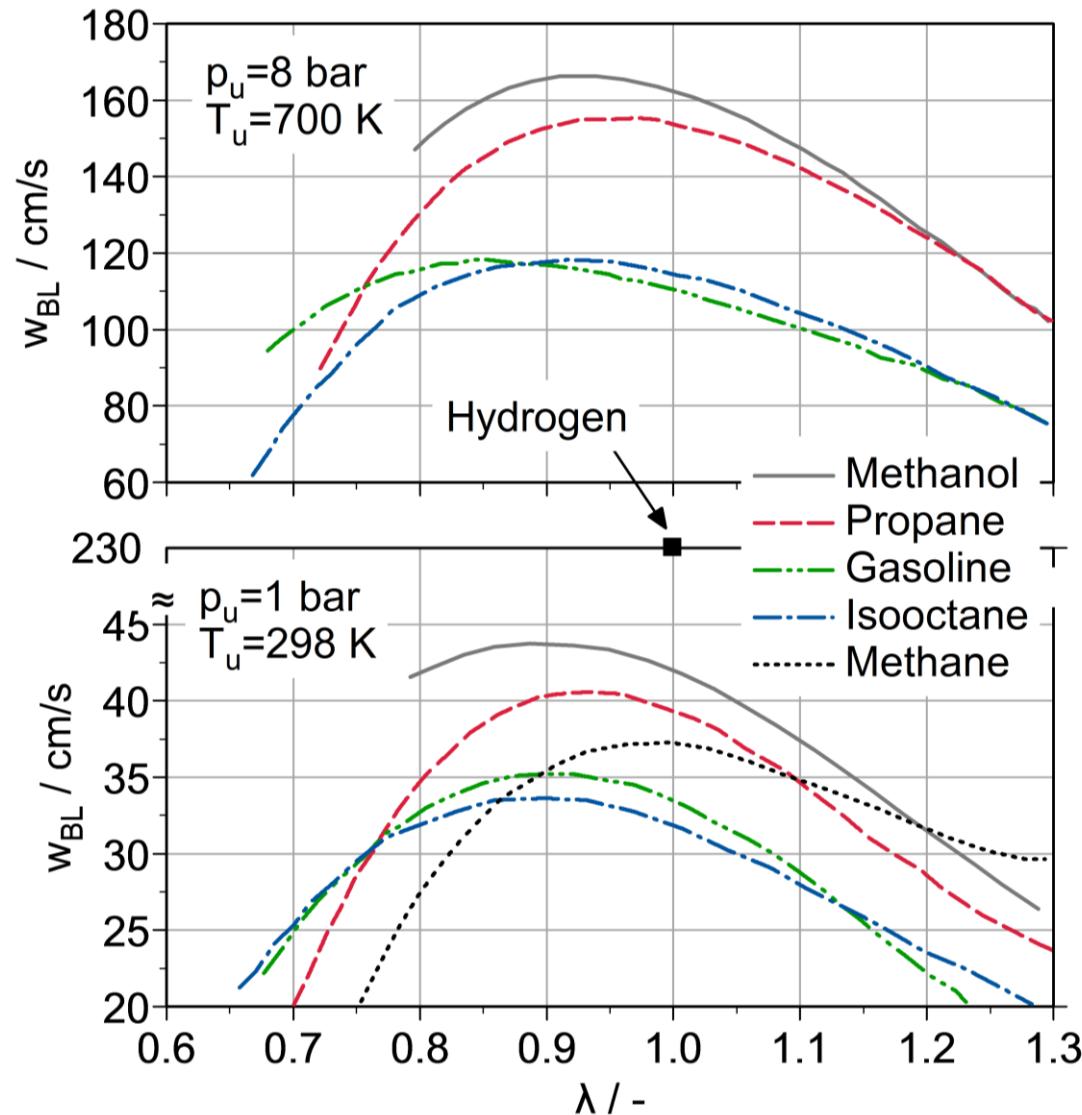


Source: Dissertation Pungs, VKA 2001

Critical Air-Fuel Ratio λ_{crit}



Laminar burning velocity w_{BL} for different fuels



Source: Metghalchi, Keck

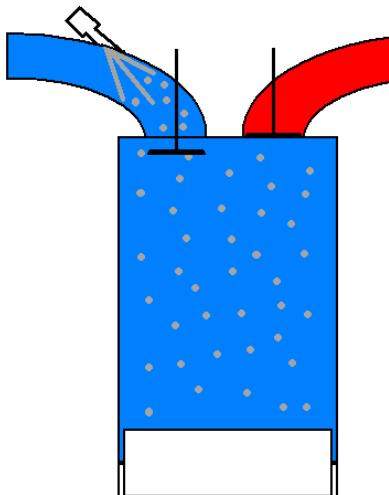
Engine ignition limits for different fuels (selection)

