

Package **AquaEnv**: an Aquatic modelling Environment in R

Andreas F. Hofmann

Centre for Estuarine and Marine Ecology
Netherlands Institute of Ecology
The Netherlands

Abstract

AquaEnv is an integrated development toolbox for aquatic chemical model generation focused on (ocean) acidification and CO₂ air-water exchange.

- It contains all elements necessary to model the pH, the related CO₂ air-water exchange, as well as aquatic acid-base chemistry in general for an arbitrary marine, estuarine or freshwater system. Also chemical batches can be modelled.
- Next to the routines necessary to calculate desired information, AquaEnv also contains a suite of tools to visualize this information.
- Furthermore, AquaEnv can not only be used to build dynamic models of aquatic systems, but it can also serve as a simple desktop tool for the experimental aquatic chemist to generate and visualize all possible derived information from a set of measurements with one single easy to use R function.
- Additionally, the sensitivity of the system to variations in the input variables can be visualized.
- AquaEnv also contains a number of example “applications” that make use of the aquatic modelling toolbox that AquaEnv provides:
 - a theoretical titration simulator
 - and a routine to determine total alkalinity ([TA]), the total dissolved inorganic carbon concentration ([\sum CO₂]), as well as additionally the electrode standard potential (E_0) and the first dissociation constant of the carbonate system ($K_{\text{CO}_2}^*$)

Keywords: aquatic modelling, pH, pH scales, dissolved inorganic carbon, total alkalinity, total alkalinity curve fitting, theoretical titration, revelle factor, omega, solubility products, CO₂, ocean acidification, estuaries, carbonate system, seawater, R.

Contents

1	Introduction	3
2	Using AquaEnv	4
2.1	Basic features	4
2.1.1	calling the “K” functions directly	4
2.1.2	Minimal <i>aquaenv</i> definition	4
2.1.3	Defining the complete aquaenv system in different ways	6
2.1.4	Cloning the aquaenv system: 1 to 1 and with different pH or TA	7
2.1.5	preparing input variables	8
2.1.6	Vectors as input variables	8

1 Introduction

AquaEnv is a toolbox for aquatic modelling that serves several purposes

- It provides functions to calculate the stoichiometric equilibrium constants (K^*) for key acid base systems in natural seawater, the Henry's constants (K_0), as well as the solubility products (K_{sp}) for calcite and aragonite. This functionality is provided via the functions `K_CO2`, `K_HCO3`, `K_BOH3`, `K_W`, `K_HSO4`, `K_HF`, `K_NH4`, `K_H2S`, `K_H3PO4`, `K_H2PO4`, `K_HPO4`, `K_SiOH4`, `K_SiOOH3`, `KO_CO2`, `KO_O2`, `Ksp_aragonite`, and `Ksp_calcite`.

- all information (which information: including revelle, full speciation, all partial derivatives of α needed for Hofmann, Meysman, Soetaert, and Middelburg (2008a) and ionization fractions as defined by Stumm and Morgan (1996) and used in Hofmann, Middelburg, Soetaert, Wolf-Gladrow, and Meysman (2008b) that can be calculated from all that is known over one single sample (tool for the experimentalist). this is done by the central function `aquaenv`. (is returned as object of class `aquaenv`) Input for `aquaenv` has to be supplied in standard SI units and in molinity (mol/kg-solution) (footnote: explain importance of molinity vs molality), This can be done with the function `convert`.

the information created with `aquaenv` is also supplied in standard SI units and in molinity and can be converted to other units with the function `convert`

from any of the variable pairs also $\sum \text{CO}_2$ can be calculated

one can use input vectors of T , S or d for `aquaenv` to obtain vectors of all possibly calculated information as function of the input vector this can be visualized in a large variety of ways using the `plot` function specially defined for objects of type `aquaenv`

objects of class `aquaenv` can be used in dynamic models to define the state of the system in each timestep of the numerical integration (done e.g. with **deSolve**). with the function `from.dataframe` it is possible to convert output of those dynamic models into objects of type `aquaenv` which allows the user to use the whole suite of visualisation tools that is provided in **AquaEnv**.

