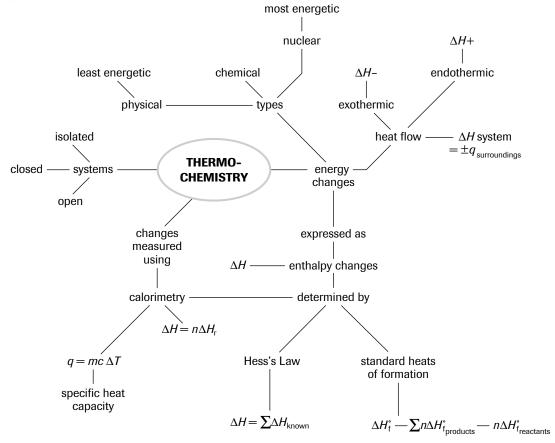
## **CHAPTER 5 SUMMARY**

## **MAKE A SUMMARY**

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## **CHAPTER 5 SELF-QUIZ**

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- 1. False: Nuclear changes generally *produce* more energy than chemical changes.
- 2. False: In exothermic reactions, the reactants have more *potential* energy than the products.
- 3. True
- 4. False: In endothermic reactions, the heat term is written on the *left* side of the equation.
- 5. True
- 6. True
- 7. False: Burning of gasoline is an example of an exothermic physical change.
- 8. True
- 9. False: Specific heat capacity is the amount of heat required to change *one gram* through 1°C.
- 10. True
- 11. (c)
- 12. (b)
- 13. (e)
- 14. (c)
- 15. (c)
- 16. (c)
- 17. (e)
- 18. (c)

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