Fuel	Ignition limits
Gasoline	$0.4 < \lambda < 1.4$ (1.7)
Hydrogen	$0.15 < \lambda < 10.5$
Natural gas (CNG)	$0.7 < \lambda < 2.1$
Methanol	$0.34 < \lambda < 2.0$
Ethanol	$0.3 < \lambda < 2.1$
LPG	$0.4 < \lambda < 1.7$

Chemical energy sources: Main properties and significance for vehicle propulsion

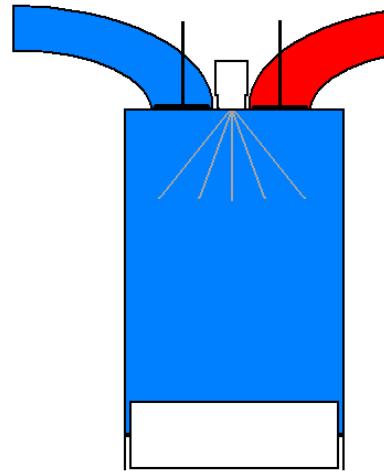
- Mixture calorific value: Important for internal combustion, indicates displacement based engine power.

external mixture formation



$$H_G = \frac{m_B * H_u}{V_G} = \frac{\rho_G * H_u}{\lambda * L_{St} + 1}$$

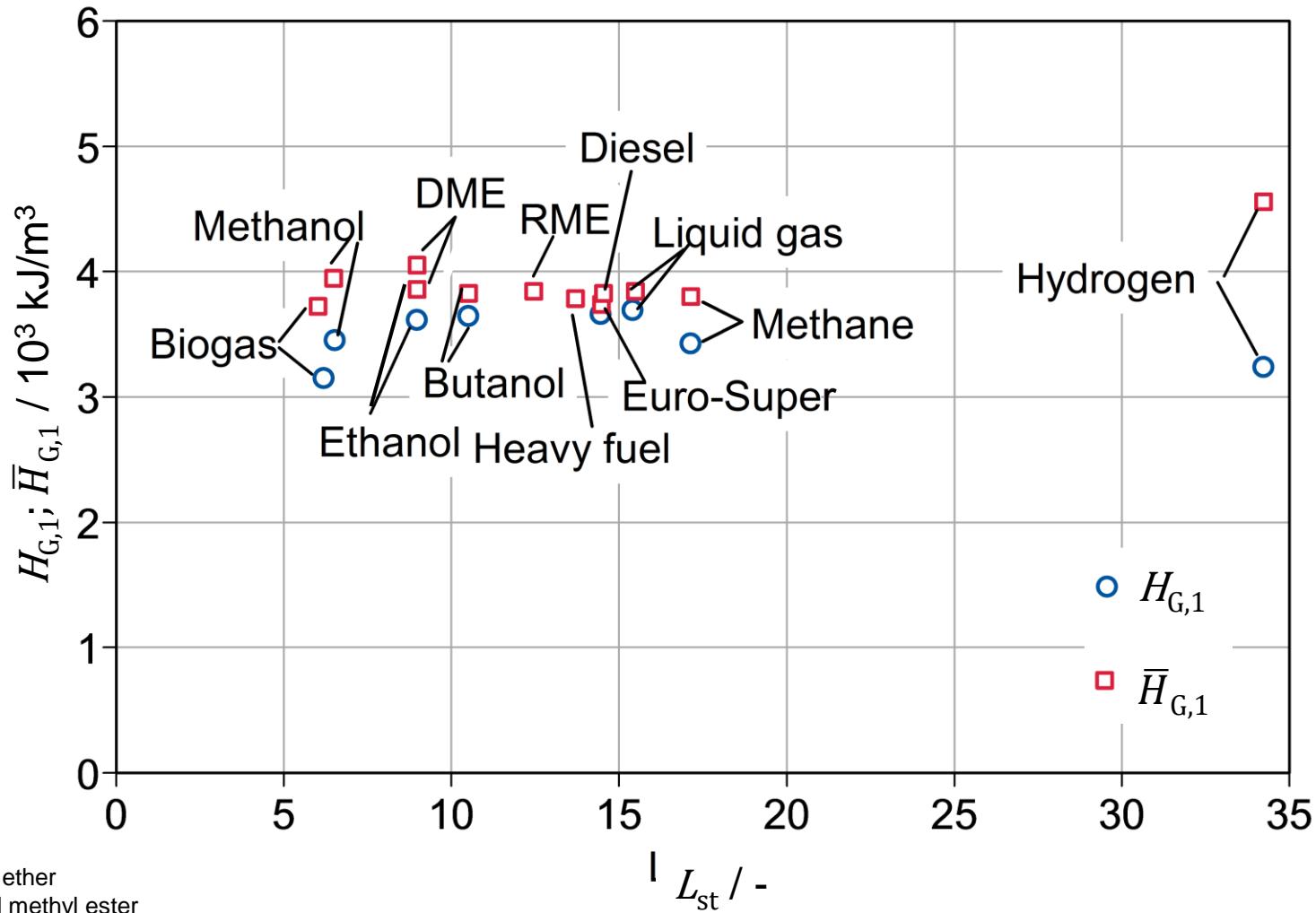
internal mixture formation



$$\bar{H}_G = \frac{m_B * H_u}{V_L} = \frac{\rho_L * H_u}{\lambda * L_{St}}$$

- Viscosity: Suitability for high pressure injection (lubrication)

