

$$\begin{aligned} \mathbf{r}(t) \\ \mathbf{v}(t) = \dot{\mathbf{r}}(t) \\ \mathbf{a}(t) = \ddot{\mathbf{r}}(t) \end{aligned}$$

$$\frac{d(\mathbf{r}(t_i) \cdot \dot{\mathbf{r}}(t_i))}{dt} = \dot{\mathbf{r}}(t_i) \cdot \dot{\mathbf{r}}(t_i) + \ddot{\mathbf{r}}(t_i) \cdot \mathbf{r}(t_i)$$

$$\sum_{i=1}^N m_i \cdot \frac{d(\mathbf{r}(t_i) \cdot \dot{\mathbf{r}}(t_i))}{dt} = \sum_{i=1}^N m_i [\dot{\mathbf{r}}(t_i)]^2 + \sum_{i=1}^N m_i \ddot{\mathbf{r}}(t_i) \cdot \mathbf{r}(t_i)$$

$$\underbrace{\left\langle \sum_{i=1}^N m_i \cdot \frac{d(\mathbf{r}(t_i) \cdot \dot{\mathbf{r}}(t_i))}{dt} \right\rangle}_{\text{isolated system.}} = \left\langle \sum_{i=1}^N \frac{1}{2} m_i [\dot{\mathbf{r}}(t_i)]^2 \right\rangle + \left\langle \sum_{i=1}^N \ddot{\mathbf{r}}(t_i) \cdot \mathbf{r}(t_i) \right\rangle$$

$$0 = 2 \langle K \rangle + \langle W \rangle$$

$\langle \text{Kinetic} \rangle \qquad \langle \text{virial} \rangle$

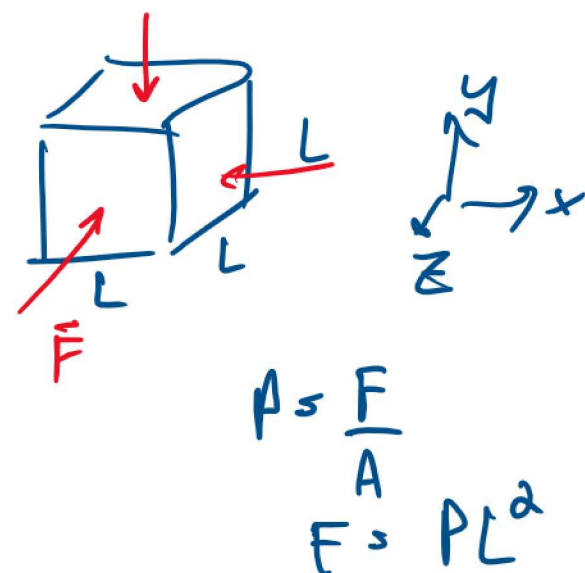
$$\langle K \rangle = \left\langle \sum_{i=1}^N \frac{1}{2} m_i v_i^2 \right\rangle = \frac{3}{2} N k_B T \rightarrow \text{Temp.}$$

$$N \text{ atoms} \rightarrow \text{each has } 3 \text{ d.o.f.} \quad \frac{1}{2} k_B T$$

$$\langle W \rangle = \left\langle \sum_{i=1}^N \mathbf{r}(t_i) \cdot \ddot{\mathbf{r}}(t_i) \right\rangle \rightarrow \text{pressure}$$

$$\ddot{\mathbf{r}}_i = \ddot{\mathbf{r}}_i^{\text{int}} + \ddot{\mathbf{r}}_i^{\text{ext}} \text{ due to pressure}$$

$\downarrow$   
all other atoms



$$\langle W \rangle = \sum_{i=1}^N \mathbf{r}(t_i) \cdot \ddot{\mathbf{r}}_i^{\text{int}} + \sum_{i=1}^N \mathbf{r}(t_i) \cdot \ddot{\mathbf{r}}_i^{\text{ext}}$$

$$\begin{aligned} \left\langle \sum_{i=1}^N \mathbf{r}(t_i) \cdot \ddot{\mathbf{r}}_i^{\text{ext}} \right\rangle &= 0 - pL^3 + 0 - pL^3 + 0 - pL^3 \\ &= -3pL^3 \\ &= -3pV \end{aligned}$$

$x=0 \quad x=L \quad y=0 \quad y=L \quad z=0 \quad z=L$

$$2 \langle K \rangle + \langle W \rangle = 0$$

$$\frac{3}{2} N k_B T + \left\langle \sum_{i=1}^N \mathbf{r}(t_i) \cdot \ddot{\mathbf{r}}_i^{\text{int}} \right\rangle - 3pV = 0$$

For gas:

$$3Nk_B T + 0 - 3pV = 0$$

$$\boxed{Nk_B T = pV} \quad *$$