

Acids and bases

The behaviour of the chemical determines if it is an acid or a base

Acids

- Molecular compounds
- Chemical formulas of acids normally start with a hydrogen
- They taste sour
- Corrosive to skin, metals and limestone
- They turn litmus paper red
- They have a pH below 7
- They are electrolytes - they conduct electricity when dissolved in water because ions are formed
- When they dissolve in water they produce hydrogen ions
 - Hydrogen ions are what make acids dangerous
- When an acid reacts with metal it produces hydrogen gas which is explosive
- When an acid reacts with carbonates it produces carbon dioxide
- Acids are made when nonmetal oxides react with water

Types of Acids

- Binary Acids
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- Oxyacids
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- Organic Acids - formula has “COOH”
 - Vinegar - CH_3COOH
 - Citric acid - $\text{C}_3\text{H}_5\text{O}(\text{COOH})_3$ - sour candies
 - Ascorbic acid - $\text{C}_5\text{H}_7\text{O}_4(\text{COOH})$ - vitamin C
 - Formic acid - HCOOH - ants

Acid Rain

- Cars and factories produce carbon dioxide nitrogen dioxide and sulfur dioxide
- These oxides react with water to produce acids (carbonic acid, nitric acid, sulfuric acid)
- Acid rain destroys statues made out of limestone (carbonate)
- Acid rain also causes your car to rust faster
- We put catalytic converters on cars to reduce the pollution and to reduce acid rain
- Factories install “scrubbers” to reduce their pollution

Bases

- Most bases are ionic compounds
 - Chemical formulas of bases normally end with a hydroxide ion
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- They taste bitter
 - They feel slippery
 - Corrosive to skin but they **do not** react with metal
 - They have a pH above 7
 - They turn litmus paper blue
 - They are also called electrolytes because they produce ions
 - They dissolve in water and produce hydroxide ions
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- Hydroxide ions are what make bases dangerous
 - Bases are made by reacting metal oxides with water
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- Bases only react with organic material, oils or with acids
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- We use bases as cleaning products
 - Examples
 - Sodium hydroxide is oven cleaner
 - Calcium hydroxide is lye which is added to soils
 - Magnesium hydroxide is found in Tums

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