## **VSEPR**

Substance	Lewis Structure	Central Atom		VSEPR Code	Molecular Shape	Is the molecule polar?Y or N
		# of lone pairs	# of atoms bonded			polar. T of iv
BeCl <sub>2</sub>						
$NH_3$						
SiCl <sub>4</sub> CO <sub>2</sub>						
$CO_2$						
$SF_4$ $CO_2^{-2}$						
$CO_2^{-2}$						
HC1						
$H_2S$						
$NI_3$						
CF <sub>4</sub>						
OCl <sub>2</sub>						
HOF						
$N_2$						
HCBr <sub>3</sub>						
SF <sub>6</sub>						
PCl <sub>5</sub>						
PCl <sub>5</sub> SO <sub>4</sub> -2 BrCl <sub>5</sub> BrO <sub>3</sub> -1						
BrCl <sub>5</sub>						
$BrO_3^{-1}$						