

Chapter 2 : Polymers

Section 2.1 – Addition Polymers

Define the following:

1. Monomer, Dimer (section 2.2), Polymer, Polymerization
2. How is the formula of a polymer represented?
3. Draw an example of a polymer, its monomer and its formula.
4. What needs to be present for an addition polymer to form?
5. Where do the monomers attach?
6. Define Plastics.
 - a. Describe the physical and chemical properties
 - b. Describe the intermolecular forces
7. Substituted groups – modifying the monomer
 - a. What are some benefits of modifying the monomer?
 - b. What problems may occur – use the carbonyl group as an example
8. Crosslinking - define
 - a. What is the role of double bonds?
 - b. How can sulfur be used
 - c. How does crosslinking influence the structure of a polymer?
9. What is the chemical differences between thermoplastics and thermoset plastics?

Condensation Reactions 2.2

1. What functional groups are required to form a condensation reaction?
2. Why do dimers form?
3. What is the difference between addition and condensation polymers?
 - a. Intermolecular forces
 - b. Physical and chemical properties
 - c. Crosslinking
4. Make an ester polymer and identify the monomers
5. Make Nylon 4,4 and identify the monomers