Chemical energy carriers and characteristics

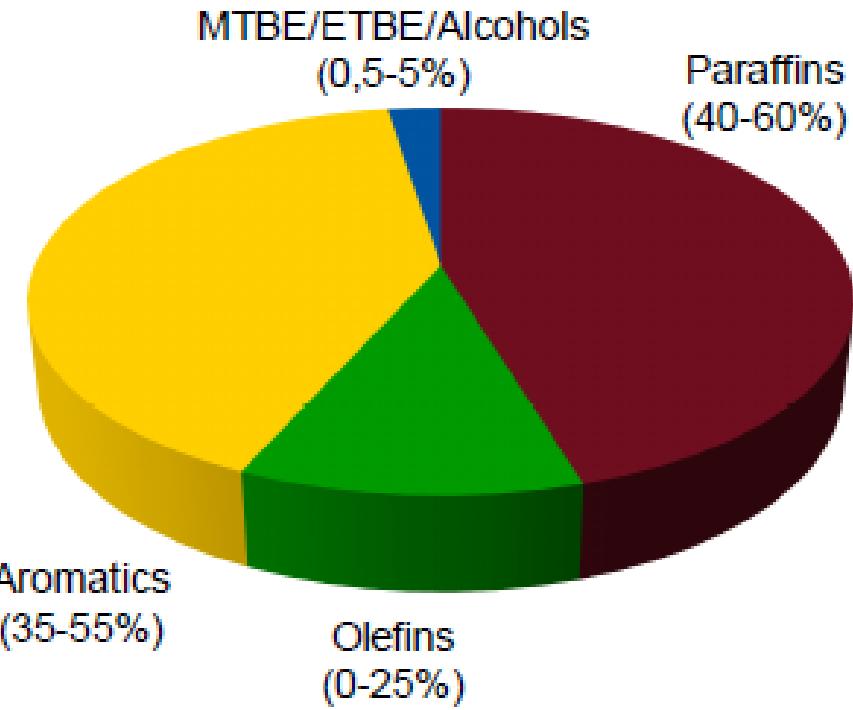


DME – dimethyl ether

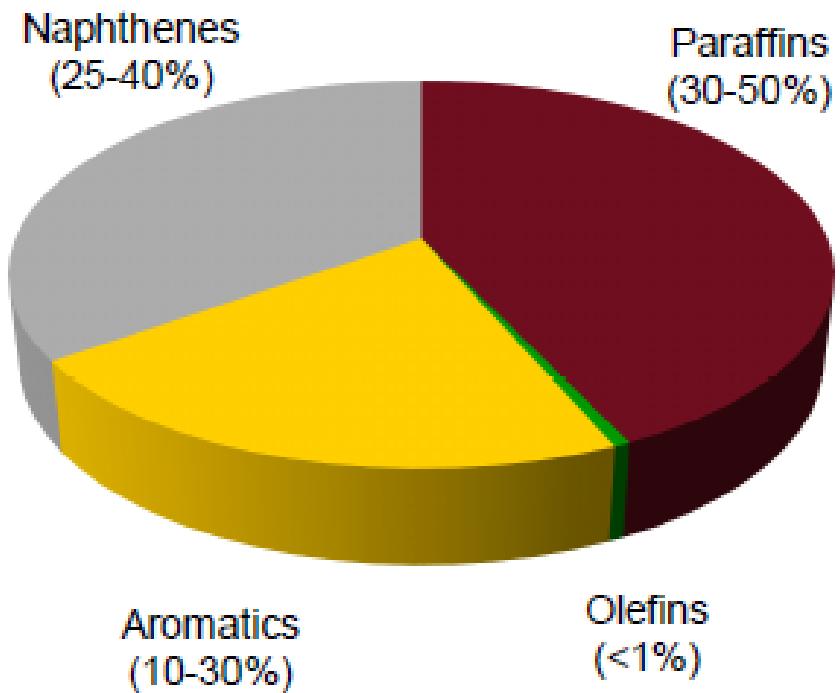
RME – Rapeseed methyl ester

Chemical energy carriers and characteristics

Gasoline (Super unleaded)



Diesel



Source: ARAL, 2010

Characteristic values of gasoline fuel

- Components C_4 to C_{12}
- $h; c: 0,14; 0,84$
- $n_H / n_C = 2,12$
- Mean molar mass $M_B \approx 98 \text{ kg/kmol}$
- Flash point $\approx -40 \text{ }^{\circ}\text{C}$ ($<-20 \text{ }^{\circ}\text{C}$)
- Hazard category A I
- Permittivity $\approx 2,0$
- Heating value $\approx 42000 \text{ kJ/kg}$
- Density ($15 \text{ }^{\circ}\text{C}$) $\approx 0,760 \text{ kg/l}$
- Heat of evaporation $\approx 420 \text{ kJ/kg}$
- Adiab. cooling ($\lambda = 1$) $20 \text{ }^{\circ}\text{C}$
- Kinem. viscosity ($20 \text{ }^{\circ}\text{C}$) $\approx 0,7 \text{ cSt}$
- Heating value of mixture H_{G1} ($20 \text{ }^{\circ}\text{C}$) 3670 kJ/m^3
- Stoichiom. air demand $L_{St} \approx 14,5 \text{ kg/kg}$
- Ignition limits in air $\approx 1,0$ to 8 Vol. %
- Ignition limits in engine $\lambda \approx 0,4$ to 1,4
- Ignition temperature $220 \dots 300 \dots 450 \text{ }^{\circ}\text{C}$

Characteristics of Diesel fuel

- $h : c = 0,137 : 0,863$
