

$$C = \frac{\text{Solute}}{\text{solution}}$$

$$C = \frac{n}{V}$$

$$C_1 V_1 = C_2 V_2$$

$$C = \frac{n}{V}$$

6.1 + 6.2 - Terms and Theory of Solutions

6.3 - Concentration calculations

6.5 - Dilution and solution preparation

7.1 - Solubility graph and rules

Solubility table

Solvay and LeBlanc Process

7.3 - Net Ionic Equations

7.5 - Qualitative - colour, precipitates

7.6 - Stoichiometry - solutions

8.1 - Characteristics of acids and bases

8.2 - pH: understand + calculate H_3O^+

8.4 - Theories of acids and bases

8.5 - Reactions and the titration

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