Hofmann *et al.* (2008a) and Hofmann *et al.* (2008b) describe methods for an “explicit” pH modelling which allows for the quantification of the influences of kinetically modelled processes on the pH. Objects of type `aquaenv` provide all needed quantities (partial derivatives of α , ionization fractions) to employ both of those methods in dynamic models. Furthermore, **AquaEnv** provides the functionality to cumulatively plot the obtained influences on the pH.

as an example of how to use the toolbox that is **AquaEnv**, two applications are provided the creation of theoretical titrations which can be used e.g. to create Bjerrum plots, something that can also be done with the `plot` function in **AquaEnv**.

a routine based on a method in DOE (1994) that makes use of that theoretical titration function and allows for determining total alkalinity ($[TA]$), the total dissolved inorganic carbon concentration ($[\sum \text{CO}_2]$), as well as additionally the electrode standard potential (E_0) and the first dissociation constant of the carbonate system ($K_{\text{CO}_2}^*$) using the Levenberg-Marquart algorithm (least squares optimization procedure) as provided in **minpack.lm**

2 Using AquaEnv

2.1 Basic features

2.1.1 calling the “K” functions directly

```
> K_CO2(15, 30)
```

```
[1] 9.089117e-07
attr(,"unit")
[1] "mol/kg-soln"
attr(,"pH scale")
[1] "free"
```

```
> K_HCO3(15, 30)
```

```
[1] 6.014934e-10
attr(,"unit")
[1] "mol/kg-soln"
attr(,"pH scale")
[1] "free"
```

```
> K0_CO2(15, 30)
```

```
[1] 0.03852158
attr(,"unit")
[1] "mol/(kg-soln*atm)"
```

```
> Ksp_calcite(15, 30, 100)
```

```
[1] 3.644549e-07
attr(,"unit")
[1] "(mol/kg-soln)^2"
```

2.1.2 Minimal *aquaenv* definition

```
> ae <- aquaenv(Tc = 15, S = 30)
> ae$K_CO2
```

```
[1] 9.089117e-07
attr(,"unit")
[1] "mol/kg-soln"
attr(,"pH scale")
[1] "free"
```

```
> ae$Ksp_calcite
```

```

[1] 3.582242e-07
attr("unit")
[1] "(mol/kg-soln)^2"

> ae$Ksp_aragonite

[1] 5.643199e-07
attr("unit")
[1] "(mol/kg-soln)^2"

> ae <- aquaenv(Tc = 15, S = 30, d = 10)
> ae$K_CO2

[1] 9.09921e-07
attr("unit")
[1] "mol/kg-soln"
attr("pH scale")
[1] "free"

> ae$Ksp_calcite

[1] 3.588429e-07
attr("unit")
[1] "(mol/kg-soln)^2"

> ae$Ksp_aragonite

[1] 5.652276e-07
attr("unit")
[1] "(mol/kg-soln)^2"

> names(ae)

[1] "Tc"           "Tk"           "S"            "Cl"
[5] "I"            "d"            "hydroP"       "density"
[9] "SumCO2"       "SumNH4"       "SumH2S"       "SumHNO3"
[13] "SumHNO2"      "SumH3PO4"     "SumSiOH4"     "SumBOH3"
[17] "SumH2SO4"     "SumHF"        "SumBr"        "ClConc"
[21] "Na"           "Mg"           "Ca"           "K"
[25] "Sr"           "molal2molin"  "free2tot"     "free2sws"
[29] "tot2free"     "tot2sws"      "sws2free"     "sws2tot"
[33] "K0_CO2"       "K0_O2"        "CO2_sat"      "O2_sat"
[37] "K_W"          "K_HSO4"       "K_HF"         "K_CO2"
[41] "K_HCO3"       "K_BOH3"       "K_NH4"        "K_H2S"
[45] "K_H3PO4"      "K_H2PO4"      "K_HPO4"       "K_SiOH4"
[49] "K_SiOOH3"     "K_HNO2"       "K_HNO3"       "K_H2SO4"
[53] "K_HS"         "Ksp_calcite"  "Ksp_aragonite"

