

1. Fill-in the table below:

Element	Se	<b>Cu</b>	<b>Kr</b>	<b>Ca</b>
Mass number	77	65	<b>79</b>	42
Number of neutrons	<b>43</b>	36	43	<b>22</b>
Number of protons	<b>34</b>	<b>29</b>	36	<b>20</b>
Number of electrons	<b>35</b>	<b>28</b>	36	18
Charge	-1	+1	<b>0</b>	+2
Isotope Notation	<b><math>^{77}_{34}\text{Se}</math></b>	<b><math>^{65}_{29}\text{Cu}</math></b>	<b><math>^{79}_{36}\text{Kr}</math></b>	<b><math>^{42}_{20}\text{Ca}</math></b>