- $n_H / n_C \approx 1,90$
- Mean molar mass $M_B \approx 190 \text{ kg / kmol}$
- Boiling temperature $\approx 170\text{-}350^\circ\text{C}$
- Electrical conductivity $< 10^{-15} \text{ S/cm}$
- Heating value $H_u \approx 42800 \text{ kJ/kg}$
- Volumetric heating value $\approx 35950 \text{ kJ/l}$
- Density $\approx 0,840 \text{ kg/l}$
- Aromatic content $\approx 26 \text{ Vol -\%}$
- Evaporation heat $\approx 300 \text{ kJ/kg}$
- Mixture heating value $H_{G1} \approx 3790 \text{ kJ/m}^3$
- Stoichiometric air demand $L_{st} \approx 14,6$
- Ignition temperature $\approx 230^\circ\text{C}$

Fuel properties I

	Unit	Euro-Super	Diesel	Crude-oil	Methanol	Ethanol	RME	LPG ¹⁾	Methane	Biogas ²⁾	Hydrogen	DME
H_u	MJ/kg MJ/l ⁴⁾	42 31.7	42.8 35.6	40.3 39.5	19.7 15.5	26.8 21.2	37.2 32.8	45.33 23.64	50.0 0.036	20.0 0.021	120 8.52	28.4 18.74
$H_{G,1}$ ³⁾ $H_{G,1}$	MJ/m ³ MJ/m ³	3.67 3.74		3.78	3.44 3.92	3.60 3.85		3.66 3.76	3.40 3.76	3.15 3.71	3.19 4.54	3.81 4.08
Ignition limit λ	-	0.4 1.4	0.48 1.35	0.5 1.35	0.34 2.0	0.3 2.1		0.4 1.7	0.7 2.1	0.7 2.3	0.5 10.5	
λ_{crit}	-	0.33	0.34	0.36	0	0.17	0.32	0.30	0.25	-	0	0.17
RON MON	-	95 85			114.4 94.6	111.4 94.0		94.5	140			
MN	-							28	100	125	0	
CN	-		min 51	20...45			54...58					55...60
EN	-	228	590				14214	589				