```

2.1.3 Defining the complete aquaenv system in different ways

```

> Tc <- 15
> S <- 30
> d <- 10
> SumCO2 <- 0.002
> pH <- 8
> TA <- 0.002140323
> pCO2 <- 0.000533576
> CO2 <- 2.055419e-05
> ae <- aquaenv(Tc, S, d, SumCO2 = SumCO2, pH = pH)
> ae$TA

[1] 0.002140800
attr(,"unit")
[1] "mol/kg-soln"

> ae <- aquaenv(Tc, S, d, SumCO2 = SumCO2, TA = TA)
> ae$pH

[1] 7.998791
attr(,"pH scale")
[1] "free"

> ae <- aquaenv(Tc, S, d, SumCO2 = SumCO2, CO2 = CO2)
> ae$pH

[1] 7.999293
attr(,"pH scale")
[1] "free"

> ae <- aquaenv(Tc, S, d, SumCO2 = SumCO2, pCO2 = pCO2,
+             dsa = TRUE, revelle = TRUE)
> ae$pH

[1] 7.999293
attr(,"pH scale")
[1] "free"

> names(ae)

[1] "Tc"          "Tk"          "S"
[4] "Cl"          "I"           "d"
[7] "hydroP"      "density"     "SumCO2"
[10] "SumNH4"      "SumH2S"      "SumHNO3"
[13] "SumHNO2"     "SumH3PO4"    "SumSiOH4"

```

[16]	"SumBOH3"	"SumH2SO4"	"SumHF"
[19]	"SumBr"	"ClConc"	"Na"
[22]	"Mg"	"Ca"	"K"
[25]	"Sr"	"molal2molin"	"free2tot"
[28]	"free2sws"	"tot2free"	"tot2sws"
[31]	"sws2free"	"sws2tot"	"K0_CO2"
[34]	"K0_O2"	"CO2_sat"	"O2_sat"
[37]	"K_W"	"K_HSO4"	"K_HF"
[40]	"K_CO2"	"K_HCO3"	"K_BOH3"
[43]	"K_NH4"	"K_H2S"	"K_H3PO4"
[46]	"K_H2PO4"	"K_HPO4"	"K_SiOH4"
[49]	"K_SiOOH3"	"K_HNO2"	"K_HNO3"
[52]	"K_H2SO4"	"K_HS"	"Ksp_calcite"
[55]	"Ksp_aragonite"	"TA"	"pH"
[58]	"pCO2"	"CO2"	"HCO3"
[61]	"CO3"	"BOH3"	"BOH4"
[64]	"OH"	"H3PO4"	"H2PO4"
[67]	"HPO4"	"PO4"	"SiOH4"
[70]	"SiOOH3"	"SiO2OH2"	"H2S"
[73]	"HS"	"S2min"	"NH4"
[76]	"NH3"	"H2SO4"	"HSO4"
[79]	"SO4"	"HF"	"F"
[82]	"HNO3"	"NO3"	"HNO2"
[85]	"NO2"	"omega_calcite"	"omega_aragonite"
[88]	"revelle"	"c1"	"c2"
[91]	"c3"	"dTAdSumCO2"	"b1"
[94]	"b2"	"dTAdSumBOH3"	"so1"
[97]	"so2"	"so3"	"dTAdSumH2SO4"
[100]	"f1"	"f2"	"dTAdSumHF"
[103]	"dTAdH"	"dTAdKdKdS"	"dTAdKdKdT"
[106]	"dTAdKdKdd"	"dTAdKdKdSumH2SO4"	"dTAdKdKdSumHF"

```
> ae <- aquaenv(Tc, S, d, SumCO2 = SumCO2, CO2 = CO2, pCO2 = pCO2)
```

```
[1] "Error! Overdetermined system: entered pCO2: 0.000533576 , calculated pCO2: 0.000533576"
[1] "Please enter only one of: pH, TA, CO2, or pCO2."
```

```
> ae <- aquaenv(Tc, S, d, SumCO2 = SumCO2, pH = pH, TA = TA)
```

```
[1] "Error! Overdetermined system: entered TA: 0.002140323 , calculated TA: 0.002140799720"
[1] "Please enter only one of: pH, TA, CO2, or pCO2."
```

```
> ae <- aquaenv(Tc, S, d, SumCO2 = SumCO2, pH = pH, CO2 = CO2)
```

```
[1] "Error! Overdetermined system: entered pH: 8 , calculated pH: 7.99929303203143"
[1] "Please enter only one of: pH, TA, CO2, or pCO2."
```

```

> ae <- aquaenv(Tc, S, d, SumCO2 = SumCO2, pH = pH, pCO2 = pCO2)

[1] "Error! Overdetermined system: entered pH: 8 , calculated pH: 7.99929301525035"
[1] "Please enter only one of: pH, TA, CO2, or pCO2."

> ae <- aquaenv(Tc, S, d, SumCO2 = SumCO2, TA = TA, CO2 = CO2)

[1] "Error! Overdetermined system: entered TA: 0.002140323 , calculated TA: 0.002140520813"
[1] "Please enter only one of: pH, TA, CO2, or pCO2."

> ae <- aquaenv(Tc, S, d, SumCO2 = SumCO2, TA = TA, pCO2 = pCO2)

[1] "Error! Overdetermined system: entered TA: 0.002140323 , calculated TA: 0.002140520806"
[1] "Please enter only one of: pH, TA, CO2, or pCO2."

