Today - orbital angular momentum, multi-electron astoms

Orbital angular numeratum: L = rxp L2 = |L|2

(For H-ation $L^{2}|47 = \frac{1}{16}l(l+1)|47$, $l = 0, 1, 2, ..., max <math>l = \frac{1}{n-1}$

Lz 147 = tm 147, m= - l,-l+1, _,0,1, _,l-1, l

general

For the hydrogen atom.

{ A, L, L, J forms a C.S.C.O. (ignoring spin)

=> label the common eigenstates by their eigenvalues or information about their eigenvalues, eg. In l m>

 $\widehat{H} | nlm \rangle = + \frac{E_1}{n^2} | nlm \rangle$ where $E_n = \frac{E_1}{n^2}$

Eg. 1s state for H atom

11007 => we know energy = E,

(1001 L, 100) = 0 1100>

L11007=01100>

{Inlm} & forms an orthonormal basis.

$$\vec{L} = \vec{r} \times \vec{p}$$

 $\hat{P}_{2m} = \frac{1}{2m} \left(\hat{P}_r + \frac{\hat{L}}{r^2} \right)$

Hydrogen others $\hat{f} = \hat{p}^{\dagger} + V(r)$

(Born-Opporheiner)

V depends only on ITI = r (central potential)

(c.f. next week's lectures).

System has rotational symmetry => Angular momentum is anserved.

- definition of vector ours product

Consequences of [Li, Li] = its Eigh Lh

uncertainty relation (WILZ)

DADB = 1 | ([A,B]> |

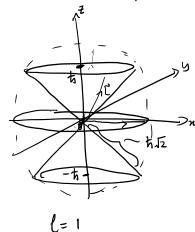
So eg.

[210,007

$$\Delta L_x \Delta L_y \geqslant \frac{1}{2} | ik \langle L_z \rangle | = \frac{k}{2} | \langle L_z \rangle |$$

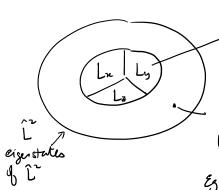
For l=0, we know L= (0.0,0)

Otherwise, it is not possible to simultaneously determine Lx, Ly, Lz.



$$L^{2} | nlm \rangle = \hbar^{2} \{(lt) | nlm \rangle = |L|$$
 $cg l = 1,$
 $L_{2} | nlm \rangle = mto | nlm \rangle$
 $is to \sqrt{2}$

An eigenstate of Lz is an eigenstate of L^2 . but not an eigenstate of Lx or Ly.



- eigenstate of both L' and Ly

eigenstate of L

but not an eigenstate of Lx, Ls, Lz.

Multi-election atoms

147 described a single particle wavefunction (TI47 = 4(T)

Tone real-space coordinate

Now 147 describes more than one particle in a multi-electron system

many election wavefunction (r., rz, ..., r, 14) = 4(r, rz, ..., rn)

14(7, 5, 5, 57) [DV. DV. DV. probability of finding

particle 1 in a volume DV, about 7, " 2 — " — OV_ " 70°,

(+Z) ,\(\)

For a multi-electron atom,

in the Born-Opporheimer approx. (\(\frac{1}{\lambda} \sum_{i=1}^{\infty} \)

$$\hat{H} = \sum_{i=1}^{z} \frac{p_{i}^{2}}{2m_{i}} - \sum_{i=1}^{z} \frac{ze^{2}}{r_{i}} + \sum_{i < j} \frac{e^{2}}{|\vec{r}_{i} - \vec{r}_{j}|}$$

T KEJe

energy

$$+ \sum_{i < j} \frac{e^{2}}{|\vec{r}_{i} - \vec{r}_{j}|}$$

repulsive potential energy between e-

makes the Itam, Itonian challenging to solve

All terms are 'huge'.

Interestingly, in the periodic table, multi-electron atoms are described quite simply in terms if s, p, d, f orbitals, Carbital angular momentum eigenstates)

> relevant for central potentials

Here, we introduce the single particle approximations / quasiparticle approximation where instead of considering $Y(r_1, \ldots, r_Z)$, we consider

effective single particle were fractions 4eff (r.)

4 eff (r2)

بوه (ب).

need an effective single particle hamiltonian

PC3130 W2L2 AY24 Page 4

Hett for yeth (r), that captures the essential physics of the many-e system.

Today, we will learn , one such Heff ,

called the central potential approximations

 $\mu^{\text{eff}} = \frac{p_i}{2m_i} + V_{e}(r)$

Let's just consider limiting cases.

r->0 close to mideus

r>>0 for from nuclais.

$$V_c(n) \sim -\frac{e^2}{r}$$
 (Shielding / Sweening)

Central potential

Solutions approx, as 1s, 2s, 2p. etc. (see above)