Organic Reactions

Show structural formulas for reactants and products.

1. 1-butene + bromine

2. 3-methyl-2-pentene + hydrogen

3. propene burned in excess oxygen

4. water reacts with 2,3-dimethyl-2-butene

5. ethyne burned in excess oxygen

6. excess hydrochloric acid added to propyne

7. one mole of hydrobromic acid added to ethyne

8. 3 moles of chlorine gas added to ethane

9. 1-butanol + propanoic acid

10. methanol + benzoic acid

11. pentanoic acid + 1-octanol

12. magnesium + ethanoic acid

13. benzoic acid + sodium hydroxide

14. potassium carbonate + oxalic acid (diethanoic acid)

15. propyne + excess iodine

16. 1-butene + hydrogen bromide

17. production of 1,1,2-trichloroethane from ethene and chlorine

18. one mole of HCl + ethyne

19. propene + hydrogen

20. cyclohexane + chlorine

21. ethanol + butanoic acid

22. 2-pentene + bromine

23. ethene + water

24. chloroethane + NaOH (dilute and Conc)

25. cyclopentene + HBr

26. cyclohexane+chlorine

27. bromobenzene + NaOH

28. methanol oxidized (1st stage)

29. 2-propanol oxidized(1st stage)

30. # 28 oxidized

31. ethanoic acid + magnesium hydroxide

32. Complete combustion of

2-methylpentanoic acid

Page 27: #1+3

Page 30: #4

Page 31: #2-3

Page 37: #6, #2

Page 44: #7-9

Page 48: #3,4,6

Page 56: #6

Page 57: #3

Page 63: #5

Page 66: #11, 12

Page 68: #3-4

Page 77: #7

Page 78: #11,#1