```

2.1.4 Cloning the aquaenv system: 1 to 1 and with different pH or TA

```

> Tc <- 15
> S <- 30
> SumCO2 <- 0.002
> TA <- 0.00214
> ae <- aquaenv(Tc, S, SumCO2 = SumCO2, TA = TA)
> aeclone1 <- aquaenv(ae = ae)
> pH <- 9
> aeclone2 <- aquaenv(ae = ae, pH = pH)
> TA <- 0.002
> aeclone3 <- aquaenv(ae = ae, TA = TA)
> ae$pH

[1] 7.99838
attr("pH scale")
[1] "free"

> aeclone1$pH

[1] 7.99838
attr("pH scale")
[1] "free"

> aeclone2$TA

[1] 0.002982758
attr("unit")
[1] "mol/kg-soln"

> aeclone3$pH

[1] 7.548174
attr("pH scale")
[1] "free"

```


2.1.5 preparing input variables

```
> Tc <- 15
> S <- 10
> pH_NBS <- 8.142777
> SumCO2molar <- 0.002016803
> pH_free <- convert(pH_NBS, "pHscale", "nbs2free", Tc = Tc,
+   S = S)
> SumCO2molin <- convert(SumCO2molar, "conc", "molar2molin",
+   Tc = Tc, S = S)
> ae <- aquaenv(Tc, S, SumCO2 = SumCO2molin, pH = pH_free)
> ae$pH
```

```
[1] 8
attr(,"pH scale")
[1] "free"
```

```
> ae$SumCO2
```

```
[1] 0.002000000
attr(,"unit")
[1] "mol/kg-soln"
```

```
> ae$TA
```

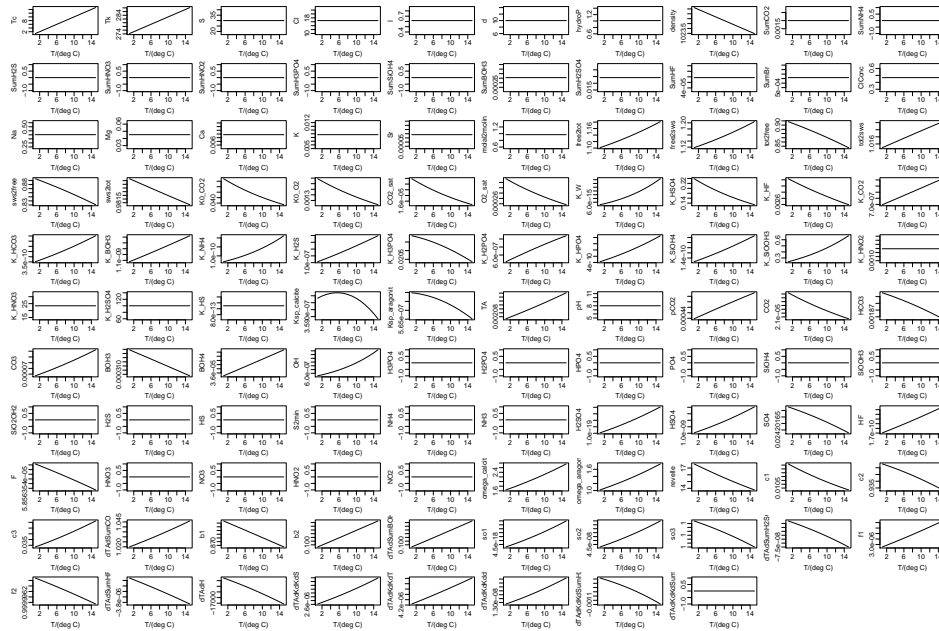
```
[1] 0.002048921
attr(,"unit")
[1] "mol/kg-soln"
```