1) Liquefied Petroleum Gas (typical winter fuel in EU: 70 Mass % Propane, 30 Mass % Butanes)

2) Gas mixture of 60 % methane, 35 % CO₂, additional quantities of steam and hydrogensulfide

3) p = 1.013 bar; T = 273 K 4) T = 20°C

Fuel properties II

	Unit	Euro-Super	Diesel	Crude-oil	Methanol	Ethanol	RME	LPG ¹⁾	Methane	Biogas ²⁾	Hydrogen	DME
<i>h</i>	-	0.14	0.14	0.11	0.12	0.13	0.12	0.18	0.25	0.09	1	0.13
<i>c</i>	-	0.84	0.86	0.86	0.38	0.52	0.77	0.82	0.75	0.44	0	0.52
<i>o</i>	-	0.02	0	0	0.5	0.35	0.11	0	0	0.47	0	0.35
<i>s</i>	-	0	0	0.03	0	0	0	0	0	0	0	0
<i>M</i>	kg/kmol	98	190	198	32	46	296	48	16	27	2	46
Boiling point	°C	30...190	170...350	175...>450	65	78	180...360	-39	-162	-128	-253	-24.9
Evaporation heat ³⁾	kJ/kg	420	300		1097	839		325	509		450	410
Vapor pressure ⁴⁾	bar	0.45...0.9			0.37	0.21		10				
Density ³⁾	kg/m ³	f 725-780	820-860	950 -1010	795	793	880	540	424		71	660
	g							2.06	0.72	1.20	0.09	
<i>L_{st}</i>	kg/kg	14.5	14.6	13.8	6.5	9.0	12.5	15.6	17.2	6.1	34.2	9.0

1) Liquefied Petroleum Gas (typical winter fuel in EU: 70 Mass % Propane, 30 Mass % Butanes)

2) Gas mixture of 60 % methane, 35 % CO₂, additional quantities of steam and hydrogensulfide

3) p = 1.013 bar and the appropriated boiling temperature

4) T = 40 °C

Lecture notes: ICE1 2.4, AVPS 2.2

E 4336-20 ICE1, AVPS

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Assessment criteria for alternative fuels

Availability of Primary Energy	Conversion to Drive Energy	Distribution of infrastructures	Vehicle Technology	Customer Benefit	Environmental Compatibility
<ul style="list-style-type: none"> - Amount - Range - Security of supply - Produktions-technik - Possibility of usage by other consumers 	<ul style="list-style-type: none"> - Usage of existing facilities - Required new systems - Possibility of usage by other consumers 	<ul style="list-style-type: none"> - Usage of existing systems - Required new facilities - Possibility of usage by other consumers 	<ul style="list-style-type: none"> - Principal suitability - Required new facilities - Usage of existing facilities - Improvement of efficiency 	<ul style="list-style-type: none"> - Refueling time - Action radius - Payload - Mileage - Comfort - Economy - Emotional acceptance 	<ul style="list-style-type: none"> - Global (Global warming) - Regional (Smog) - Local (Toxicity, noise, Property damage)

Security / costs / total energy and ecological balance / avoidance of environmental costs / economic impact / cost-benefit analysis / technology advantage for industry



"Social benefit" for alternative fuel X

Alternative fuels

Atom	Name	Formula	Note
H	Hydrogen	H ₂	LH ₂ /GH ₂
C,H	Methane	CH ₄	Natural gas (CNG, LNG)
C,H	Liquefied petroleum gas	C ₃ H ₈ , C ₄ H ₁₀	Propane, Butane (main component LPG)
C,H,O	Methanol	CH ₃ OH	R-OH (Alcohol)
C,H,O	Ethanol	C ₂ H ₅ OH	R-OH (Alcohol)
C,H,O	Butanol	C ₄ H ₉ OH	R-OH (Alcohol)
C,H,O	Dimethylether	CH ₃ OCH ₃	CH ₃ (Methyl) R ₁ -O-R ₂ (Ether)
C,H,O	Vegetable oil methyl ester		R ₁ -C(=O)-O-R ₂ (Ester)
C,H,O	Mixtures (e.g. biogas)	CH ₄ +CO ₂	