Table 1. The Various Shapes of Molecules and Polyatomic Ions

Note: central atom = A

bonding atom (i.e. bonding EPD) = B

lone pair of electrons (or non-bonding EPD) on central atom =E

total # of EPD = bonding e-pairs + lone (non-bonding) electron pairs

Formula	Total # of EPD	# bonding pairs of e-	# non-bonding (or lone) pairs of electrons	Lewis Structure (single bonds only)	electronic geometry (geometric arrangement)	molecular geometry (shape)	Bond Angle(s)
AB <sub>2</sub>	2	2	0		linear	linear	
AB <sub>3</sub>	3	3	0		trigonal planar	trigonal planar	
AB <sub>2</sub> E	3	2	1		trigonal planar	angular or bent	
AB <sub>4</sub>	4	4	0		tetrahedral	tetrahedral	
AB₃E	4	3	1		tetrahedral	pyramidal	
$AB_2E_2$	4	2	2		tetrahedral	angular or bent	
AB <sub>5</sub>	5	5	0		trigonal bipyramidal	trigonal bipyramidal	

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Formula	Total # of EPD	# bonding pairs of e-	# non-bonding (or lone) pairs of electrons	Lewis Structure (single bonds only)	electronic geometry (geometric arrangement)	molecular geometry (shape)	Bond Angle(s)
AB₄E	5	4	1		trigonal bipyramidal	see-saw	
AB <sub>3</sub> E <sub>2</sub>	5	3	2		trigonal bipyramidal	T-shape	
AB <sub>2</sub> E <sub>3</sub>	5	2	3		trigonal bipyramidal	linear	
AB <sub>6</sub>	6	6	0		octahedral	octahedral	
AB₅E	6	5	1		octahedral	square- based pyramidal	
$AB_4E_2$	6	4	2		octahedral	square planar	

Table 2. Complete the following table. Leave the column entitled "Polarity" blank for the time being.

Table	2. Complete the follow	ring table.	LCave tric c	Oldiffil Citatic	d i clarity blain	r ioi the time being.	
Formula	Lewis Structure	Total EPD on central atom	# bonding EPD on central atom	# non- bonding EPD on central atom	electronic geometry	molecular geometry	Polar or Non- polar? (exclude ions)
CH₄							
NH <sub>3</sub>							
BF <sub>3</sub>							
SCI <sub>2</sub>							
SF <sub>6</sub>							

Formula	Lewis Structure	Total EPD on central atom	# bonding EPD on central atom	# non- bonding EPD on central atom	electronic geometry	molecular geometry	Polar or Non- polar? (exclude ions)
TiCl <sub>4</sub>							
XeF <sub>2</sub>							
NO <sub>3</sub> -							
NO <sub>2</sub> -							

Formula	Lewis Structure	Total EPD on central atom	# bonding EPD on central atom	# non- bonding EPD on central atom	electronic geometry	molecular geometry	Polar or Non- polar? (exclude ions)
CS <sub>2</sub>							
AsF <sub>3</sub>							
MoF <sub>6</sub>							

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