2.1.6 Vectors as input variables

(only ONE input variable may be a vector)

(with full output: including the Revelle factor and the DSA properties)

```
> SumCO2 <- 0.002
> pH <- 8
> Tc <- 1:15
> S <- 30
> d <- 10
> ae <- aquaenv(Tc, S, d, SumCO2 = SumCO2, pH = pH, reveille = TRUE,
+   dsa = TRUE)
> plot(ae, xval = Tc, xlab = "T/(deg C)", newdevice = FALSE)
```



extra examples

```
> Tc <- 15
> S <- 1:30
> ae <- aquaenv(Tc, S, d, SumCO2 = SumCO2, pH = pH, revelle = TRUE,
+   dsa = TRUE)
> plot(ae, xval = S, xlab = "S")
> S <- 30
> d <- seq(1, 1000, 100)
> ae <- aquaenv(Tc, S, d, SumCO2 = SumCO2, pH = pH, revelle = TRUE,
+   dsa = TRUE)
> plot(ae, xval = d, xlab = "depth/m")
> TA <- 0.0023
> Tc <- 1:15
> S <- 30
> d <- 10
> ae <- aquaenv(Tc, S, d, SumCO2 = SumCO2, TA = TA, revelle = TRUE,
+   dsa = TRUE)
> plot(ae, xval = Tc, xlab = "T/(deg C)")
> Tc <- 15
> S <- 1:30
> ae <- aquaenv(Tc, S, d, SumCO2 = SumCO2, TA = TA, revelle = TRUE,
+   dsa = TRUE)
> plot(ae, xval = S, xlab = "S")
> S <- 30
> d <- seq(1, 1000, 100)
> ae <- aquaenv(Tc, S, d, SumCO2 = SumCO2, TA = TA, revelle = TRUE,
+   dsa = TRUE)
> plot(ae, xval = d, xlab = "depth/m")
```

References

- DOE (1994). *Handbook of Methods for the Analysis of the Various Parameters of the Carbon Dioxide System in Sea Water*. ORNL/CDIAC-74.
- Hofmann AF, Meysman FJR, Soetaert K, Middelburg JJ (2008a). “A step-by-step procedure for pH model construction in aquatic systems.” *Biogeosciences J1 - BG*, **5**(1), 227–251. URL <http://www.biogeosciences.net/5/227/2008/L1-http://www.biogeosciences.net/5/227/2008/bg-5-227-2008.pdf>.
- Hofmann AF, Middelburg J, Soetaert K, Wolf-Gladrow DA, Meysman F (2008b). “pH modelling in aquatic environments: from an alkalinity-based to a proton-based view.” *in preparation*.
- Stumm W, Morgan JJ (1996). *Aquatic Chemistry: Chemical Equilibria and Rates in natural Waters*. Wiley Interscience, New York.

Affiliation:

Andreas F. Hofmann
Centre for Estuarine and Marine Ecology (CEME)
Netherlands Institute of Ecology (NIOO)
4401 NT Yerseke, Netherlands E-mail: a.hofmann@nioo.knaw.nl
URL: <http://www.nioo.knaw.nl/ppages/ahofmann>